
EGR Catalyst for Cooler Fouling Reduction

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Acknowledgements

- Jimi Tjong and his dynamometer group at the Ford Canada Essex Engine Plant ran the cooler fouling engine tests
- Oak Ridge National Lab performed some of the analyses shown – Scott Sluder, John Storey, Sam Lewis
- Johnson Matthey Environmental Catalysts and Technologies provided the catalysts used in these experiments
- Emitec, Inc. provided the metallic catalyst substrates used in these experiments

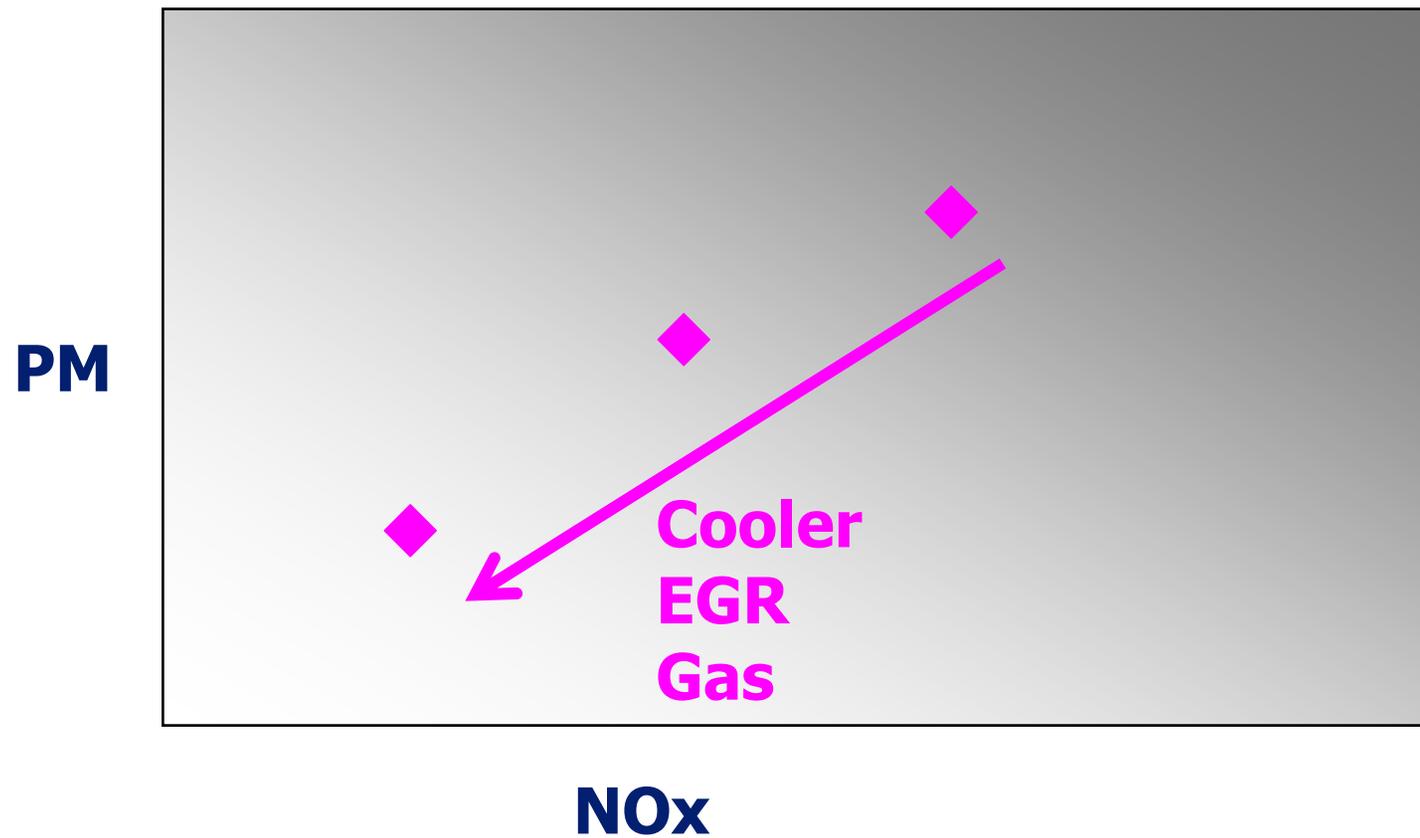


Increased EGR Cooling required

- Future emission standards: lower NO_x, PM
- Reduced charge temperature helps (see next slide)
 - EGR is cooled
 - Bigger coolers
 - Lower temperatures



Improved EGR Cooling Reduces NOx and PM

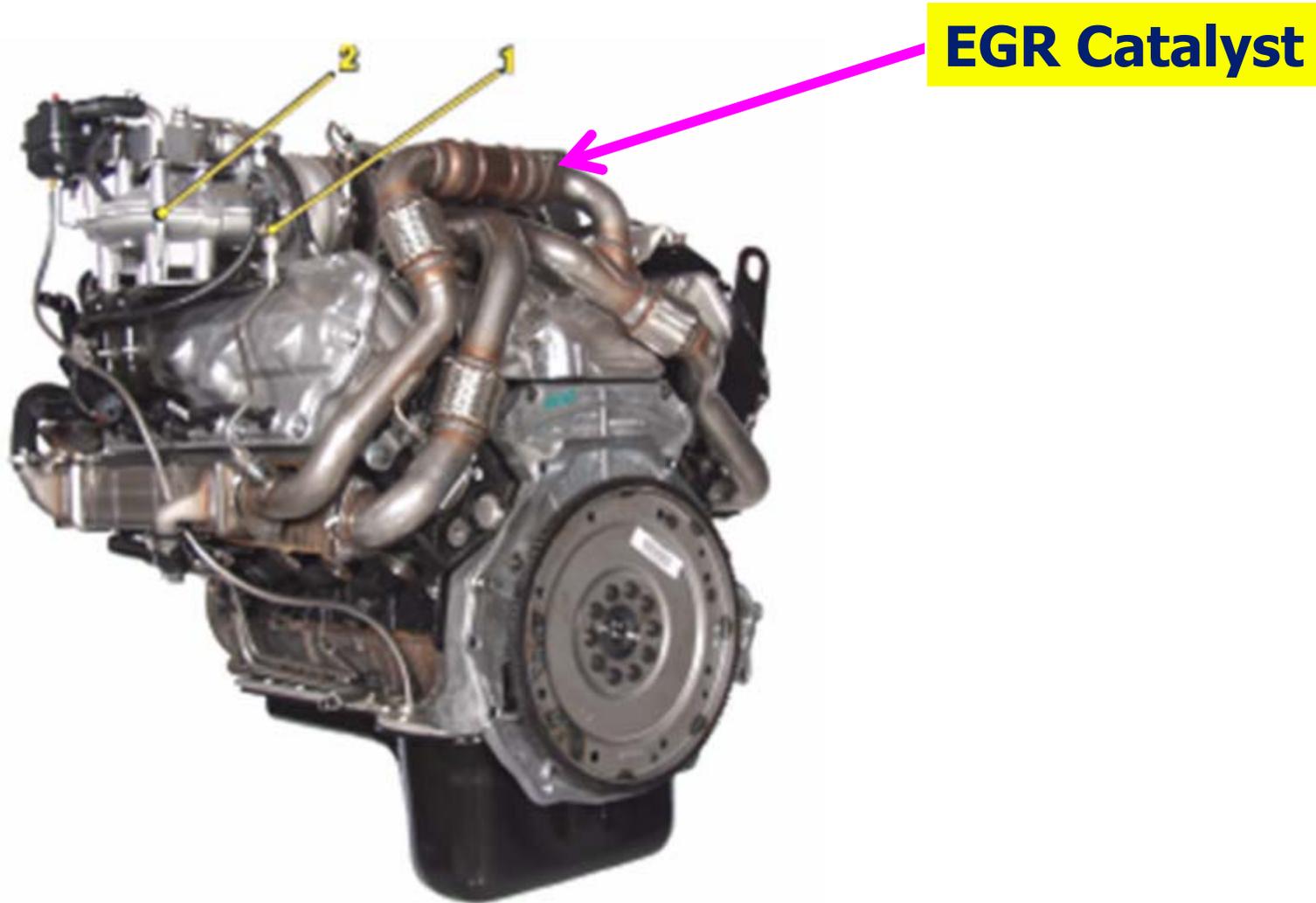


Deposits

- Cool surfaces exposed to gas collect deposits:
 - Soot - thermophoresis
 - Hydrocarbons - condensation
 - Partially oxidized and pyrolyzed HC
 - Acids – sulfuric, nitric, formic, acetic
- Deposit concerns are worse when
 - Wall temperatures are low
 - “Heavy Wet PM” – more likely at low-NO_x calibrations
- Likely to get worse with future calibrations for very low NO_x levels!



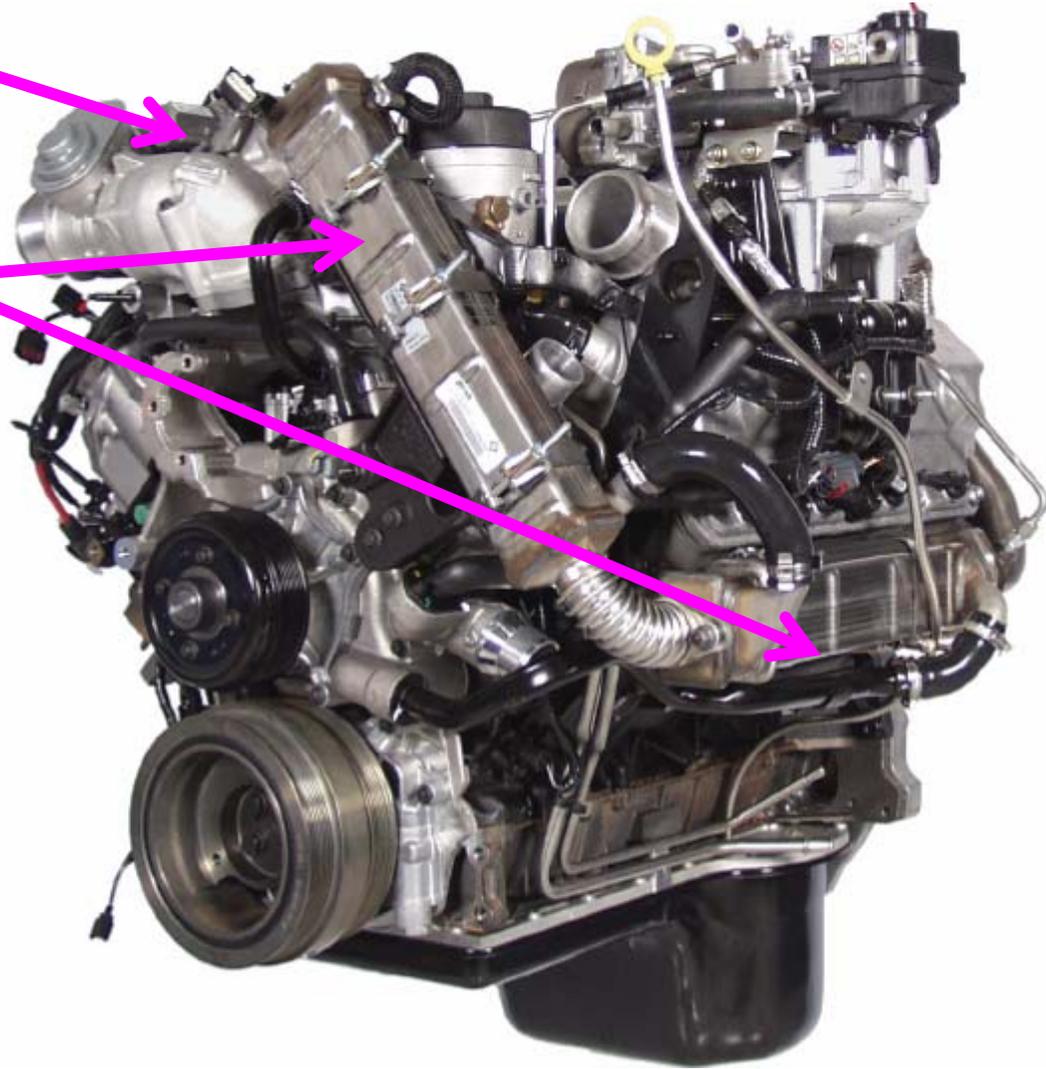
Test Engine – Rear View



Test Engine – Left Front View

EGR Valve

EGR Coolers



Engine Dynamometer Cooler Fouling Test Cycle

<i>Mode</i>	<i>RPM</i>	<i>Load ft-lb</i>	<i>T_{inlet} °C</i>	<i>Space Velocity khr⁻¹</i>
Idle	700	50	150	127
A25	2100	166	250	503
HSV	2300	300	350	1147

- **Two hours at each point in order**
- **Repeat until effectiveness stabilizes**



Response Variables

- Effectiveness

$$\varepsilon = \frac{q_{\text{actual}}}{q_{\text{max-theoretical}}} = \frac{m_{\text{exh}} C_{p,\text{exh}} (T_{\text{exh},i} - T_{\text{exh},o})}{m_{\text{exh}} C_{p,\text{exh}} (T_{\text{exh},i} - T_{\text{coolant},o})} = \frac{(T_{\text{gas in}} - T_{\text{gas out}})}{(T_{\text{gas in}} - T_{\text{coolant in}})}$$

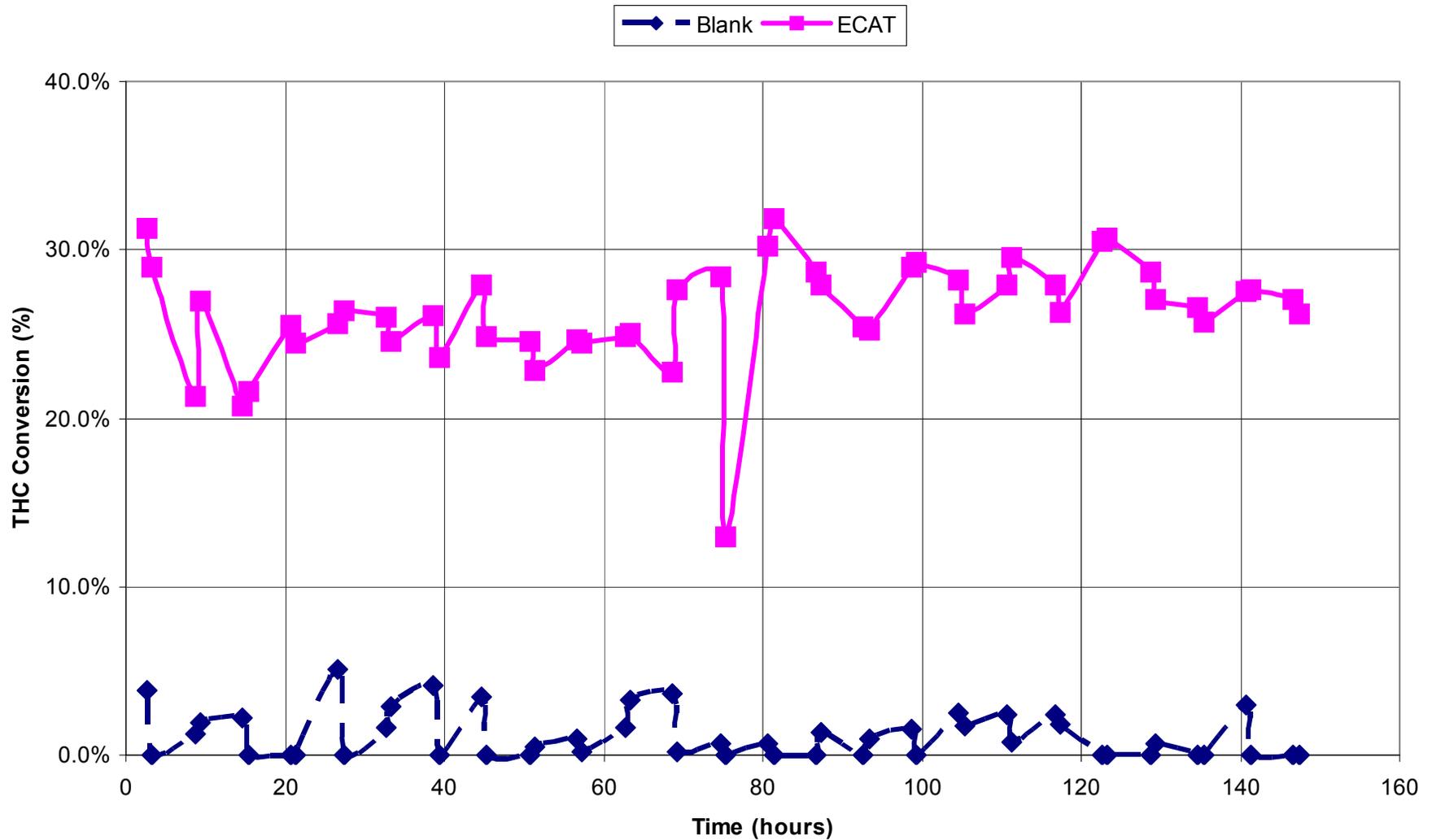
- ECAT HC conversion efficiency

Fouling Test – ECAT Effect

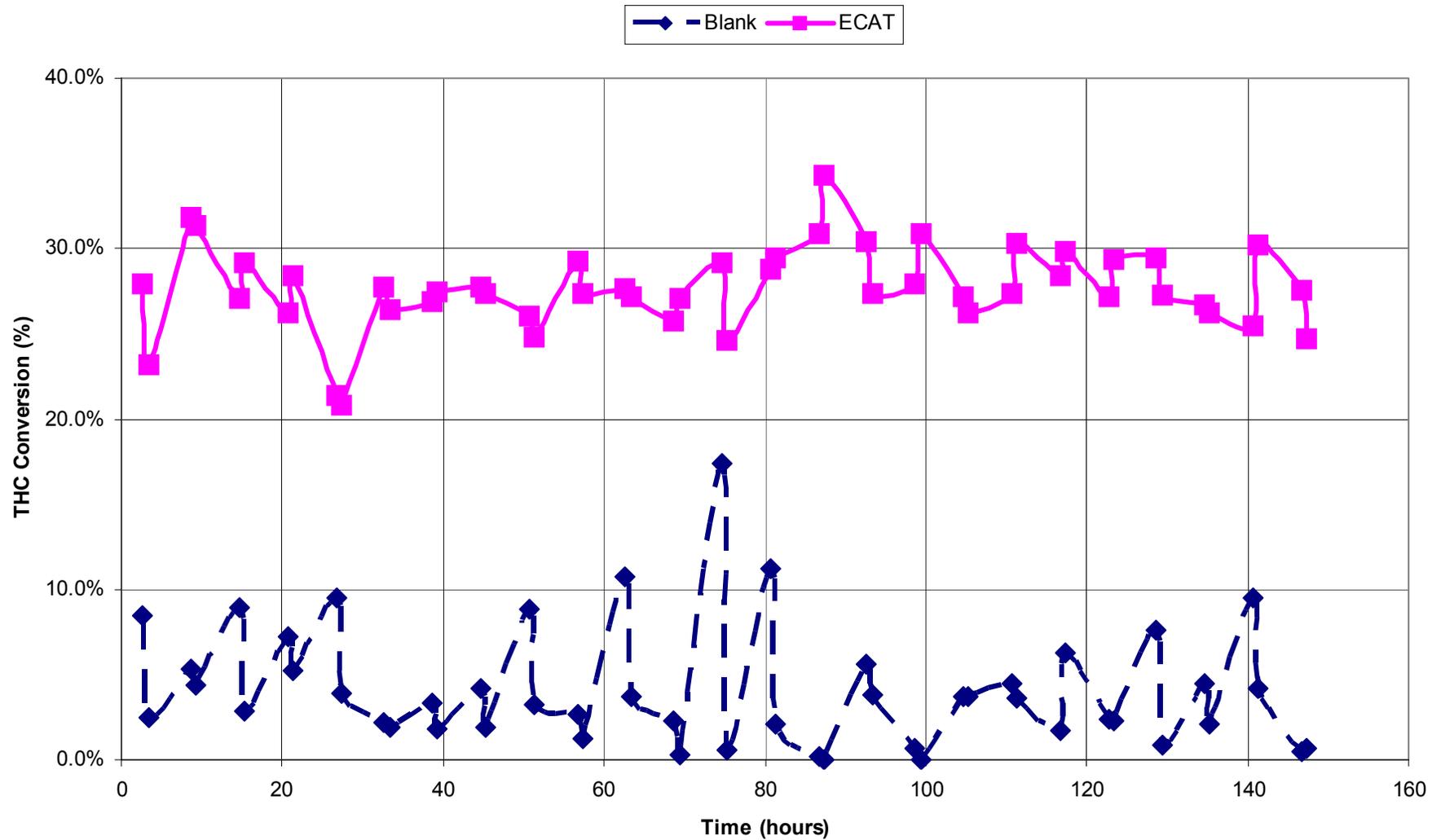
- Cooler fouling test run with ECAT, and with blank ECAT
- ECAT
 - 200 cpsi metallic substrate
 - 60 mm diameter by 90 mm long, 0.24 L volume
 - Oxidation catalyst formulation
 - Compared to coated monolith without PGM
 - Samples prepared for Ford by Johnson Matthey on Emitec substrates
- Test fuel
 - Canadian market 2005-2006 fuel
 - ~400 ppm sulfur
 - 25-30% aromatics



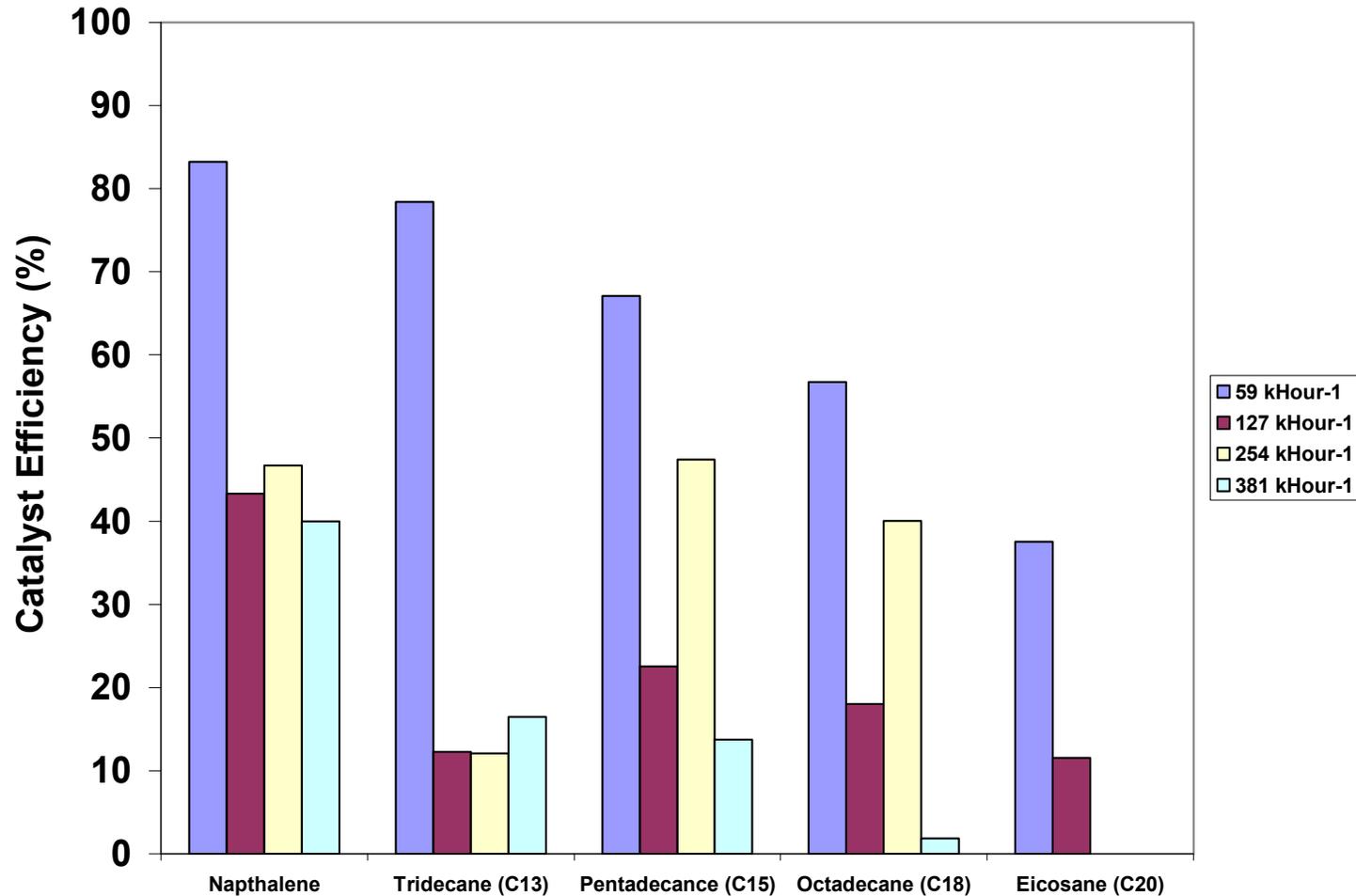
ECAT HC Conversion – A25



ECAT HC Conversion – HSV



ECAT Conversion for Different Species and Space Velocity

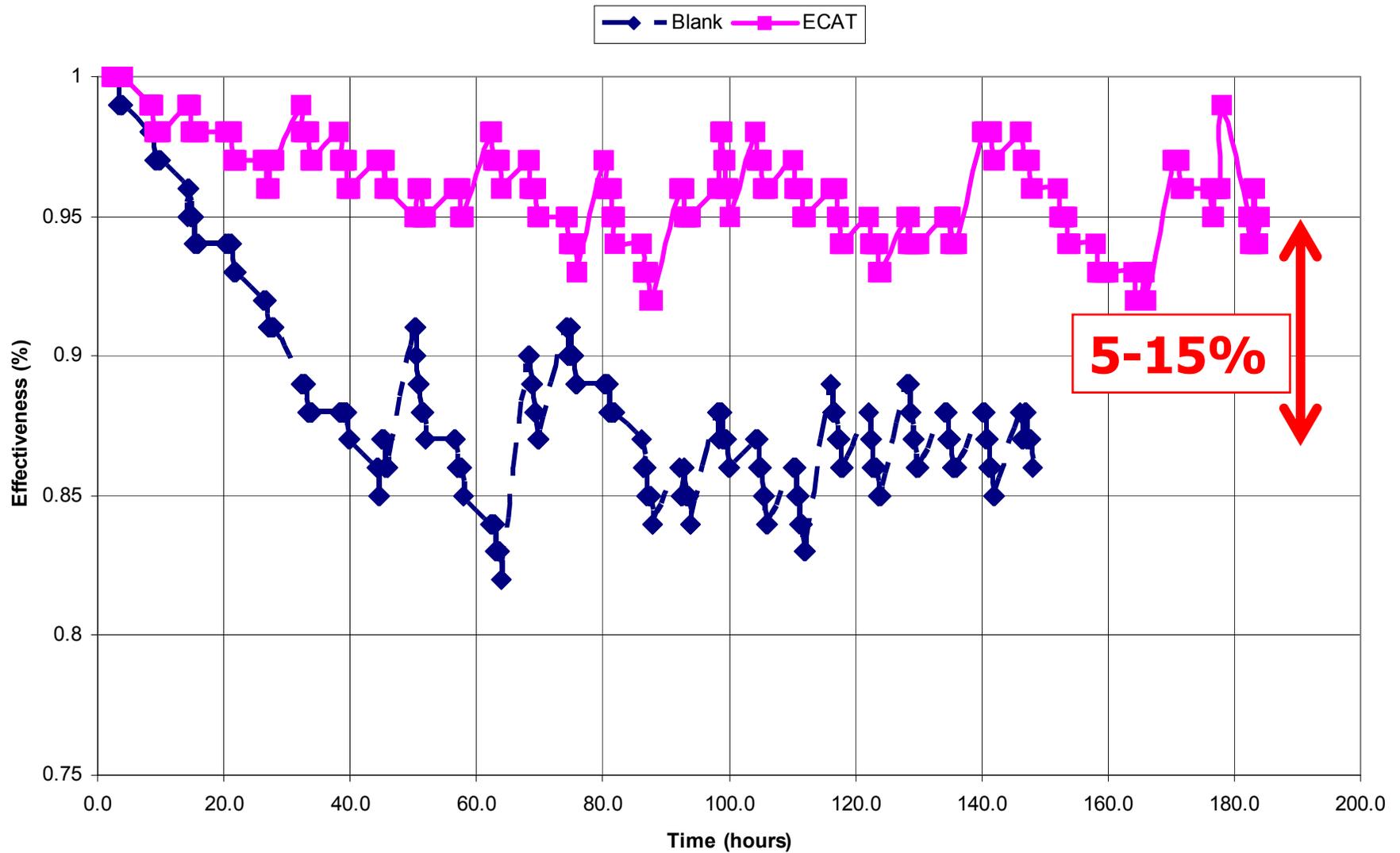


- Reference catalyst and engine
- Testing at ORNL



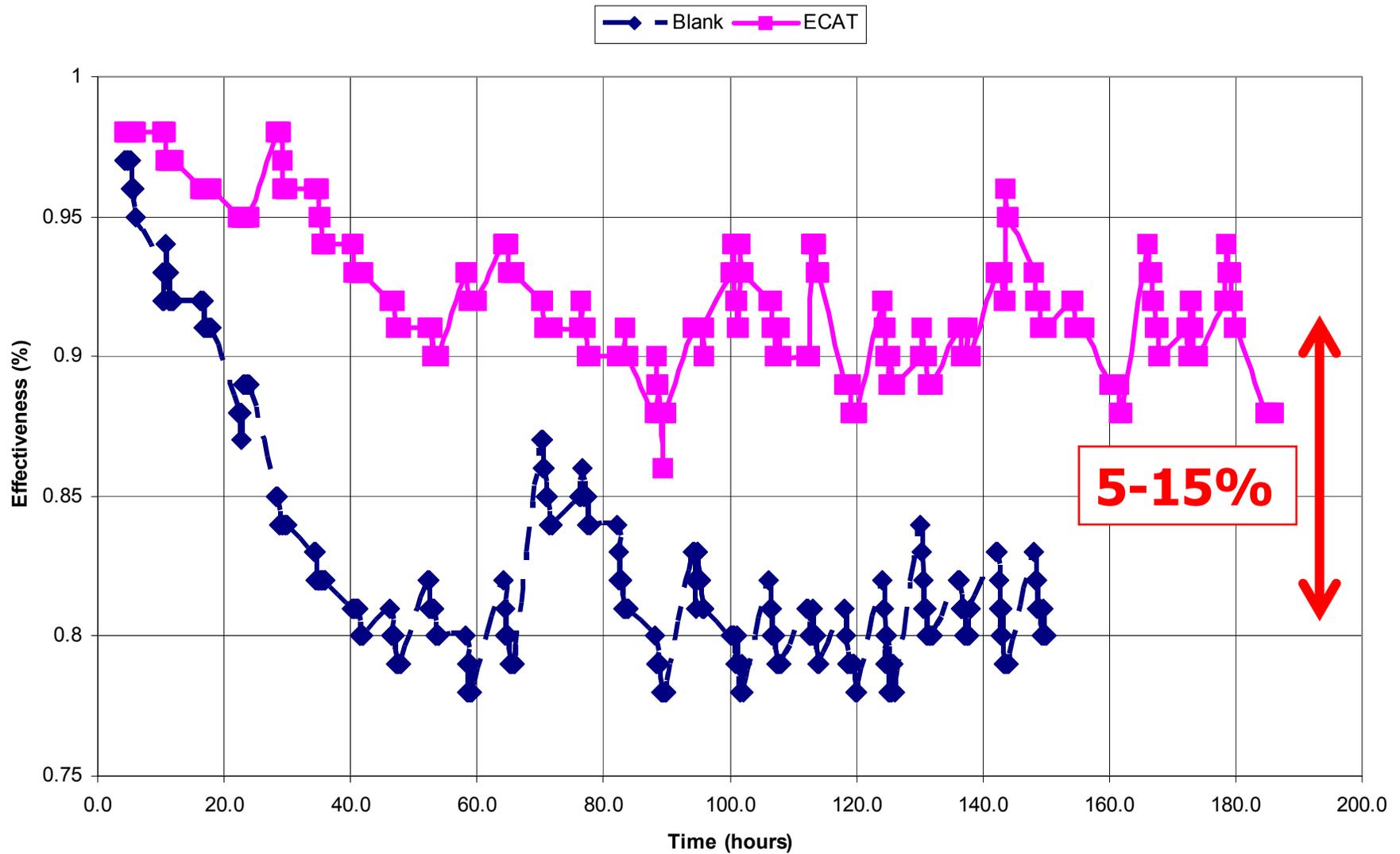
ECAT Effect on Effectiveness – A25

EGR Cooler System Effectiveness @ A25



ECAT Effect on Effectiveness - HSV

EGR Cooler System Effectiveness @ HSV



Cooler Deposit Analyses

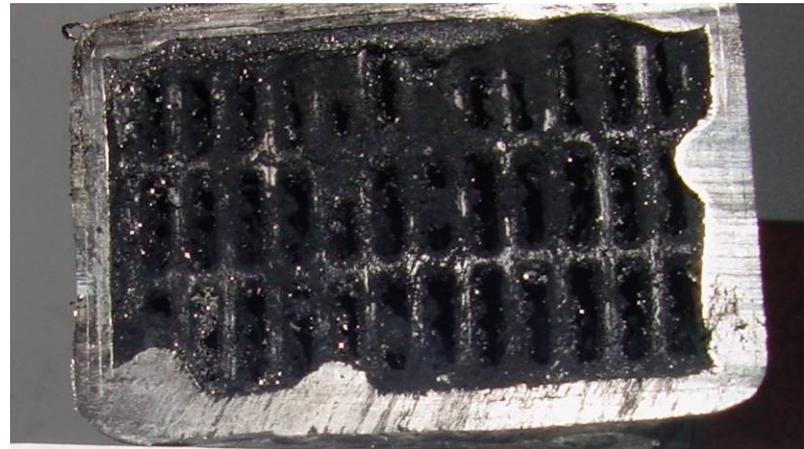
- Following a fouling test with ECAT
- Deposits were analyzed at ORNL



Deposits



Cooler 1 Inlet

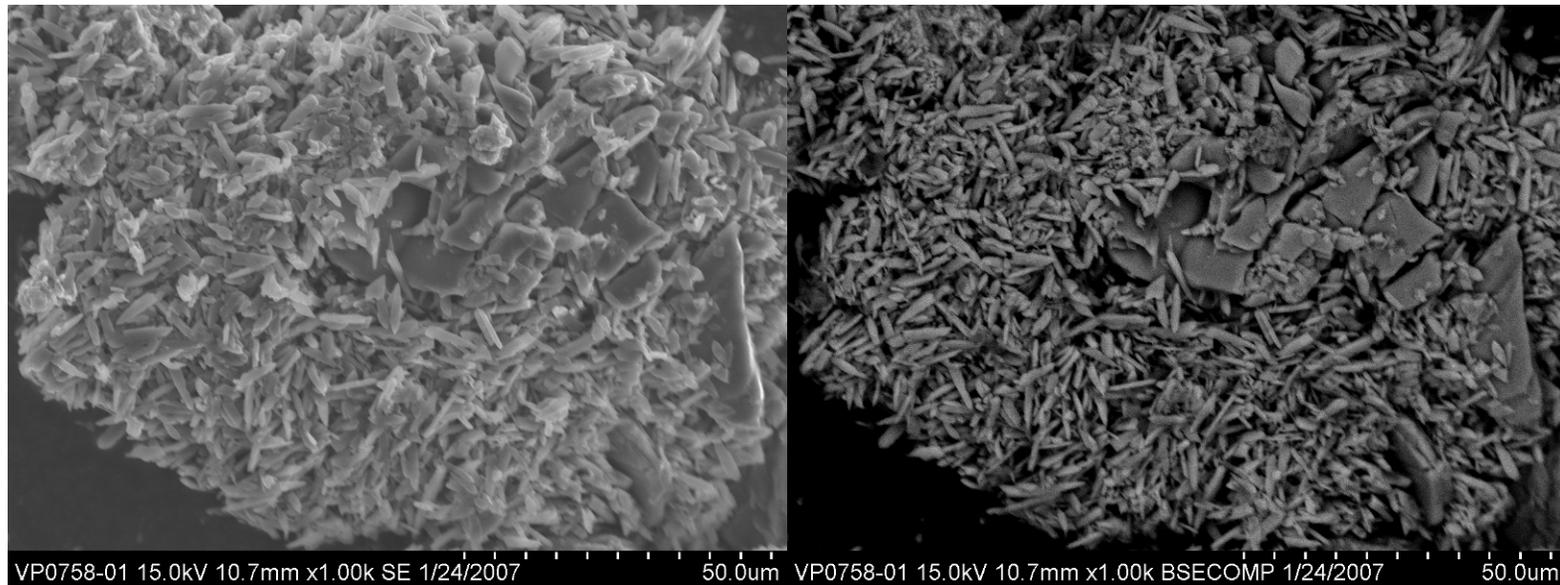


Cooler 2 Inlet

Electron Microscopy of "Ash" Particles Showed Significant Sulfate Fraction

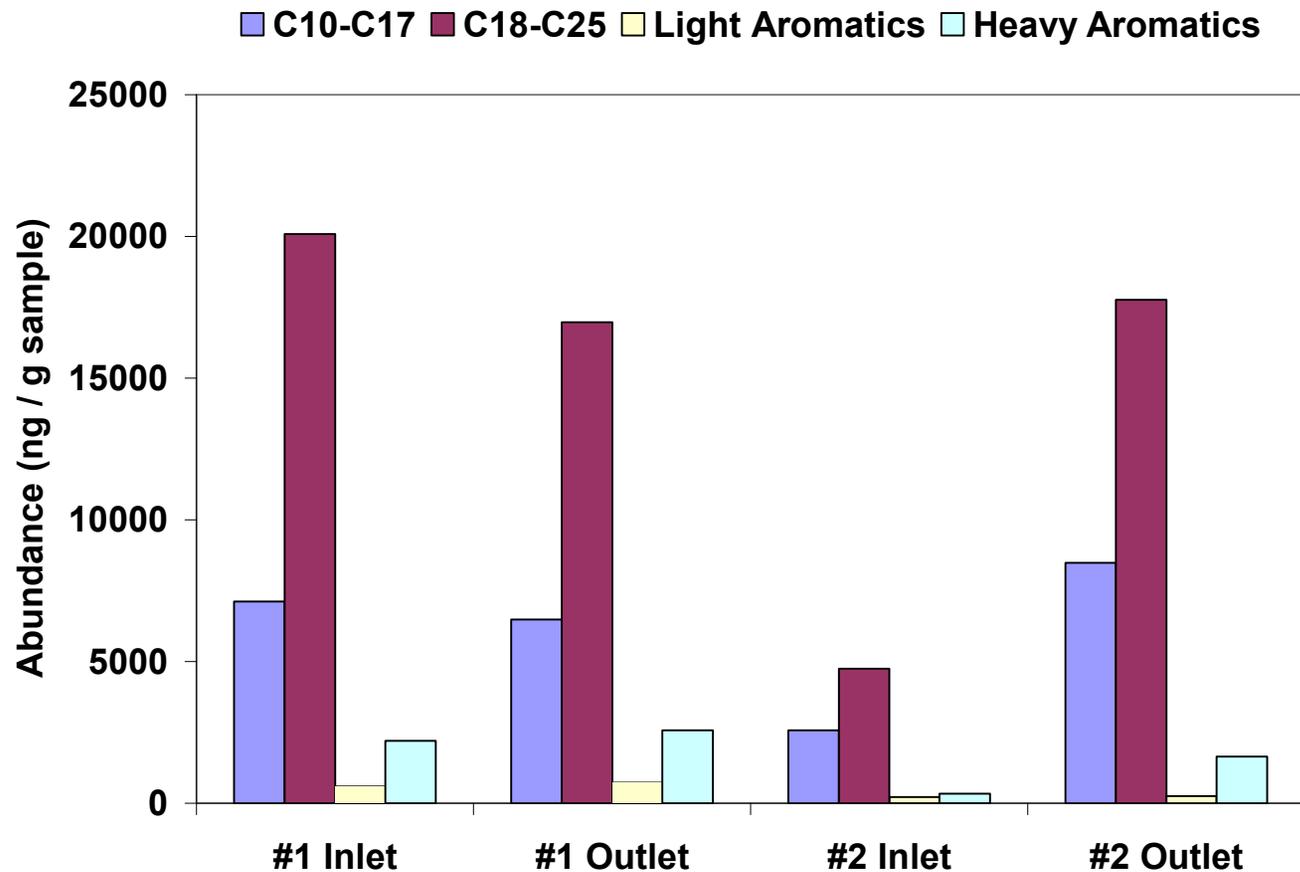
Secondary Electron Image

Back-Scattered Electron image

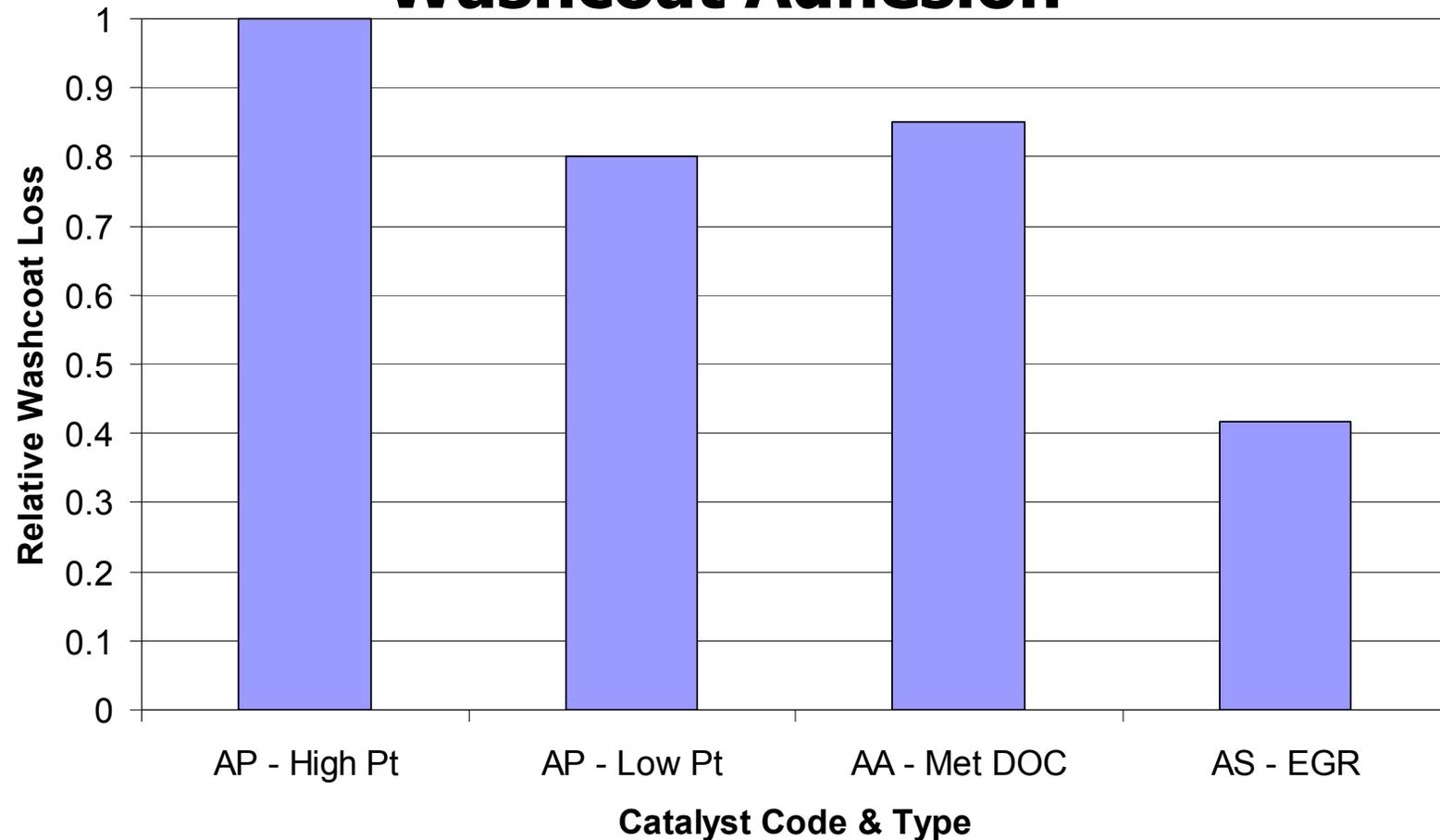


- A significant percentage of the soot was composed of a sulfate phase shown here. The sulfate appeared like grains of rice approximately 5 to 10 microns long and 2 microns wide.
- Consistent with oxidation of fuel sulfur by ECAT; also consistent with fuel sulfur content.

Chemical Extraction and Analysis Showed that Deposit HCs were Dominated by the Heavy Fraction.



Washcoat Adhesion



- Washcoat loss puts powder into engine
- Development improved adhesion
 - "AS" versus earlier designs



Conclusions

- An ECAT has reduced the rate of EGR cooler fouling
- This can be accomplished with a remarkably high SV
- Washcoat adhesion improved



Thanks For Your Attention.....

- Questions?

