

High Energy Lithium-Sulfur Cathodes

Yi Cui

Stanford University

June 9, 2015

Project ID
#ES230

Overview

Timeline

- Start: August 1, 2013
- End: July 31, 2016
- Percent complete: 60%

Budget

- Total project funding
\$900k from DOE
- Funding received in FY13
\$300k

Funding for FY14
\$300k

- Funding for FY15
\$300k

Barriers

Barriers of batteries

- High cost (A)
- Low energy density (C)
- Short battery life (E)

Targets: cost-effective and high-energy electrode materials and batteries

Partners

- Collaboration
 - BATT program PI's
 - SLAC: In-situ X-ray
 - Amprius Inc.
 - Beihang Univ, China
 - Zhejiang Univ of Technology, China

Project Objective and Relevance

Objective

- Develop lithium-sulfur batteries to power electric vehicles (HEV/PEV/EV) and decrease the high cost of batteries.
- Develop sulfur cathodes with high capacity and stability to generate high energy lithium-sulfur batteries with long cycle life.
- Design and fabricate novel nanostructured sulfur cathode with multifunctional coatings to overcome the materials challenges that lead to short battery life, including volume expansion, active material loss and low conductivity of sulfur cathode.
- Develop scalable low-cost methods for the synthesis of nanostructured sulfur cathode.
- Project contents are directly aimed at the listed barriers: high cost, low energy density and short battery life.

Milestones for FY14 and 15

Month/year	Milestones
1/2014	Develop low-cost and scalable sulfur cathode coated with one type of polymer and one type of inorganic material with stable cycling (completed)
4/2014	Develop surface coating with several types of polymers; Understand amphiphillic interaction of sulfur and sulfide species (completed)
7/2014	Demonstrate sulfur cathodes with 200 cycles with 80% capacity retention and $>0.3 \text{ mAh/cm}^2$ capacity loading; Modify the separator with conductive coating to enhance the capacity and cycling stability of the sulfur cathode (completed)
12/2014	Demonstrate sulfur cathodes capped by layered metal disulfides; Demonstrate high areal capacity of 3 mAh/cm^2 under high mass-loading conditions ($5.3 \text{ mg Li}_2\text{S/cm}^2$) (completed)
4/2015 Go-no go	Identify the interaction mechanism between sulfur species and different types of sulfides/oxides/metals, and discover/select the optimal material to improve the capacity and cycling of sulfur cathode (on track)
7/2015	Develop sulfur cathodes with high rate capability and volumetric energy density at high mass loading (on track)

Approach/Strategy

Advanced nanostructured sulfur cathodes design and synthesis

- 1) Engineer empty space into sulfur cathode to solve the problem of electrode volume expansion.
- 2) Develop novel sulfur nanostructures with multi-functional coatings for the confinement of sulfur/lithium polysulfides to address the issues of active materials loss and low conductivity.
- 3) Develop/discover optimal nanostructured materials that can capture the polysulfide dissolved in the electrolyte.
- 4) Develop space efficiently packed nanostructured sulfur cathode to increase the volumetric energy density and rate capability.

Structure and property characterization

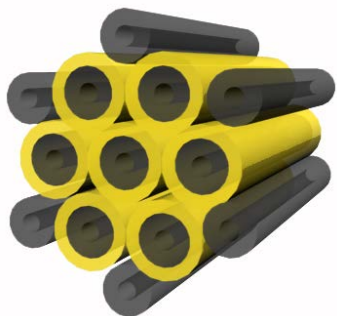
- 1) Ex-situ transmission electron microscopy
- 2) Ex-situ scanning electron microscopy
- 3) Inductively Coupled Plasma elemental analysis
- 4) In operando X-ray diffraction and transmission X-ray microscopy

Electrochemical testing

- 1) Coin cells and pouch cells
- 2) A set of electrochemical techniques

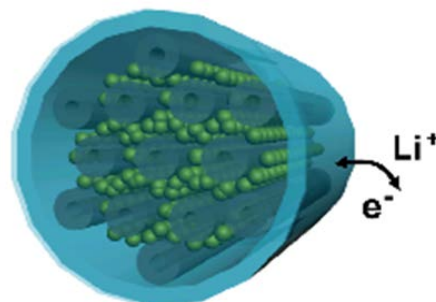
Previous Accomplishments on Sulfur Cathodes

Mesoporous carbon/S



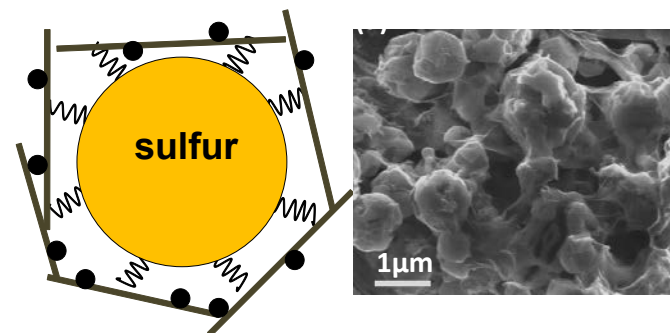
Nano Letters 10, 1486 (2010)

PEDOT/PSS-coated mesoporous carbon/S



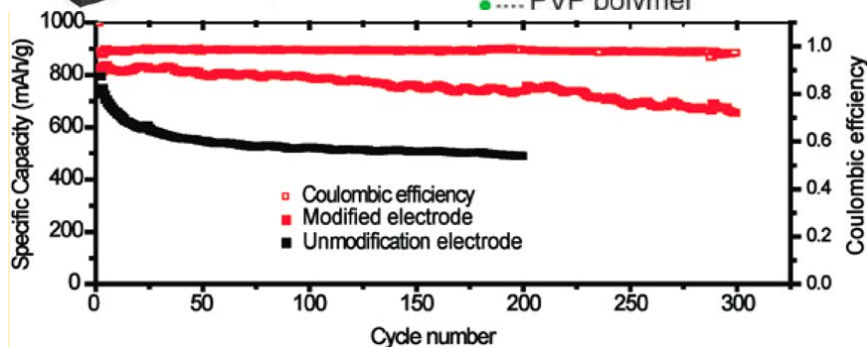
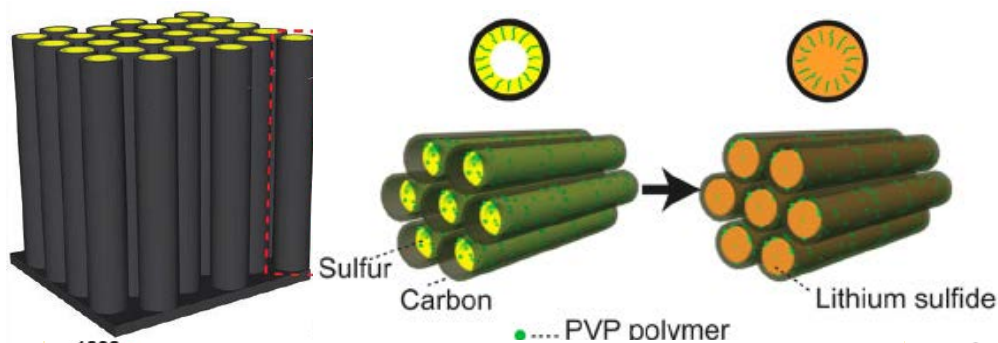
ACS Nano 5, 9187 (2011)

Graphene-coated S particles



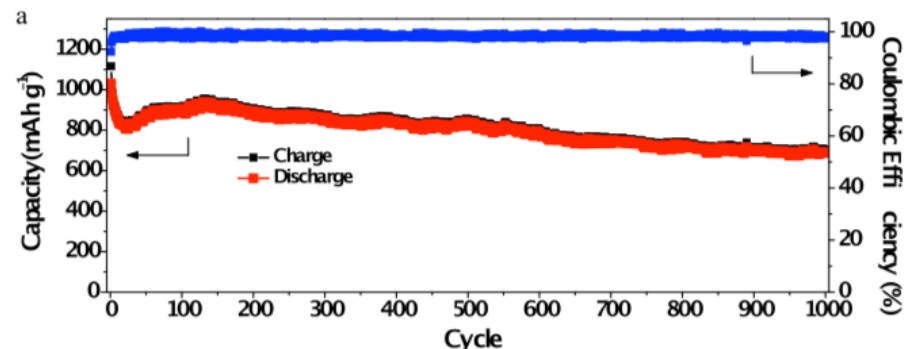
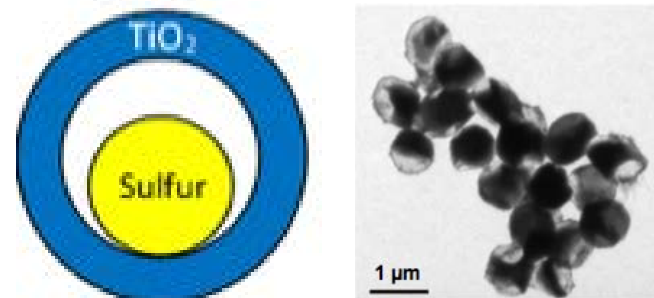
Nano Letters 11, 2644 (2011)

Hollow Carbon Fiber Encapsulated S



Nano Letters 11, 4462 (2011) *Nano Letters* 13, 1265 (2013)

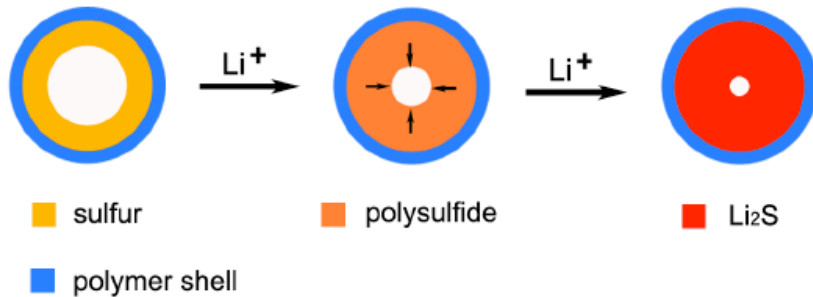
Yolk-Shell S-TiO₂ Nanoparticles



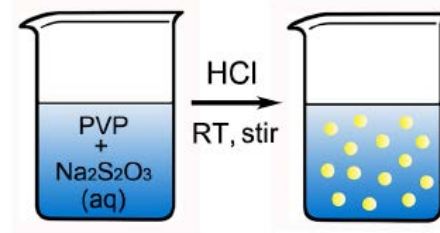
Nature Communication 4: 1331 (2013)

Accomplishment

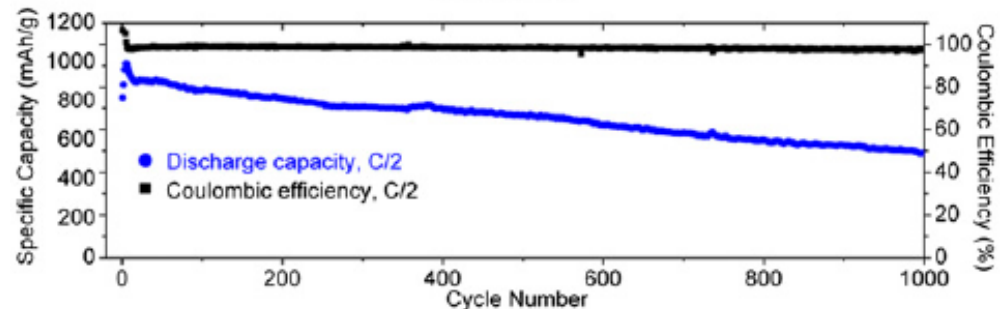
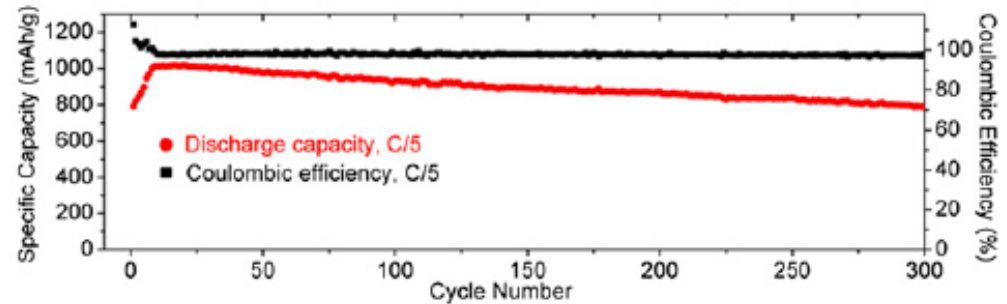
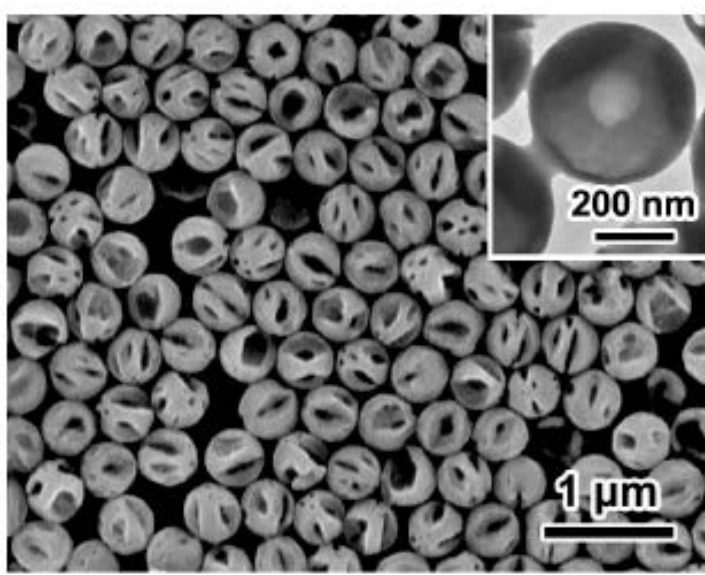
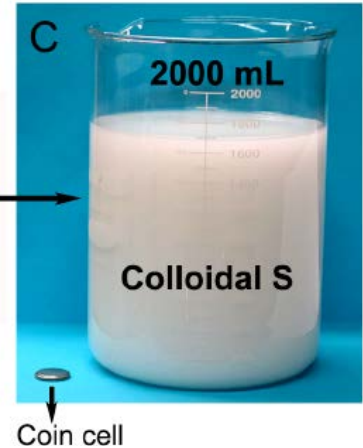
Hollow S-Amphiphilic Polymer Nanoparticles



B



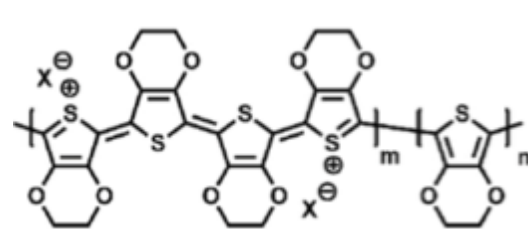
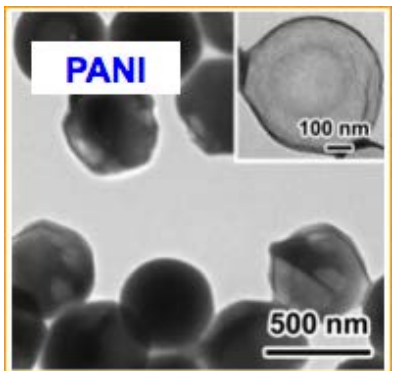
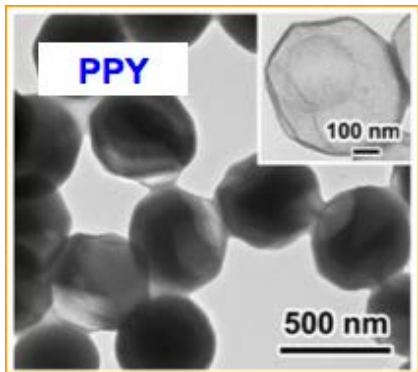
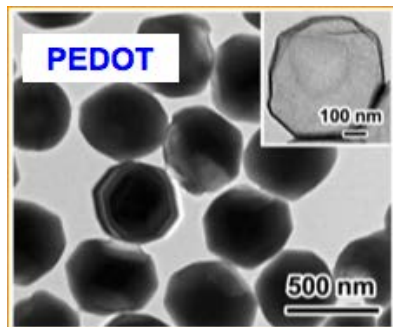
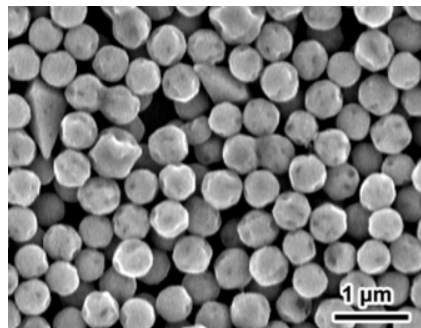
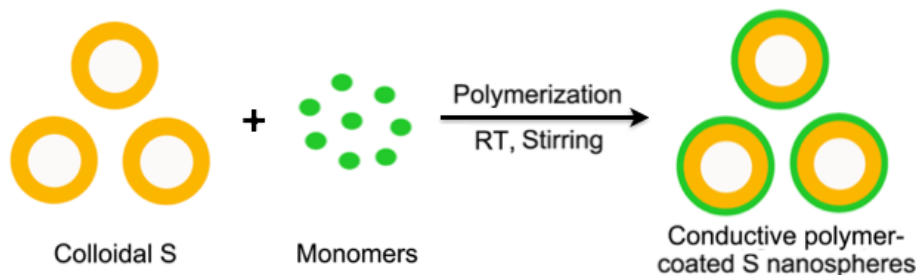
C



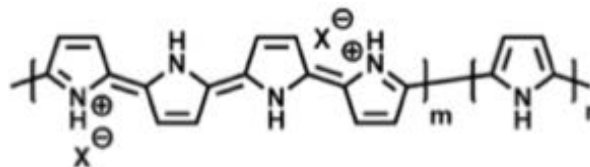
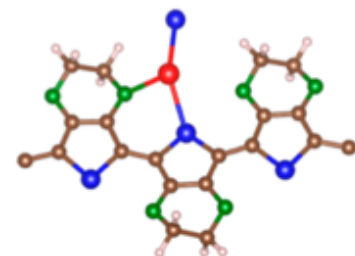
Cui group, *PNAS* 110, 7148 (2013)

Accomplishment

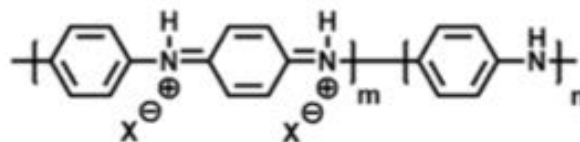
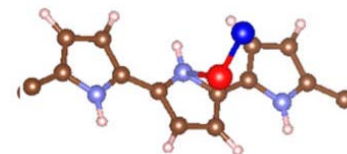
Conductive polymer-coated hollow sulfur cathodes -Synthesis, morphology and simulation



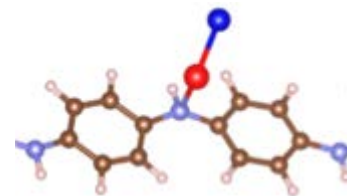
PEDOT-LiS• (1.22 eV)



PPY-LiS• (0.64 eV)

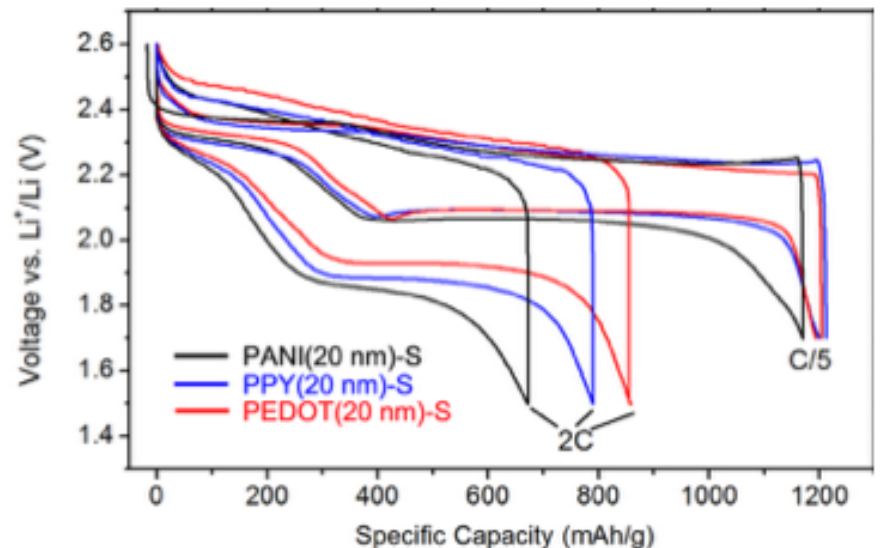
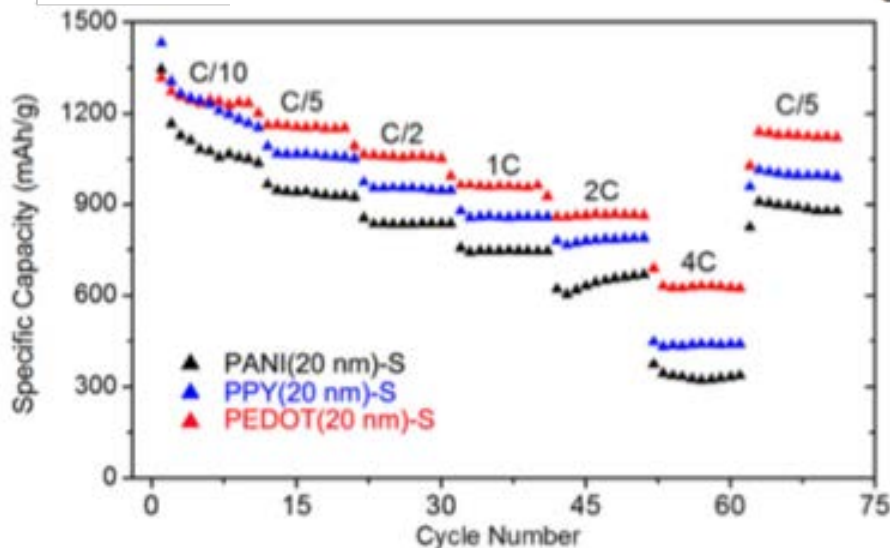
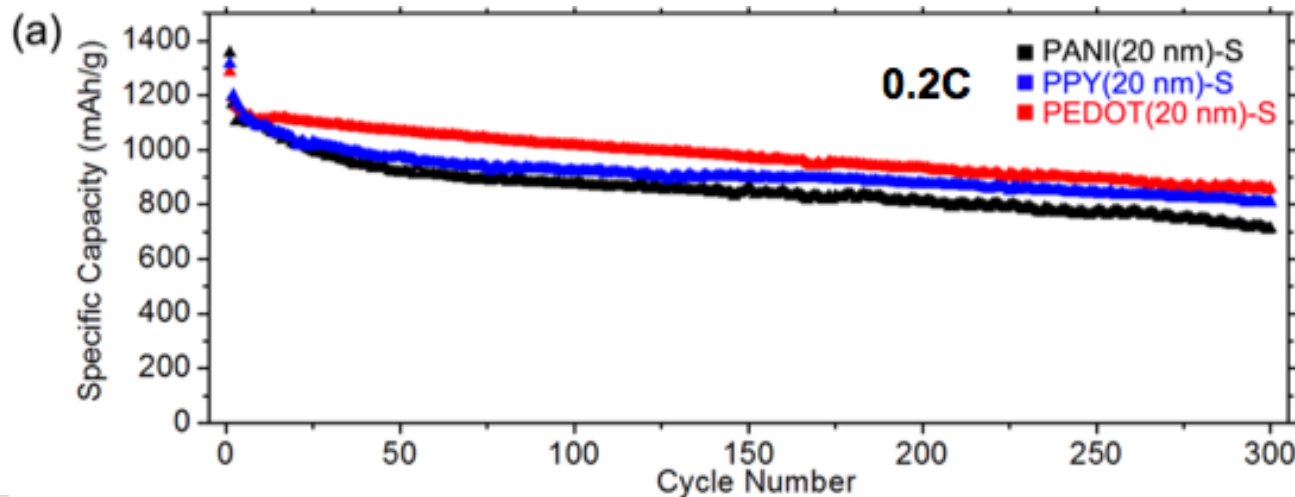


PANI-LiS• (0.67 eV)



Accomplishment

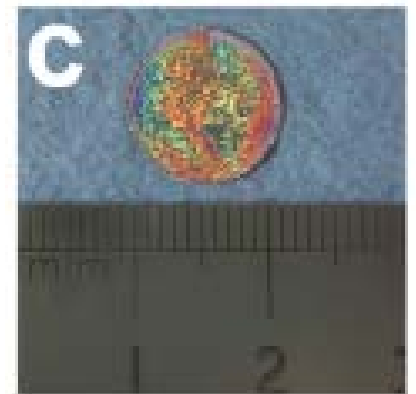
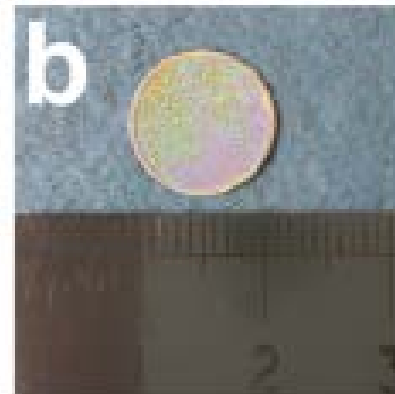
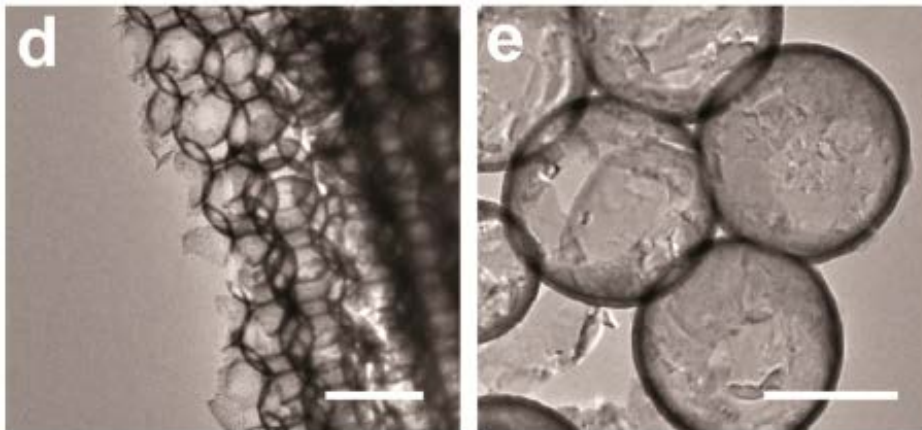
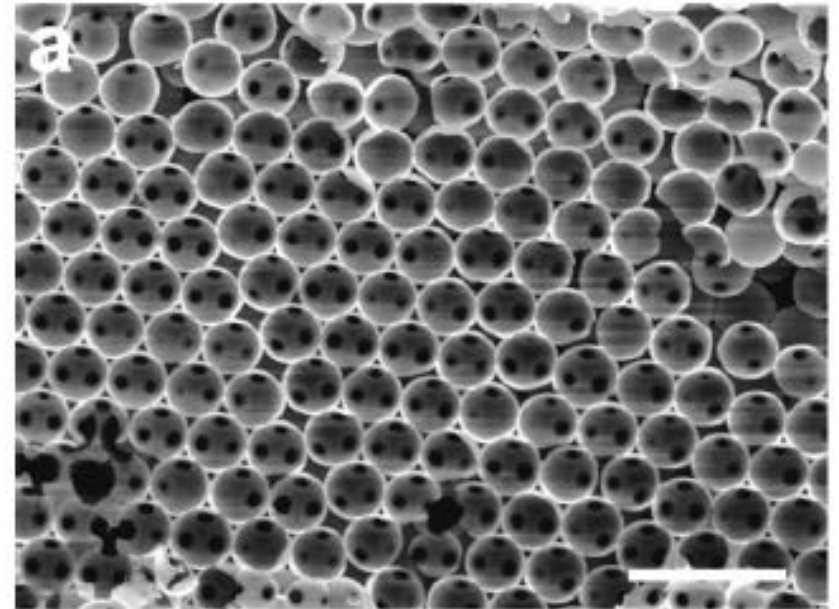
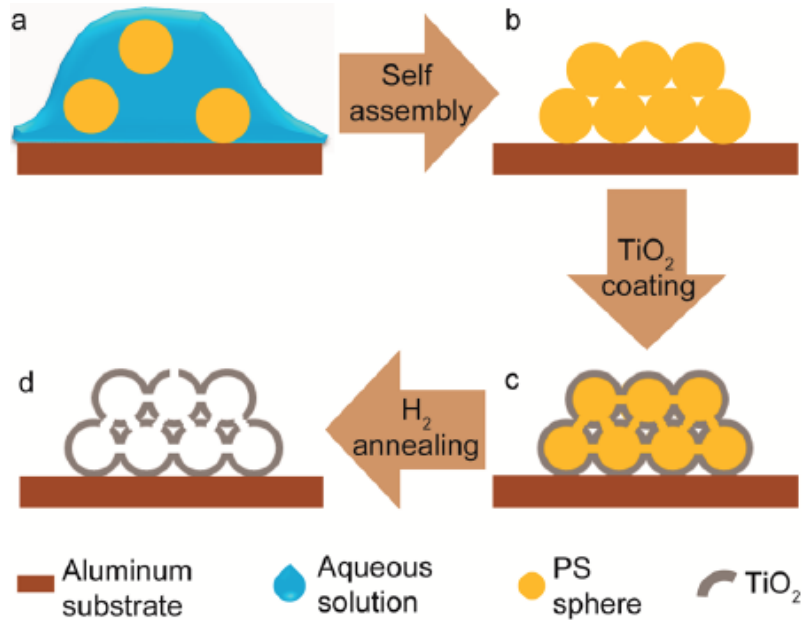
Conductive polymer-coated hollow sulfur cathodes
-Battery performance: excellent rate capability



Cui group, *Nano Letters*, 13, 5534 (2013)

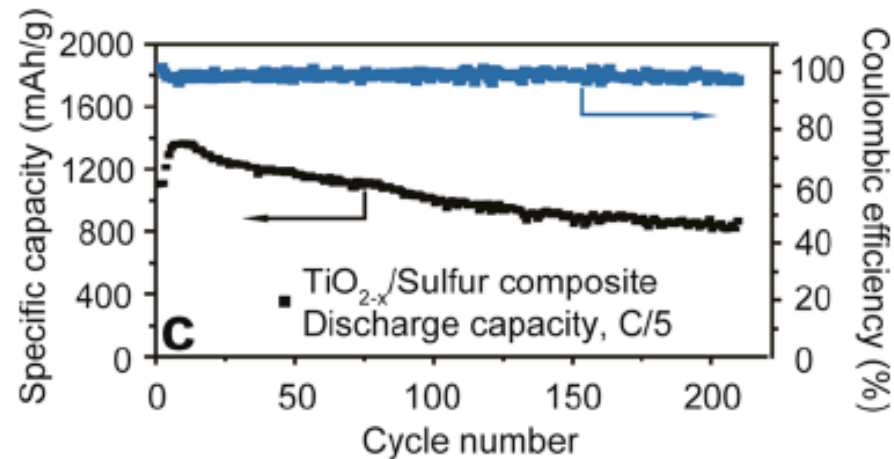
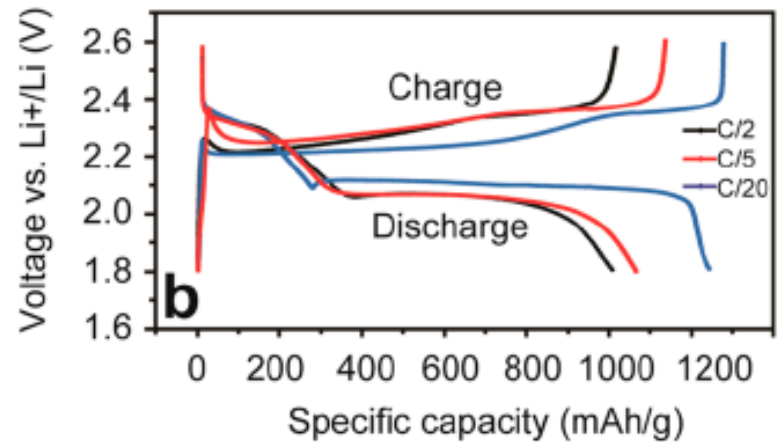
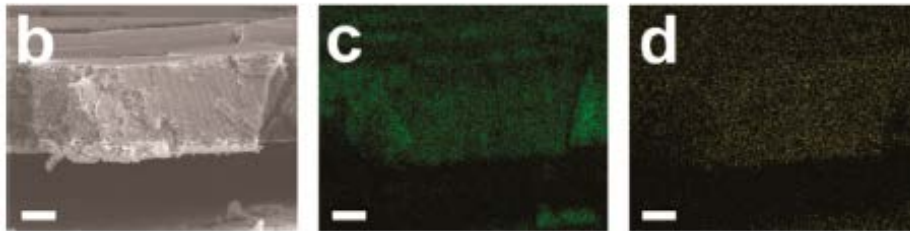
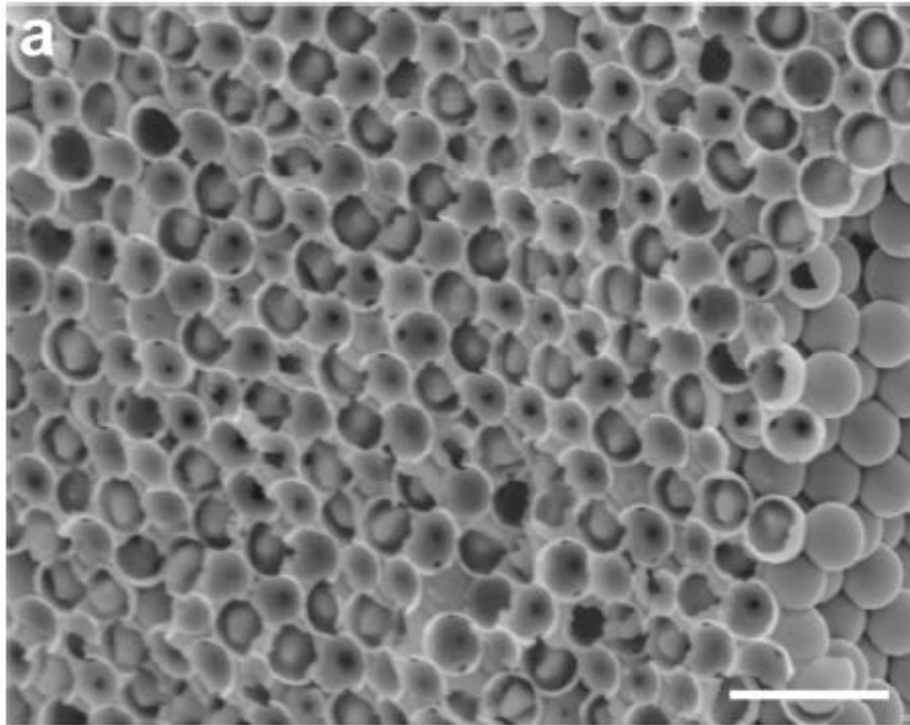
Accomplishment

Hydrogen Reduced TiO_{2-x} Inverse Opal- synthesis and morphology



Accomplishment

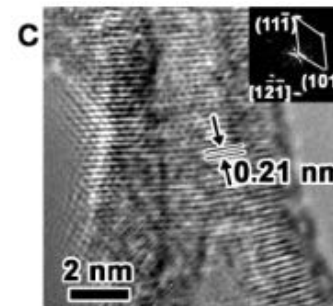
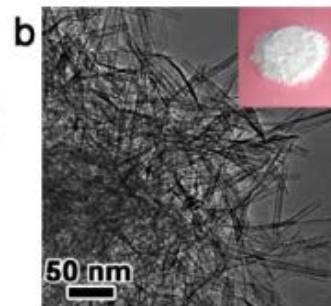
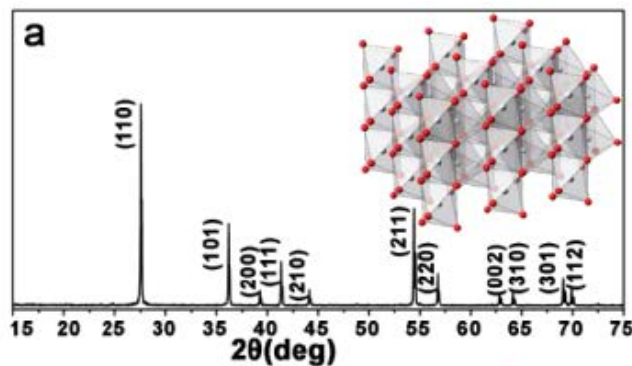
Hydrogen Reduced TiO_{2-x} Inverse Opal Sulfur- Battery performance



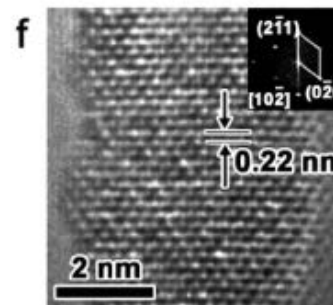
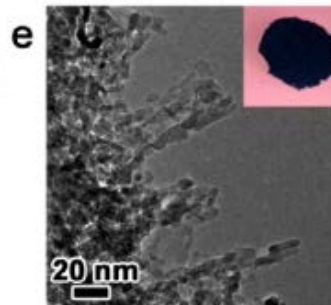
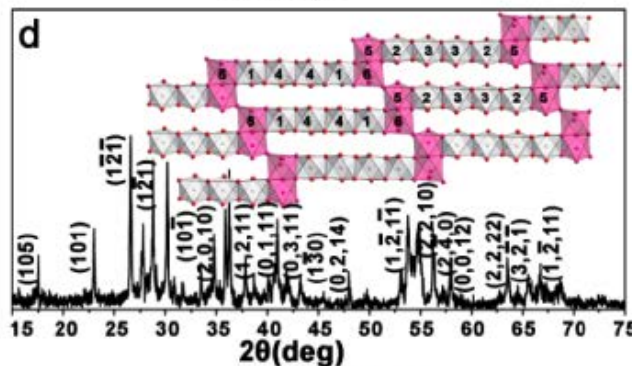
Accomplishment

Magnéli-Phase $\text{Ti}_n\text{O}_{2n-1}$ Nanomaterials for S Cathodes

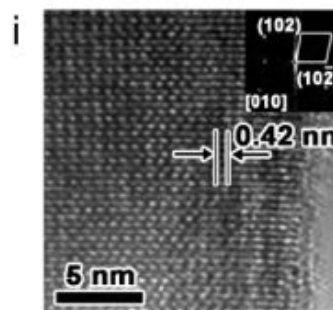
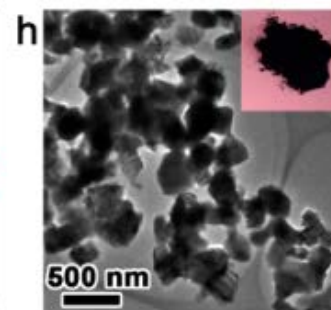
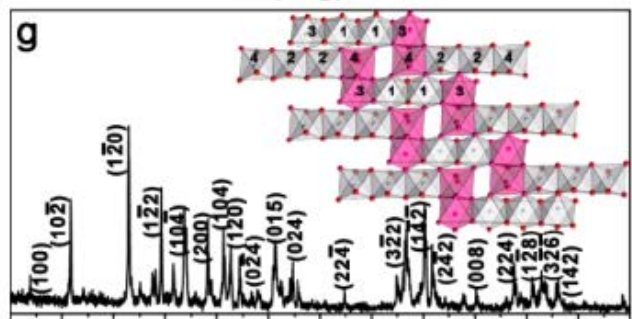
TiO_2



Ti_6O_{11}

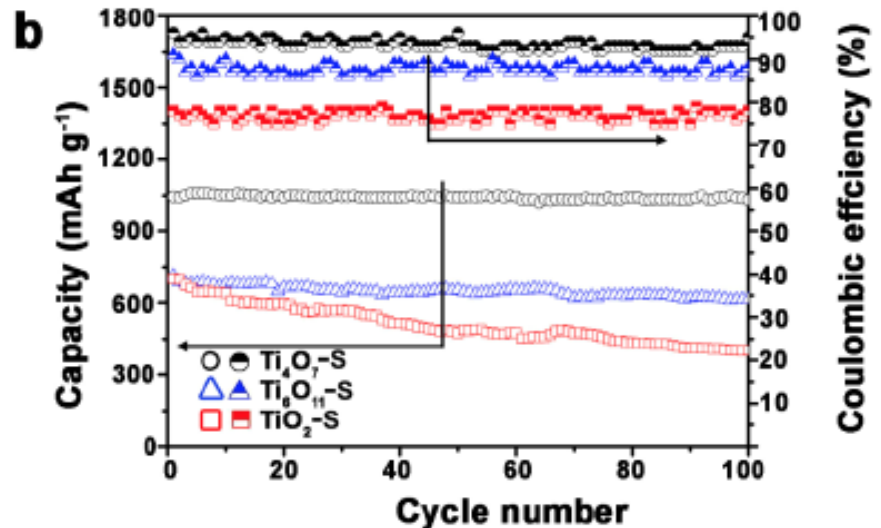
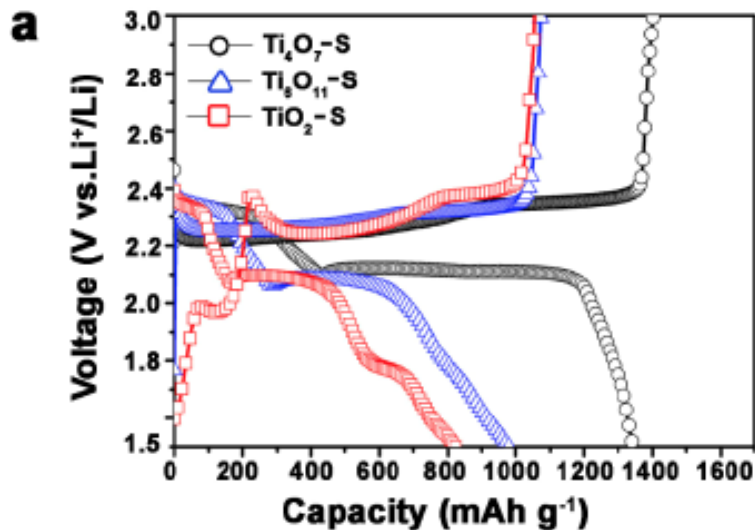
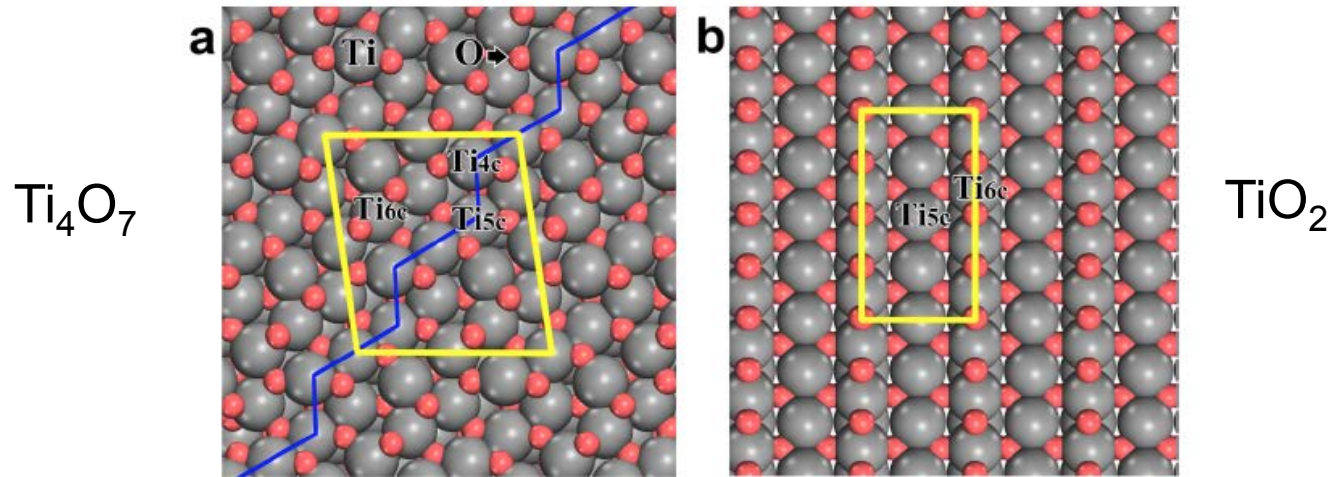


Ti_4O_7



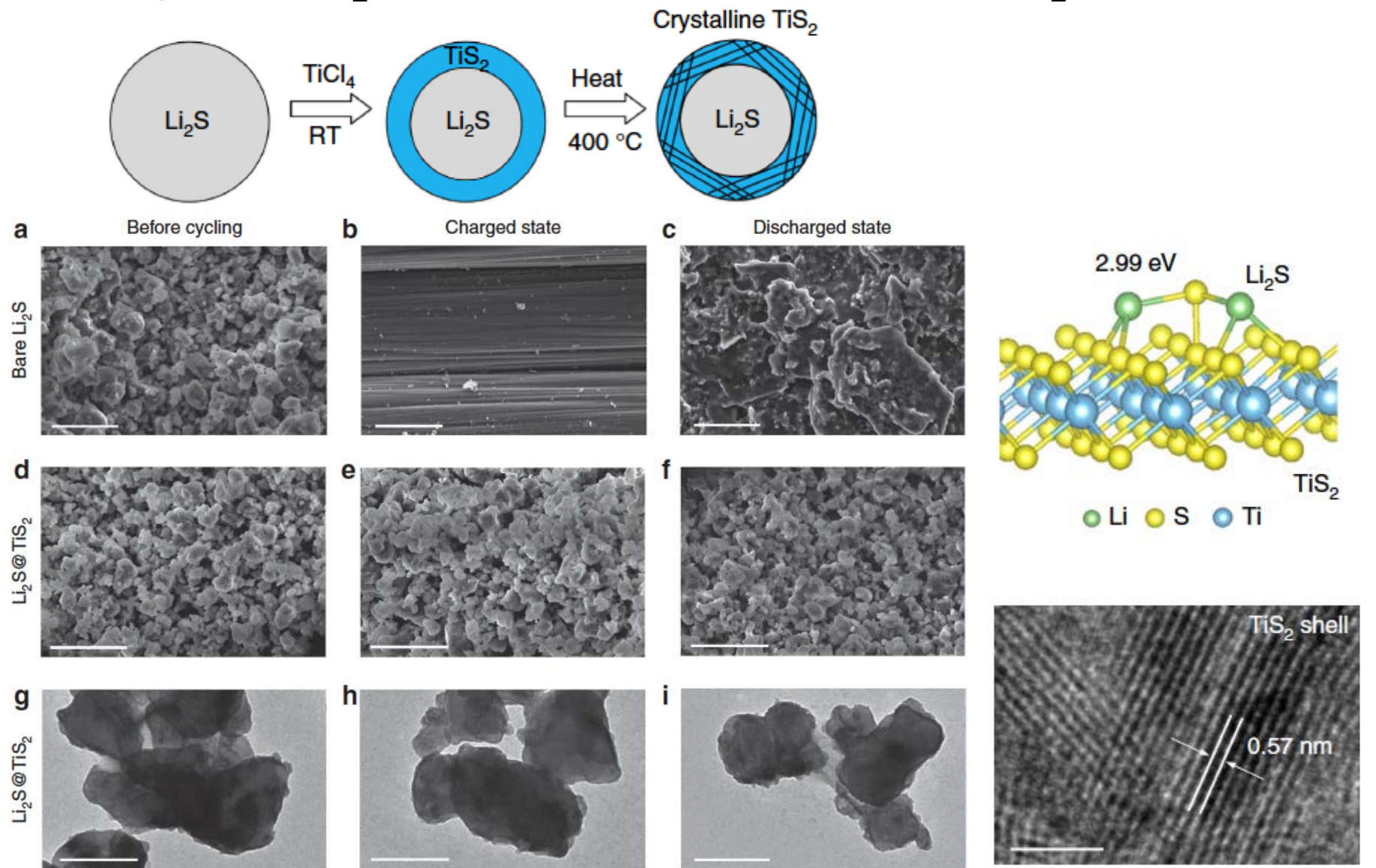
Accomplishment

Strong sulfur binding with conductive magnéli-phase Ti_4O_7 nanoparticles:
Magnéli-Phase has high concentration of O vacancies



Accomplishment

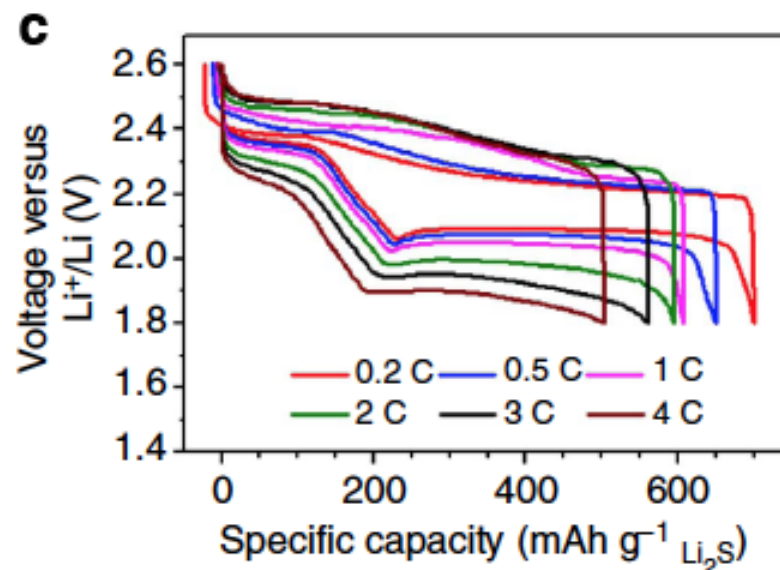
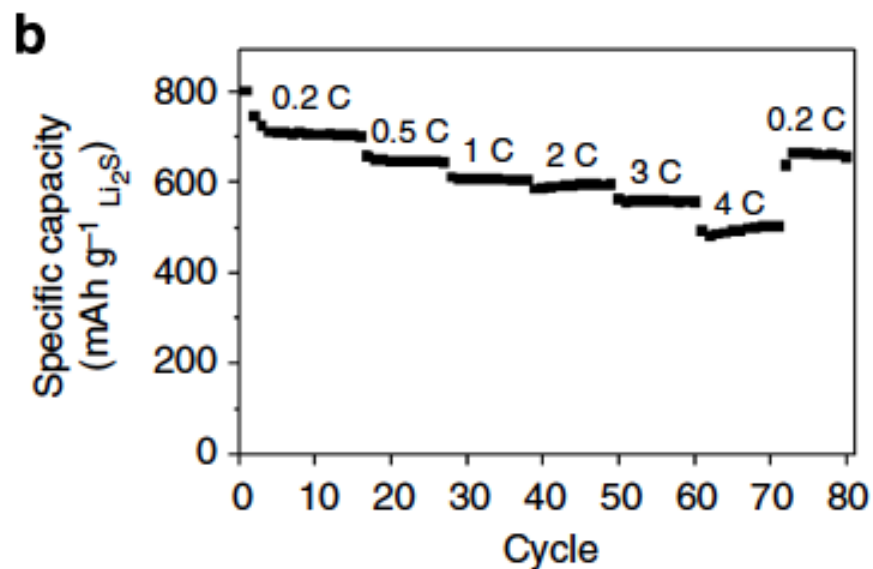
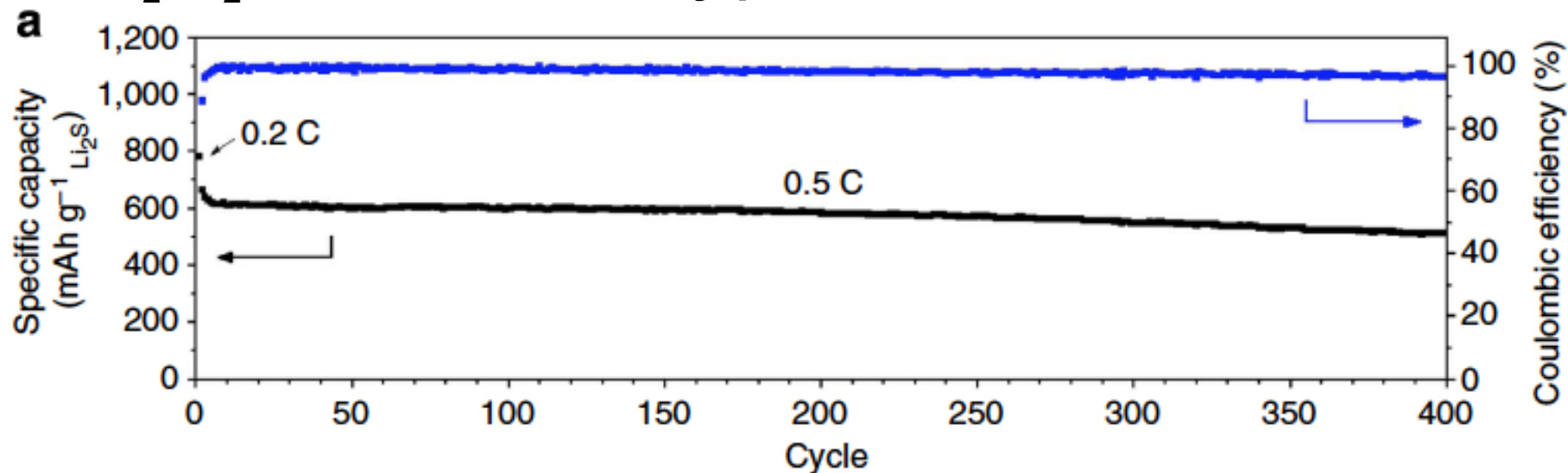
2D layered MS₂ for Effective Encapsulation of Li₂S Cathodes



Cui group, *Nature Communications* 5:5017 (2014)

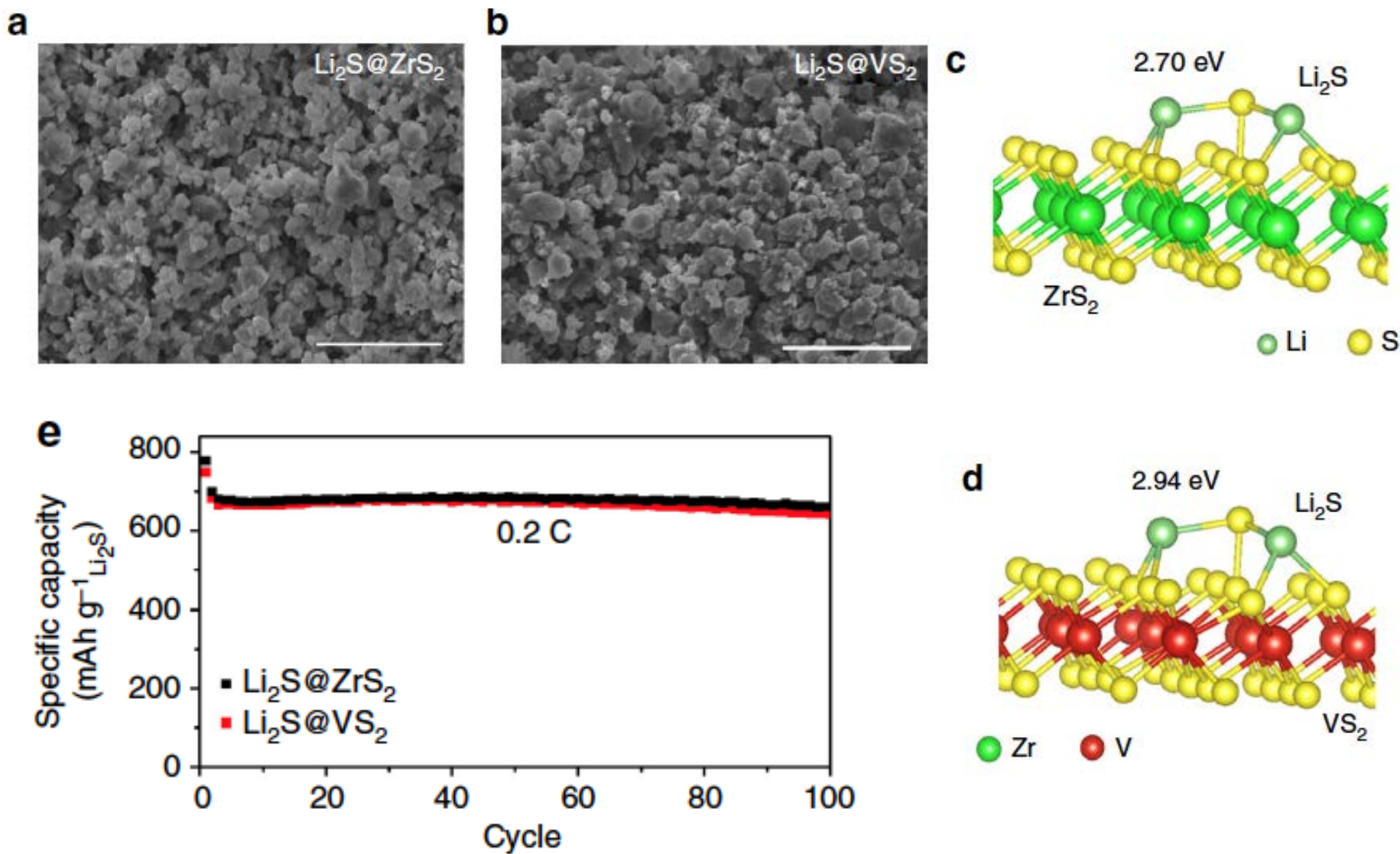
Accomplishment

TiS₂-Li₂S Cathodes: battery performance



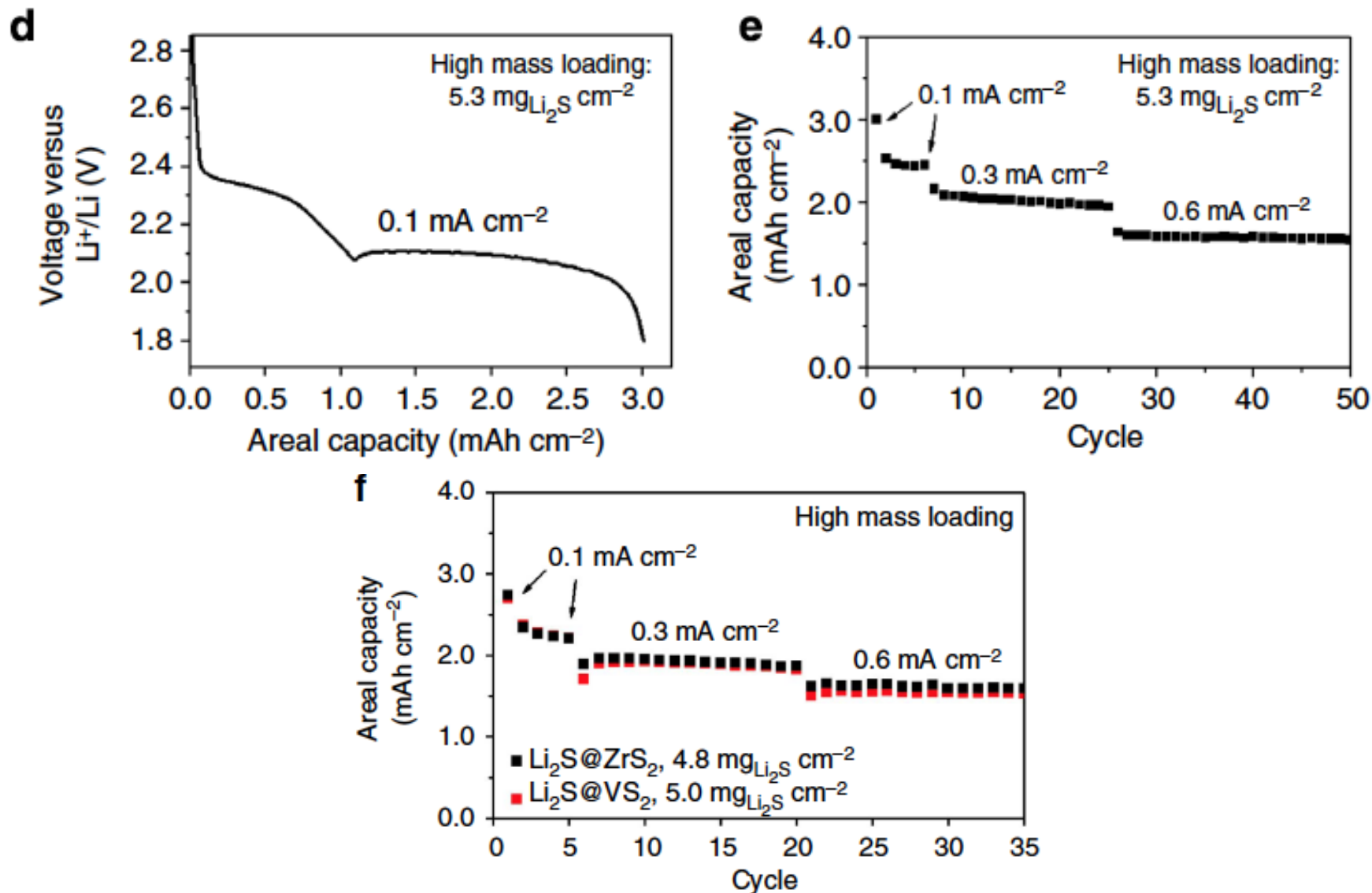
Accomplishment

ZrS₂/VS₂-Li₂S Cathodes: morphology and battery performance



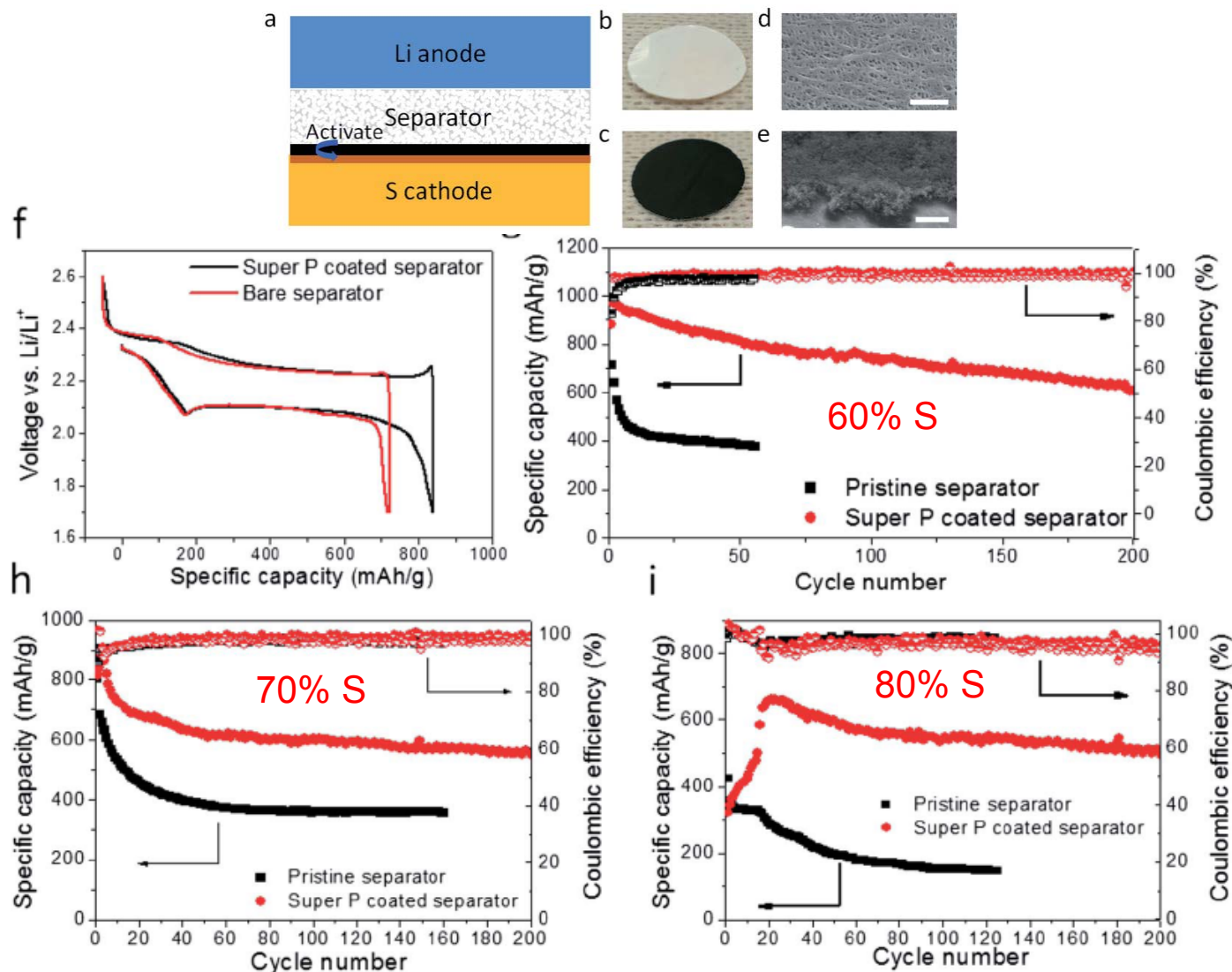
Accomplishment

High Areal Capacity Loading of $\text{TiS}_2/\text{ZrS}_2/\text{VS}_2\text{-Li}_2\text{S}$ Cathodes



Accomplishment

Improved Li-S batteries with a conductive coating on the separator
-Activate the sulfur cathode surface



Cui group, *Energy & Environmental Science*, 7, 3381 (2014)

Responses to Previous Year Reviewers' Comments

Not applicable

Collaboration and Coordination



NATIONAL
ACCELERATOR
LABORATORY

SLAC: In-situ X-ray, Prof. Mike Toney



Beihang University, China:
Ab initio simulations, Prof. Qianfan Zhang



Zhejiang University of Technology, China:
Development of conductive magnéli-phase
 Ti_4O_7 nanoparticles, Prof. Wenhui Zhang



Companies: Amprius Inc.

Remaining Challenges and Barriers

- It is difficult to maintain high capacity and excellent cycling stability of lithium-sulfur batteries while increasing the mass loading of active sulfur in the cathode.
- It is challenging to improve the rate capability (performance of battery at high current densities) of lithium-sulfur batteries.
- It is difficult to fully prevent all the active sulfur species from diffusing into the electrolyte.
- The volumetric energy density of lithium-sulfur batteries needs to be further increased.
- The lithium dendrites grown on the lithium metal surface is a concern for the safety of lithium-sulfur batteries that use lithium metal as anodes.

Proposed Future Work

- To understand the interaction between sulfur/sulfide species and different metals/oxides/sulfides, and select the optimal materials to re-capture the active sulfur species diffused in the electrolyte.
- To develop space efficiently packed nanostructured sulfur cathode to increase the volumetric energy density.
- To improve the interparticle contact and conductivity of sulfur nanostructures to increase the kinetics and thus improve the rate capability.
- To test sulfur cathodes with high areal mass loading up to 2-3 mg/cm² at high current densities.
- To develop approaches to prevent the lithium dendrites growth on lithium metal anodes in lithium-sulfur batteries
- To combine lithium sulfide cathodes with non-lithium anodes, such as silicon, to assemble full batteries to eliminate the safety concern of using lithium metal.

Summary

- **Objective and Relevance:** The goal of this project is to develop stable and high capacity sulfur anodes from the perspective of nanomaterials design to enable high energy lithium-sulfur batteries to power electric vehicles, highly relevant to the VT Program goal.
- **Approach/Strategy:** This project combines advanced nanomaterials synthesis, characterization, battery assembly and testing, which has been demonstrated to be highly effective.
- **Technical Accomplishments and Progress:** This project has produced many significant results, meeting milestones. They include identifying the key issues in lithium-sulfur batteries, using rational materials design, synthesizing and testing, and developing scalable and low-cost methods. The results have been published in top peer-reviewed scientific journals. The PI has received numerous invitations to speak in national and international conferences.
- **Collaborations and Coordination:** The PI has established a number of highly effective collaborations.
- **Proposed Future Work:** Rational and exciting future has been planned.