

# Overview and Progress of the Batteries for Advanced Transportation Technologies (BATT) Activity

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Project ID: **ES108**

## **Perform cutting-edge research on new materials and address fundamental chemical and mechanical instability issues.**

### **Challenges**

- ❑ Conduct research and development on the next-generation of battery anodes, cathodes, and electrolytes.
- ❑ Understand failure mechanisms to enable higher energy, longer lasting, and less expensive batteries.
- ❑ Conduct comprehensive modeling of cell and material behavior.

## Battery/Energy Storage R&D Funding (\$, M)

<b>FY 2012* Enacted</b>	\$90
<b>FY 2013** Full Year CR</b>	\$88
<b>FY 2014*** (request)</b>	\$170.5

\*FY 2012 SBIR/STTR removed.

\*\*FY 2013 full year CR inclusive of SBIR/STTR.

\*\*\* FY 2014 budget request inclusive of SBIR/STTR.

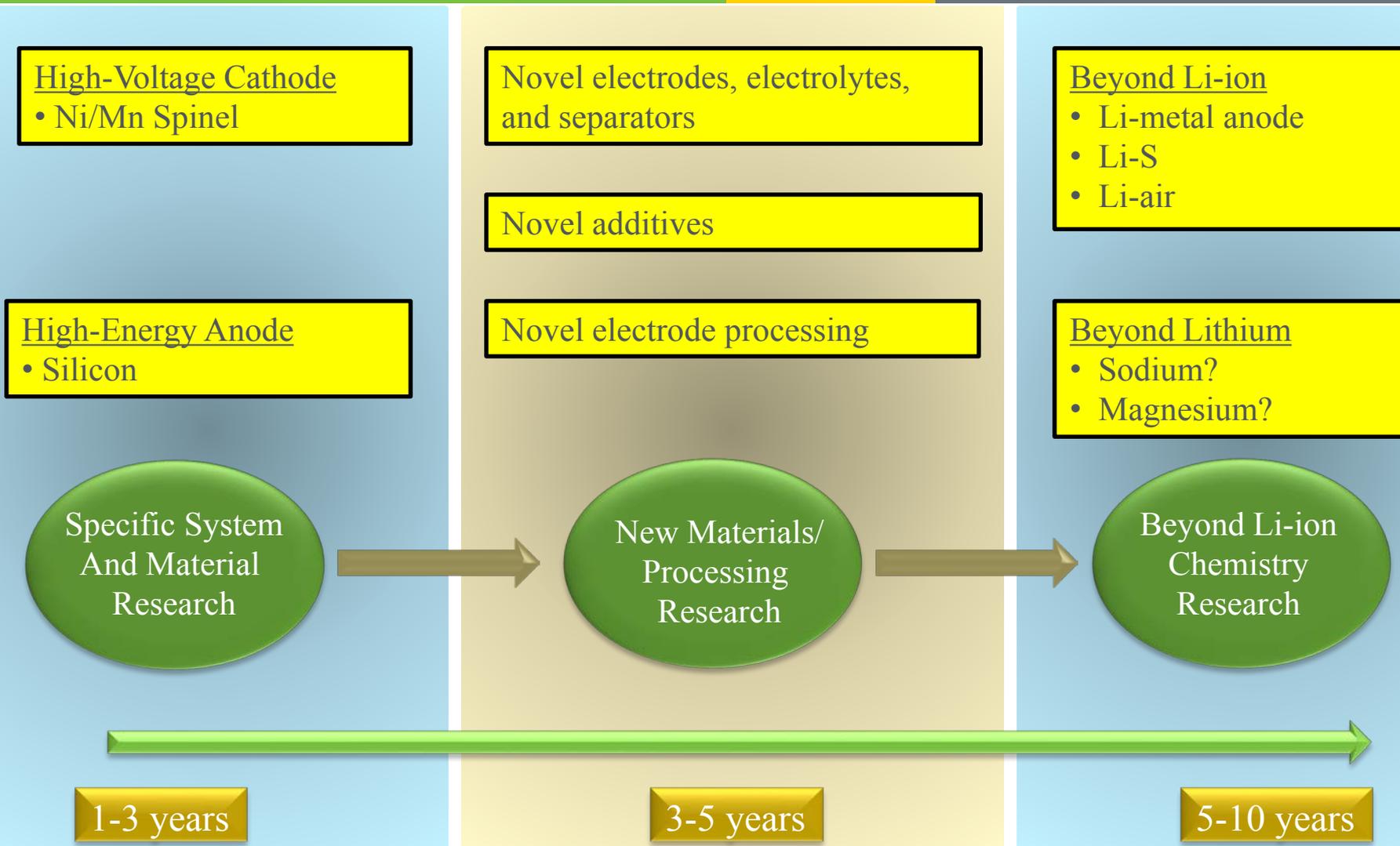
## FY 2013 Energy Storage R&D Budget\*\* (\$88M)



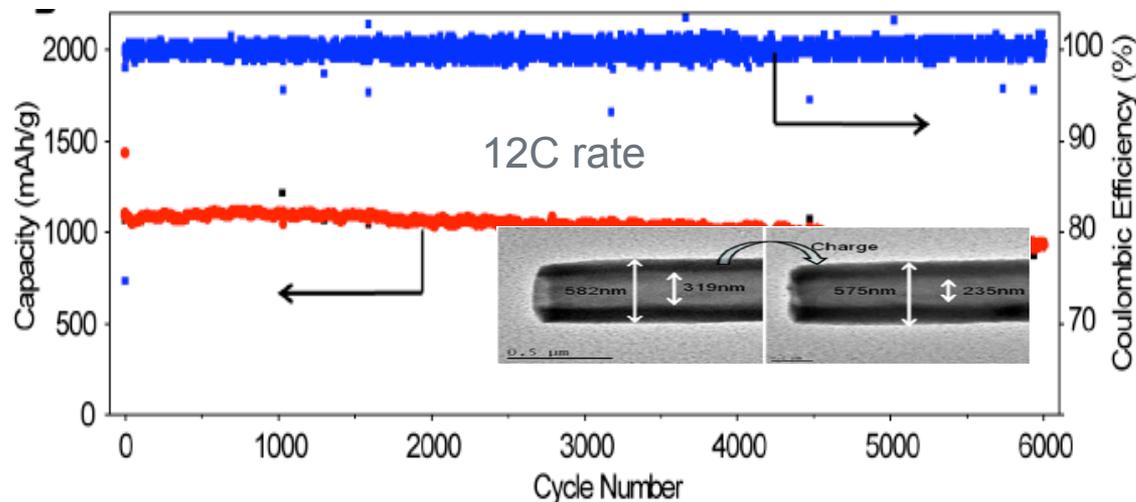
# Participants



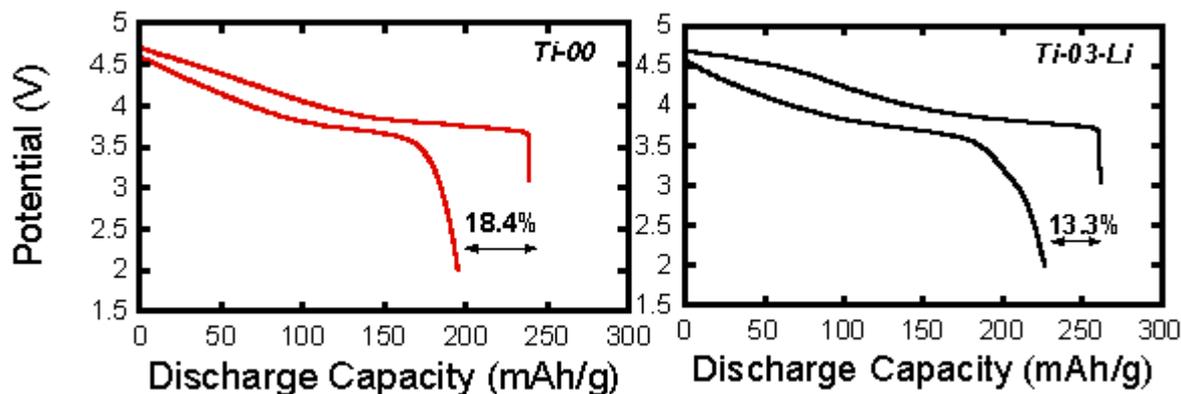
# Current BATT Portfolio



## Electrode Materials

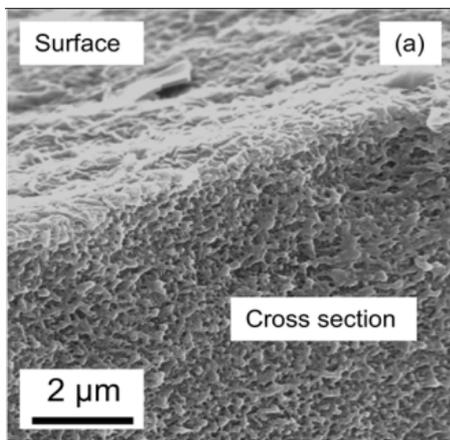


- Silicon nanotube electrodes show potential to eliminate cracking and exhibit excellent cycling capability (Cui, Stanford)



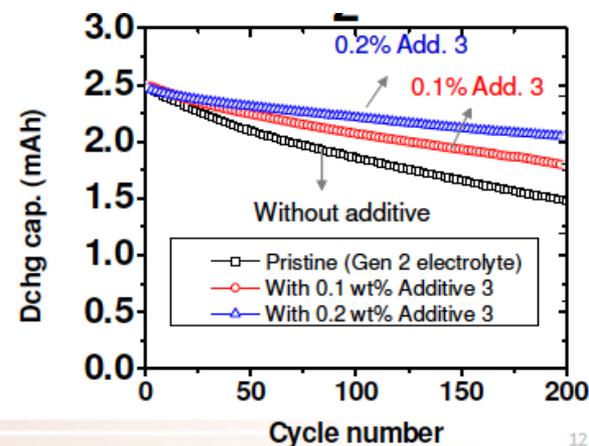
- Ti-doped NMC cathodes show promise with capacities of 225 mAh/g with improved cycling at higher voltages (Doeff, LBNL)

## Electrolytes, Additives, Separators



- Self-assembled separators using a low-cost wet process show promise. Ionic conductivities in membrane comparable to Celgard. (Balsara, LBNL)

- 3-oxabicyclohexane-2,4-dione additive, predicted by modeling, shows promise in improving cyclability by forming a better SEI. (Amine/Curtiss, ANL)



## Advanced Diagnostics, Modeling, and Assembly of Battery Materials and Electrodes

- ❑ 170 white papers received
- ❑ 54 full proposals requested
- ❑ 15 proposals awarded

### Cell Development

Investigator	Institution	Project
Yet-Ming Chiang, Antoni Tomsia	MIT LBNL	<i>Design and Scalable Assembly of High Density Low Tortuosity Electrodes</i>
Gao Liu	LBNL	<i>Hierarchical Assembly of Inorganic/Organic Hybrid Si Negative Electrodes</i>
Karim Zaghib	Hydro-Québec	<i>Assembly of Battery Materials and Electrodes</i>

## Modeling

Investigator	Institution	Project
Gerbrand Ceder	MIT	<i>First Principles Calculations of Existing and Novel Electrode Materials</i>
Kristin Persson	LBNL	<i>Understanding and Predicting Novel Electrode Materials from First-Principles</i>
Dean Wheeler, Brian Mazzeo	Brigham Young University	<i>Predicting Microstructure and Performance for Optimal Cell Fabrication</i>
Yue Qi, Xingcheng Xiao, Huajian Gao, Brian Sheldon, Yang-Tse Cheng	General Motors Brown University University of Kentucky	<i>A Combined Experimental and Modeling Approach for the Design of High Current Efficiency Si Electrodes</i>
Perla Balbuena, Jorge Seminario, Kevin Leung, Susan Rempe	Texas A&M University Sandia National Laboratories	<i>First Principles Modeling of SEI Formation on Bare and Surface/Additive Modified Silicon Anode</i>

## Diagnostics

Investigator	Institution	Project
Clare Grey	University of Cambridge	<i>NMR and Pulse Field Gradient Studies of SEI and Electrode Structure</i>
Robert Kostecki	LBL	<i>In-situ Characterization of Interfaces and Interphases in Electrical Energy Storage Systems with Far- and Near-Field Multiprobe Techniques</i>
Guoying Chen	LBL	<i>Optimization of High-energy Electrode Materials</i>
Nitash Balsara, David Prendergast Jordi Cabana	University of California, Berkeley LBL	<i>Active Binders for High-energy Density Cathode</i>
Xiao-Qing Yang, Kyung-Wan Nam	BNL	<i>Advanced in-situ Diagnostic Techniques for Battery Materials</i>
Gabor Somorjai, Phillip Ross	University of California, Berkeley LBL	<i>Analysis of Film Formation Chemistry on Silicon Anodes by Advanced in-situ and operando Vibrational Spectroscopy</i>
Shirley Meng	UC San Diego	<i>Optimization of Ion Transport in High-energy Composite Cathodes</i>

## □ **Basic Energy Sciences**

- Energy Frontier Research Centers
- Joint Center for Energy Storage Research
  - Multivalent Intercalation
  - High-throughput calculations and datamining of advanced electrolytes

## □ **ARPA-E**

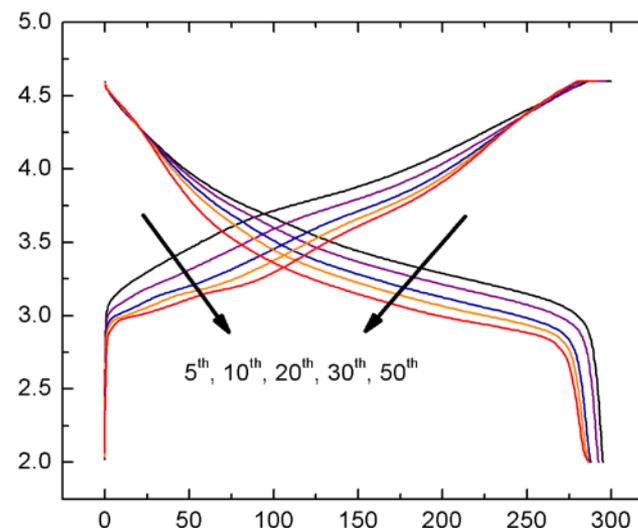
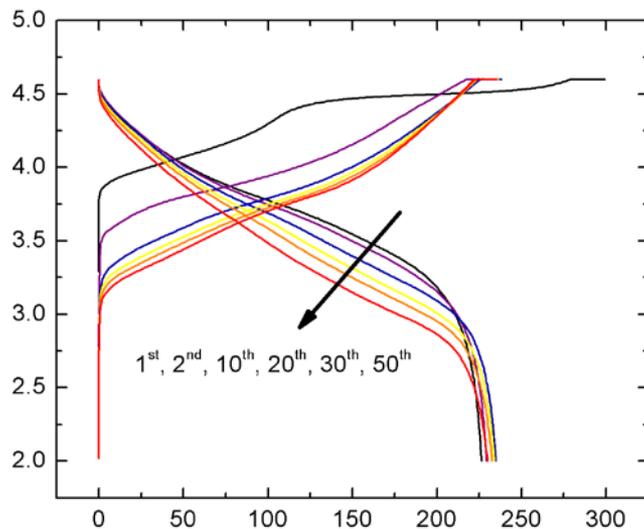
- Lithium/Air and Lithium/Sulfur chemistries

- Solicit new proposals for Novel Electrolytes and Additives
  - Primary focus on single ion conducting membranes
  - Request for concepts papers announced: January 2014

## Issues

- ❑ Advanced anode materials (Sn and Si) have large storage capacities ( $>1,000$  mAh/g), but they undergo a significant volume change
  - Have large irreversible capacity loss.
  - Consume electrolyte due to a continuing forming of the SEI layer.
- ❑ Stability of Li-rich, layered-cathode ( $x\text{Li}_2\text{MnO}_3 \cdot (1-x)\text{LiMO}_2$ ), spinels and phosphates at high operating voltages remains problematic
  - Current electrolytes are not stable above 4.3V.
  - Metal dissolution resulting in poor performance and short cycle life.
  - Structure of the material changes during cycling causing voltage fade.

Voltage profiles of  $\text{Li}/0.5\text{Li}_2\text{MnO}_3 \cdot 0.5\text{LiNi}_{0.44}\text{Co}_{0.25}\text{Mn}_{0.31}\text{O}_2$  cells cycled at C/15 rate



## Approach

- Current research includes new synthesis routes, development of novel electrolytes and additives, and applying coatings on particles and electrodes.

Resolving these issues would make feasible an advanced lithium-ion cell with specific energies of 300–350 Wh/kg

Beyond Li-ion chemistries would likely require use of metallic lithium

## Li-Metal Anode Issues

- ❑ Dendrite formation resulting in loss of lithium and possible safety hazard
- ❑ Solvent reduction also resulting in loss of lithium and electrolyte

## Approach

- ❑ Decouple lithium metal from cathode chemistry
  - Block co-polymers
  - Single-ion conducting ceramics
  - Multiple ceramic/polymer stabilization layers

Protection of lithium metal surface from chemical interactions is critically important for beyond Li-ion batteries

## Not many options for high-capacity cathodes – still the limiting electrode

- ❑ Non-intercalation, low-cost, and high-capacity cathode materials
  - $\text{LiV}_3\text{O}_8$ : 280 mAh/g,  $\text{V}_2\text{O}_5$ : 441 mAh/g, and others (TBD)
  - Sulfur: 1,675 mAh/g
  - $\text{O}_2$ : still have reservations on air electrode

## Sulfur Cathode Issues

- ❑ Dissolution of lithium polysulfides leading to high self-discharge
- ❑ Insoluble sulfur species such as  $\text{Li}_2\text{S}_2$  and  $\text{Li}_2\text{S}$  could passivate the electrodes
- ❑ Practical specific energy is 2 times better, but energy density is comparable to Li-ion battery

## Approach

- ❑ Confining the polysulfides in core shell or yolk shell nanostructures
- ❑ New solvents, additives and dual electrolyte systems