



# Fuel Properties to Enable Lifted Flame Combustion

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FT017

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# Overview



## Timeline

- Start Date - October 2011
- End Date - December 2014
- Percent complete – 80%

## Budget

- \$1.875M
  - DOE Share: \$1.5M
  - Ford Share: \$375k
- \$406k in FY2013
- \$694k in FY2014

## Barriers

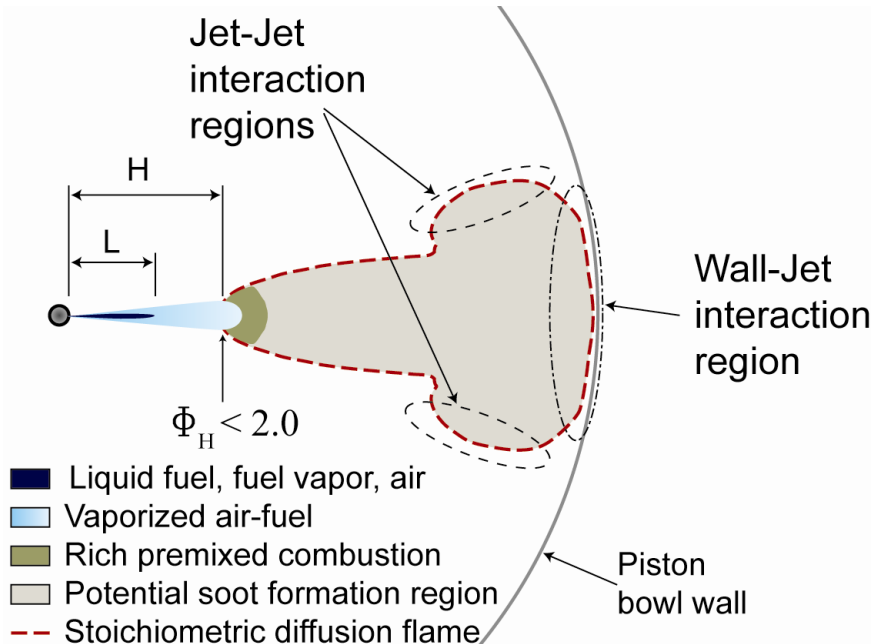
- Understand fuel impact on combustion & emissions
- Improve predictive tools
- Enable advanced combustion regimes - LLFC

## Partners

- Interactions/collaborations
  - Sandia National Labs: sprays, optical engine
  - University of Wisconsin-Madison: Advanced model development
  - Lawrence Livermore National Labs: Kinetics model development
- Project lead: Ford

# Relevance: Objective

Project objective: Identify fuel properties that can be used to enable controllable, non-sooting Leaner Lifted Flame Combustion (LLFC)



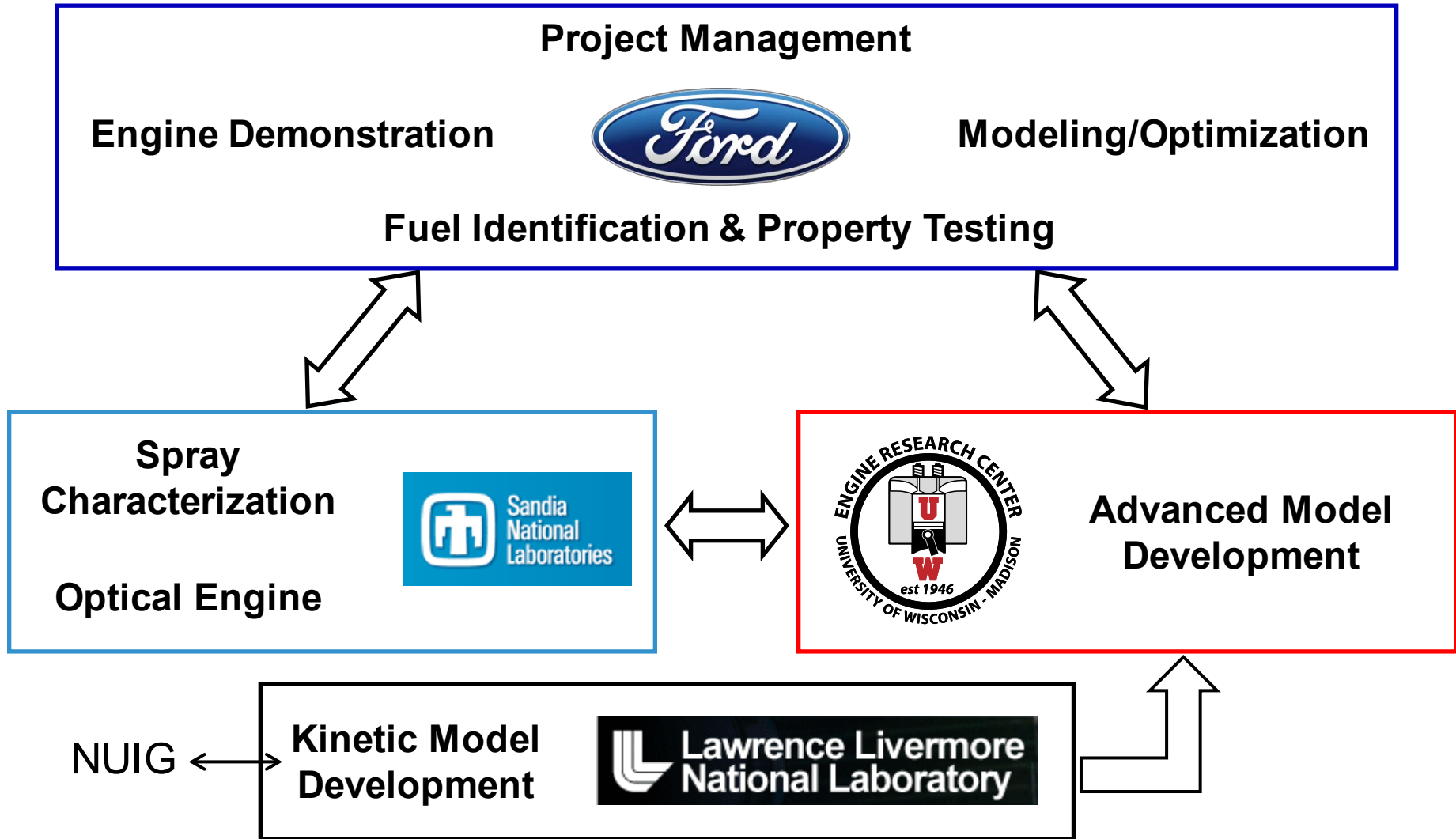
- LLFC benefit: non-sooting diffusion flame
- Requirements:
  - Short liquid penetration
  - Lean mixture ( $\Phi < 2$ ) where the flame starts
- Methods:
  - Increase lift-off length
  - Increase air entrainment
  - Increase fuel oxygen

Investigate fuel effects on the ability to achieve LLFC

- Improve fundamental understanding – relatively new combustion mode
- Enhance predictive tools – develop models to capture observed fuel effects
- Improve efficiency – potential to eliminate the DPF
- Reduce petroleum consumption – application for renewable fuels



# Collaboration: Roles & Responsibilities



# Approach: Timeline



Project Management



Fuel Property Testing



Spray Characterization



Modeling



Engine Testing

BP1

BP 2

BP 3

Existing data

Optical SCE

Metal SCE



## Budget Period 1 (2011/2012)

- Initiate modeling projects
- Identify, select & test fuels
- Initiate spray characterization

## Budget Period 2 (2013)

- Complete spray characterization
- Initiate optical engine studies
- Models ready for used in BP3

## Budget Period 3 (2014)

- Explore fuel effects with models
- Optimize SCE for LLFC
- Complete optical engine studies
- Demonstrate LLFC on SCE

The project is structured to progress from fundamental investigations to more applied studies over the duration of the project.

# Approach: Milestones

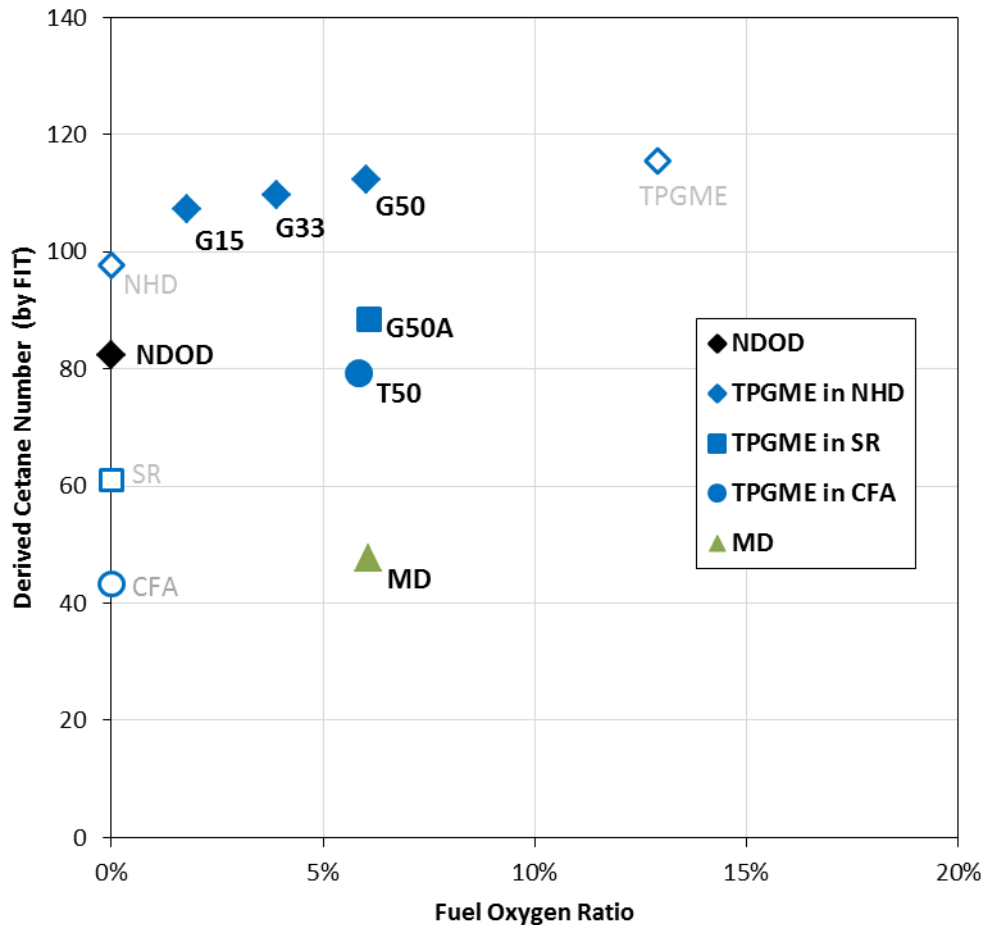


Year	Milestone	Status
2011/2012	Develop a list of potential oxygenated fuels to use in the study	Complete ✓
	Select fuels for spray studies	Complete ✓
	Initiate modeling studies	Complete ✓
2013	Complete spray characterization of different fuels identifying BCs for LLFC	Complete ✓
	Select fuels for optical engine studies	Complete ✓
	Validate CFD models based on engine and spray studies	Complete ✓
	Improve TPGME kinetic mechanism	Complete ✓
2014	Complete optical engine study	Ongoing
	Optimize SCE combustion system for LLFC demonstration	Ongoing
	Select fuels for SCE demonstration	Q3
	Demonstrate LLFC on SCE	Q3/Q4

# Accomplishments: Test Fuels



Presumed desired fuel: sufficient oxygen content (clean) and quick transition to diffusion burn (controllable)



- Line of similar cetane rating
- Line of similar oxygen content
- Two hydrocarbon base fuels
  - Aromatics vs. aromatic free
- Two oxygenate types
  - Ether vs. ester

**NDOD** = n-dodecane

**G15** = 15%v TPGME in NHD

**G33** = 33%v TPGME in NHD

**G50** = 50%v TPGME in NHD

**G50A** = 50%v TPGME in SR

**MD** = methyl decanoate

**T50** = 50%v TPGME in CFA

TPGME = tri(propylene glycol) methyl ether

NHD = n-hexadecane

SR = 23%v m-xylene in n-dodecane

CFA = No. 2 diesel certification fuel



# Accomplishments: Transient luminosity and lift-off imaging for hydrocarbon vs. oxygenated fuels

n-dodecane  
22.8 kg/m<sup>3</sup> – 850 K

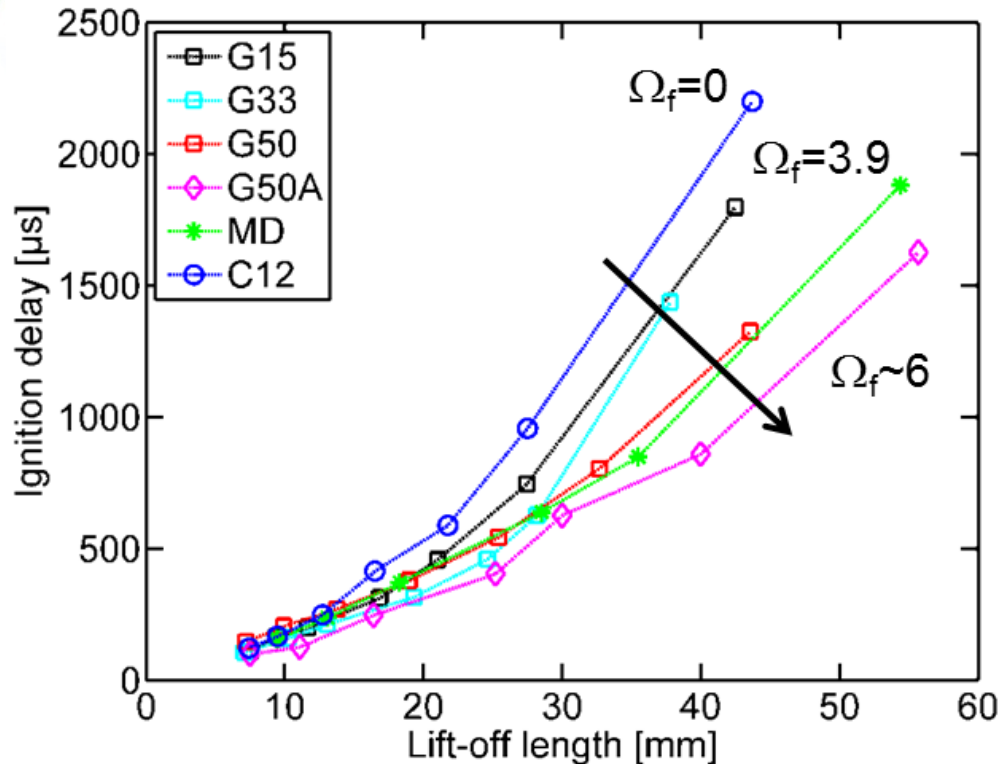
G50  
22.8 kg/m<sup>3</sup> – 850 K



New method developed allows the ability to generate quantitative, high-speed, soot measurements with dual wavelength diffused back-illumination extinction imaging



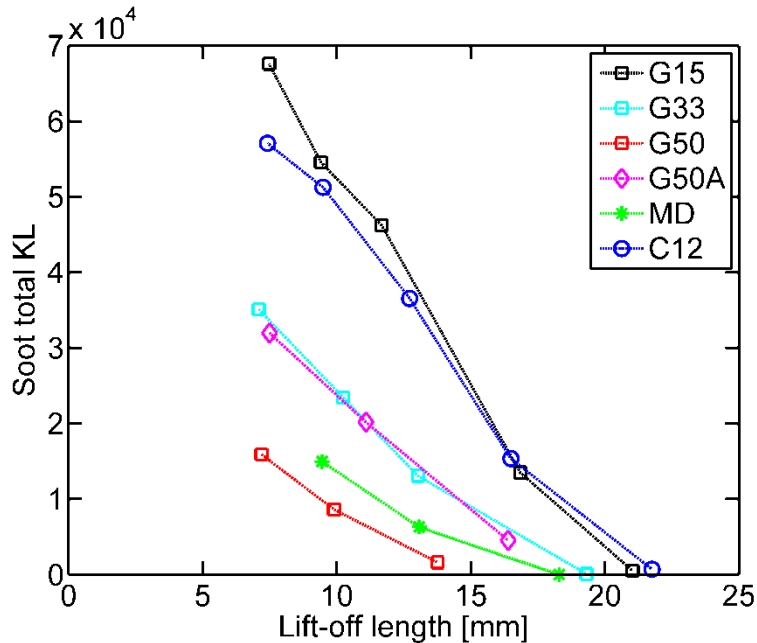
# Accomplishments: Ignition delay as a function of lift-off length



- N-dodecane (pure alkane) shows the shortest lift-off lengths of all fuels at a specific ignition delay time
- Increasing the amount of oxygen in the fuel produces longer lifted flames for a given ignition delay (only noticeable at lower ambient temperatures)
- The mixture of m-xylene (aromatic), n-dodecane and TPGME provides the longest lift-off length for a given ignition delay, despite a cetane number that is higher than methyl decanoate

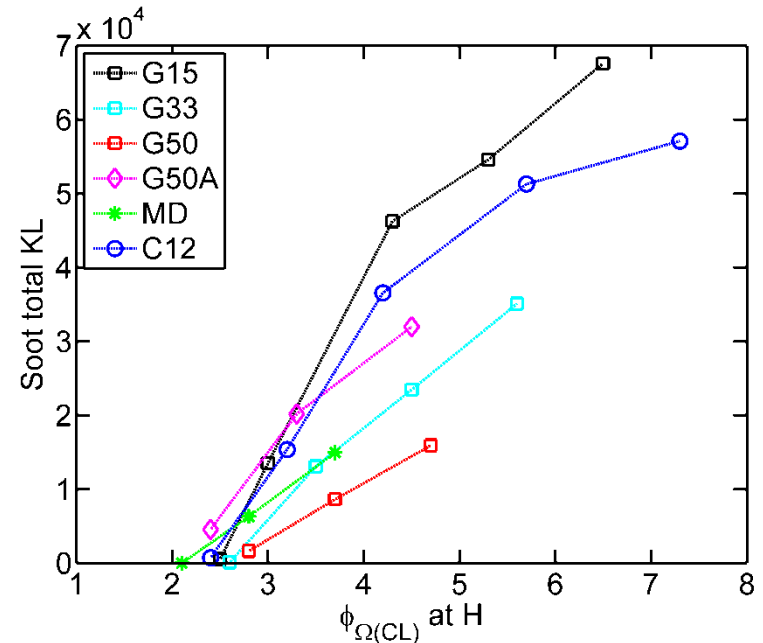
Oxygenated fuels appear to be naturally well suited for LLFC – lower air entrainment requirement & longer flame lift-off for a given ignition delay

# Accomplishments: Flame lift-off and soot



- Normalizing the fuel to an oxygen adjusted equivalence ratio at lift-off length highlights the benefit of highly oxygenated fuels

- Fuels producing lower amount of soot at short lift-off distances offer clear advantages
  - Low soot production while good engine/combustion control



- G50 appears to enable non-sooting combustion at richer mixtures.
- Aromatics may lead to a leaner equivalence ratio requirement for LLFC.

# Accomplishments: Modeling Fuel Effects

- Two primary fuel properties of interest in LLFC: CN,  $\Omega_f$
- Single-component fuel models cannot capture CN range

### THREE FUELS TESTED

- High-CN: 57 CN, Aromatics ~2%
- Mid-CN: 44 CN, Aromatics ~29%
- Low-CN: 41 CN, Aromatics ~32%

### Hydrocarbon Class

### Hydrocarbon Species

### Physical Surrogate

### Chemistry Surrogate (GCR)

lower MW  
normal paraffins

higher MW  
normal paraffins

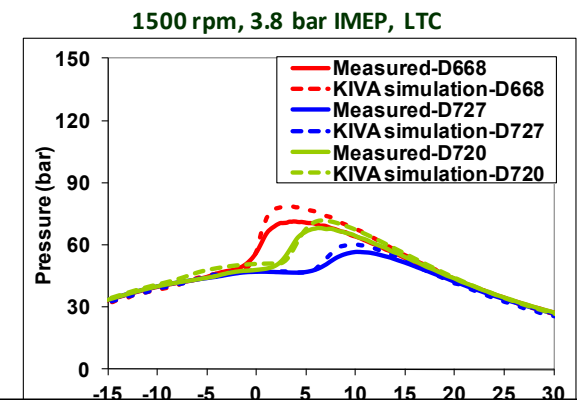
naphthenes

aromatics

Experimental Fuel

Hydrocarbon species	Mass fractions		
	High-CN	Mid-CN	Low-CN
n-decane	0.110	0.160	0.110
n-dodecane	0.092	0.000	0.000
n-tridecane	0.090	0.000	0.000
n-tetradecane	0.273	0.179	0.171
n-hexadecane	0.081	0.171	0.170
n-octadecane	0.171	0.081	0.041
n-heneicosane	0.090	0.090	0.094
Decalin	0.070	0.020	0.070
Naphthalene	0.005	0.017	0.000
Phenanthrene	0.000	0.049	0.048
m-xylene	0.013	0.050	0.041
n-pentylbenzene	0.000	0.040	0.010
Tetralin	0.000	0.052	0.122
n-heptylbenzene	0.000	0.078	0.104

Hydrocarbon species	Mass fractions		
	High-CN	Mid-CN	Low-CN
n-heptane	0.1150	0.1730	0.1290
n-tetradecane	0.7970	0.5210	0.4764
cyclohexane	0.0700	0.0200	0.0700
n-pentylbenzene	0.0180	0.2860	0.3246



Fuel effects on combustion and emissions were well captured over a wide range of operating conditions & parametric studies.

# Accomplishment: Oxygenate Modeling

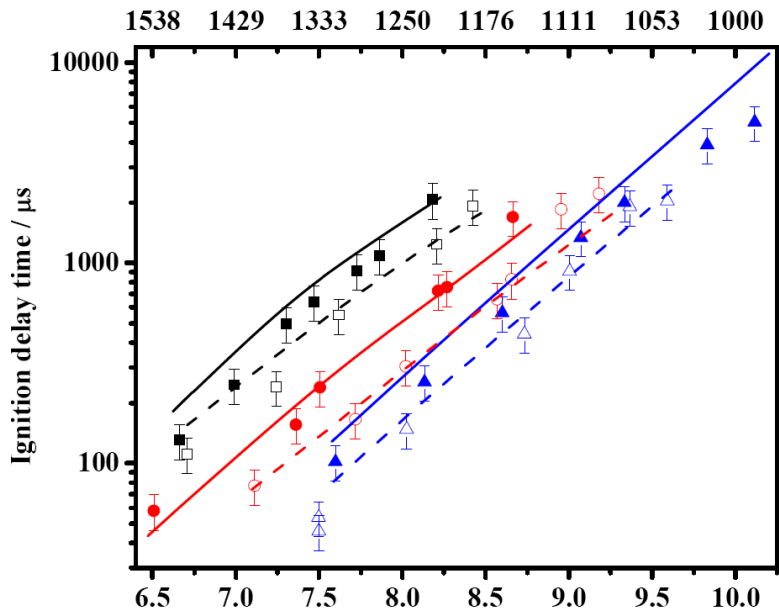
TPGME is the primary fuel being used in spray and engine testing in this project

Issue: increased ignition delay predicted with TPGME, misfire @ <1000 K ( $CN_{TPGME} > 60!$ )

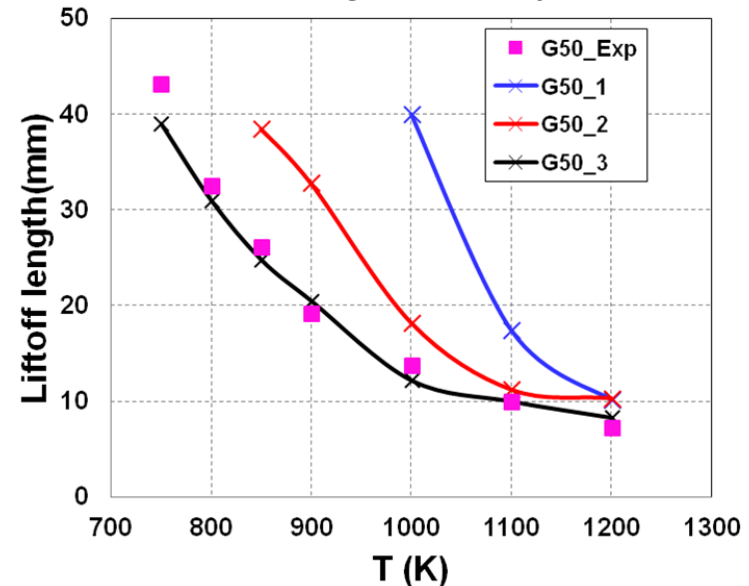
Conclusion: low temperature chemistry not well developed

Solution: engage Bill Pitz at LLNL to improve detailed TPGME kinetics model

Shock-tube ignition of TPGME/O<sub>2</sub>/Ar mixtures:



UW Modeling of Spray Data



- New full kinetic model shows reasonable agreement with shock tube data
- UW has reduced and refined this model – excellent agreement with Sandia spray data (ignition delay and lift-off length)



# Remaining Challenges and Barriers

- Spray studies provided a good foundation to understand the effect of fuel properties on the ability to achieve non-sooting conditions in a well controlled environment.
- Additional work is needed to understand how the complexity of the engine environment alters this picture.



# Planned Future Work

- Conduct optical engine studies to understand LLFC in an engine using T50.
- Leverage models to further understand the results of the spray studies.
- Improve combustion system to facilitate LLFC - implement changes on the Ford metal single-cylinder engine.
- Demonstrate LLFC on the Ford single-cylinder engine evaluating several different fuel blends with varying fuel oxygen ratio.
- Project is planned to be completed by December 2014.



# Summary

**Relevance:** Project is aimed at understand the role fuels can play to enable LLFC.

**Approach:** Through collaboration with Sandia & UW, use a suite of tools including spray vessel, optical engine, metal engine and CFD to improve fundamental understanding and model capability.

## **Technical Accomplishments:**

- Found increasing flame lift-off length with fuel oxygenation – new finding.
- Demonstrated LLFC over a wide set of conditions with  $\Omega_f \sim 6$ .
- Demonstrated CFD models that are able to capture the effect of CN.
- Validated CFD model for TPGME blends – ready to use.

**Collaboration:** Core team includes researchers from Ford, Sandia National Labs, UW-Madison and LLNL with support from NIUG.

## **Future Work:**

- Use CFD models to help better explain spray results.
- Complete optical and metal single-cylinder experiments to demonstrate LLFC.



# QUESTIONS?

Note: This project was not reviewed at the 2013 AMR.





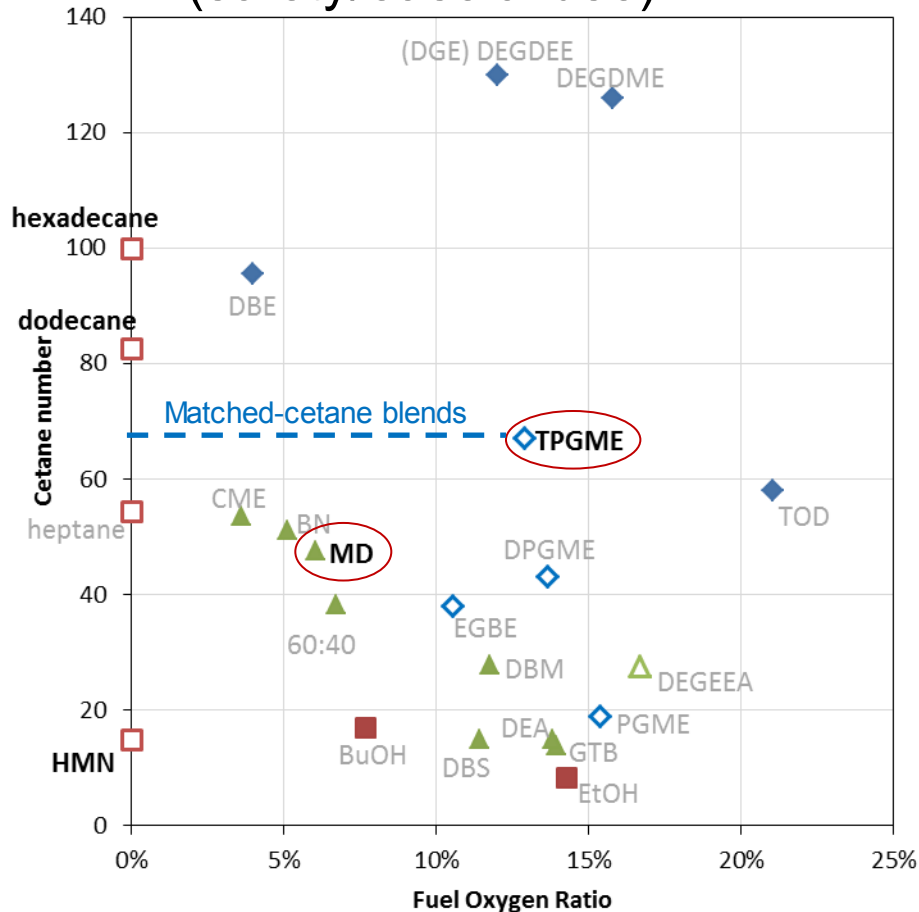
# Technical Back-Up Slides

(Note: please include this “separator” slide if you are including back-up technical slides (maximum of five technical back-up slides). These back-up technical slides will be available for your presentation and will be included in the DVD and Web PDF files released to the public.)

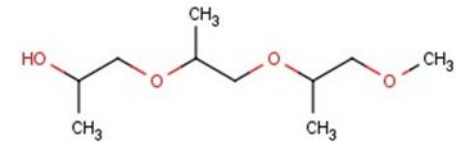
# Oxygenates Considered



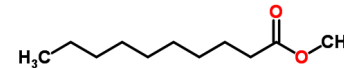
## Screened list of oxygenates (safety/ease of use)



- Tri(propylene glycol) methyl ether (TPGME, 8 isomers, C<sub>10</sub>H<sub>22</sub>O<sub>4</sub>)



- Methyl decanoate (C<sub>11</sub>H<sub>22</sub>O<sub>2</sub>)

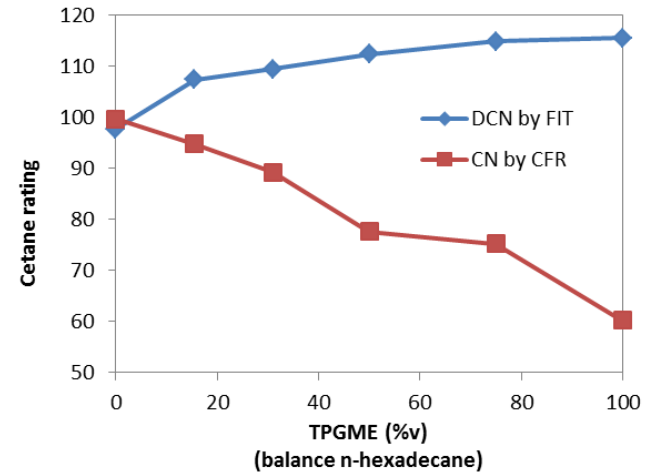


- ◆ ethers
- ◇ glycol ethers
- △ ester ethers
- ▲ esters
- ketones
- alcohols
- n-alkanes

# Cetane Rating Measurements

- Attempted to create “matched-cetane” blends of TPGME in PRF
  - PRFs are blends of NHD (n-hexadecane) and HMN (2,2,4,4,6,8,8-heptamethylnonane)
  - PRFs have method-defined CNs on CFR engine
- Different TPGME cetane ratings by CFR engine, FIT, and IQT

	Meas.	Literature
CFR (CN)	64 ± 4	63 (blend)
FIT (DCN)	108 ± 8	n/a
IQT (DCN)	79	74 (neat)

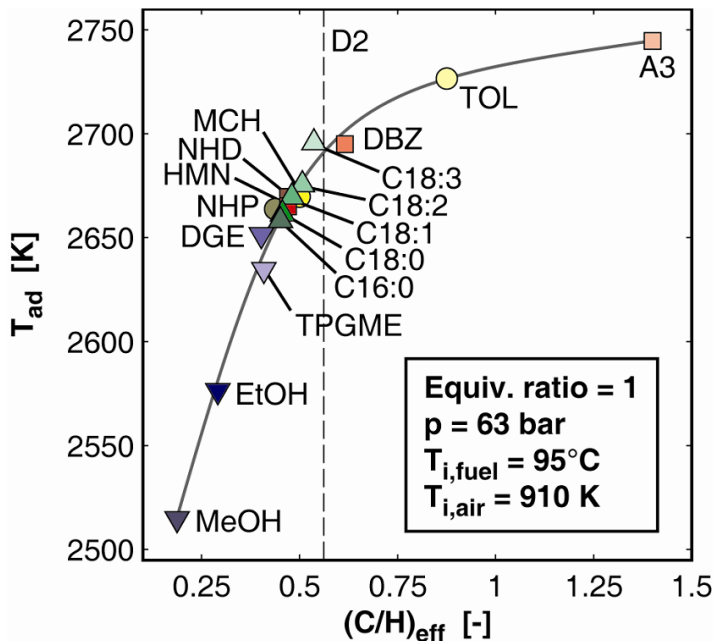


- NHD selected as base fuel (DCN=CN=100)
  - Matched DCN of TPGME (by FIT)
  - Simple composition
- Similar behavior in blends with NHD and a surrogate reference fuel.
- Cetane rating methods give substantially different results for these fuels. Explanations being investigated.*
- Sandia spray lab testing later indicated ignition delays for TPGME-NHD blends were similar.*

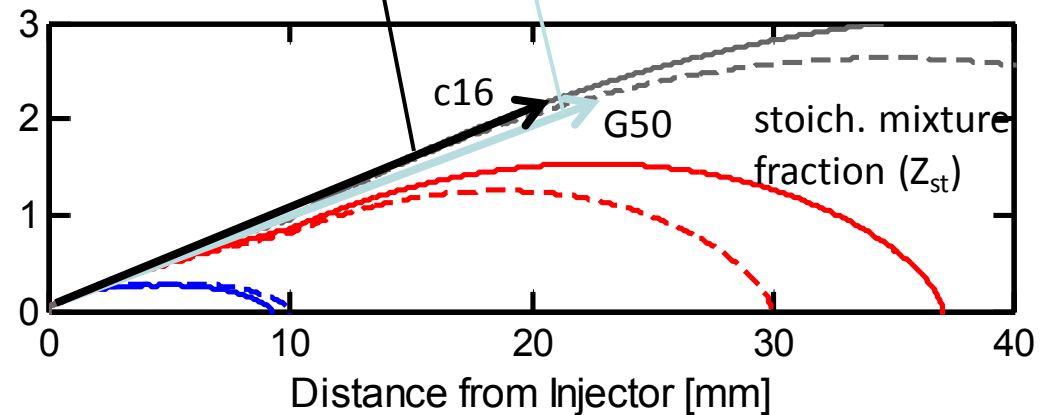
# Lift-off length: Equivalence ratio and flame temperature

- Due to the presence of oxygen in the fuel, oxygenated fuels produce igniting mixtures shifted toward the center of the jet
  - Because of the different stoichiometric contour locations, lift-off length is expected to be further downstream for oxygenated fuels

A fuel parcel following a stoichiometric contour mixing path, travels farther downstream for G50 compared to C16 over the same ignition delay



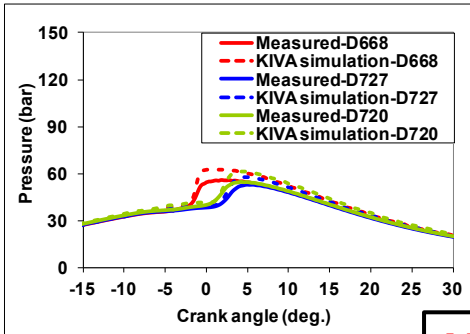
Mueller et al. SAE paper 2009-01-1792



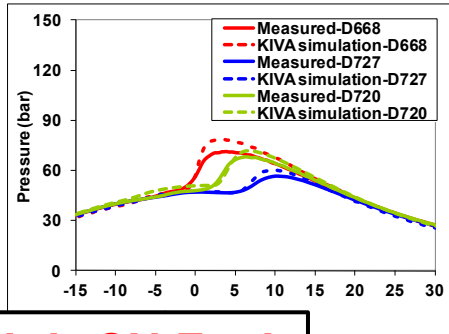
- Stoichiometric adiabatic flame temperature may also play a role in the explanation towards longer lifted flames with oxygenated blends
  - Lower adiabatic flame temperature fuels are expected to have decreased reaction rates at lift-off length which might extend lift-off length

# CN - Surrogate Model Validation

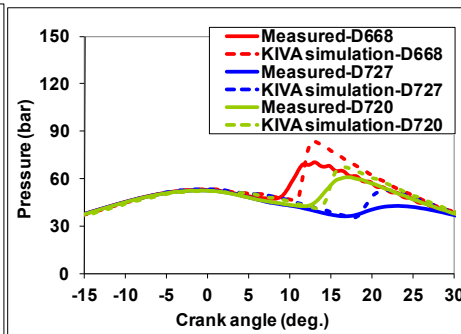
750 rpm, 2.2 bar IMEP, LTC



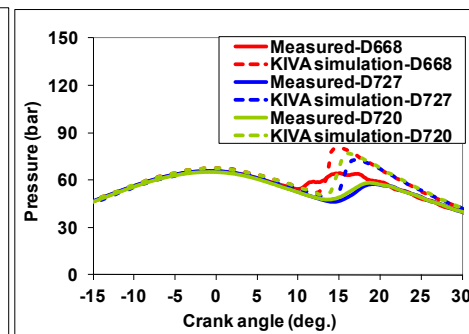
1500 rpm, 3.8 bar IMEP, LTC



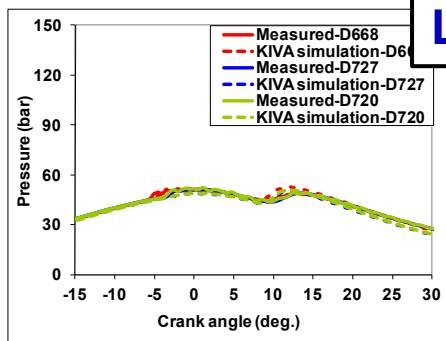
1500 rpm, 6.8 bar IMEP, LTC



2500 rpm, 5.4 bar IMEP, LTC

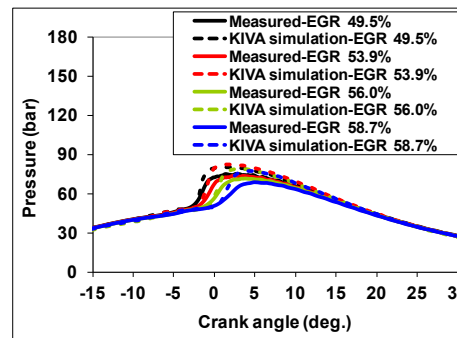


1500 rpm, 3.7 bar IMEP, conventional

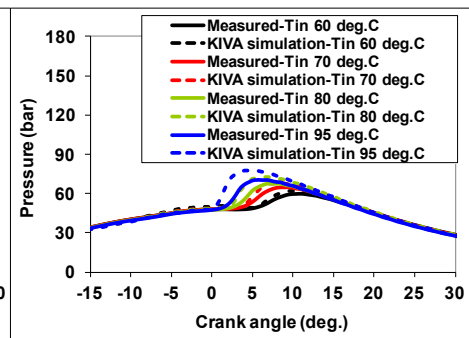


**High-CN Fuel**  
**Mid-CN Fuel**  
**Low-CN Fuel**

EGR Sweep-1500rpm, 3.7 bar IMEP

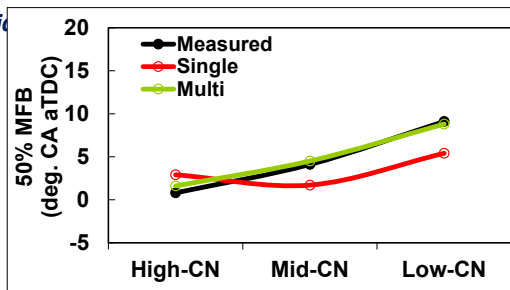


T<sub>intake</sub> Sweep-1500rpm, 3.7 bar IMEP

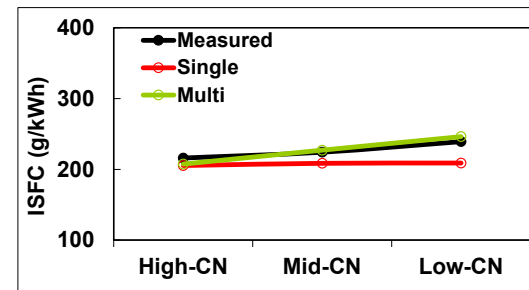


Note: D668 – High-CN Diesel Fuel, D727-Low-CN Diesel Fuel, D720-Mid-CN Diesel Fuel

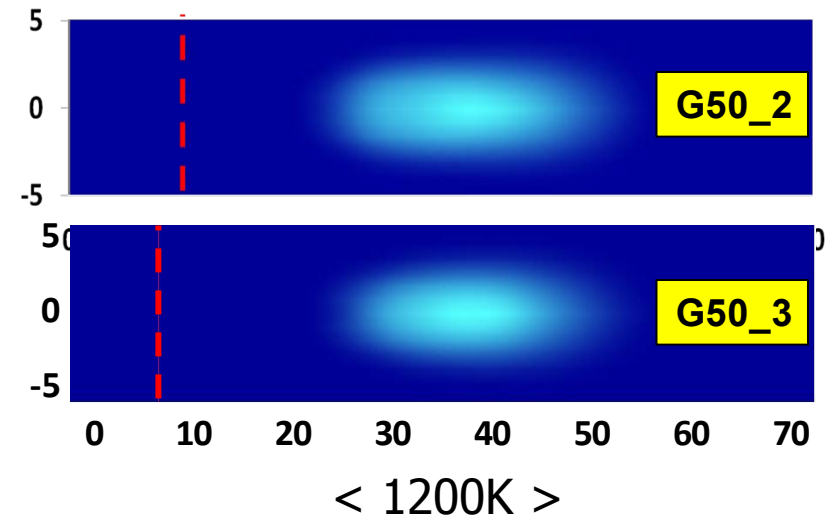
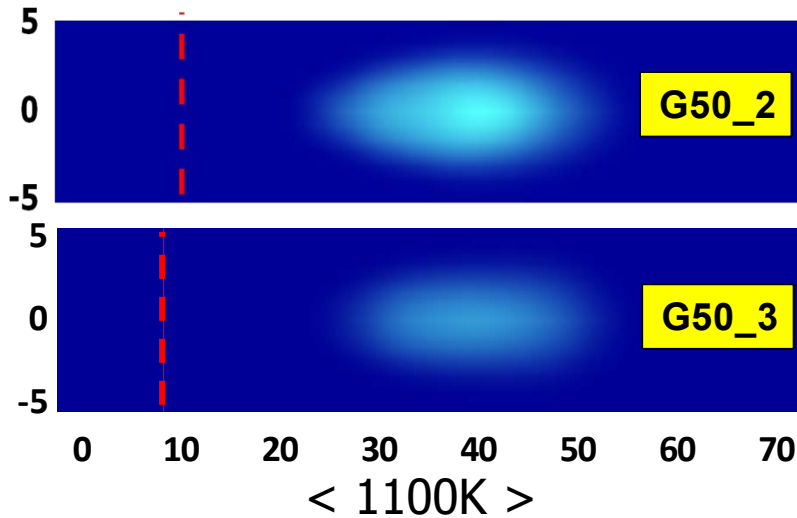
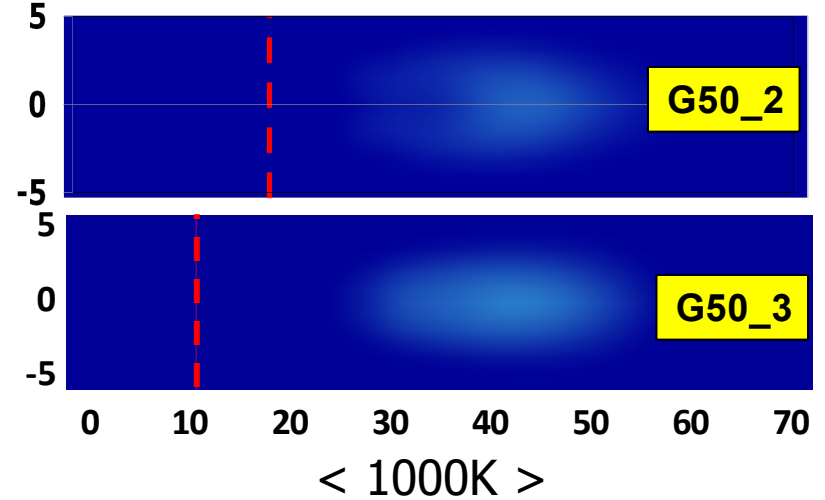
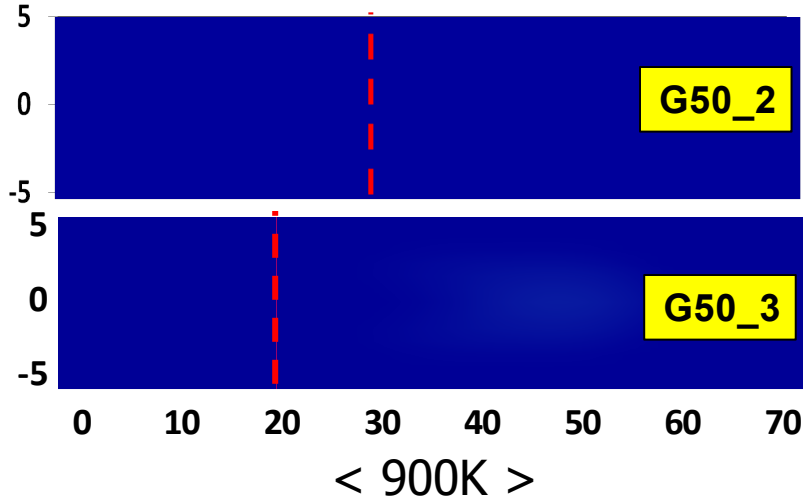
**Validated model at various speed-load points, EGR, intake temperatures and combustion modes.**



D720



# G50\_2 & G50\_3 Soot Prediction Comparisons



\* Red dashed line = Liftoff length

**Distance from injector orifice (mm)**