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Overview of Applied Battery Research

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Chemical Sciences & Engineering Division

Tuesday, February 26, 2008

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Vehicle Technologies Program



Topics Covered

- Key Energy Storage Requirements (HEV & PHEV)
- Overall Program Objective & Approach
- Focus Areas & National Lab Collaborations
 - Calendar Life Activities & Past Examples
 - Abuse Tolerance Activities & Past Examples
 - Low-Cost Materials Activities & Past Examples
 - Low-Temperature Performance Activities & Past Examples
- Major Reports & Presentation Sequence
- Acknowledgements

FreedomCAR Energy Storage Goals (HEVs)

	42-Volt		Power Assist HEV	
	M-HEV	P-HEV	Min.	Max.
Discharge Power, kW	13 (2s)	18 (10s)	25 (10s)	40 (10s)
Regen Power, kW	8 (2s)	18 (2s)	20 (10s)	35 (10s)
Available Energy	0.3	0.7	0.3	0.5
Cold Cranking Power*, kW	8 @ 21V minimum		5	7
Calendar Life, years	15			
Production Price**, \$	260	360	500	800
Operating Temperature, °C	-30/52			

* Three 2s pulses at -30°C with 10s rest between pulses

** Price based on 100,000 batteries/year production level

Also, batteries need to conform with all FMVSS!

FreedomCAR Energy Storage Goals (PHEVs)

	Short-Term	Long-Term
	SUV	Car
Discharge Power, kW	45	38
Regen Power, kW	30	25
Available Energy (CD), kWh	3.4	11.6
Available Energy (CS), kWh	0.5	0.3
Cold Cranking Power*, kW	7	
Calendar Life, years	15	15
Production Price**, \$	1,700	3,400
Operating Temperature, °C	-30 to 52	

* Three 2s pulses at -30°C with 10s rest between pulses

** Price based on 100,000 batteries/year production level

Also, batteries need to conform with all FMVSS!

Program Objective & Approach

Objective:

Assist the industrial developers of high-power & high-energy Li-Ion batteries to overcome the cost, life, abuse tolerance, and low-temperature performance barriers for this promising technology!

Approach (material & cell level studies):

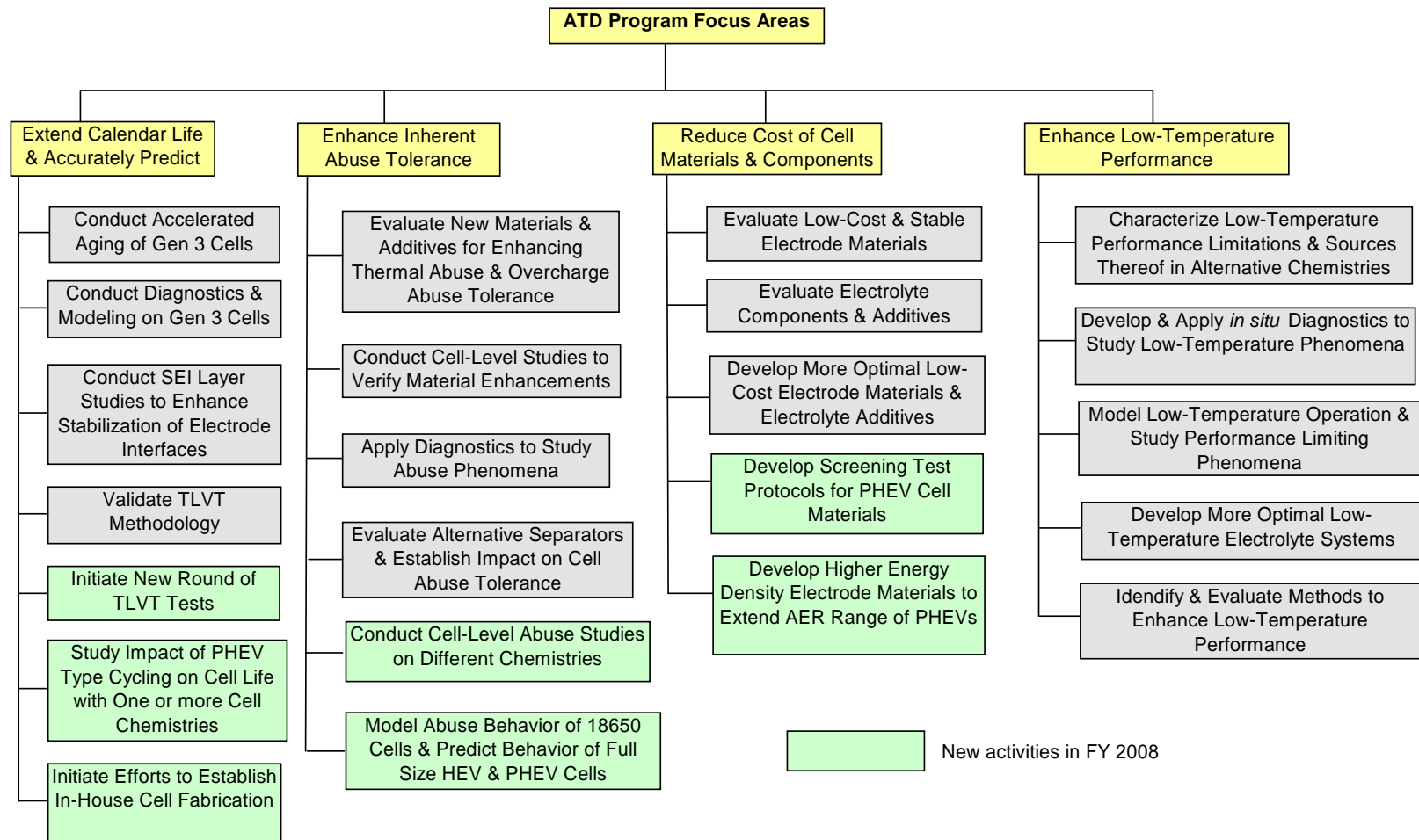
- Understand factors that control life, inherent abuse tolerance, & low-temperature performance of alternative Li-Ion cell chemistries
- Use knowledge to identify & develop low-cost cell materials that are more stable structurally, chemically, electrochemically, & thermally
- Demonstrate aging, abuse tolerance, & low-temperature performance characteristics of advanced chemistries in sealed cells

Focus Areas

(High-Power HEV & High-Energy PHEV)

- Understand factors that limit calendar life of Li-Ion cell chemistries, identify & develop cell materials that extend the calendar life, study factors that affect SEI stability, & validate TLVT methodology
- Understand factors that limit the abuse tolerance of Li-Ion cell chemistries and evaluate approaches to enhance inherent abuse tolerance of Li-Ion cells
- Search for and develop low-cost cell materials, components, and technologies with enhanced stability that perform better at low-temperature
- Understand factors that limit low-temperature performance of Li-Ion cell chemistries and evaluate approaches to enhance performance at low temperature

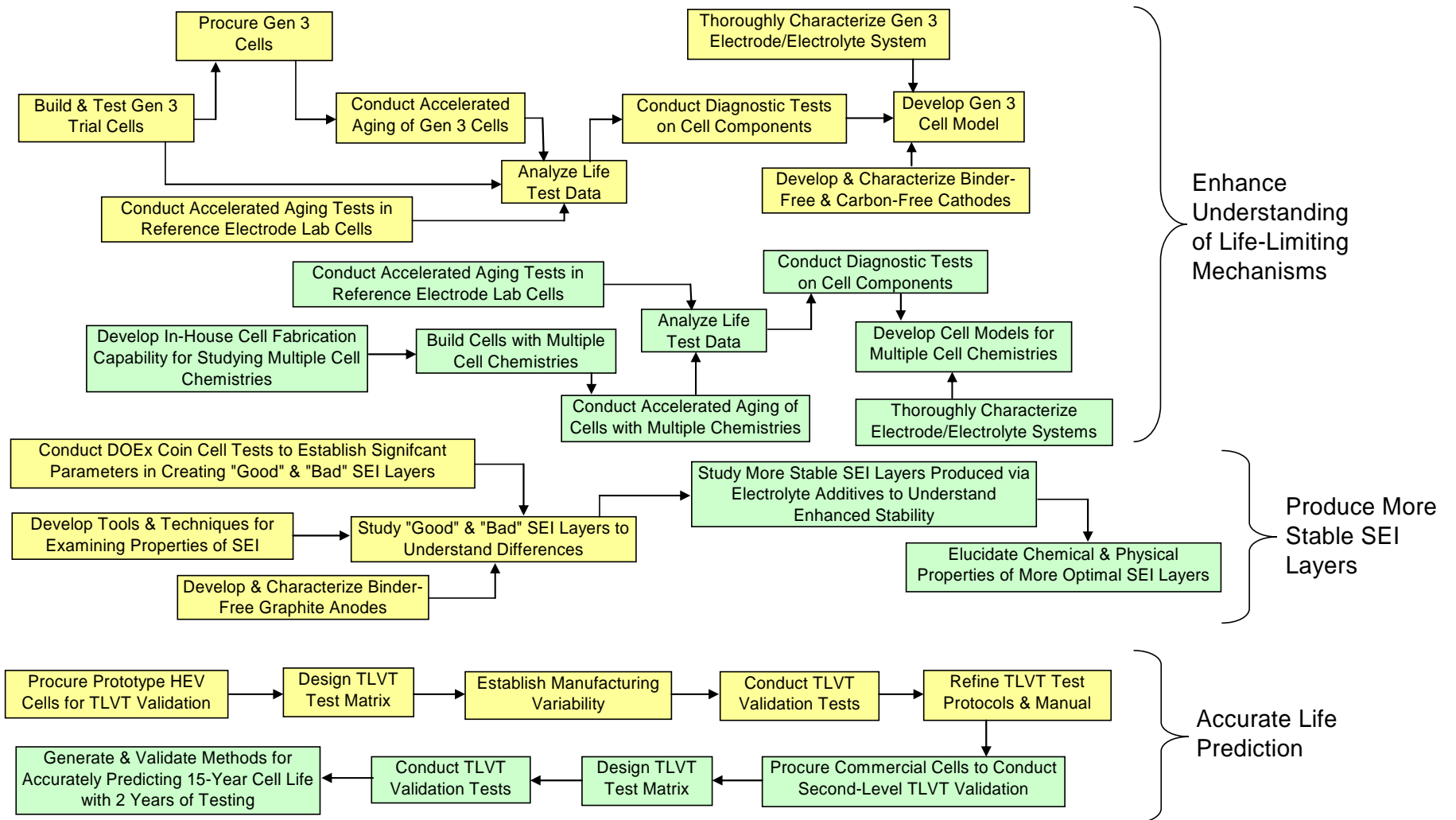
FY2008 Focus Area Activities



ATD Program Multi-Laboratory Collaborations

Lab	Calendar Life				Abuse Tolerance	Low Cost	Low-Temp Perform
	TLVT	SEI	Aging	Diagnostics			
ANL	X	X	X	X	X	X	X
BNL				X			X
INL	X	X	X				X
LBNL	X	X		X			
SNL	X				X		

Calendar Life



Recent Changes to Calendar Life Studies

- Study aging mechanisms in multiple cell chemistries—establishing capability to fabricate cells within the program to facilitate this
- Study impact of PHEV type (deeper discharge) cycles vs. HEV type cycles on life
- Expand SEI/formation studies to investigate SEI layers formed via the use of electrolyte additives
- Conduct second-level TLVT tests using commercial cells to further validate TLVT test protocols

Cell Chemistries Studied

Large quantities of Gen 1 & Gen 2 Cells were built & tested

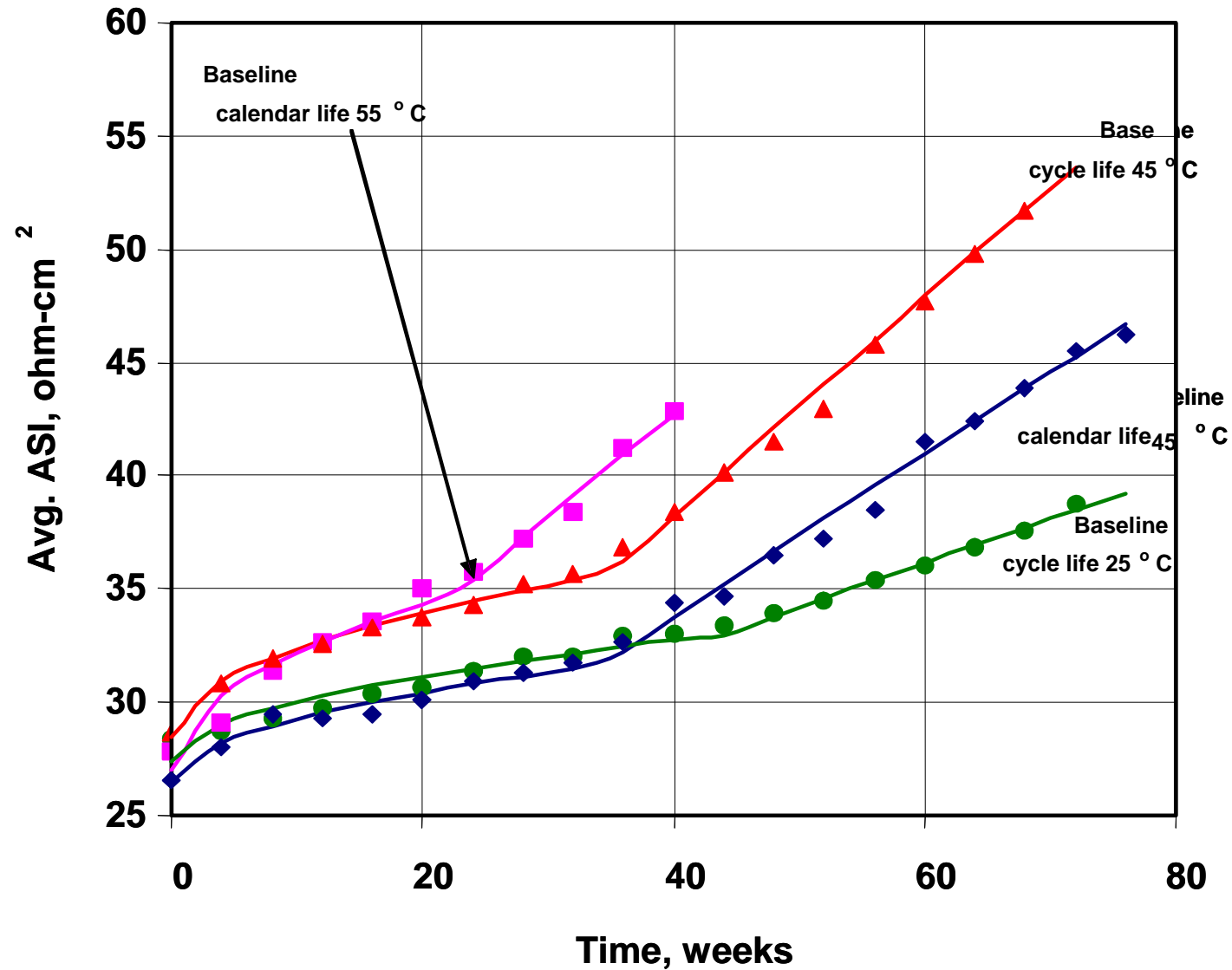
	Gen 1 Baseline	Gen 2 Baseline
Positive Electrode	8 wt % PVDF binder 4 wt % SFG-6 graphite 4 wt % carbon black 84 wt % $\text{LiNi}_{0.8}\text{Co}_{0.2}\text{O}_2$	8 wt % PVDF binder 4 wt % SFG-6 graphite 4 wt % carbon black 84 wt % $\text{LiNi}_{0.8}\text{Co}_{0.15}\text{Al}_{0.05}\text{O}_2$
Negative Electrode	9 wt % PVDF binder 16 wt % SFG-6 graphite 75 wt % MCMB-6 graphite	8 wt % PVDF binder 92 wt % MAG-10 graphite
Electrolyte	1M LiPF_6 in EC:DEC (1:1)	1.2M LiPF_6 in EC:EMC (3:7)
Separator	25 μm thick PE	25 μm thick PE

Small quantities of Gen 3 cells were built by numerous cell manufacturers

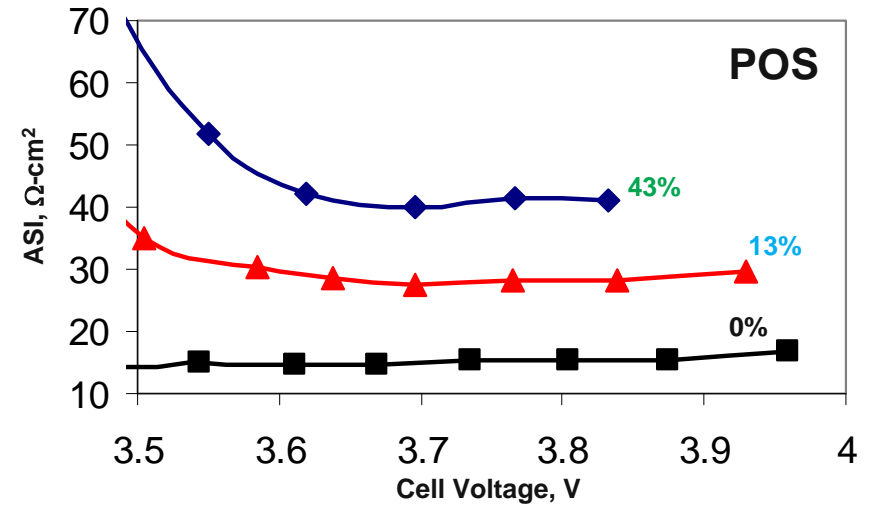
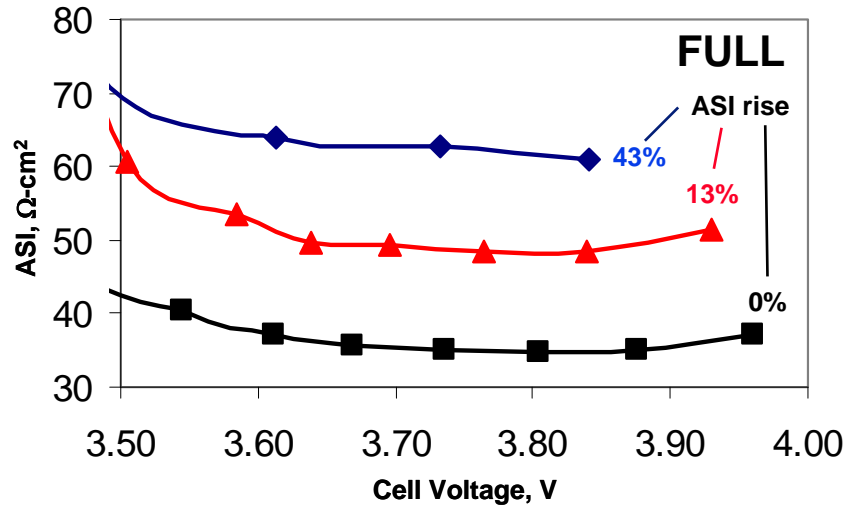
	Gen 3 Baseline	Gen 3 Variant A
Positive Electrode	84% Seimi L-333 (B) 8% TB5500 Carbon Black 8% PVDF Binder (KF#1120)	84% Seimi L-333 (B) 8% TB5500 Carbon Black 8% PVDF Binder (KF#1120)
Negative Electrode	92% MCMB 10-28 graphite 8% PVDF Binder (KF#9130)	92% MCMB 10-28 graphite 8% PVDF Binder (KF#9130)
Electrolyte	1.2M LiPF_6 in EC:EMC (3:7)	1.2M LiPF_6 in EC:EMC (3:7) with 2-3% $\text{LiC}_2\text{O}_4\text{BF}_2$ additive
Separator	25 μm thick PE	25 μm thick PE

Note: L333 = NMC = $\text{Li}[\text{Li}_x(\text{Ni}_{1/3}\text{Mn}_{1/3}\text{Co}_{1/3})_{1-x}]\text{O}_2$ where $X=0.05$

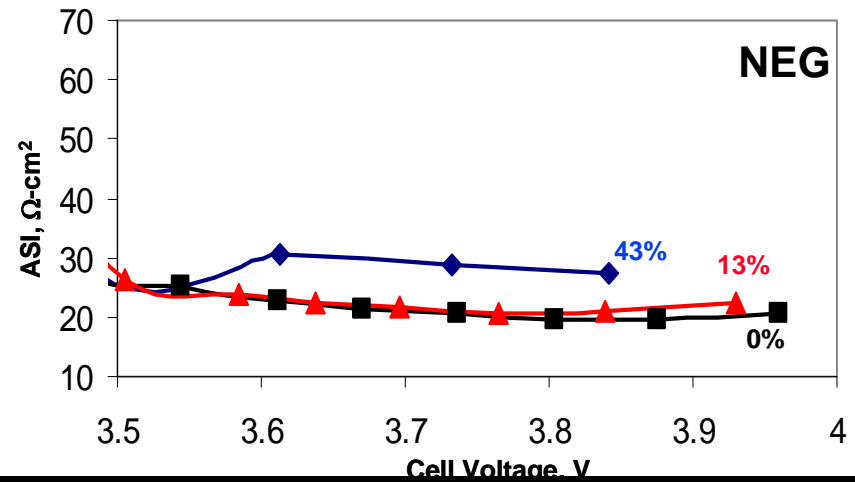
Gen 2 Cell Aging Data



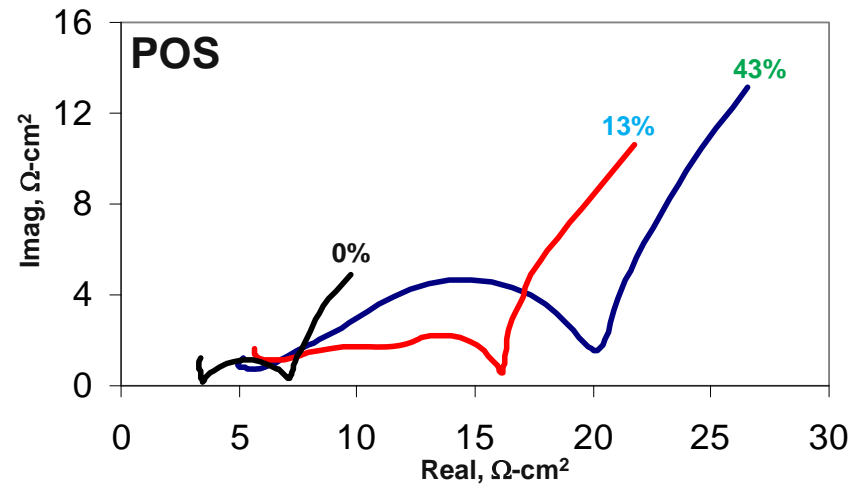
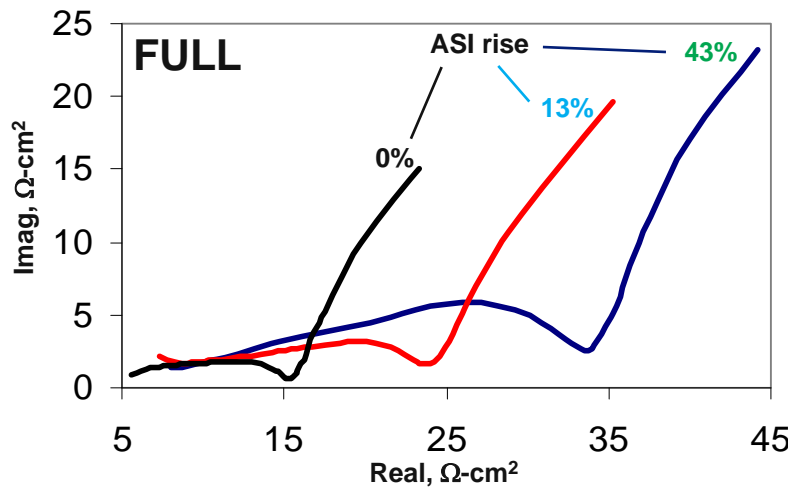
Reference Electrode Cell Data on Electrodes Harvested from 18650-Cell Aged at 55°C



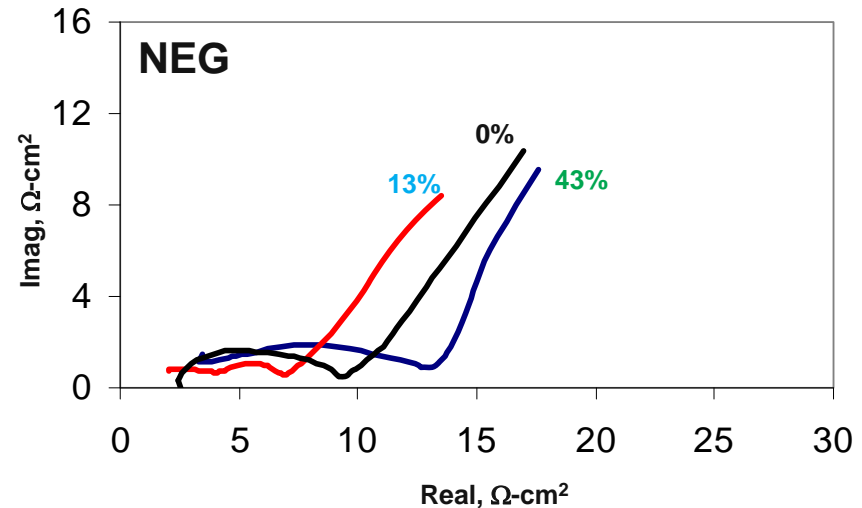
Cell impedance rise is dominated by positive electrode impedance increase.



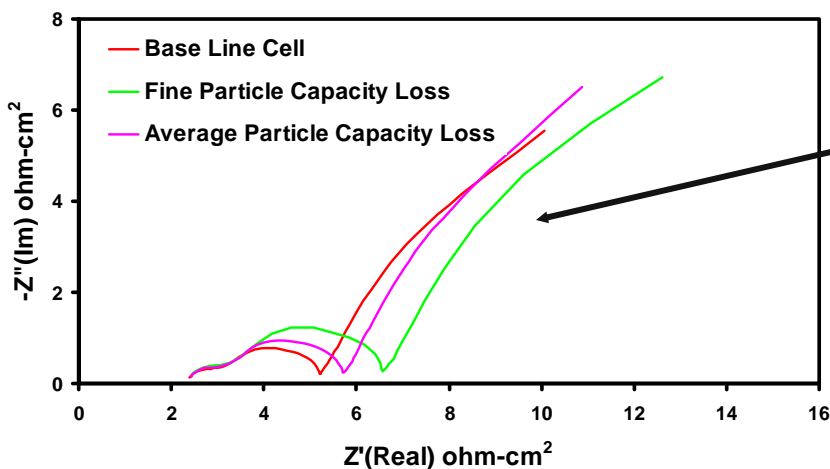
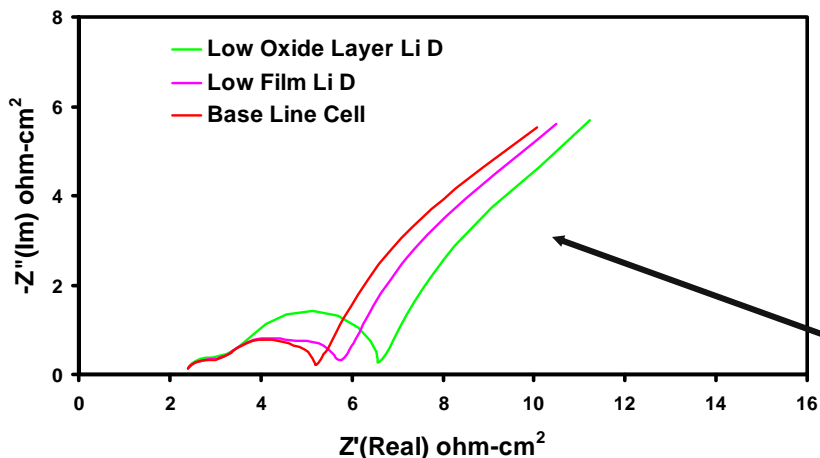
Reference Electrode Cell Data on Electrodes Harvested from 18650-Cell Aged at 55°C



Major source of impedance rise is interfacial impedance at the positive electrode



Electrochemical Cell Modeling -Positive Electrode-



Results indicate that 2 most probable mechanisms for explaining positive electrode EIS data are:

- 1. Reduced interfacial diffusion due to changes in surface films or in oxide surface layer**
- 2. Preferential isolation of small active material particles**

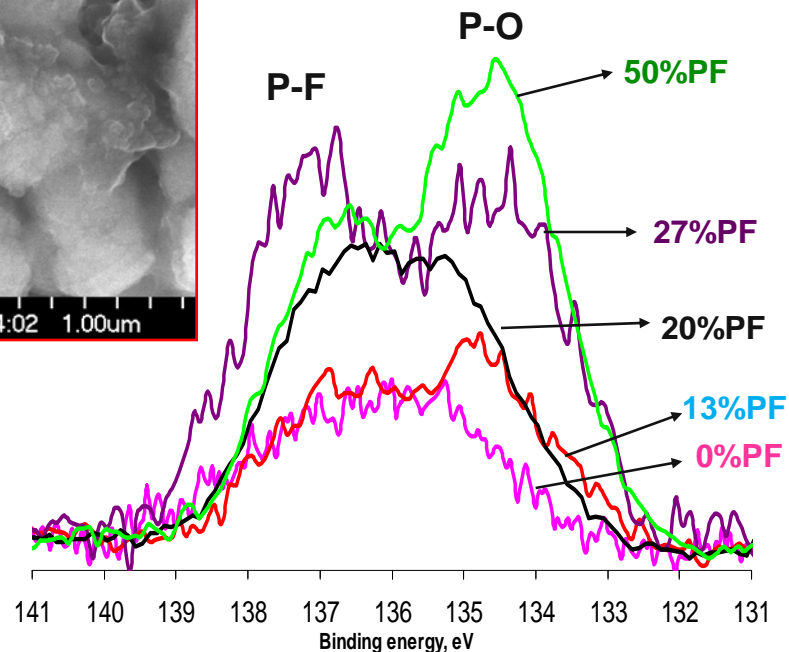
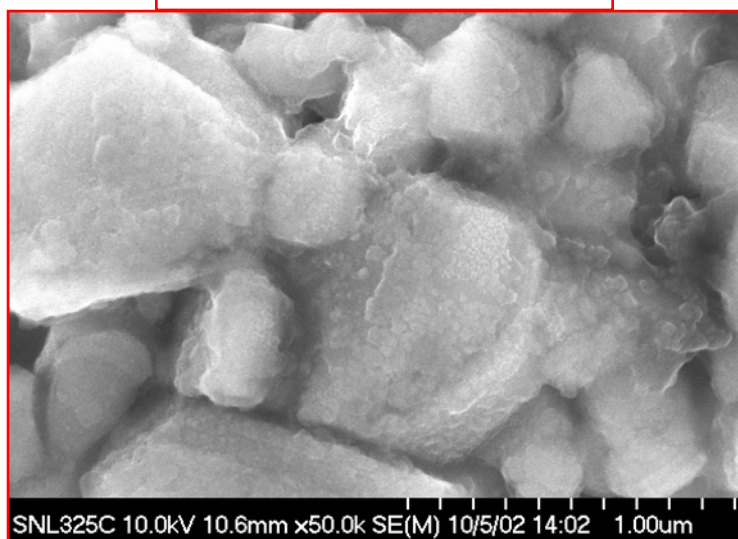
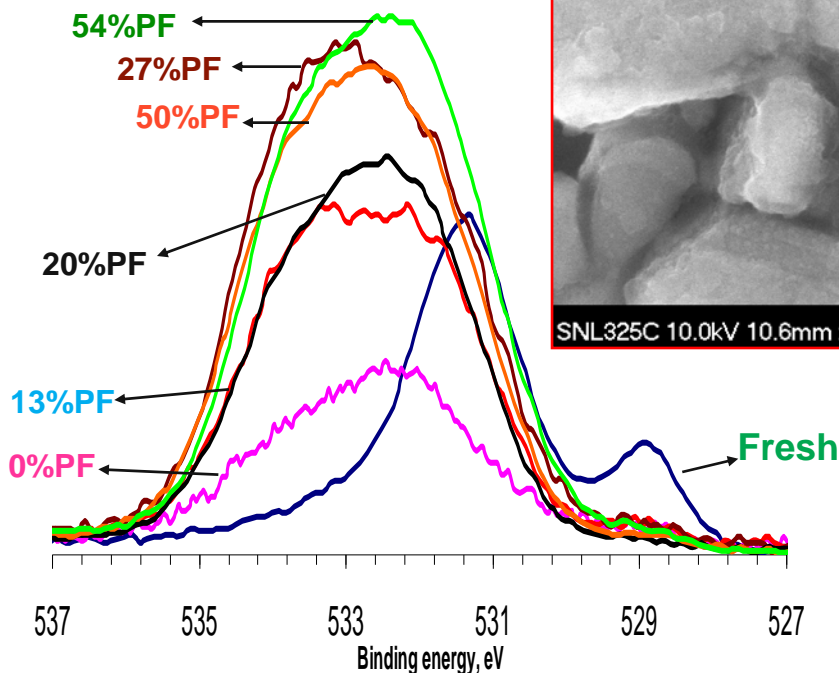
EXAMPLE: Cathode Surface Film Changes during Aging

XPS analysis area: 1 sq. mm.

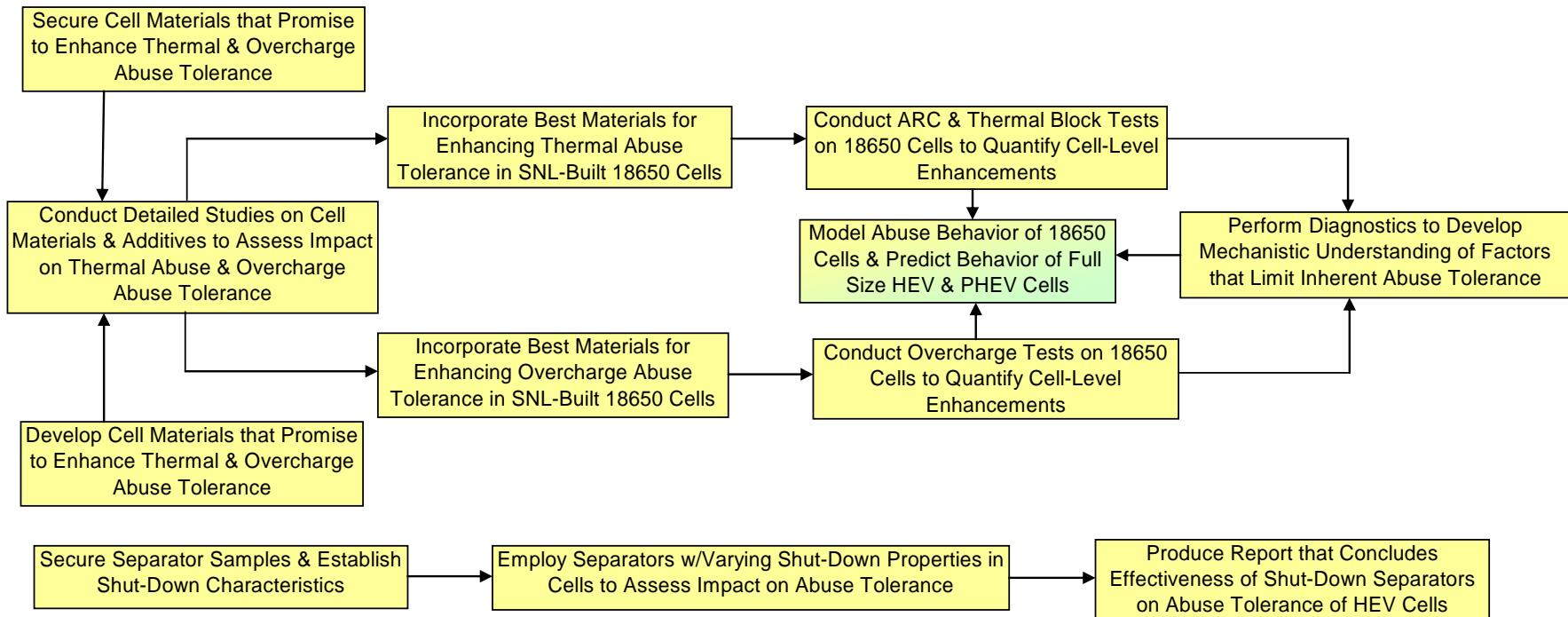
O1s XPS spectra show ROR/ROLi peak growth on aging

SEM image
SNL325, 45° C, 80%SOC

P2p XPS spectra show P-O bond changes on aging



Abuse Tolerance

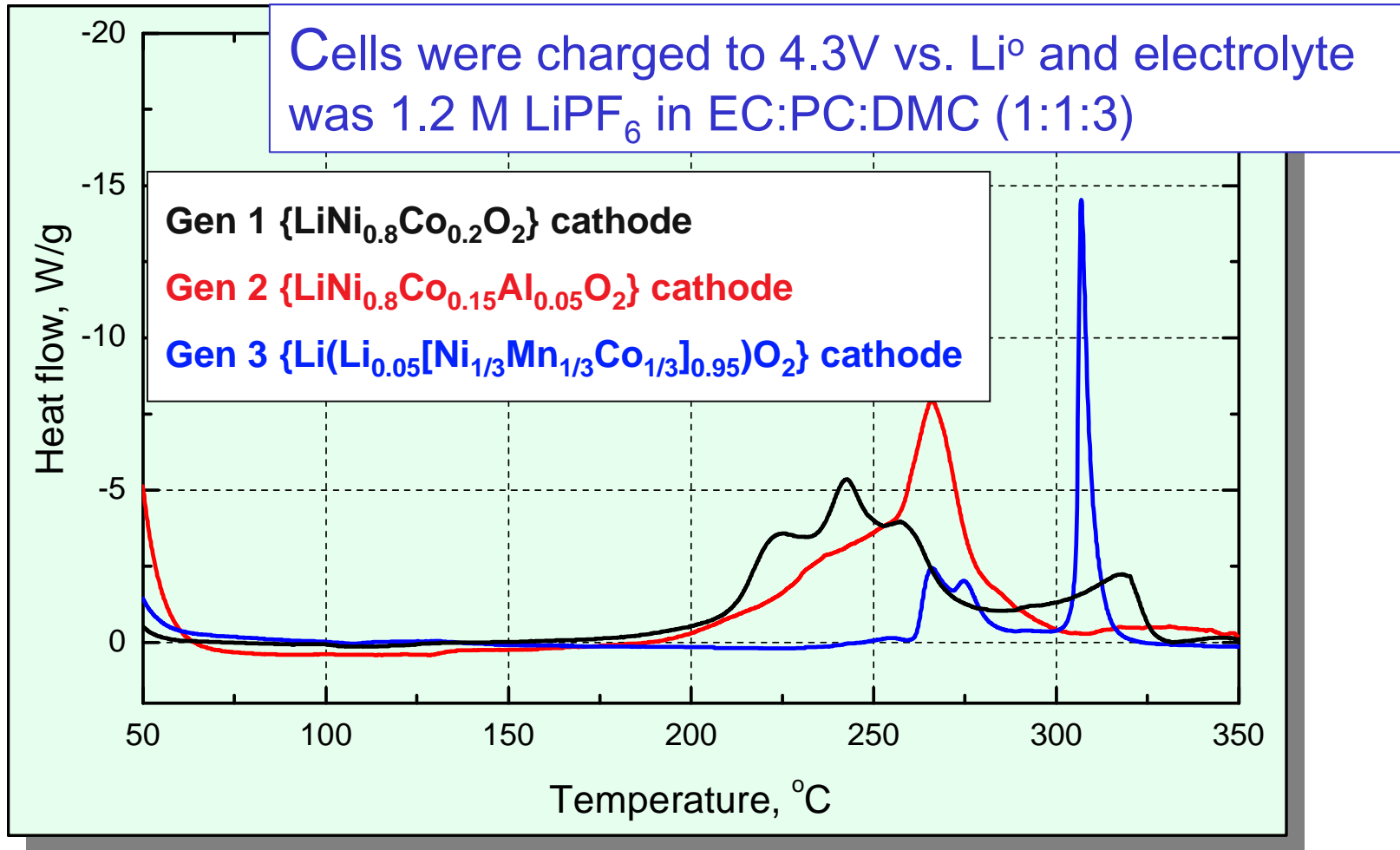


18650 cell-level abuse tests are being extended to include multiple baseline cell chemistries and electrode coating thicknesses representative of those used in PHEV batteries.

Changes to Abuse Tolerance Studies

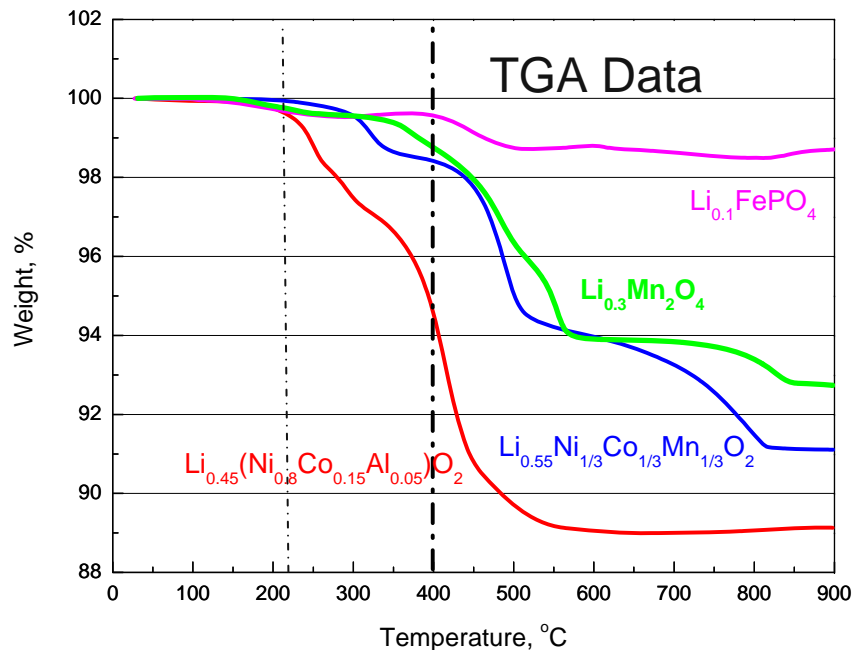
- Study abuse tolerance characteristics of multiple cell chemistries
- Study impact of going to thicker electrode coatings (lower P/E ratio) representative of PHEV cells
- Model abuse behavior of 18650 cells & predict behavior of full-size HEV & PHEV cells

Comparative DSC Data on Cathodes

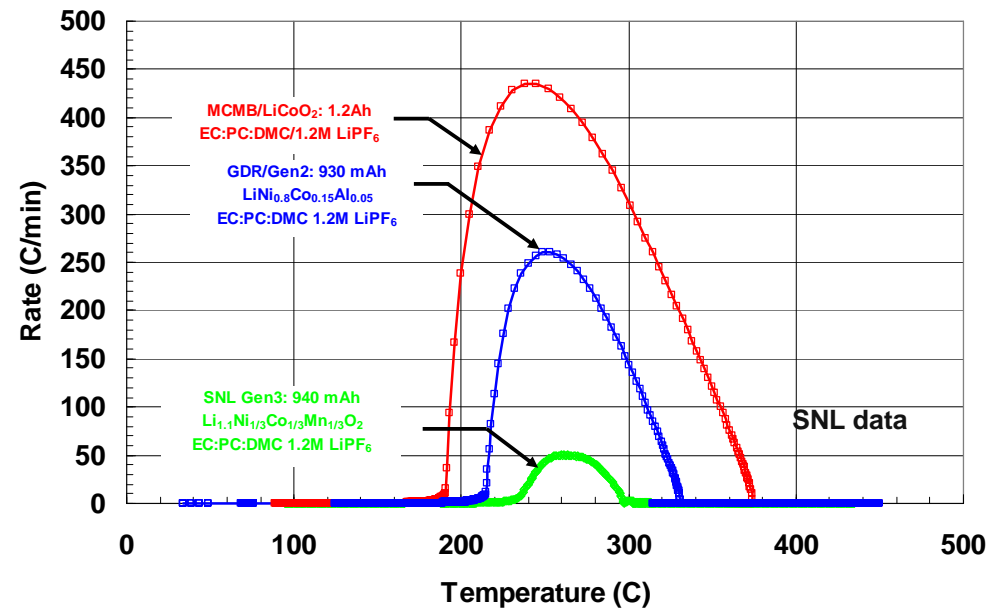


Relative Reactivity of Alternative Cathode Materials (Major component of thermal runaway)

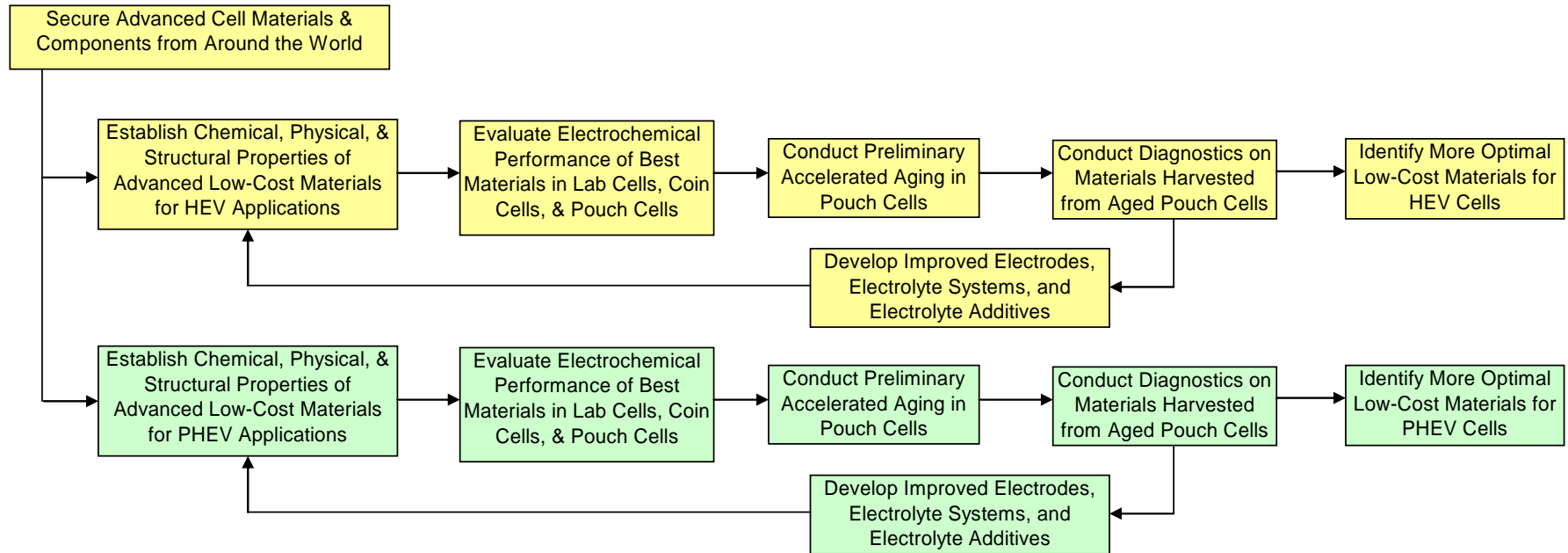
- ❑ Oxygen released from delithiated metal oxide cathodes due to decomposition
- ❑ Released oxygen reacts with organic carbonate solvents in the electrolyte generating heat



ARC Data on 18650 cells



Low-Cost Cell Materials & Components

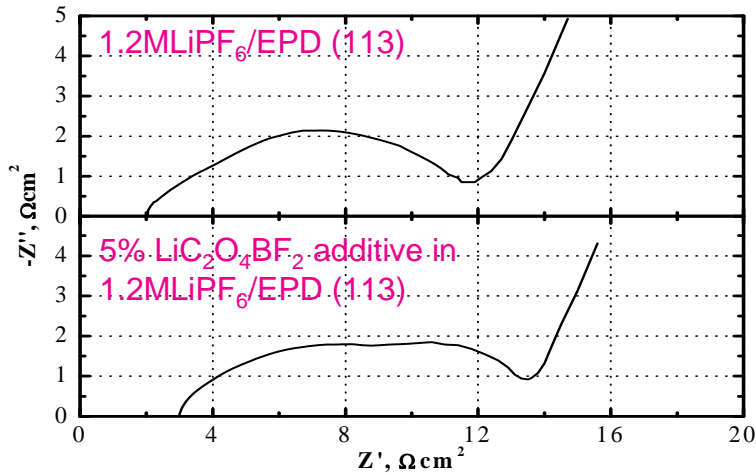


Changes to Low-Cost Materials Study

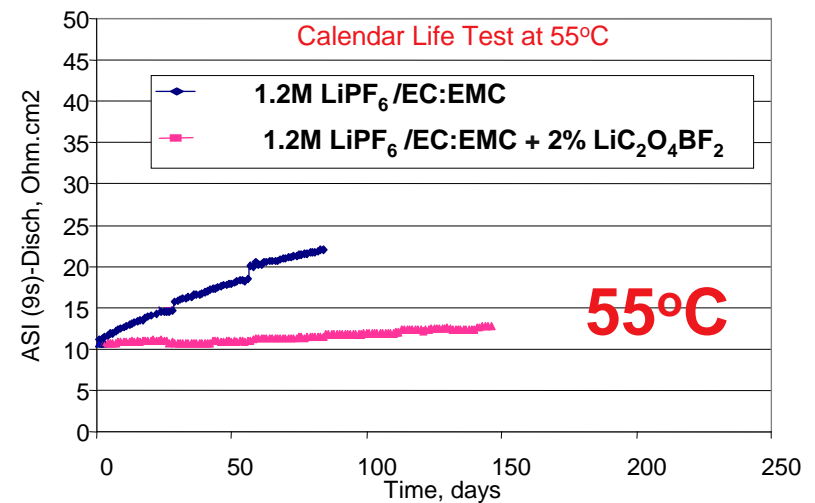
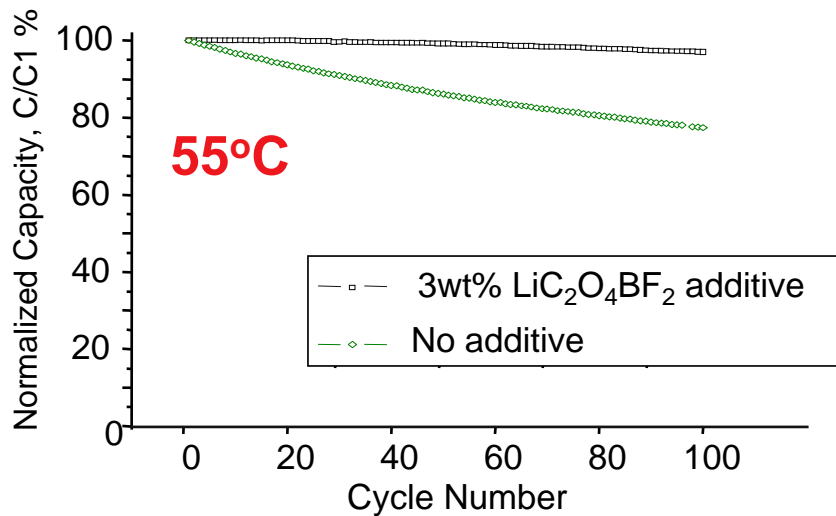
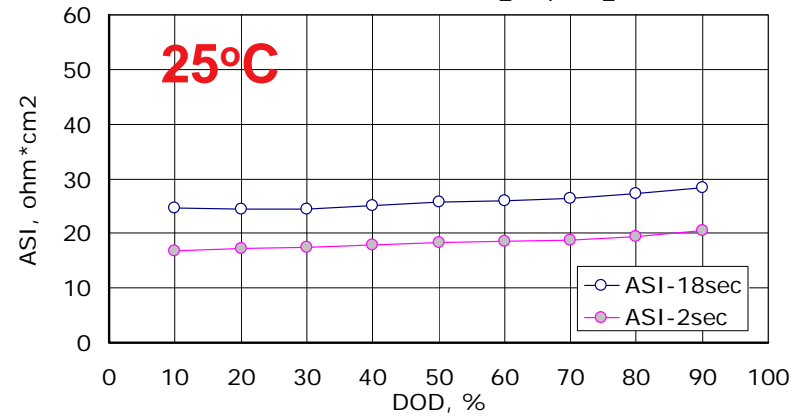
- Develop screening test protocols for PHEV application
- Expand screening tests to include materials applicable for use in PHEV cells
- Expand material development activities to include materials applicable for use in PHEV cells (e.g. electrode materials with higher specific capacity)

Electrolyte Additive Stabilizes Graphite/L333 (Gen 3) Chemistry

EPD: EC/PC/DMC (1:1:3)



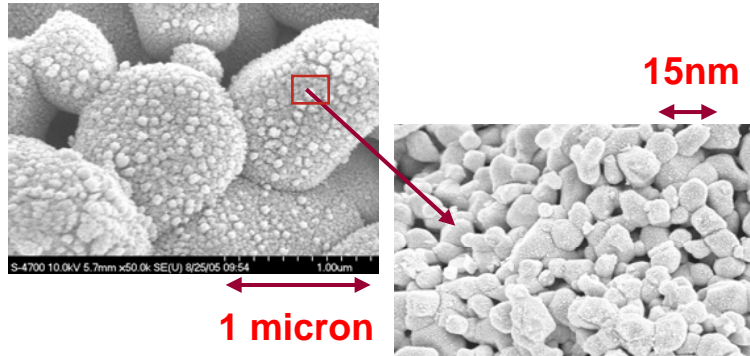
HPPC at 10C rate of graphite/L333 + EC/EMC with 3% LiC₂O₄BF₂ additive



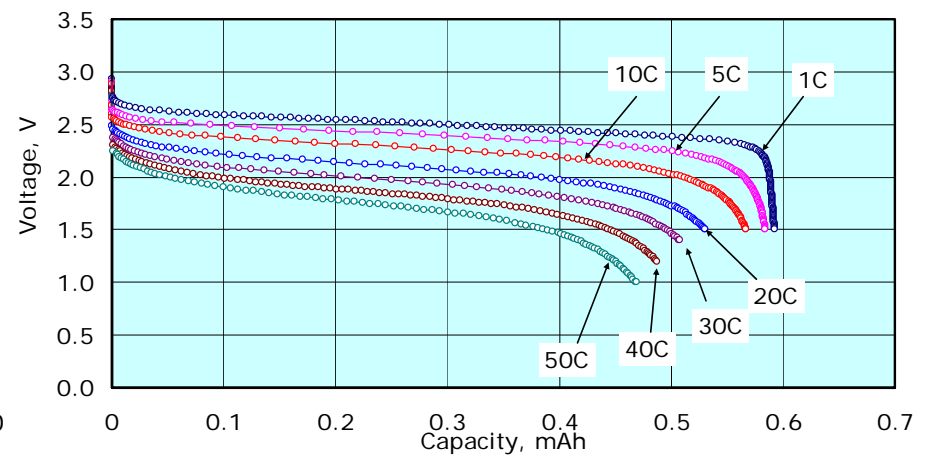
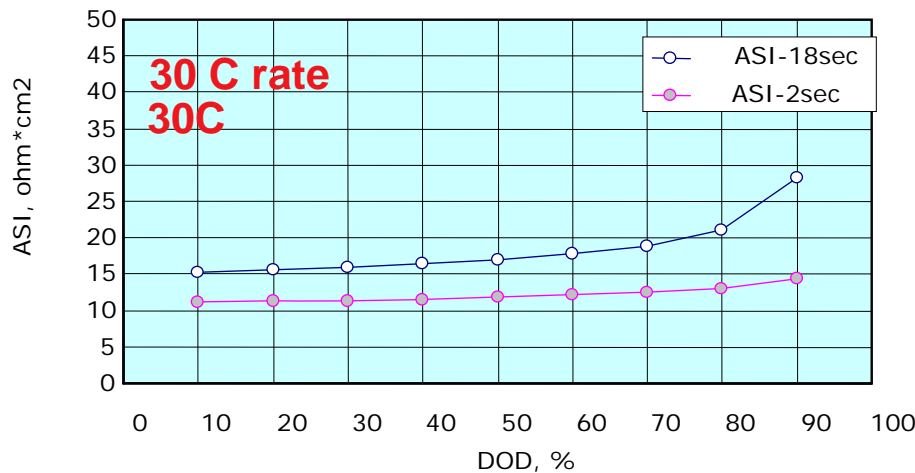
Demonstrated Feasibility of New High-Power Cell Chemistry

Nano-phase $\text{Li}_4\text{Ti}_5\text{O}_{12}/\text{Li}_{1.06}\text{Mn}_{1.94}\text{O}_4$ spinel

Argonne's nano-phase $\text{Li}_4\text{Ti}_5\text{O}_{12}$

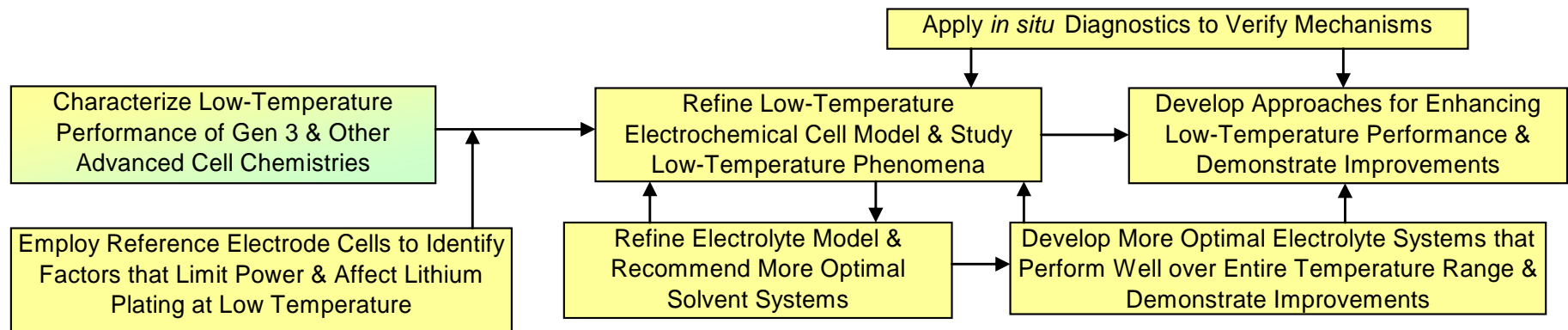


- Very high rate capability
- No lithium plating during regen at low temperature
- No SEI & resulting heat from decomposition
- Limited surface reactivity with electrolyte
- Zero volume change—no structural stress



These studies led to a FreedomCAR sponsored HEV battery program with EnerDel

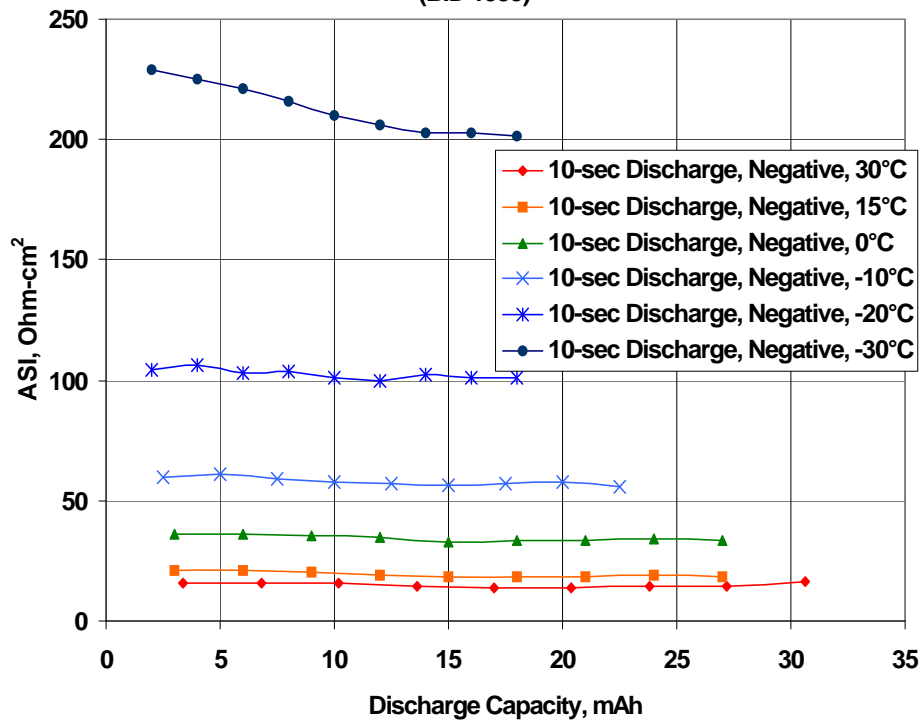
Low-Temperature Performance



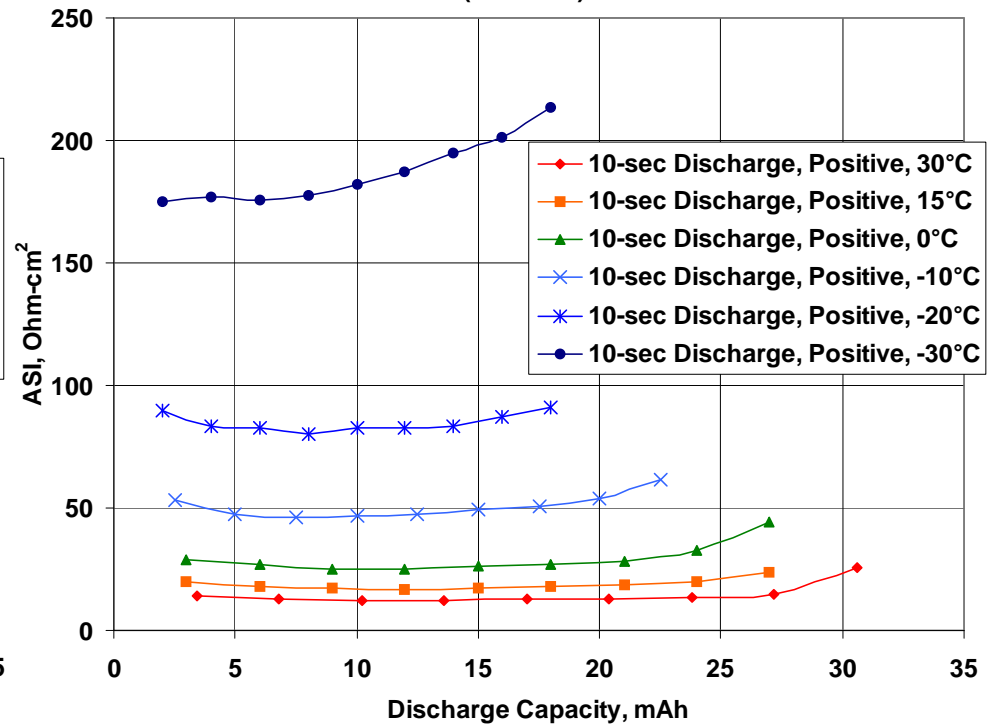
Note: No major changes are being made to the low-temperature studies except to expand to multiple cell chemistries beyond Gen 2 & Gen 3

Gen 2 Performance at Low Temperature

Gen2 Cell (MAG10), HPPC 10-sec Discharge ASI, Negative Electrode (BID 1935)



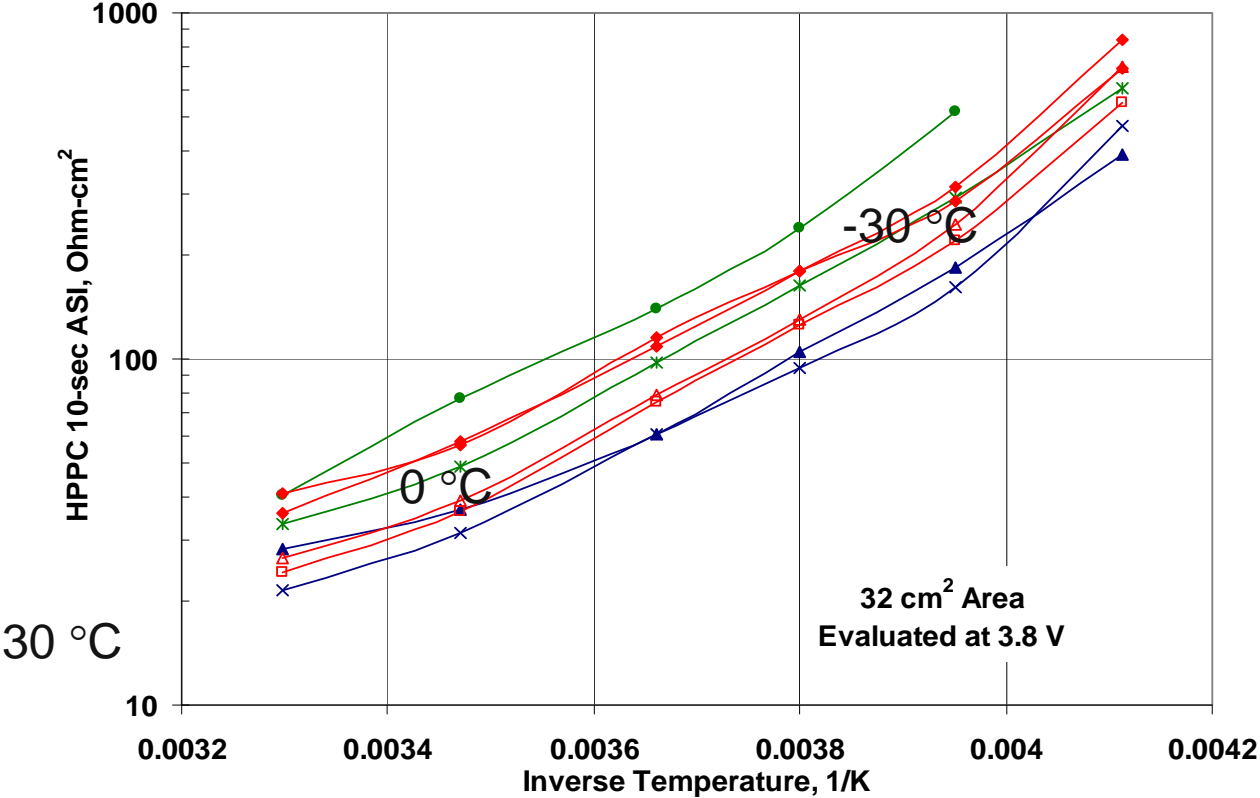
Gen2 Cell (MAG10), HPPC 10-sec Discharge ASI, Positive Electrode (BID 1935)



Impedance increases abruptly below -10° C.
Negative and positive electrodes share the impedance rise almost equally.
EIS shows the impedance to be interfacial in nature

Studied Many Different Electrolyte Systems

Comparison of Gen2 and Gen3 Electrodes in Various Electrolytes



- ▲— Gen2 Electrodes and 1.2M LiPF6 in EC:EMC (3:7 w/w) (BID 1935), 4.1V, 3 Sep.
- ×— Gen2 Electrodes and 1.2M LiPF6 in EC:EMC (3:7 w/w) (ANJ005), 4.0V, 1 Sep.
- *— Gen3-F Electrodes and 1.2M LiPF6 in EC:EMC (3:7 w/w) (BID 2067), 4.1V, 3-Sep.
- Gen3-F Electrodes and 1.2M LiPF6 in EC:PC:DMC (1:1:3 w/w) (BID 2003), 4.1V, 3-Sep.
- △— Gen3-D Electrodes and 1.2M LiPF6 in EC:EMC (3:7 w/w) (BID 2127), 4.0V, 1 Sep.
- Gen3-D Electrodes and 1M LiPF6 in EC:DEC:DMC:EMC (1:1:1:3 v/v) (BID 2126), 4.0V, 1 Sep.
- ◆— Gen3-D Electrodes and 1.1M LiPF6 in EC:GBL:EP (1:1:3 v/v) (BID 2128), 4.0V, 1 Sep.
- ◇— Gen3-D Electrodes and 1.1M LiPF6 in EC:GBL:DMC:EP (4:3:2:11 v/v) (BID 2129), 4.0V, 1 Sep.

JPL →
 K.Gering →
 K.Gering →

Major Technical Reports

1. **Materials Cost Evaluation Report for High-Power Li-Ion Batteries**; G. Henriksen, K. Amine, J. Liu, and P. Nelson; ANL-03/5, December 2002.
2. **Gen 2 Performance Evaluation Interim Report**; J. Christophersen, C. Motloch, I. Bloom, V. Battaglia, E.P. Roth, and T. Duong; INEEL/EXT-03-00095, February 2003.
3. **Screening Report on Cell Materials for High-Power Li-Ion HEV Batteries**; J. Liu, A. Kahaian, I. Belharouak, S-H Kang, S. Oliver, G. Henriksen, and K. Amine; ANL-03/16, April 2003.
4. **Handbook of Diagnostic Techniques**; contributions from ANL, BNL, and LBNL; LBID-2464, April 2003.
5. **Low-Cost Flexible Packaging for High-Power Li-Ion Batteries**; A. Jansen, K. Amine, and G. Henriksen; ANL-04/09, June 2004.
6. **Battery Technology Life Verification Test Manual**; H. Haskins, V. Battaglia, J. Christophersen, I. Bloom, G. Hunt, and E. Thomas; INEEL/EXT-04-01986, February 2005.
7. **Diagnostic Examination of Gen 2 Lithium-Ion Cells and Assessment of Performance Degradation Mechanisms**; contributions from ANL, BNL, LBNL, and UIUC; edited by D. Abraham; ANL-05/21, July 2005.
8. **Gen 2 Performance Evaluation Final Report**; J. Christophersen, I. Bloom, E. Thomas, K. Gering, G. Henriksen, V. Battaglia, & D. Howell; INL/EXT-05-00913, November 2005.
9. **Comparative Costs of Flexible Package Cells and Rigid Cells for Lithium-Ion HEV Batteries**; P. Nelson and A. Jansen; ANL-06/43, June 2006.

Note: Work performed under DOE's ATD Program has produced a large number of peer reviewed journal articles and conference papers, some of which will be identified by individual PIs in their presentations.

Presentation Sequence

Calendar Life

- 4 presentations on Gen 3 cells (builds, aging, diagnostics, & modeling)
- 4 presentations on SEI layer studies (characterization/diagnostics)
- 1 presentation (with 2 speakers on TLVT studies)

Abuse Tolerance

- 2 presentations (materials-level & cell-level studies)

Low-Cost Materials

- 1 presentation on advanced material screening
- 3 presentations on advanced material development

Low-Temperature Performance

- 3 presentations (characterization, cell modeling, & electrolyte modeling)

Acknowledgements

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 - David Howell
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 - LBNL
 - SNL
- Comments and suggestions from members of the FreedomCAR Energy Storage Technical Team

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- University of Illinois Urbana-Champaign
- University of Rhode Island