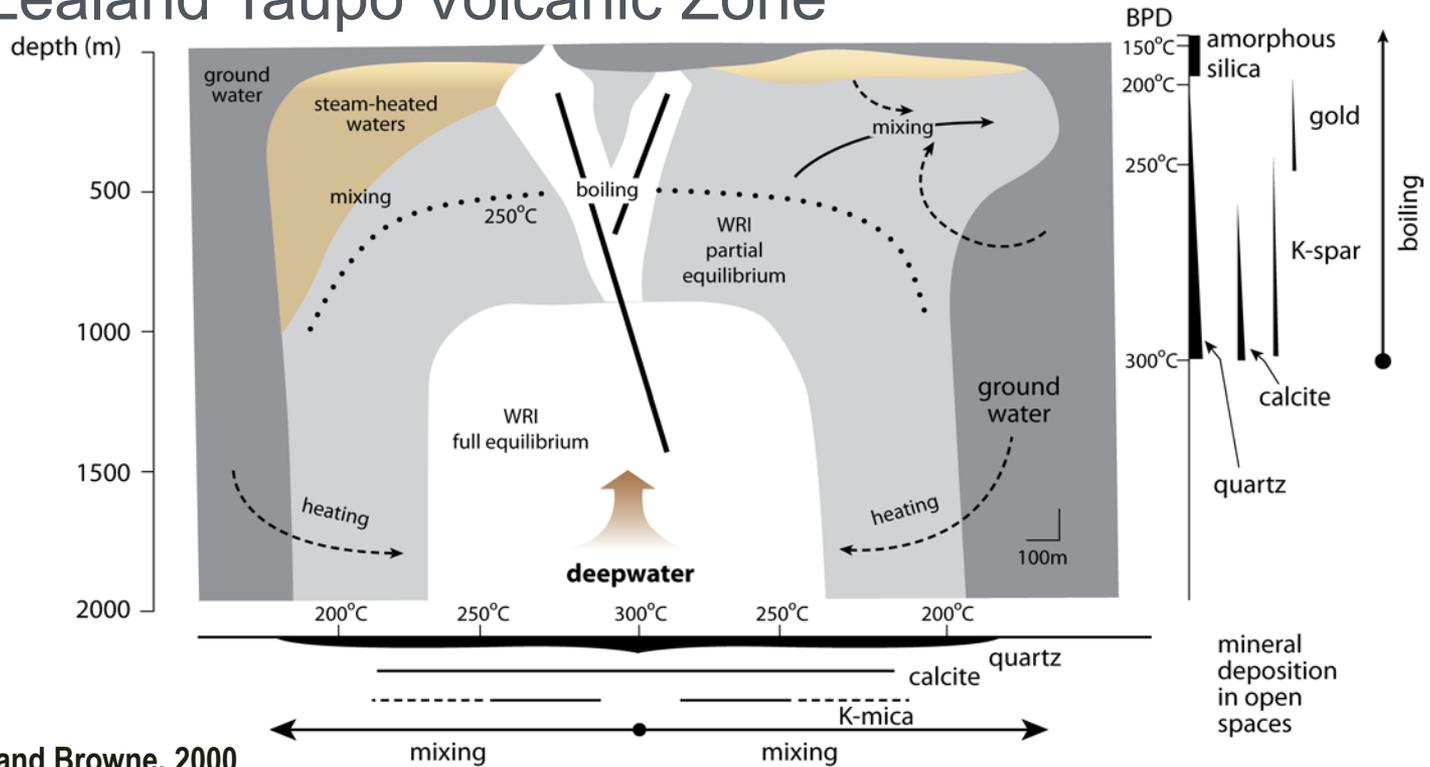


New Zealand Taupo Volcanic Zone



Simmons and Browne, 2000

Chemical Impact of Elevated CO₂ on Geothermal Energy Production

May 18-20, 2010

Principal Investigator

Susan Carroll

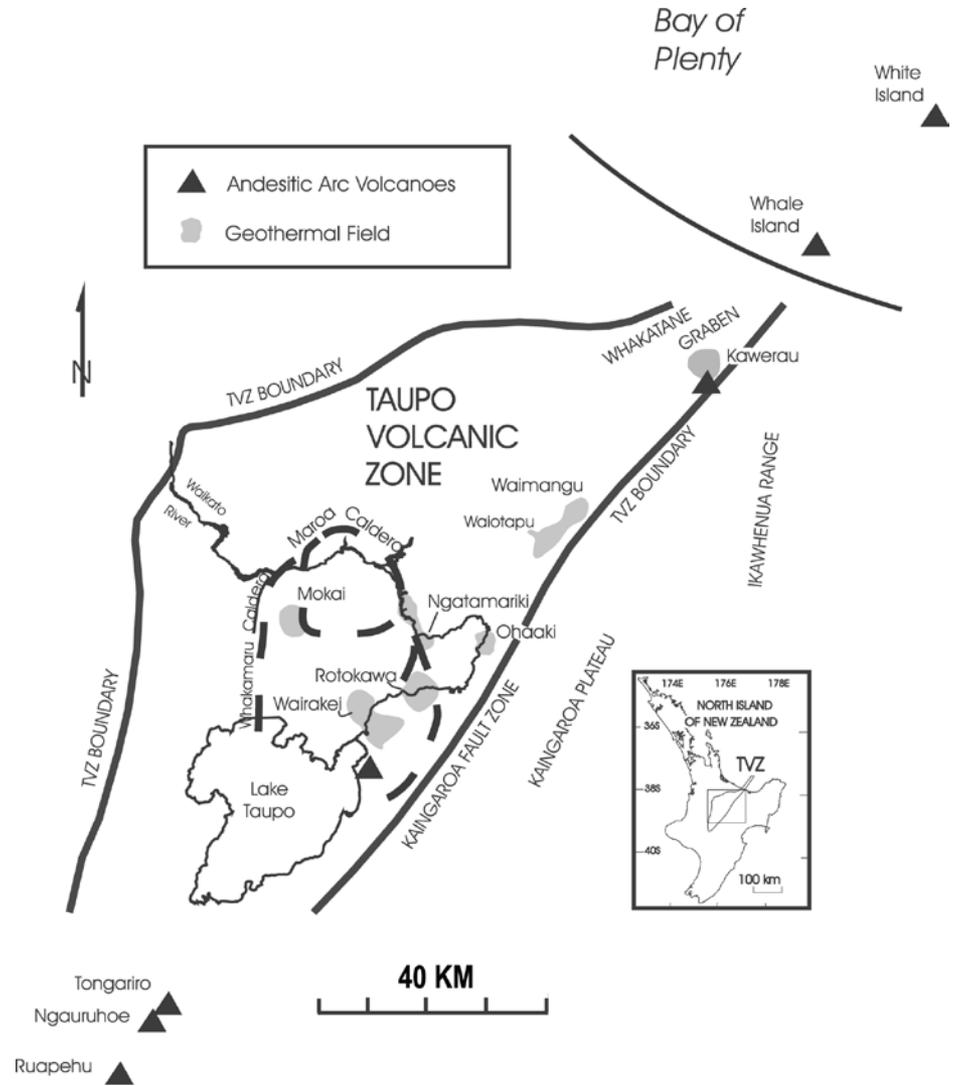
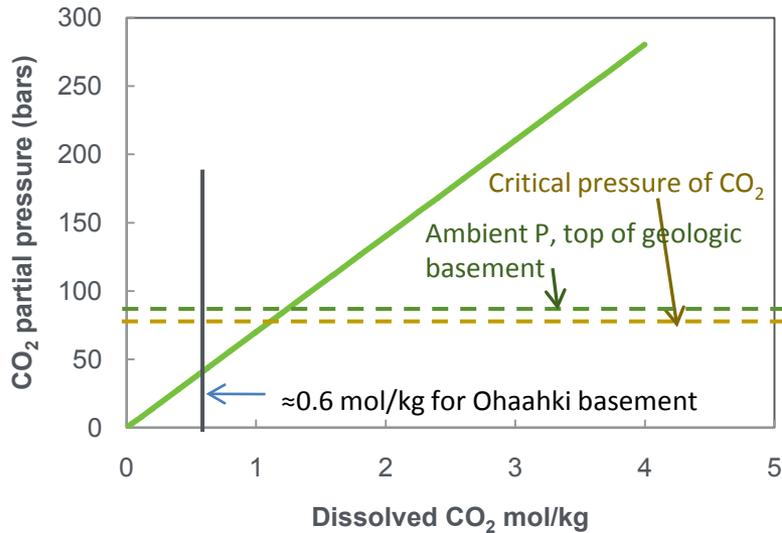
Lawrence Livermore National Lab

Track Name

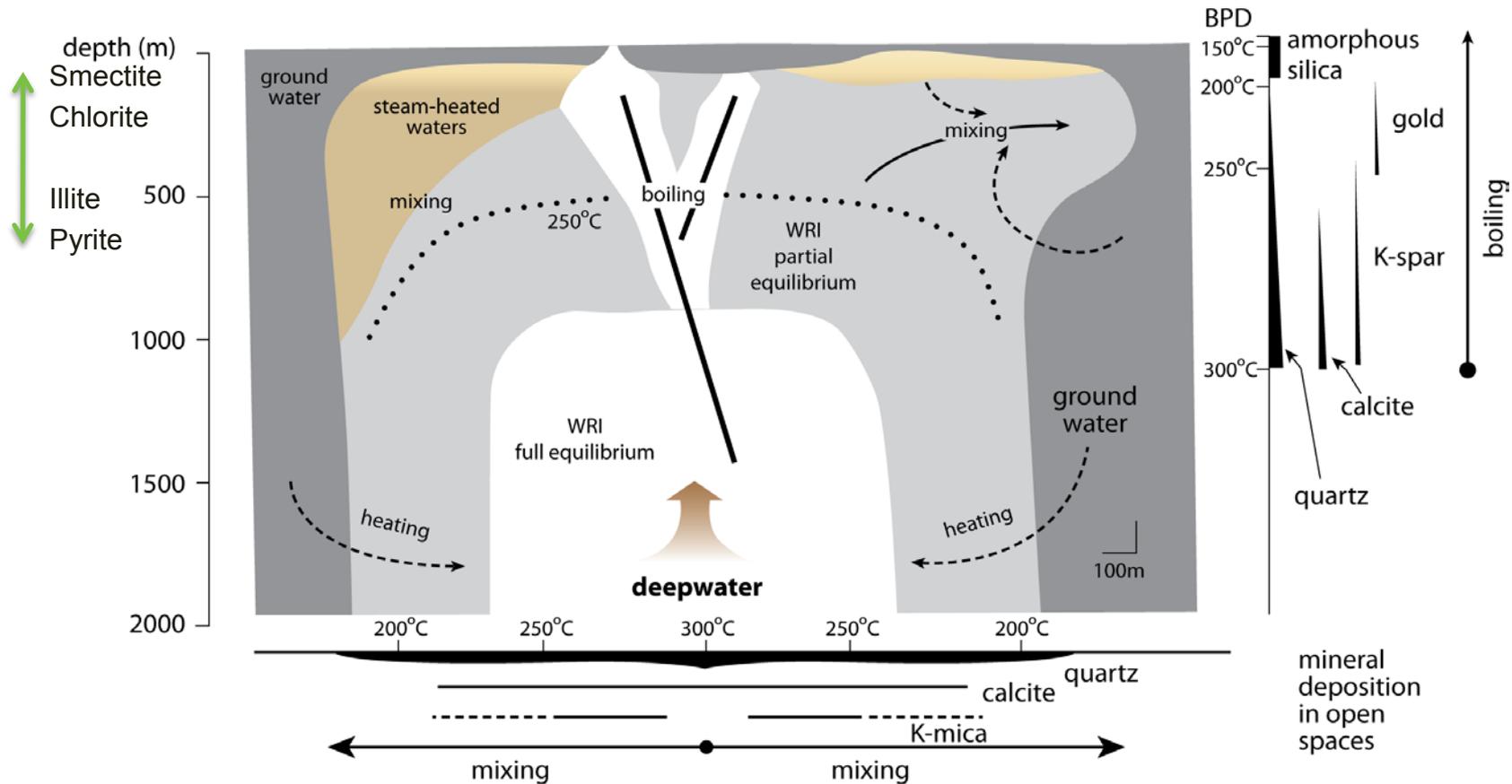
- **Timeline**
 - Start Date October 1, 2009
 - End Date June 30, 2012
 - No cost extension beyond FY11 requested for new postdoc
 - 10%
- **Budget**
 - Total project and received funding \$1024,800
 - DOE share \$1024,800
 - In kind services for Dr. Ed Mroczek, GNS Science, New Zealand
- **Barriers**
 - Difficult field experiments (boiling, gas phase separation)
- **Partners**
 - Dr. Ed Mroczek, GNS Science, New Zealand

- *Hypothesis:* The use of supercritical CO₂ for enhanced production of geothermal energy will not be compromised by mineral dissolution and precipitation reactions.
- *Proposed Work:* We propose a two phase project to assess the geochemical impact of CO₂ on geothermal energy production by
 - analyzing the geochemistry of existing geothermal fields with elevated natural CO₂
 - measuring realistic rock-water rates for geothermal systems using laboratory and field-based experiments to simulate production scale impacts (if any)
- The project will enable the effective deployment of EGS-CO₂
- Innovative aspects include combination of lab and field experiments with simulation.

- Geothermal Field Site:
Broadlands-Ohaaki
Geothermal Field
- Temperature to 300° C
- Aqueous CO₂ ≈ 0.6 molal in deep fluid
- Partners with GNS

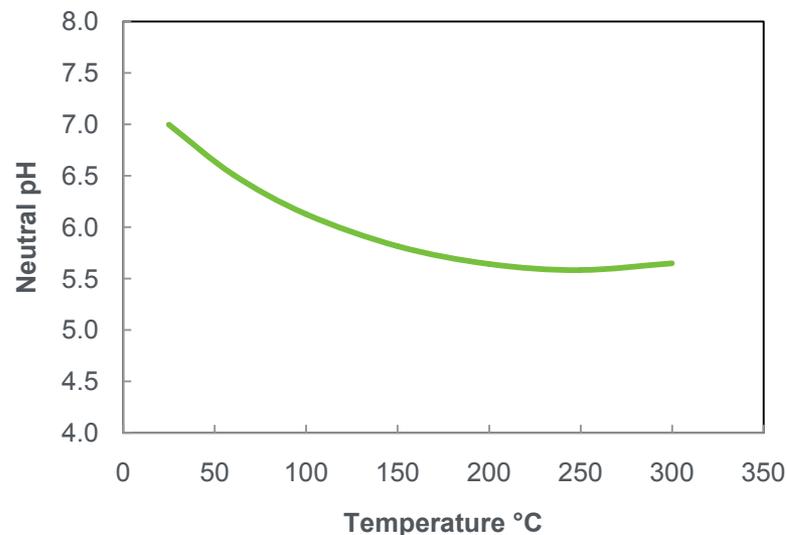
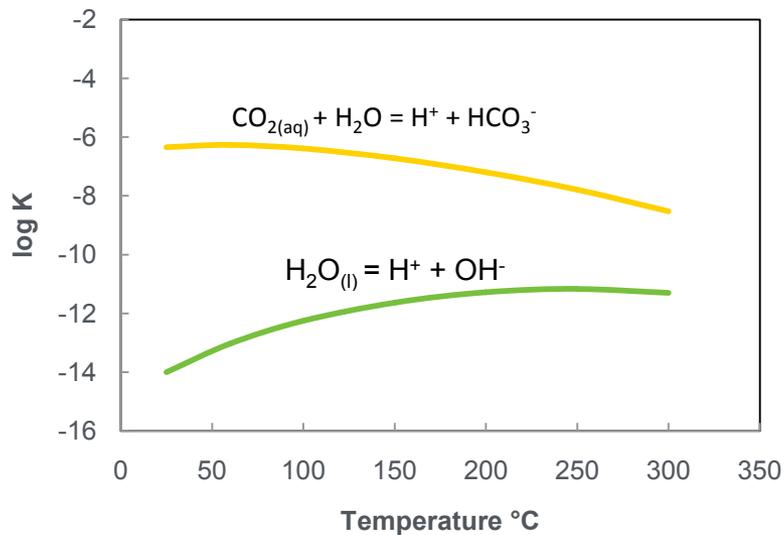
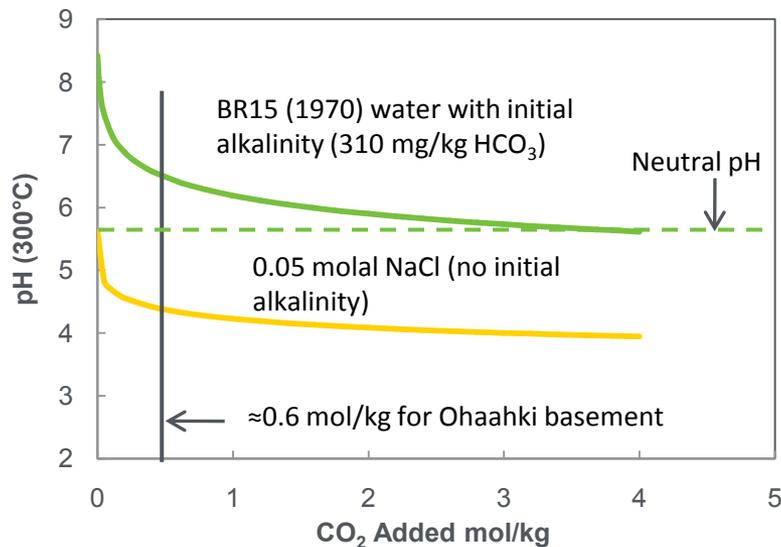


- Broadlands-Ohaaki Geothermal Field (Christensen et al., 2002; Simmons and Browne, 2000; Hedenquist, 1990)

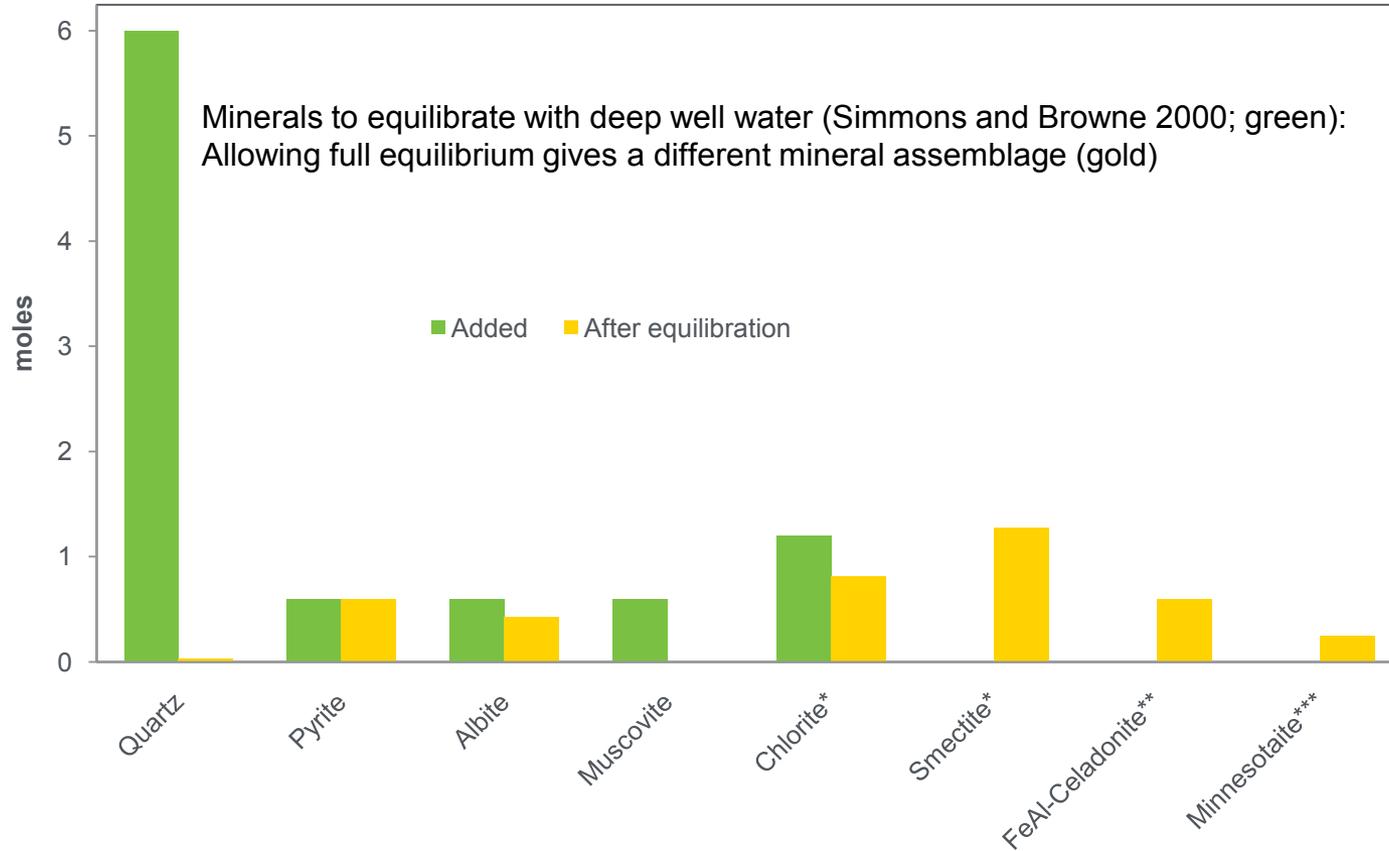


Geochemical Model

- Add CO₂ back into degassed well water
- CO₂ is a weaker acid at temperatures above 100° C

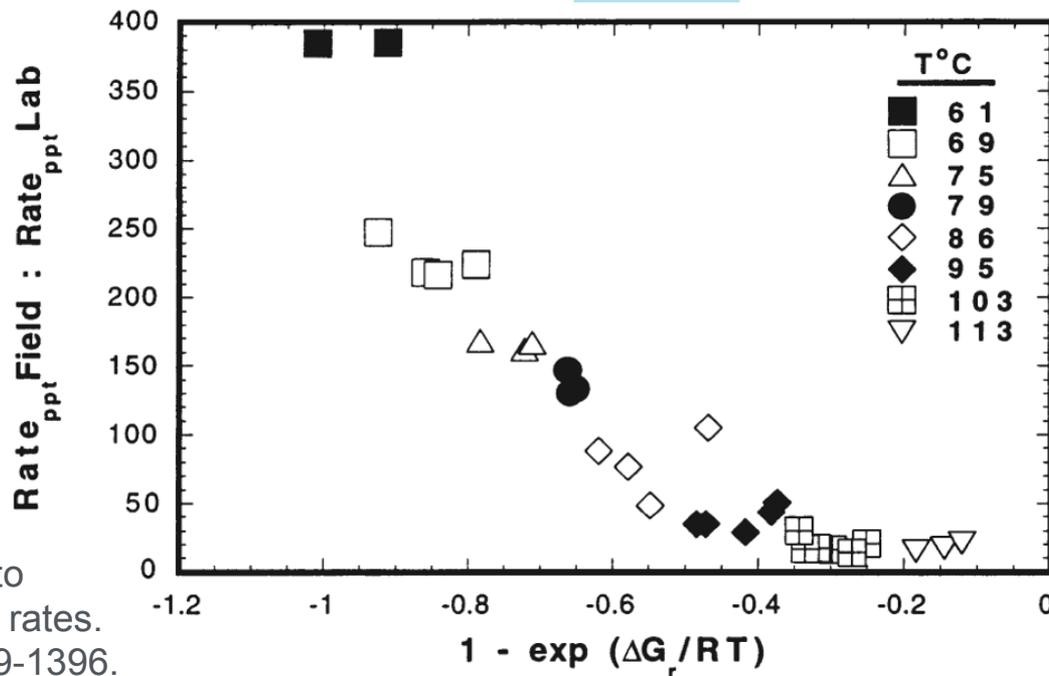
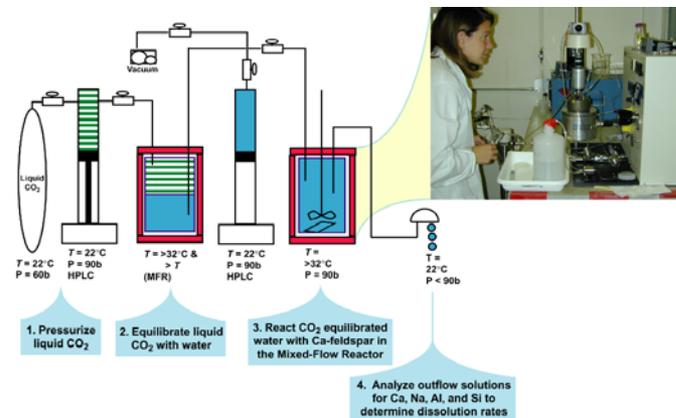
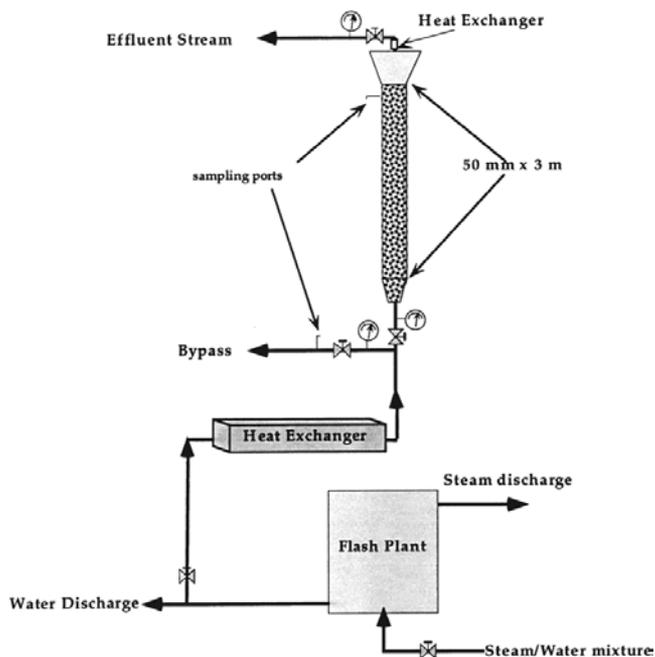


Assessment of basement fluid and mineral geochemistry



*Two chlorite compositions were added, a third one resulted
 **Ferroaluminoceladonite ($\text{KFeAlSi}_4\text{O}_{10}(\text{OH})_2$), similar to illite
 *** $\text{Fe}_3\text{Si}_4\text{O}_{10}(\text{OH})_2$

- Field and Laboratory Studies



Carroll S. A., Mroczek E., Alai, M., Ebert M. (1998) Amorphous silica precipitation (60° to 120° C): Comparison of laboratory and field rates. *Geochimica et Cosmochimica Acta* **62**, 1379-1396.

- Overview projects:
 - Selected Ohaaki-Broadlands Geothermal Field as a natural analog for CO₂-Rock-Water Interaction
 - Started simulations to bound experimental pressure, temperature, and pCO₂ for lab and field studies
 - Collaborating with Dr. Ed Mroczek, geothermal geochemist, GNS Science, New Zealand
 - Hired Dr. Megan Smith, geochemist, June 1, 2010 start date.

Table 1. Timeline for Chemical Impact of Elevated CO₂ on Geothermal Energy Production

	Year 1	Year 2	Year 3
Evaluate geothermal systems	Task 1		
Select geothermal and volcanic sites for geochemical analysis			
Finish geochemical analysis			
Report on the role of CO ₂ on geochemistry in geothermal environments			
Laboratory Measured Rates		Task 2	
Design experimental protocol based on field sites			
Conduct Laboratory experiments			
Field Measured Rates		Task 3	
Select low and high CO ₂ geothermal sites			
Design and assemble field reactor			
Finish Field Experiments			
Derive rate law from Lab and Field experiments			
Report on the comparison of lab and field rates			
			Task 4
Final Report on geochemical impact of supercritical CO₂ on geothermal energy production			

- Spend Plan

Table 2. Revised spend plan in \$K.

	<i>FY10</i>	<i>Fy11</i>	<i>FY12</i>
Oct	12.3	42.4	15.0
Nov	12.3	42.4	15.0
Dec	12.3	42.4	15.0
Jan	12.3	42.4	15.0
Feb	12.3	42.4	15.0
Mar	12.3	42.4	15.0
Apr	51.2	42.4	15.0
May	51.2	42.4	15.0
Jun	51.2	42.4	15.0
Jul	51.2	42.4	
Aug	51.2	42.4	
Sep	51.2	42.4	
FY Total	380.8	509.0	135.0
Project Total	\$1024.8K		

FY10

- Task 1: Finish Geochemical Analysis of the New Zealand Taupo Volcanic Zone
- Task 2: Design laboratory experiments based on field sites and conduct laboratory experiments
- Task 3: Design and fabricate the field reactor

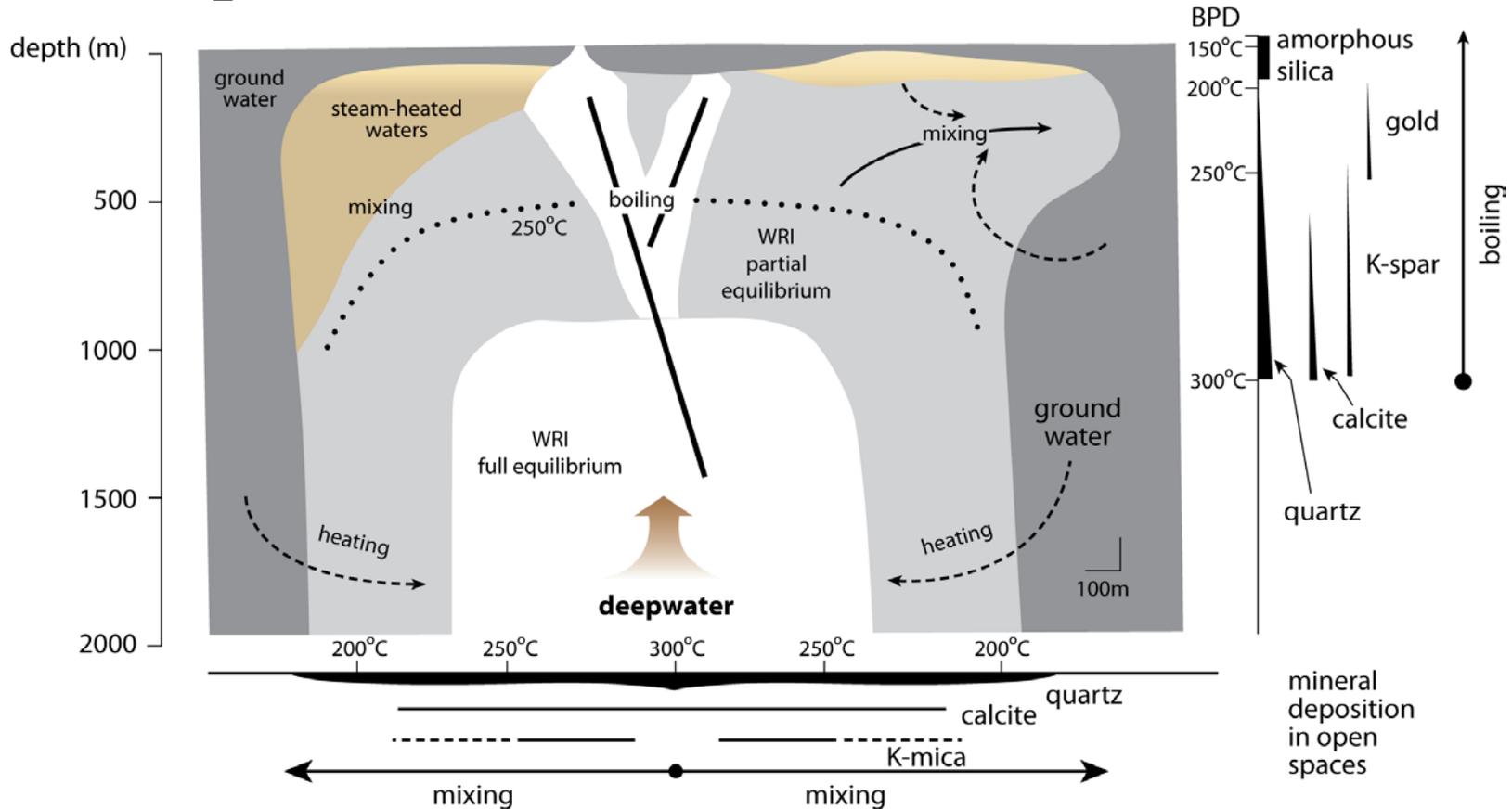
• FY11

- Task 2 and 3: September 30, 2011: Finish lab and field experiments
- Derive rate laws from field and laboratory experiments
- Report comparing field and laboratory rates

• FY12

- Task 4: Final Report on the geochemical impact of CO₂ on EGSCO₂.

- We are investigating the geochemical impact of CO₂ on EGSCO₂.



Supplemental Slides

- Wolery, TJ and Carroll, SA (2010) **CO₂-Rock Interactions in EGSCO₂: New Zealand TVZ Geothermal Systems as Natural Analogs** GRC's 34th Annual Meeting, October 24-27, 2010, Sacramento

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