# TOWARDS ROOM TEMPERATURE SODIUM BATTERIES: PROGRESS IN HIGH CAPACITY SODIUM ION BATTERIES

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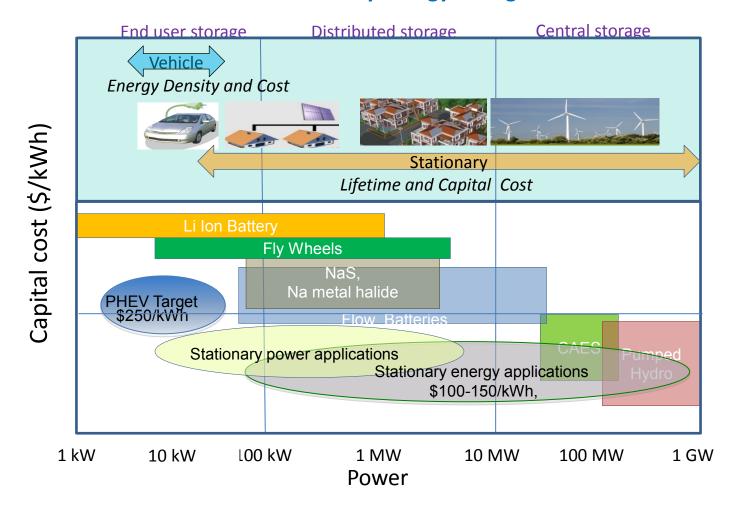


### **Outline**

- Motivation and background
- □ Objective
- ☐ New high capacity storage mechanism for cathode
- ☐ Progress in anodes
- ☐ Summary and future work



# Significant challenges for meeting the low term low cost and reliability requirement for stationary energy storage.

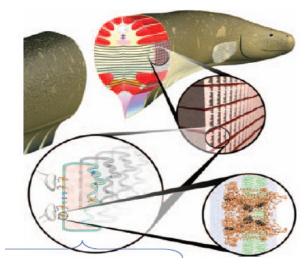


#### Alternative, low cost technologies are desired!

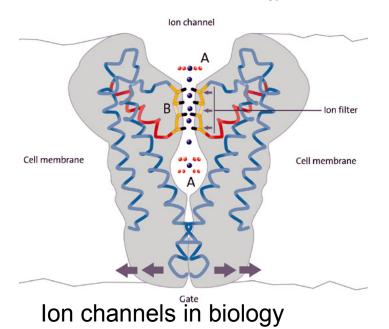
J. Liu et al, Advanced Functional Materials, Special Issue on energy storage, 2012.



#### Biology stores energy with Na, K, Ca ions, not Li ions (electrical eels).



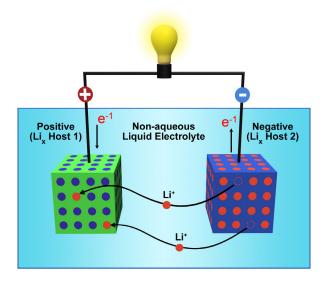
J. Xu, D. A. Lavan, Nature Nanotechnology 2008, 3, 666.



Storing large amount of energy using NaCl?





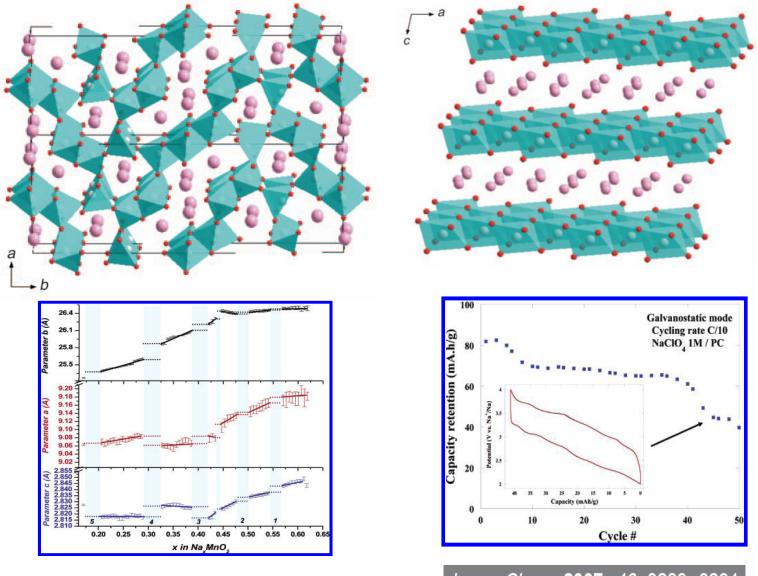


Can we develop high capacity Naion batteries like Li-ion batteries?

#### Objective

Develop new room temperature Na storage mechanisms with capacity and stability close to Li-ions

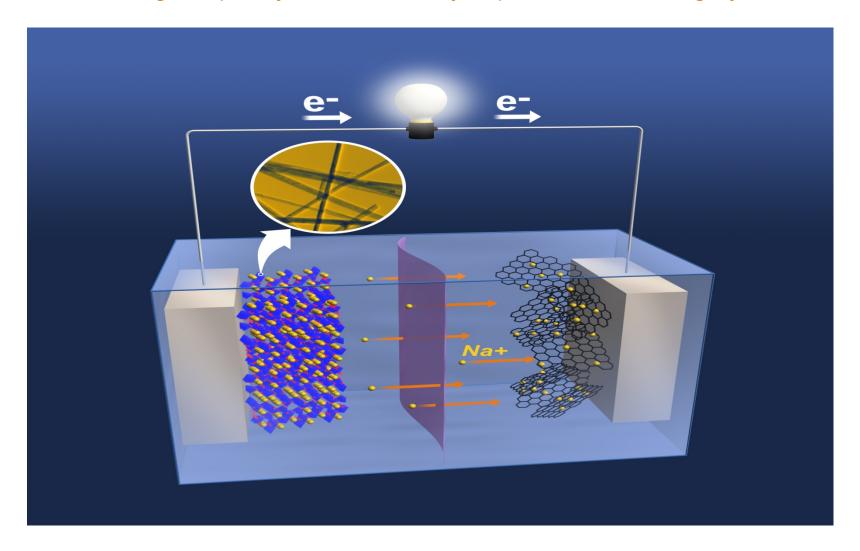
Many groups have explored Na-ion storage materials, but the capacity and stability remain a challenge.



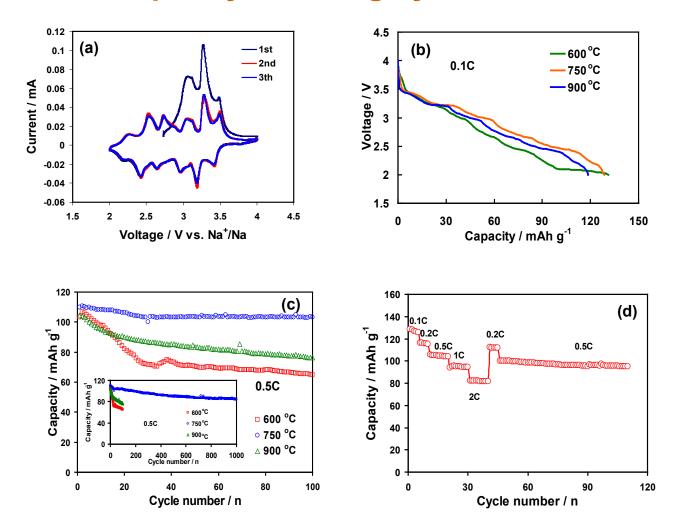
F. Sauvage, L. Laffont, J.-M. Tarascon, and E. Baudrin

Inorg. Chem. 2007, 46, 3289-3294

PNNL's breakthrough in both cathode and anode materials: demonstrated that high capacity Na-ion battery is possible with long cycle life.

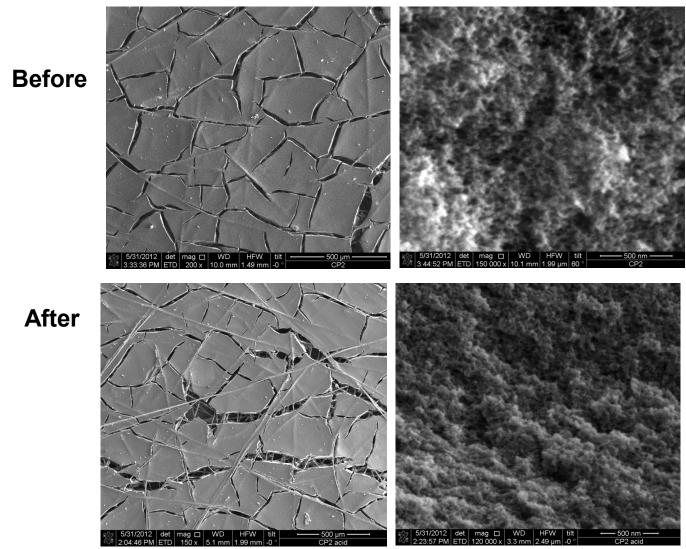


# The nanowire based Na-ion battery has good capacity and long cycle life

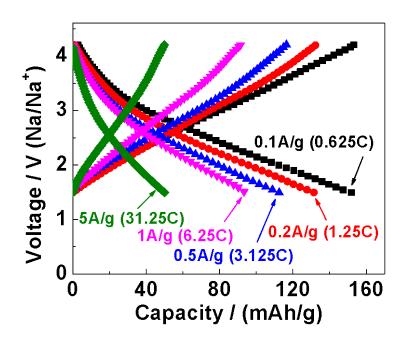


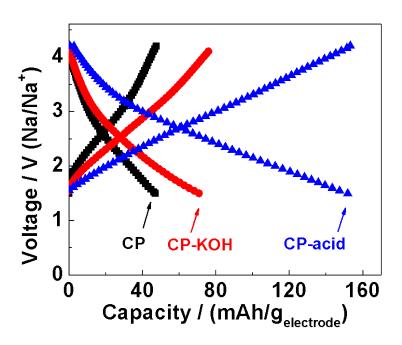
However, in general the cathode capacity is limited as compared with Li-ions.

# New monolith carbon electrode materials as the cathode for Na-ion Batteries



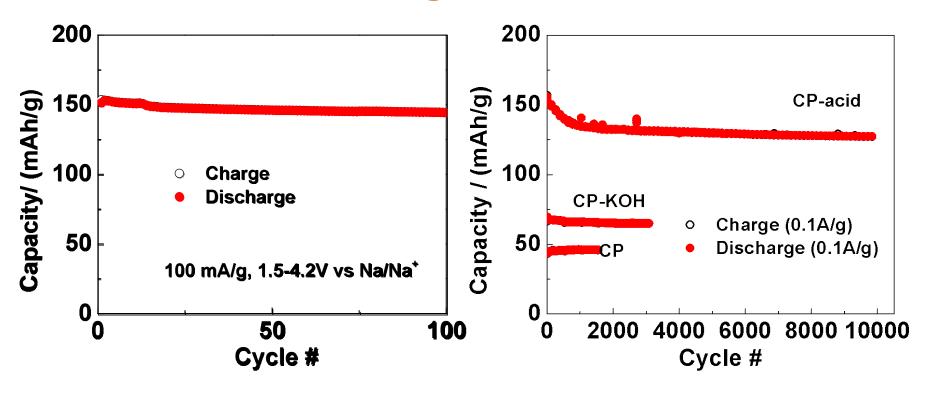
## Discharge/charge





Specific capacity of functionalized carbon =155mAh/g, double that of KOH activated carbon

# Rate and cycling stability: significantly exceeded the targets in FY12 milestone.

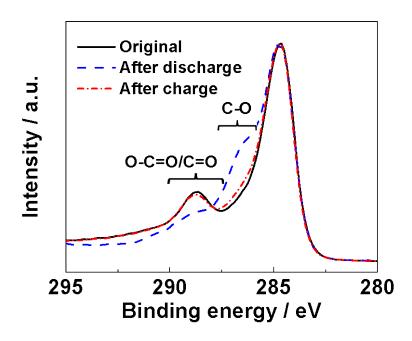


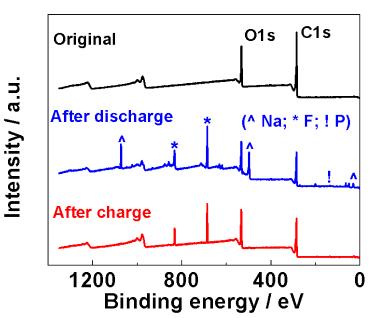
Very small capacity fading for over 100 cycles

>80% capacity retention in 10000 cycles



## **Surface chemistry**





Reversible reaction of carbonoxygen double bonds and single bonds Na content increases significantly after discharge and turns back to ~0 after charge

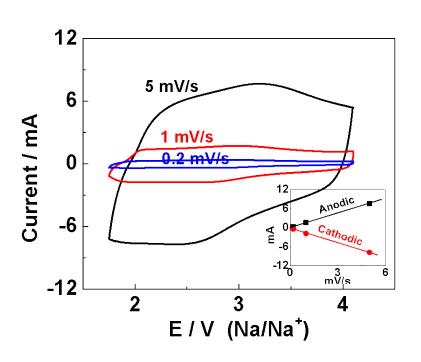
Na energy storage mechanism: Surface redox reaction between Na+ and >C=O

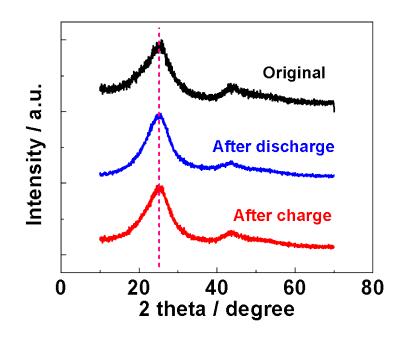


#### **Hypothesis:**

Surface redox reaction between Na+ and oxygen-carbon functional groups

$$Na^+ + > C = O + e - \leftrightarrow > C - O - Na$$





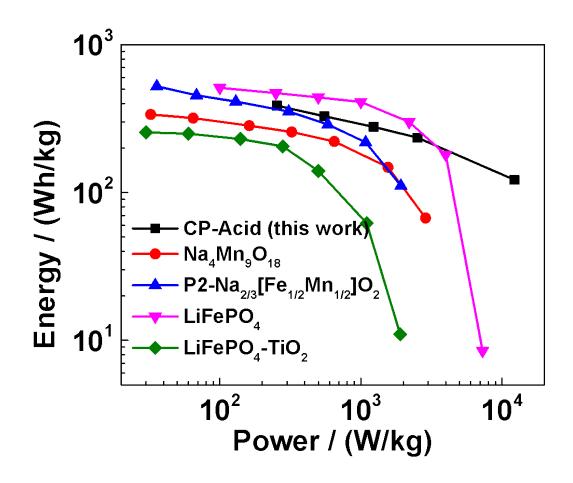
**Surface-confined redox reaction** 

No intercalation reaction

[Double layer capacitance not excluded (~1/4)]

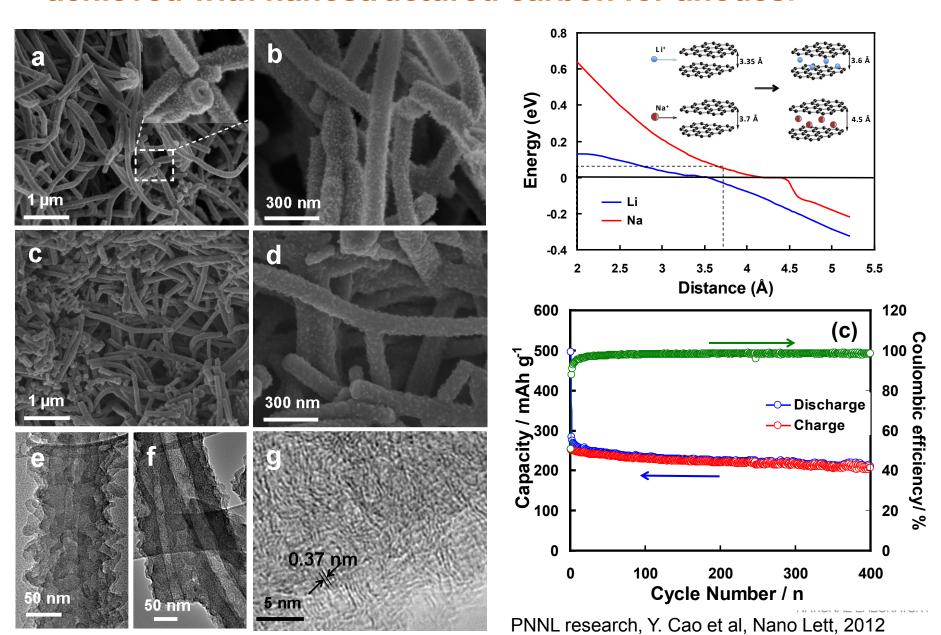


# The storage capacity of the new mechanism could be comparable to what can be achieved for Li-ions

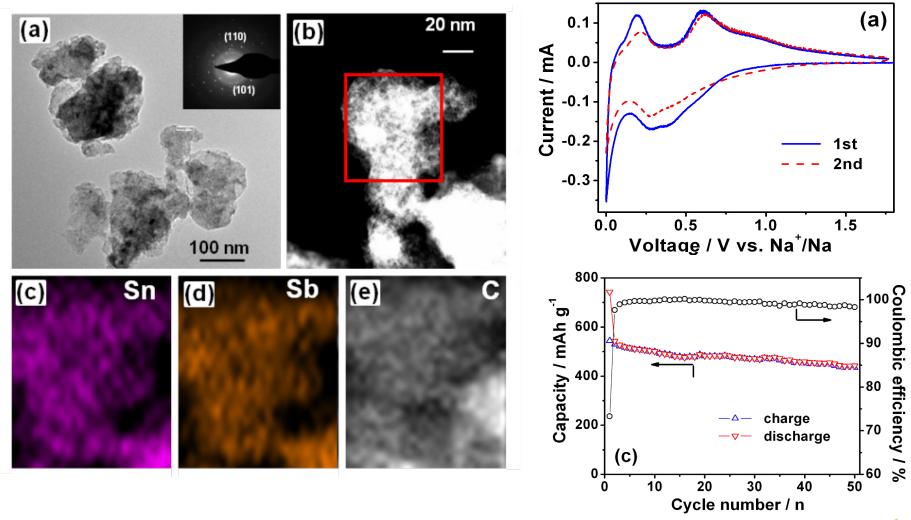




# High capacity and long cycling stability have been achieved with nanostructured carbon for anodes.



#### First demonstration of reversible Na ion storage in high capacity alloys



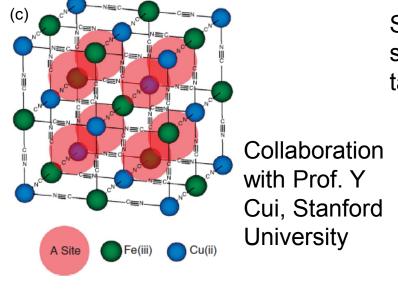
PNNL research, L. Xiao et al Chem Comm, 2012.

## **Summary**

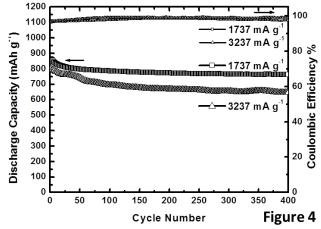
Significant progress has been made in both cathodes and anodes with capacities and stabilities close to Li-ions.

#### **Future Work**

- Understand and explore the new storage mechanisms in simple inexpensive inorganic and organic materials;
- Design new full cells to demonstrate the full cell performance.



Some new cell concepts have already been successfully tested in other batteries and will be tailored and explored for Na storage



Prediction?

Professor JM Tarascon: "High capacity Na ion batteries will be commercially available in about fives years."

