

# Design of PHEVs and Electrolyte Properties

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# Outline

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- Electrolyte Properties (with Sarah Stewart)
  - Transport and thermodynamic property measurements
  - Collaborators: Vince Battaglia
- Design of PHEVs (with Paul Albertus and Jeremy Coutts)
  - Collaborator: Venkat Srinivasan

# Purpose of Work

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- Measure transport and thermodynamic properties of electrolytic solutions
  - Use a variety of appropriate experimental techniques
- Explore how battery size and capacity usage depend on
  - Cell chemistry and design
  - Separator area and driving distance

# Barriers

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- Electrolyte properties
  - Poor transport properties
  - A lack of knowledge of thermodynamic properties
  - A need for improved experimental methods for characterizing physical properties
- Design of HEVs and PHEVs
  - Cost (\$/system, or \$/kWh)
  - Limited capacity usage, large battery size
  - A lack of simple guidelines for evaluating different chemistries

# Approach

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- Develop improved experimental methods for measuring transport and thermodynamic properties.
  - Restricted diffusion to measure diffusion coefficients
  - AC impedance for conductivity
  - Melting-point depression for activity coefficients
- Develop mathematical models to improve understanding of limitations in cell performance
  - A simplified, 0-D model, for basic insights
  - A detailed battery and vehicle model for examination of the complexities

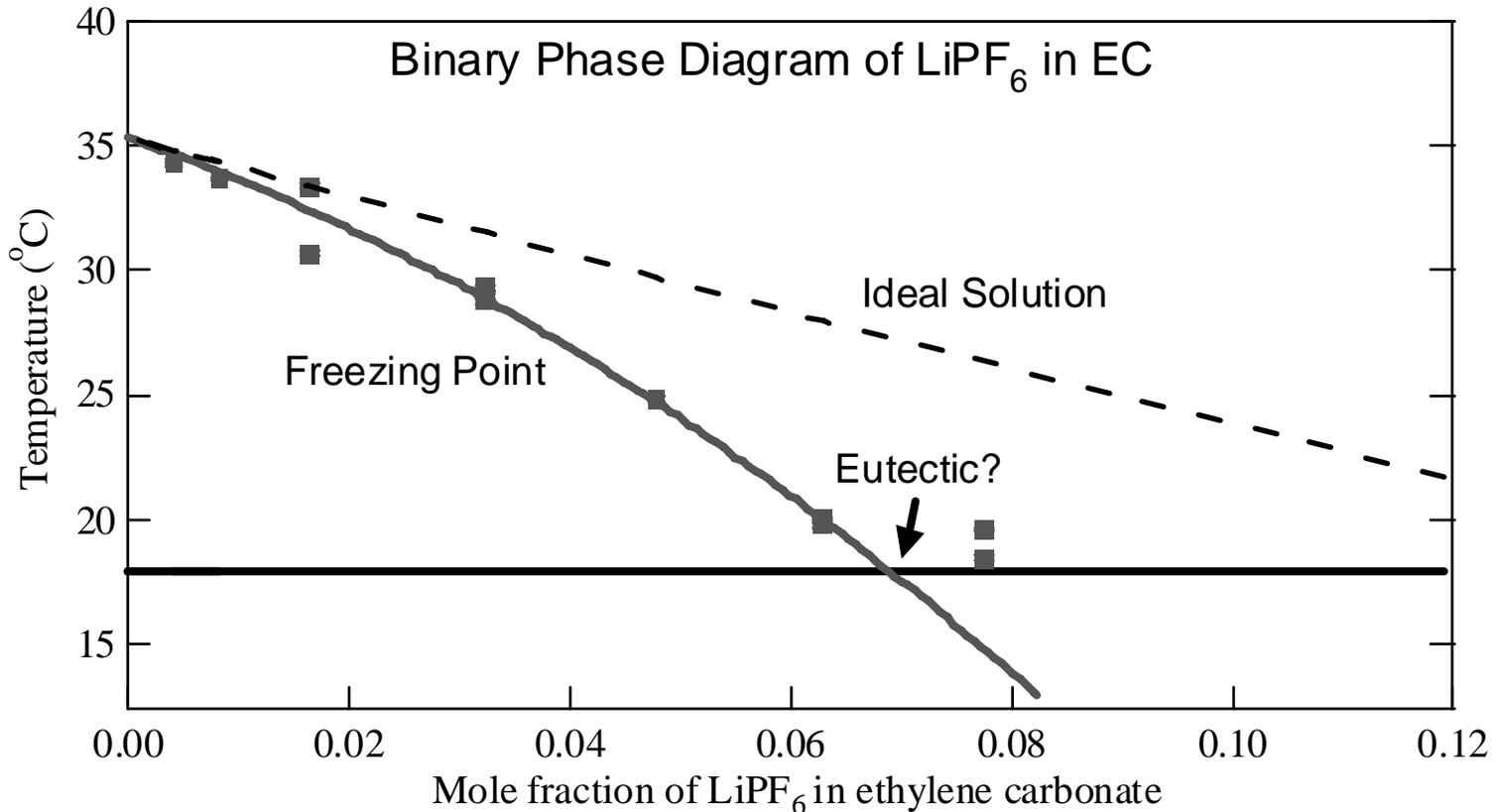
# Performance Measures

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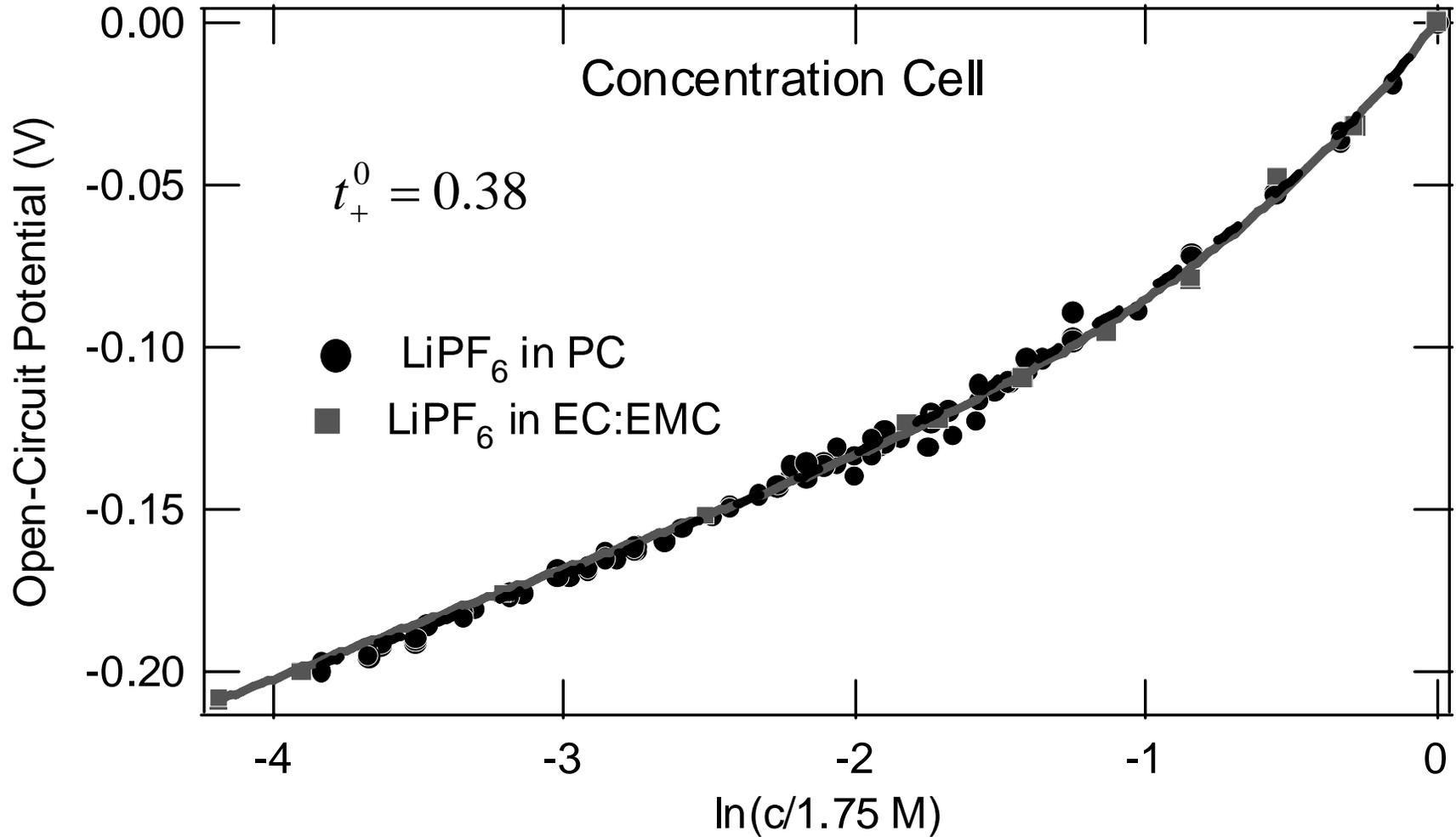
- Transport and thermodynamic properties
  - Pulse discharge and regenerative power
  - Round-trip energy efficiency
- Design of HEVs and PHEVs
  - Available energy goals
  - Discharge throughput
  - Cycle life

# Results – Electrolyte Properties

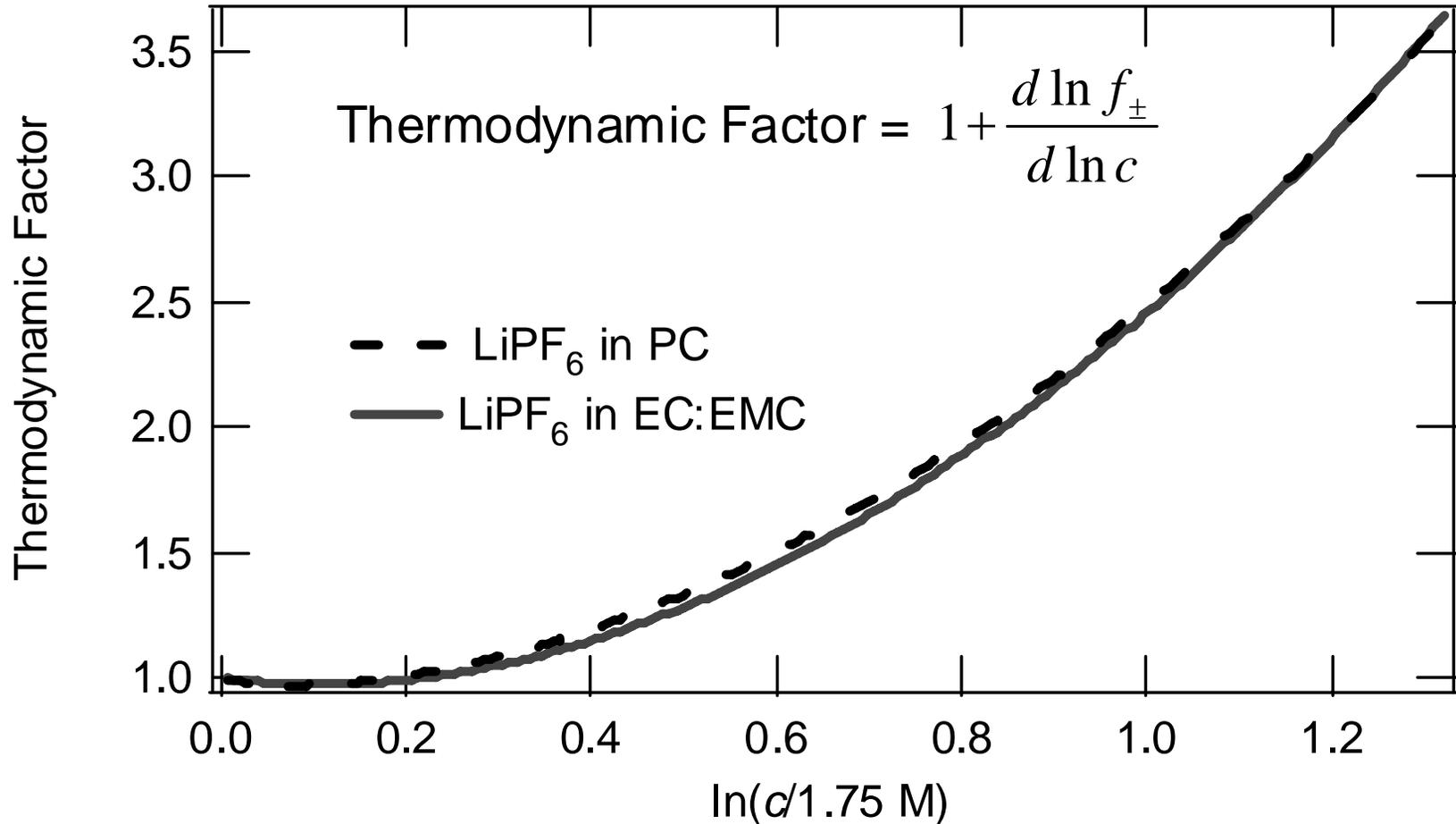
- Previous work focused on measuring diffusion coefficients using a restricted diffusion cell, and conductivity using AC impedance
- Recent work has focused on measuring the activity coefficients through the use of melting-point depression experiments



# Results - Electrolyte Properties



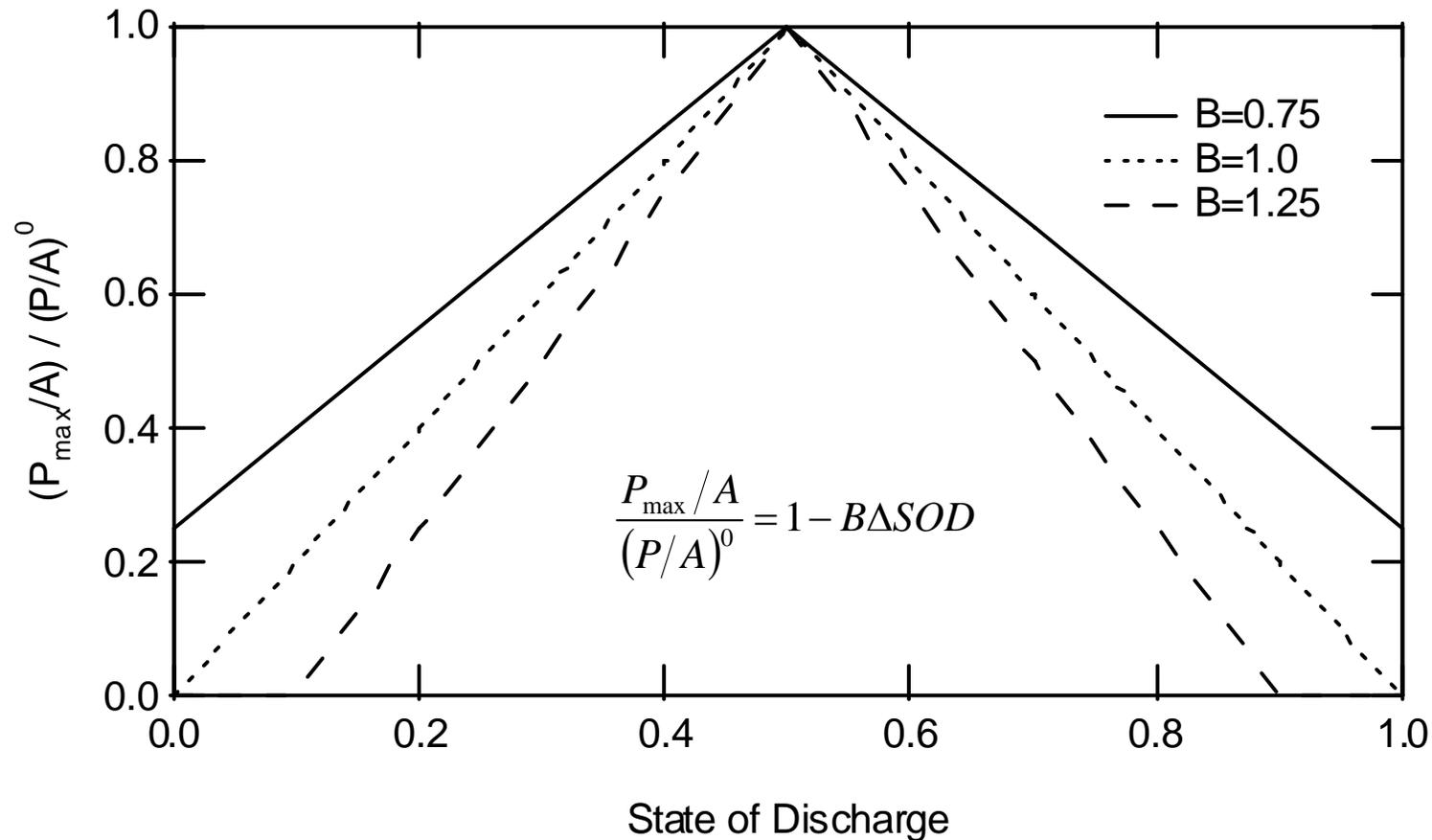
# Results – Electrolyte Properties



- The thermodynamic factor is a measure of solution nonideality

# Results – Design of PHEVs

- Develop a simplified model to characterize performance in HEVs and PHEVs

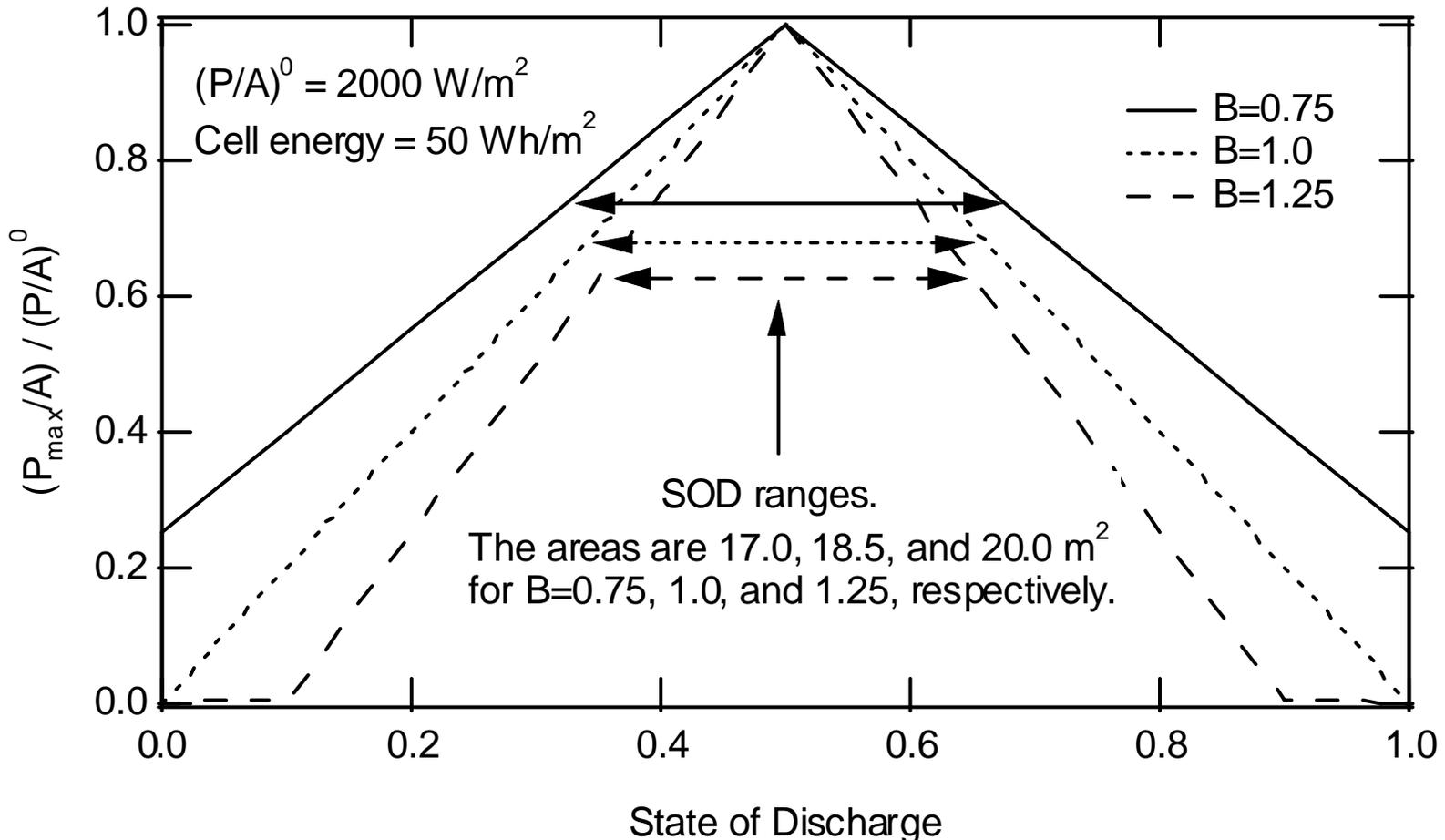


$P_{\max}$  is the maximum-power requirement

B is a parameter to describe the shape of the pulse-power capability

# Results – Design of HEVs

- HEV battery with a power requirement of 25 kW and an energy requirement of 300 Wh ( $P/E = 83 \text{ h}^{-1}$ )



# Results – Design of PHEVs

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Power capability

$$\frac{P_{\max}/A}{(P/A)^0} = 1 - B\Delta SOD$$

Energy requirement

$$E = E'L = A \langle V \rangle Q\Delta SOD$$

Combine and solve

$$B\Delta SOD = \frac{x}{1+x}$$

$$\frac{A}{P_{\max}/(P/A)^0} = 1+x$$

$$x = \frac{E'LB(P/A)^0}{P_{\max}Q\langle V \rangle}$$

$x$  is a dimensionless energy-to-power ratio

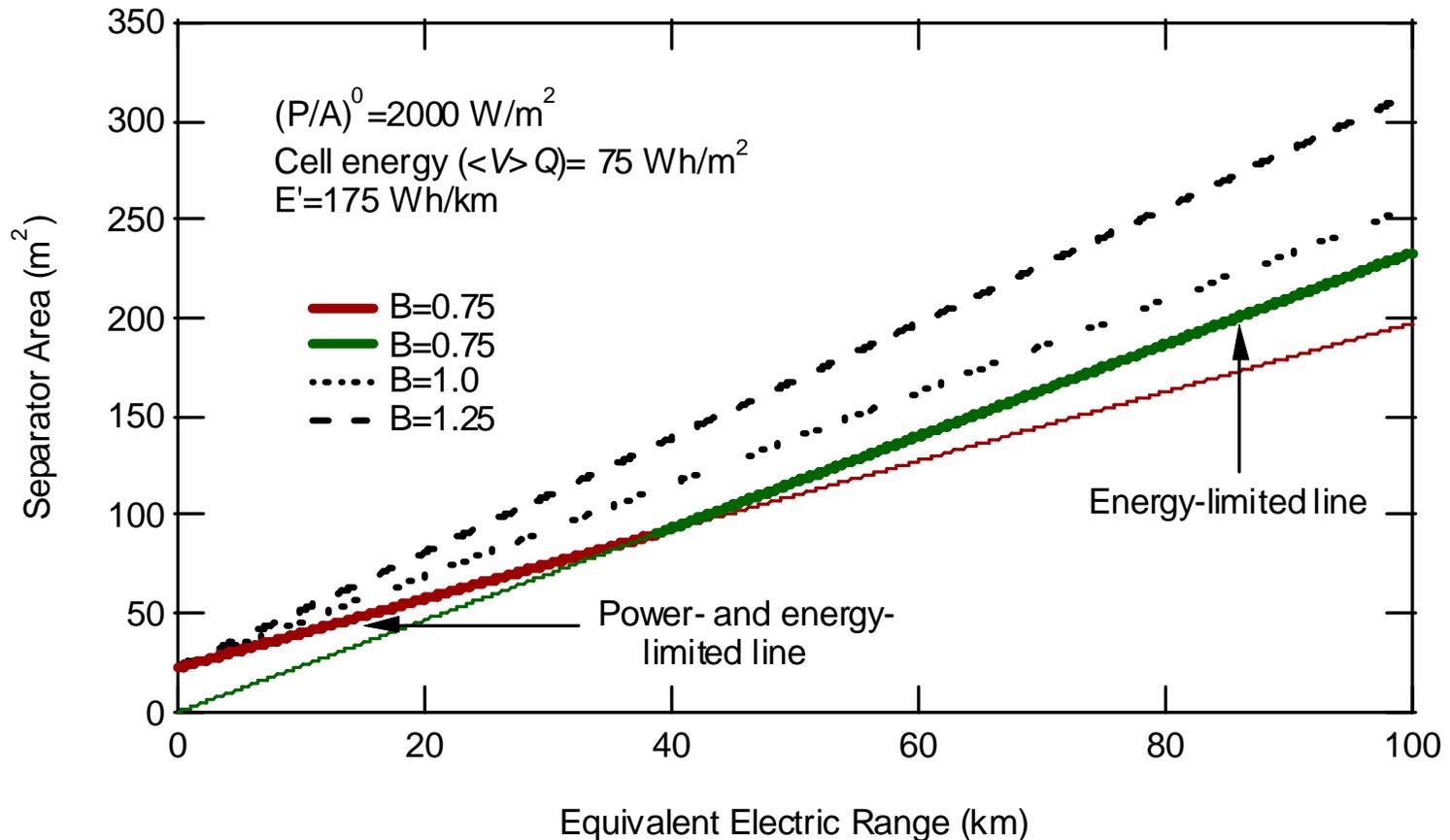
# Results – Design of PHEVs

Power capability

$$\frac{P_{\max}/A}{(P/A)^0} = 1 - B\Delta SOD$$

Energy requirement

$$E = E'L = A \langle V \rangle Q \Delta SOD$$



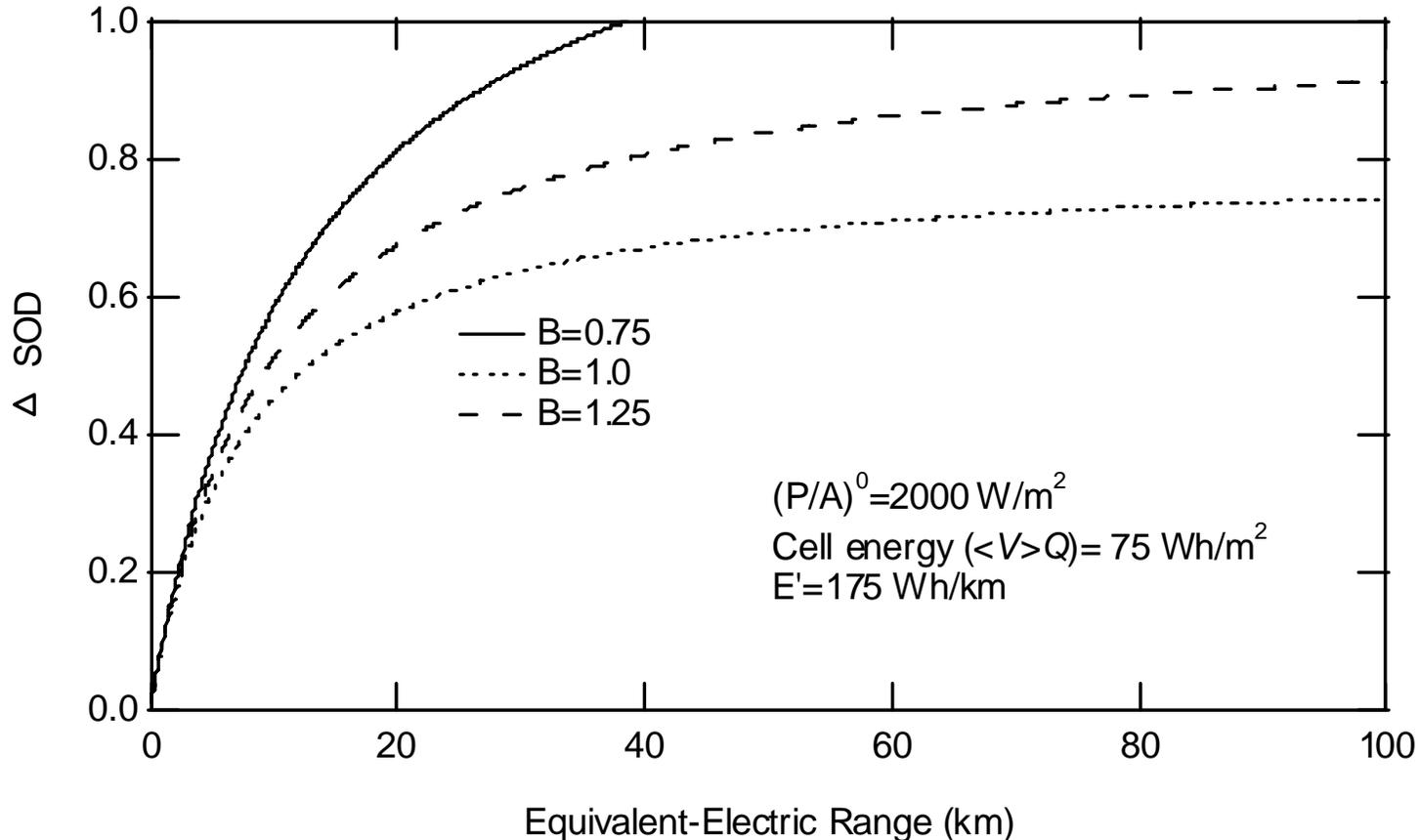
# Results – Design of PHEVs

Power capability

$$\frac{P_{\max}/A}{(P/A)^0} = 1 - B\Delta SOD$$

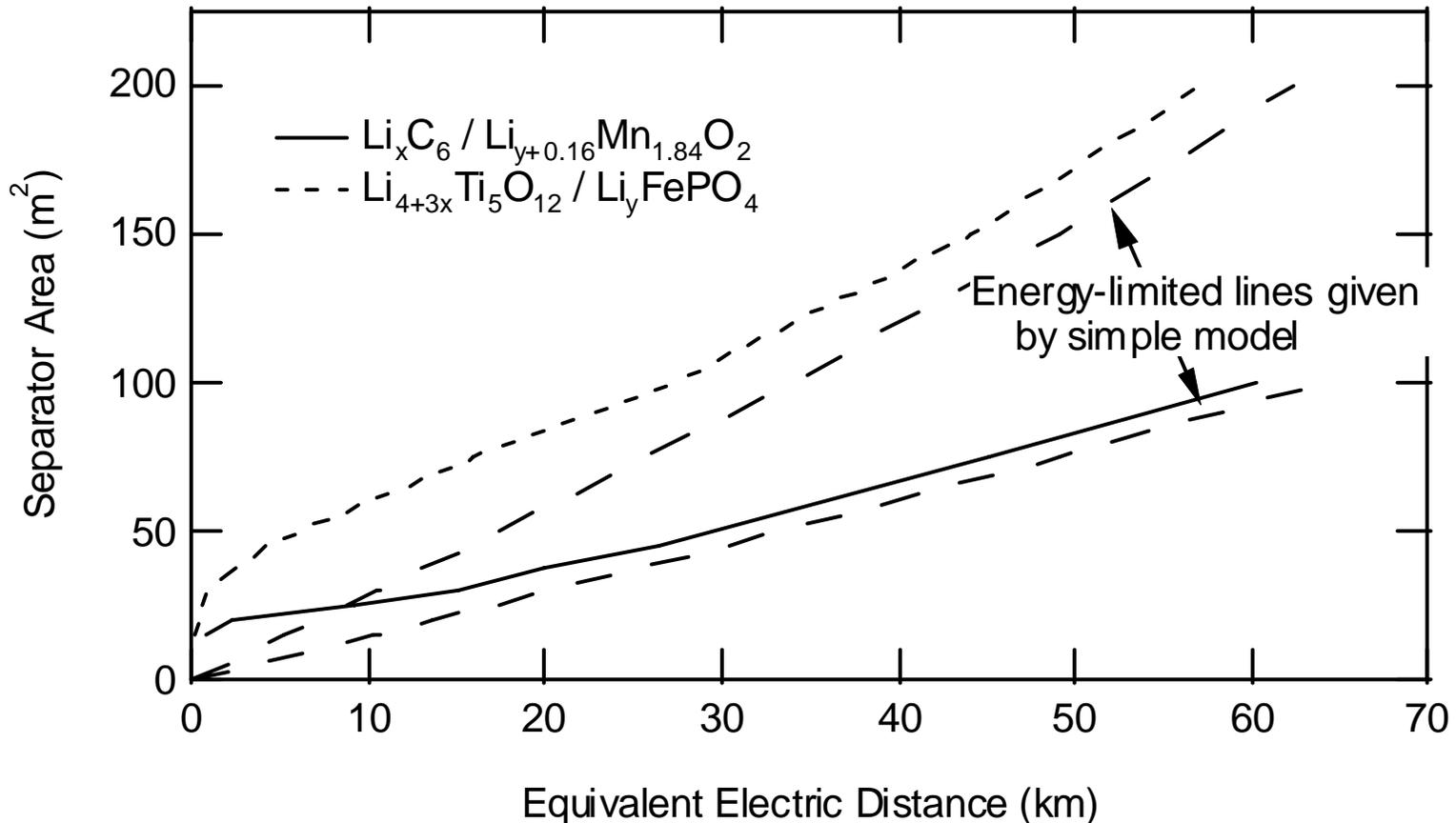
Energy requirement

$$E = E'L = A \langle V \rangle \Delta SOD Q$$



# Results – Design of PHEVs

## Detailed Model (Dualfoil and Vehicle)



- Combined model captures complexities of real chemistries

# Future Work

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- Finalize modeling work and finish write-up of the HEV and PHEV work
- Modeling spurious lithium deposition
  - Lithium deposition at the edges of electrodes
  - Use 1-D and 2-D models to understand the conditions in which deposition may occur

# Summary

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- Measurements of the transport and thermodynamic properties of  $\text{LiPF}_6$  in several solvents have been carried out
- A simplified model has been developed to provide insights on the design of HEVs and PHEVs
- A combined battery and vehicle model has been used to provide accurate and comprehensive prediction of behavior in HEVs and PHEVs, for design

# Publications

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- Stewart, S., Newman, J., “The Use of UV/vis Absorption to Measure Diffusion Coefficients in  $\text{LiPF}_6$  Electrolytic Solutions,” J. Electrochem. Soc., 155 (1) F13-F16.
- Stewart, S., Newman, J. “Measuring the Salt Activity Coefficient in Lithium-Battery Electrolytes,” J. Electrochem. Soc., Accepted.
- Stewart, S. “Determination of Transport Properties and Optimization of Lithium-Ion Batteries.” Ph.D. Dissertation, University of California, Berkeley, 2007
- Albertus, P., Newman, J.. “I. A Simplified Model for Determining Capacity Usage and Battery Size for HEVs and PHEVs,” In preparation.
- Albertus, P., Coutts, J., Srinivasan, V., Newman, J. “II. A Combined Model for Determining Battery Size and Capacity Usage of the  $\text{Li}_x\text{C}_6 / \text{Li}_{y+0.16}\text{Mn}_{1.84}\text{O}_4$  and  $\text{Li}_{4+3x}\text{Ti}_5\text{O}_{12} / \text{Li}_y\text{FePO}_4$  Chemistries for HEV and PHEV Applications,” In preparation.