

Optimization of Ion Transport in High Energy Composite Cathodes

P.I. and Presenter: Dr. Shirley Meng
University of California San Diego
June 2014

Project ID ES216

Overview

Timeline

- April 1st, 2013
- March 31st, 2017
- Percent complete: 25%

Budget

- Total project funding
 - US\$ 899,999
- Funding received in FY13
 - US\$ 225,000
- Funding for FY14
 - US\$ 225,000

Barriers

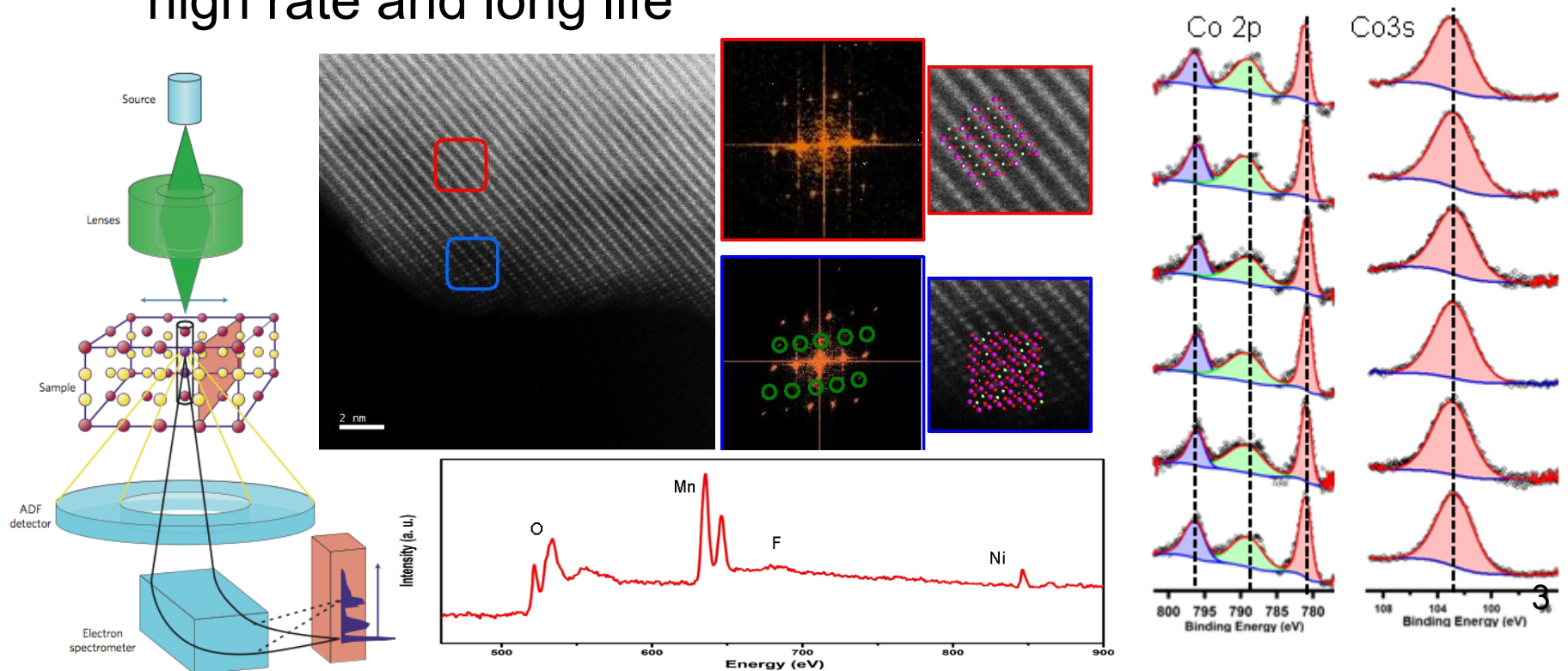
- Barriers addressed
 - Low rate
 - Poor voltage stability

Partners

- Interactions/ collaborations
 - Envia Systems
 - Oak Ridge National Lab
- BATT team: Jordi Cabana, Robert Kosteki and Kristin Persson

Objectives

- Probe and control the atomic-level kinetic process that govern the rate and stability in high energy (high voltage) cathode materials
- Establish quantitative diagnosis methods to determine the optimum bulk composition and surface characteristics for high rate and long life



Milestones

- Q1 Establish the initial suite of surface and interface characterization tools, including STEM/EELS, XPS and FP computation (12/31/13 - **completed**)
- Q2 Identify the surface coated materials (AlF_3 and Li_3PO_4) electrochemical performance matrices, including first cycle irreversible capacity, discharge energy density, voltage stability upon cycling and rate capabilities (3/31/14 - **completed**)

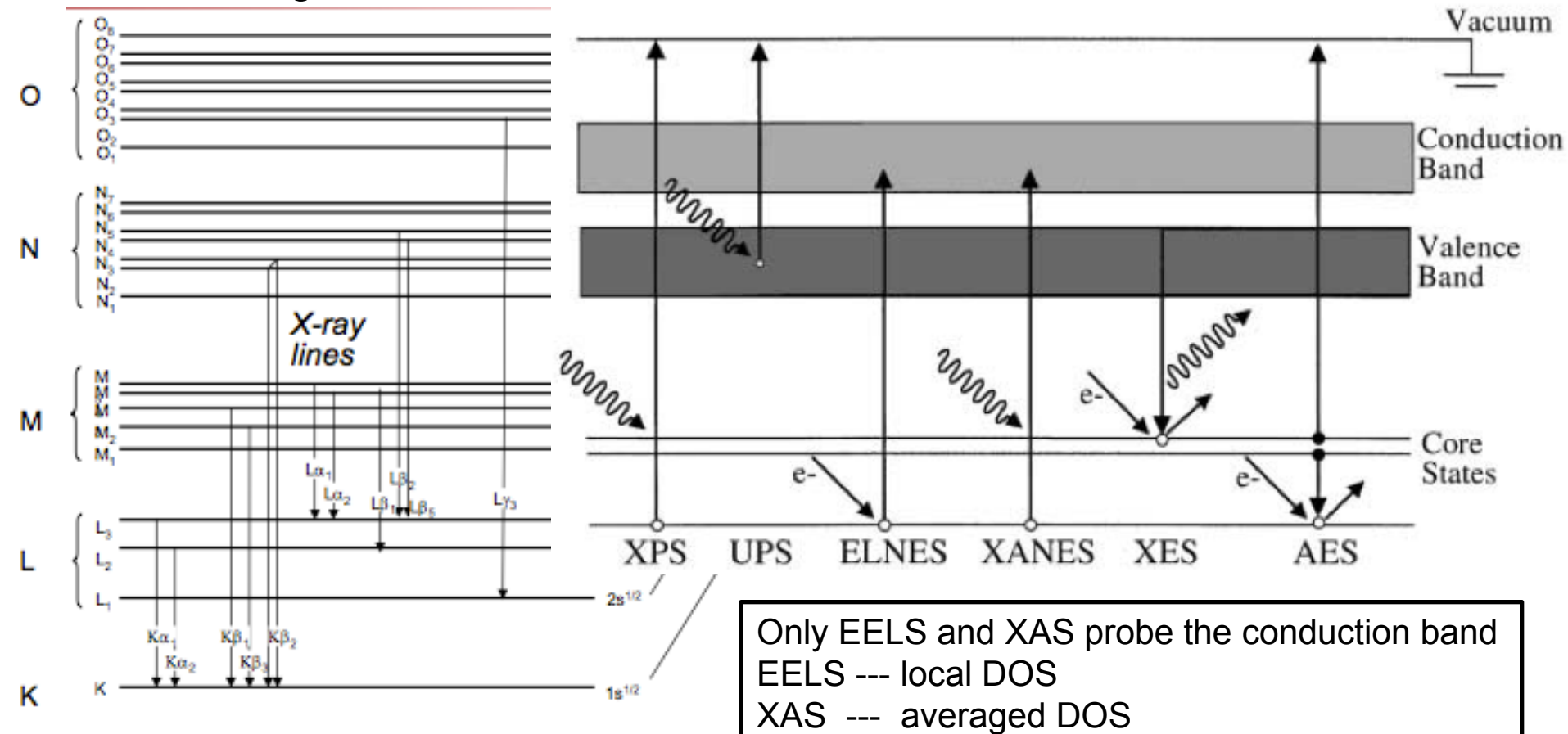
Go-No Go: Stop and change the coating materials if the improvements for rate capability and voltage stability are not significant (3/31/14)

Milestones

- Q3 Characterize coated samples - coating thickness, compositions and morphology before, during and after cycling. (6/30/14 - **completed**)
- Q4 Identify ways to extend the STEM/EELS, XAS and XPS techniques for anode materials, such as silicon anode. (9/30/14 – **on going**)

Approaches

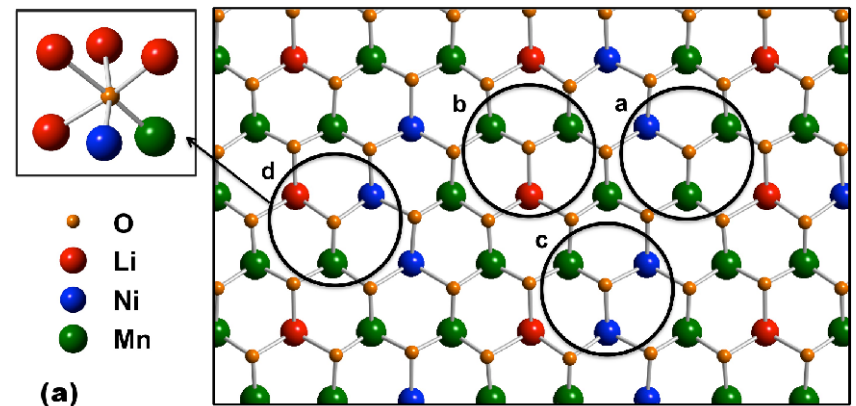
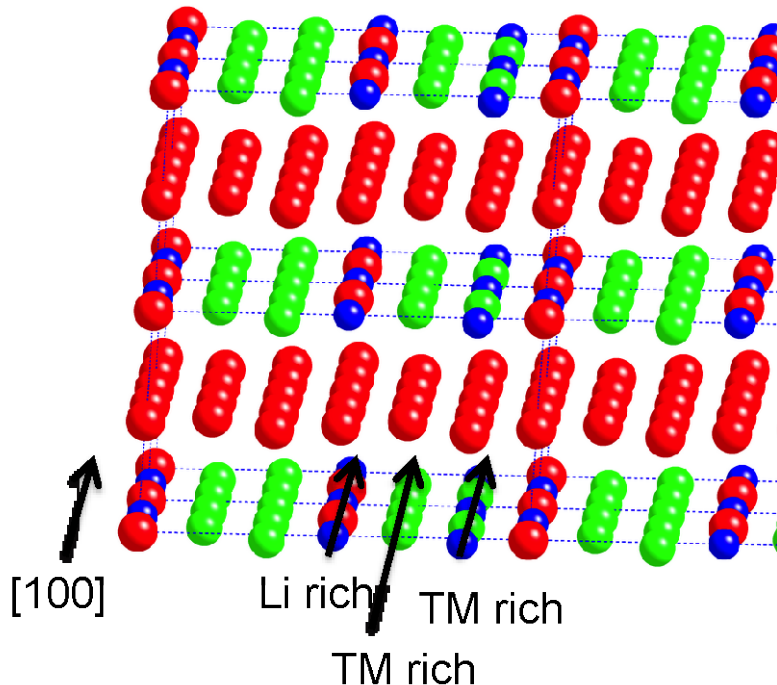
Combines atomic resolution scanning transmission electron microscopy (a-STEM) & Electron energy loss spectroscopy (EELS), X-ray photoelectron spectroscopy (XPS), X-ray absorption spectroscopy (XAS) and first principles (FP) computation to elucidate the dynamic changes of the bulk and surface changes



Approaches

Combines atomic resolution scanning transmission electron microscopy (a-STEM) & Electron energy loss spectroscopy (EELS), X-ray photoelectron spectroscopy (XPS), X-ray absorption spectroscopy (XAS) and first principles(FP) computation to elucidate the dynamic changes of the bulk and surface changes

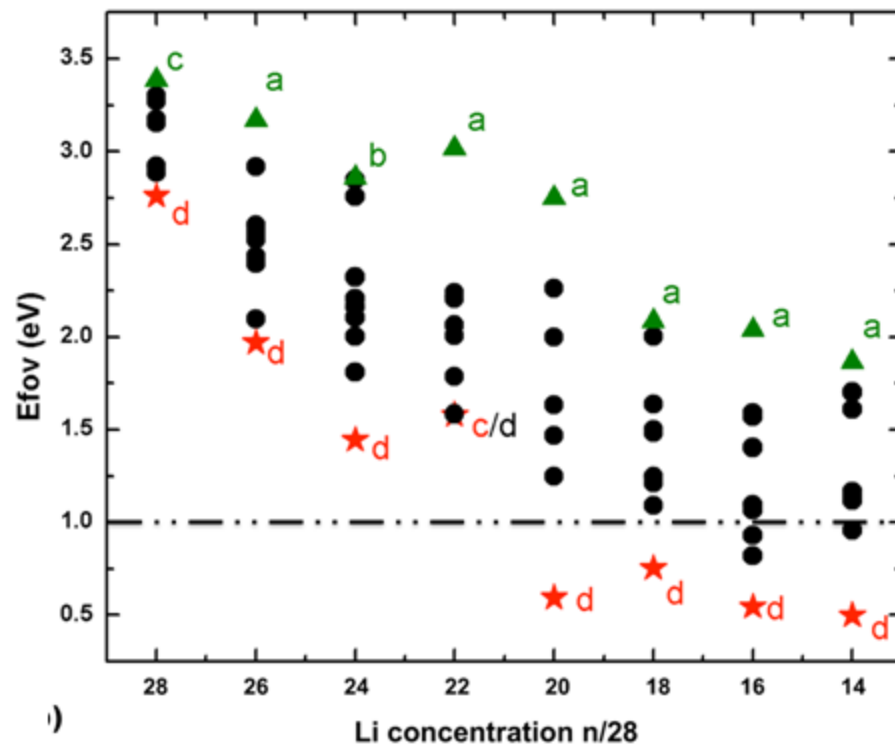
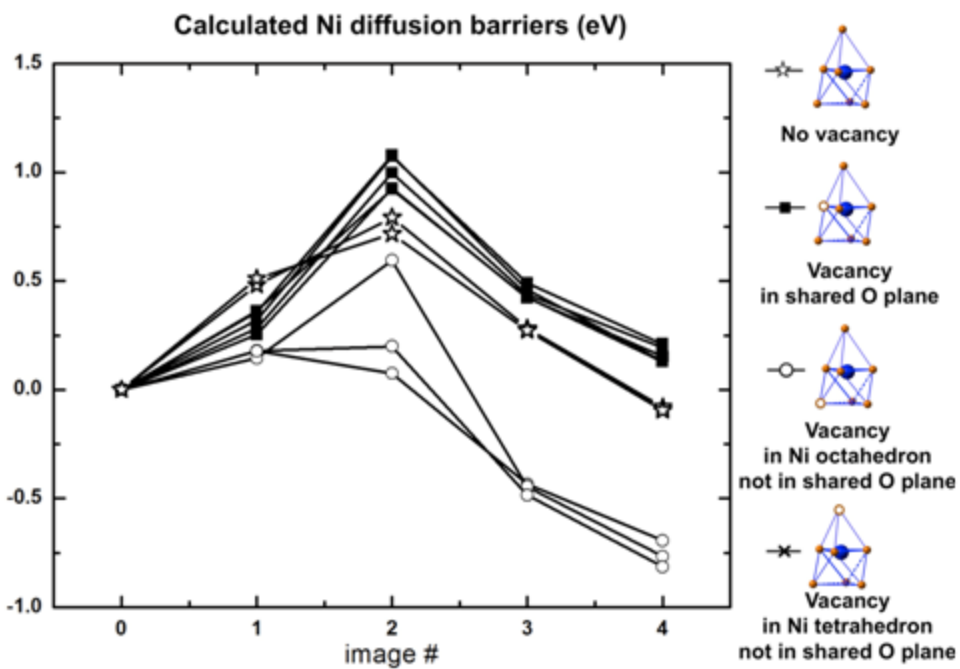
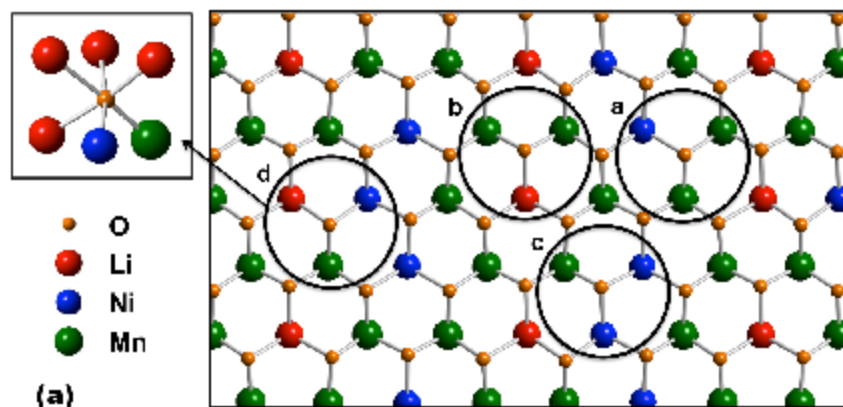
Dilute O vacancy calculation, a 96 atom unit cell



➤ 4 different types of local environment :

- ✓ 8 of type a: $2\text{Mn}1\text{Ni}3\text{Li}$
- ✓ 10 of type b: $2\text{Mn}4\text{Li}$
- ✓ 4 of type c: $1\text{Mn}2\text{Ni}3\text{Li}$
- ✓ 2 of type d: $1\text{Mn}1\text{Ni}4\text{Li}$

Technical Achievements

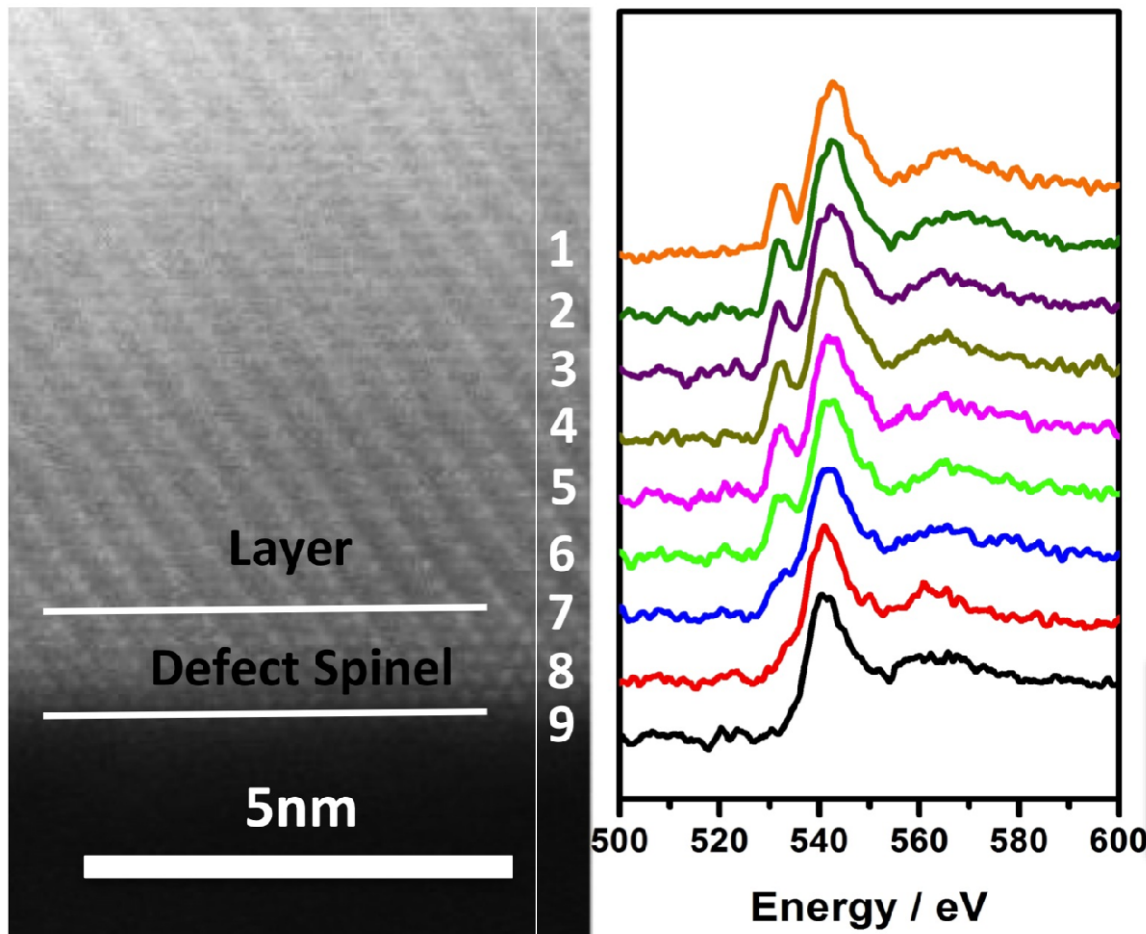


At the presence of O vacancy, both Ni and Mn migrate to Li layer

D.Qian et.al submitted, 2014

Technical Achievements

EELS was taken $\sim 0.6\text{nm}$ per step
from surface to bulk



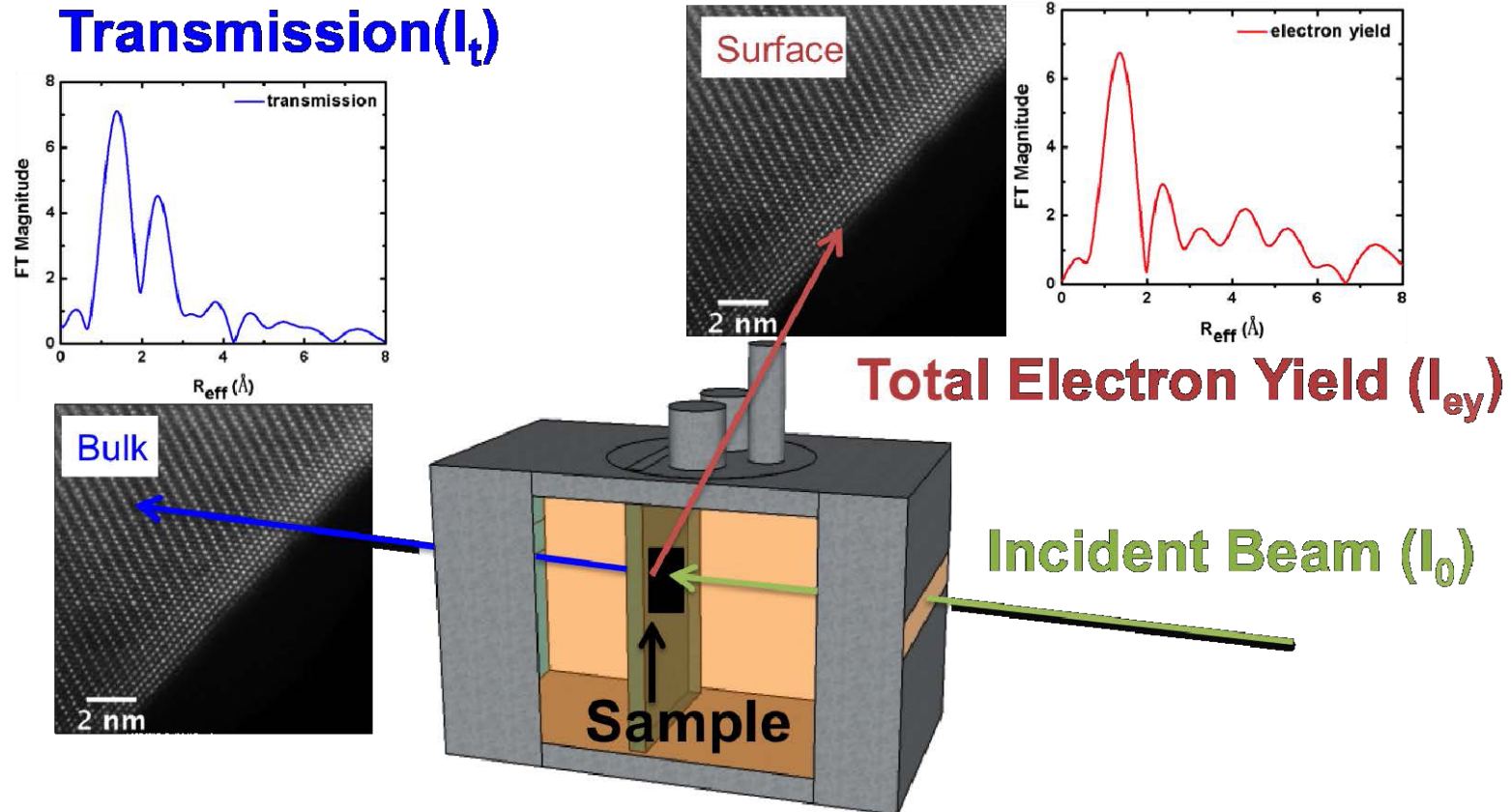
Surface phase transformation
in Li-excess materials

Defect spinel phase

Pre-peak disappearance:
1. the neighbour TM being
reduced
2. oxygen vacancy formation

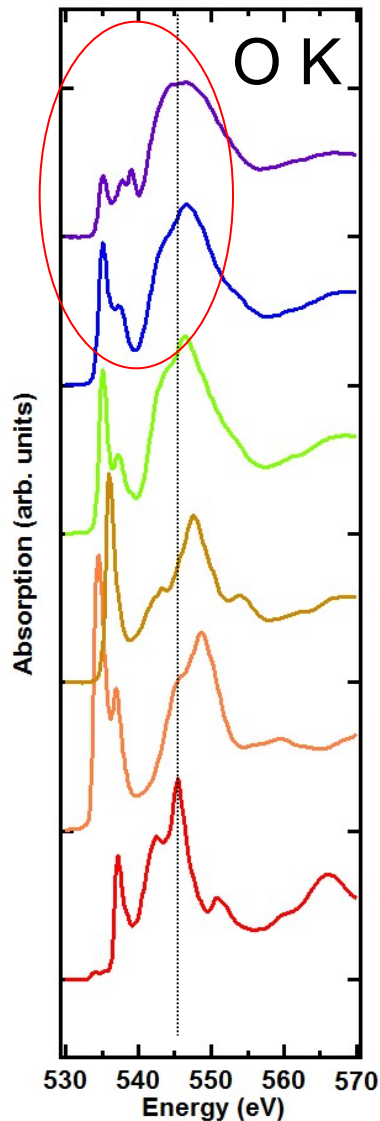
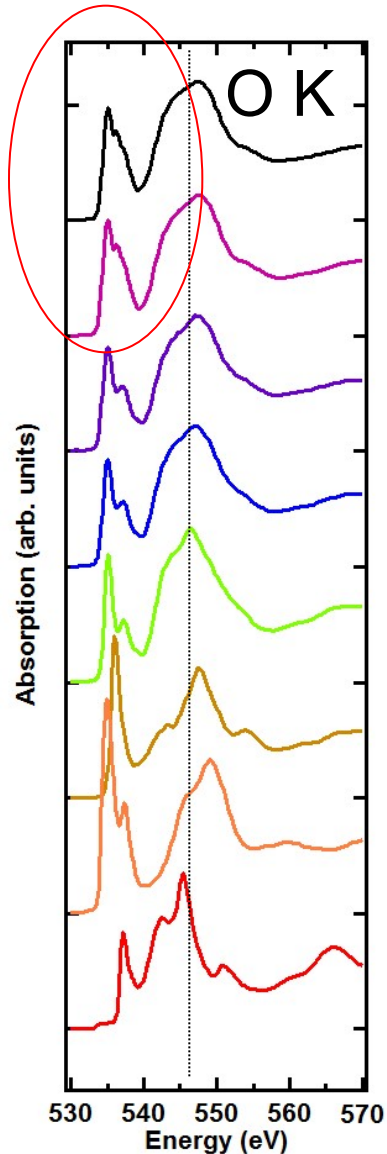
**TM ion migration happens
most in the region where the
oxygen vacancy presented**

Depth Resolved Spectroscopy



Kyler Carroll etc. PCCP, 2013

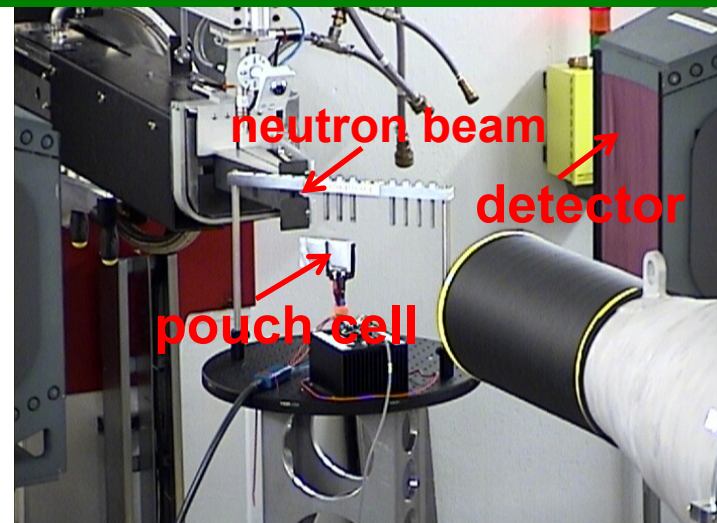
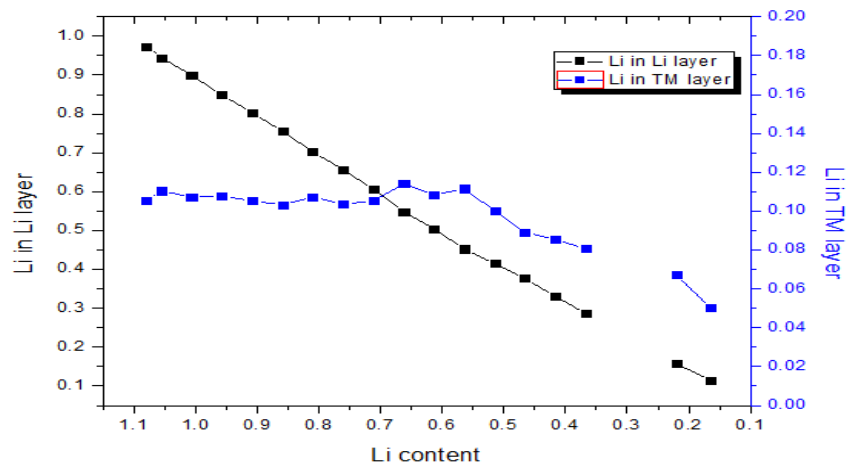
Soft XAS



Surface O environment definitely changed after 1 cycle

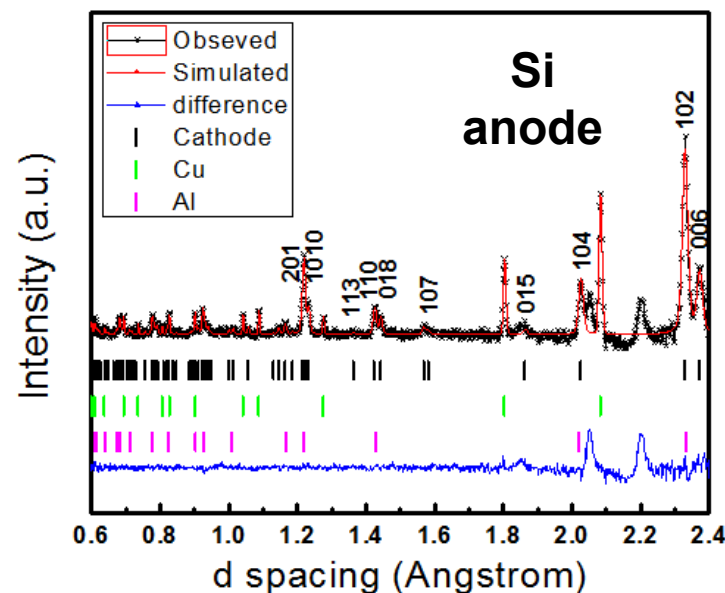
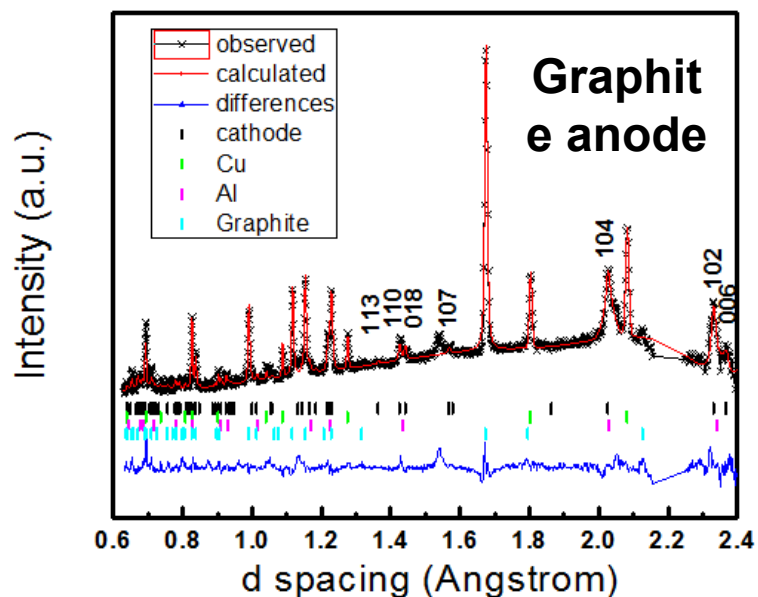
Most significant changes occur at discharge

In Situ Neutron Diffraction

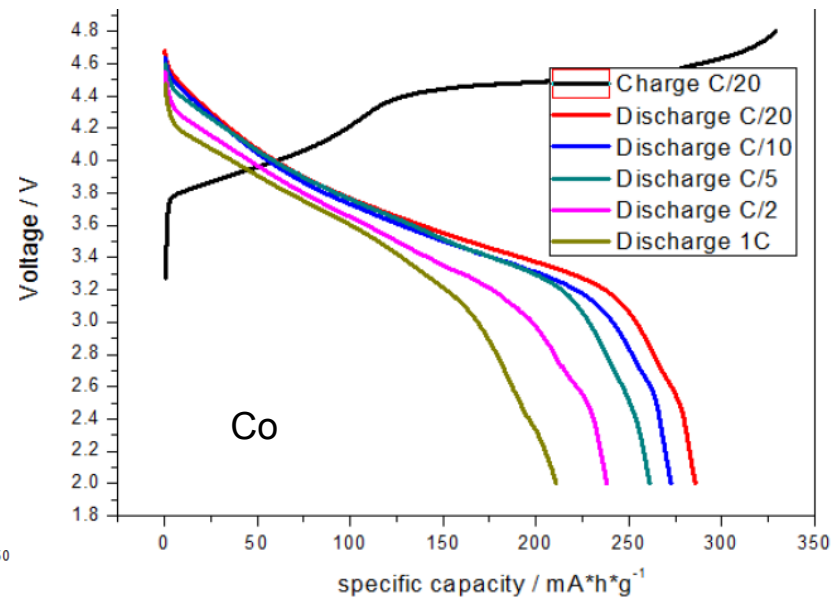
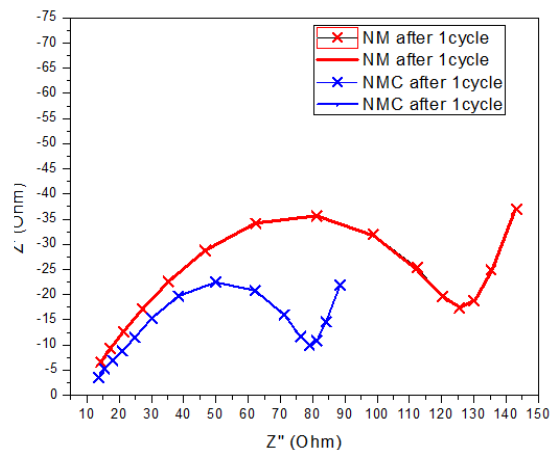
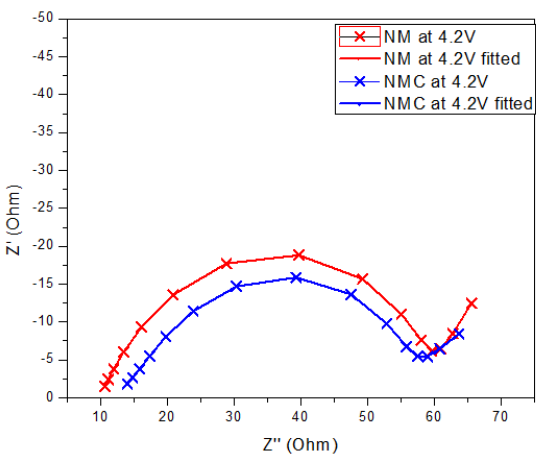
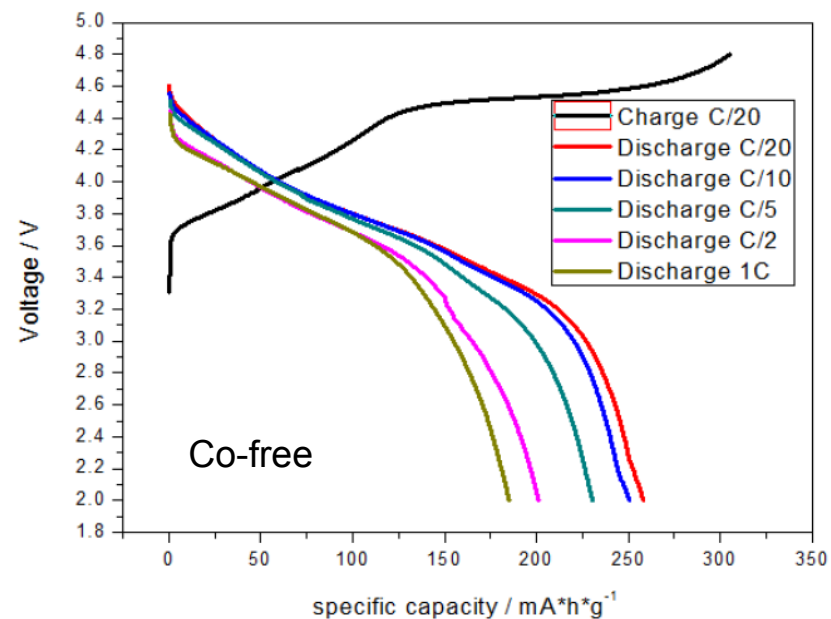
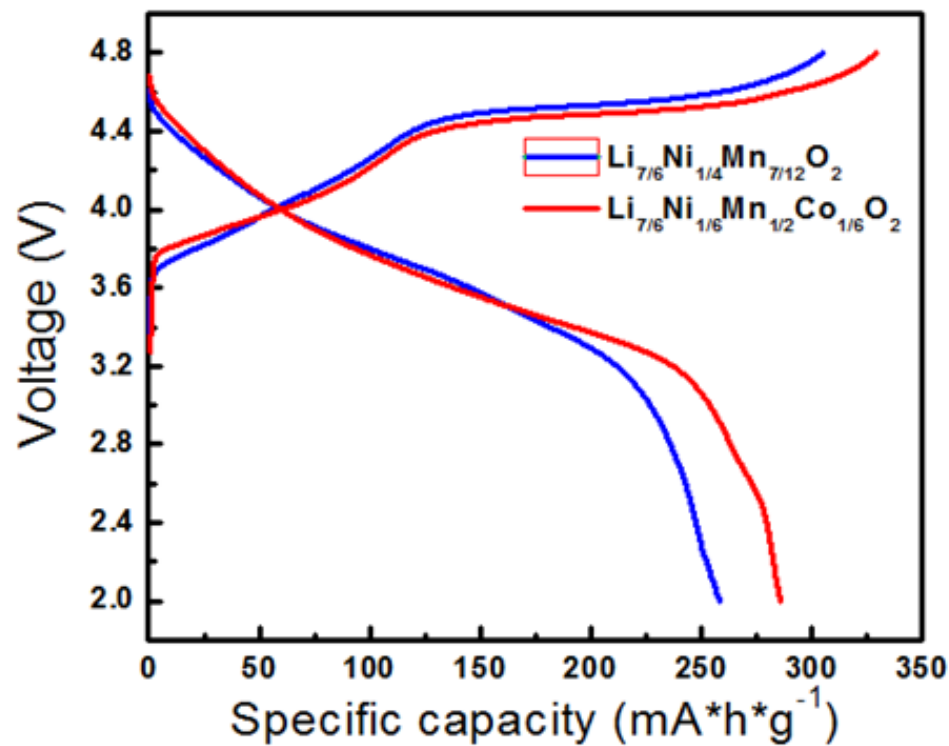


$$R_{F2} = 11.81\% \quad R_{wp} = 1.2\% \quad \chi^2 = 1.702$$

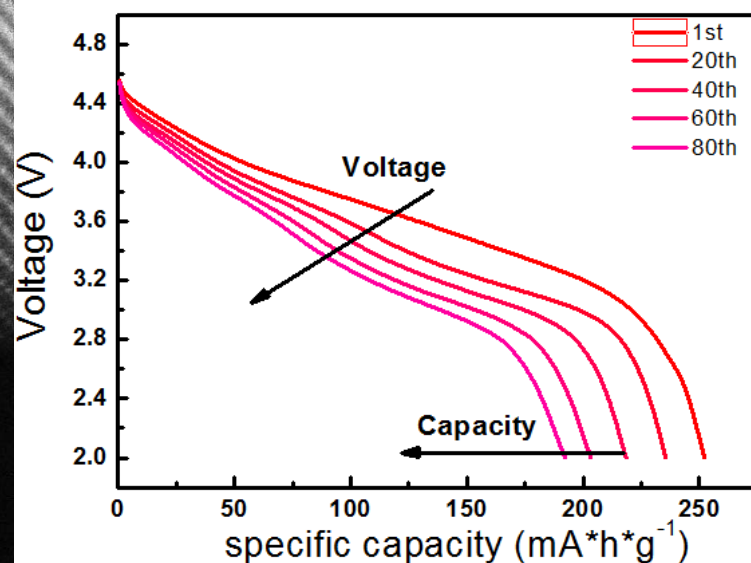
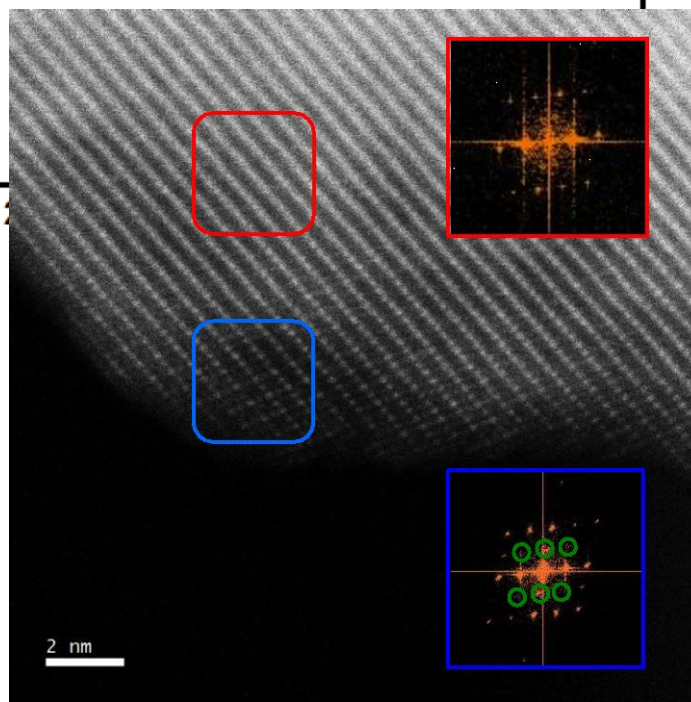
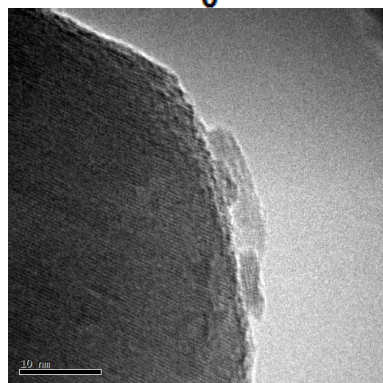
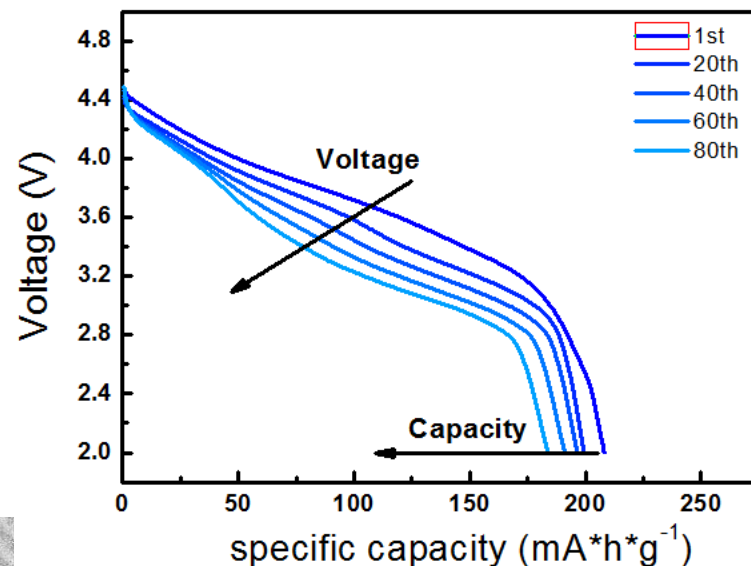
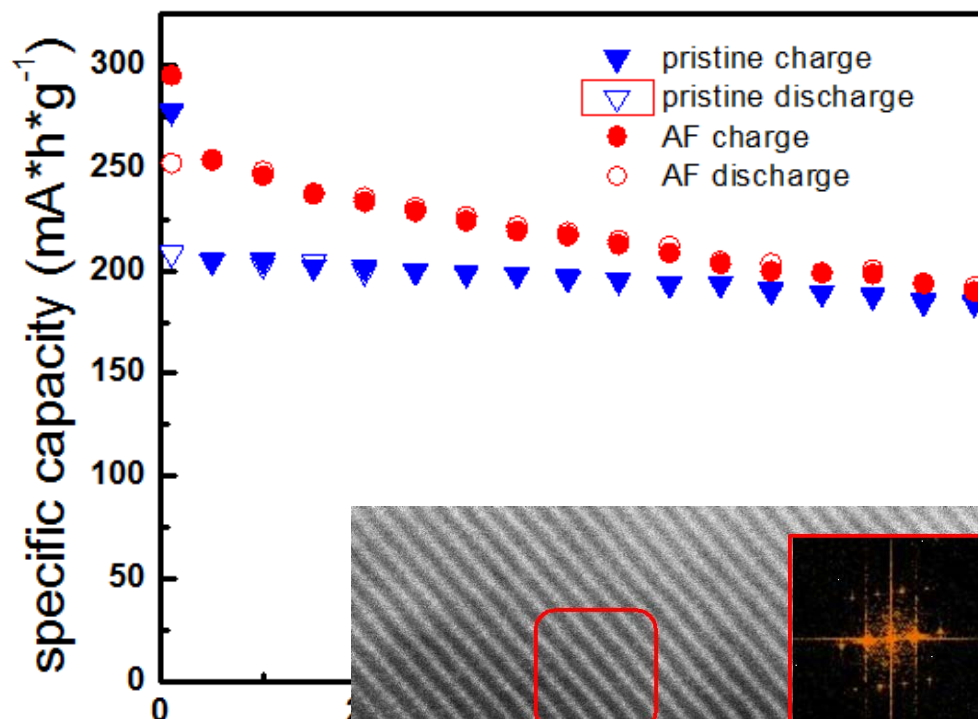
$$R_{F2} = 9.27\% \quad R_{wp} = 1.16\% \quad \chi^2 = 1.358$$



Co-Substitution

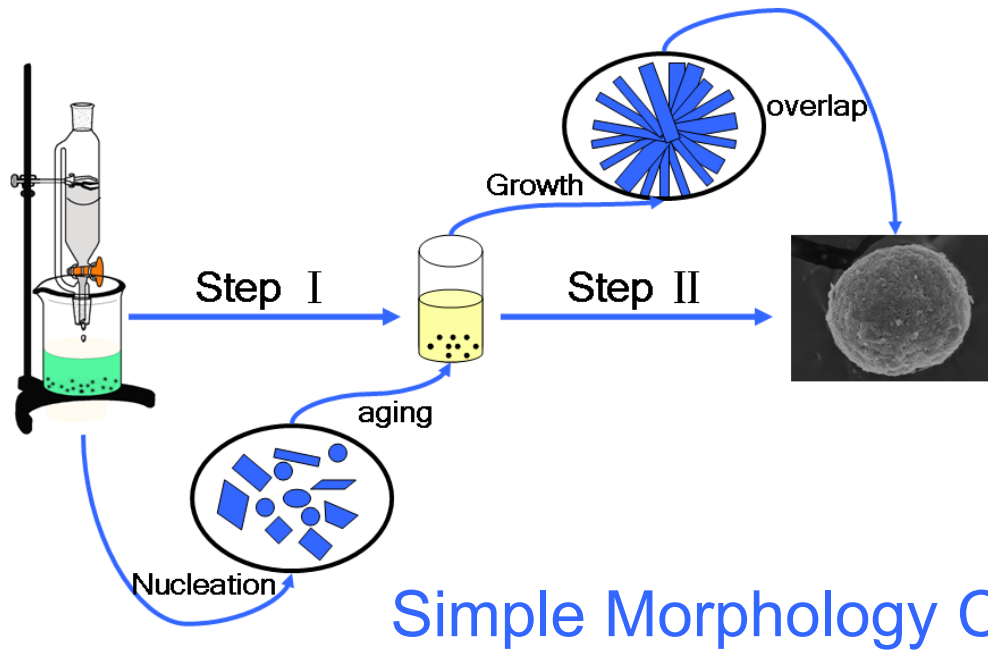


Coatings – AlF_3

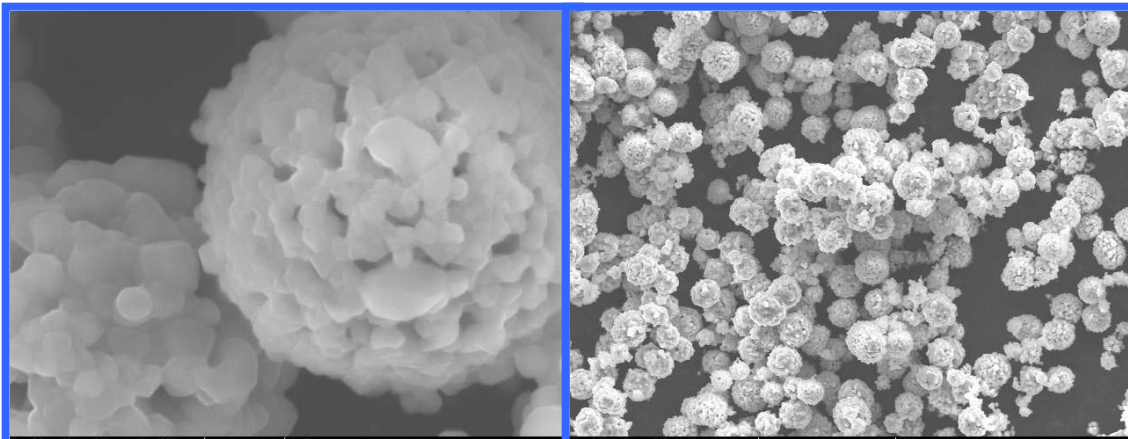
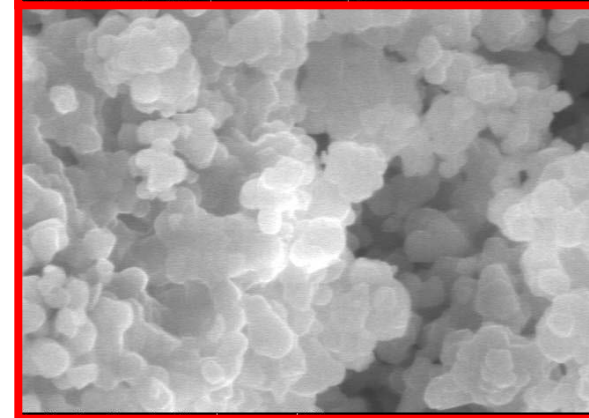
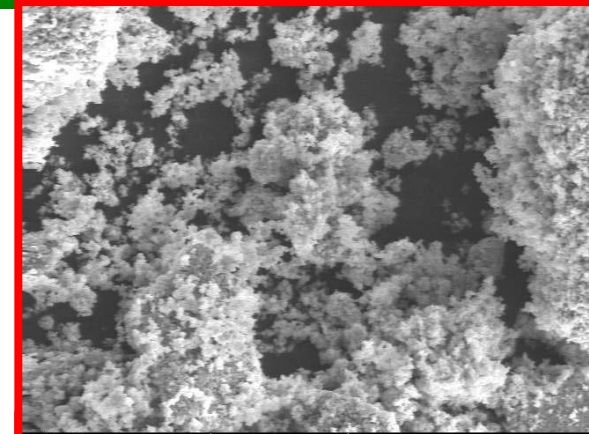


Morphology Control

Non- Morphology Control



Simple Morphology Control



Impact

A Dual Layer Complex Interphase Model

EELS/XAS

Mn^{3+} during charge

Mn^{4+} discharge

O^{2-} to O^- oxidation

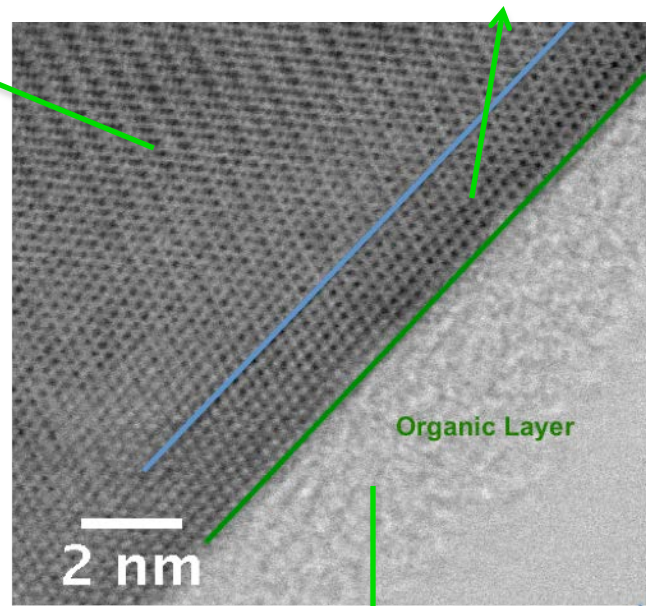
$\text{Ni}^{2+} \rightarrow \text{Ni}^{4+}$ charge

$\text{Ni}^{4+} \rightarrow \text{Ni}^{2+}$ discharge

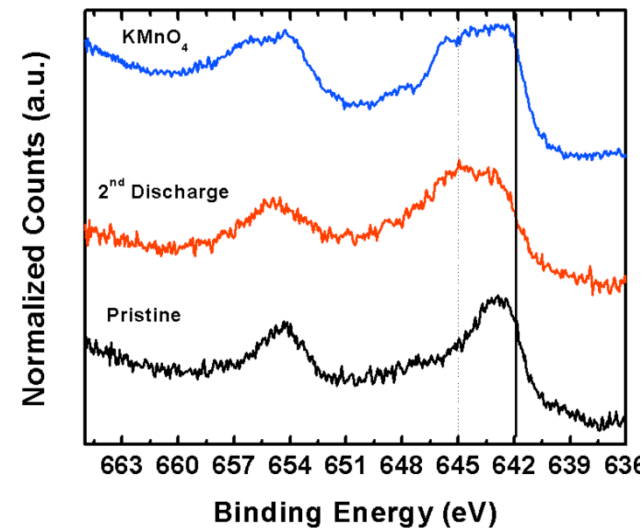
EELS/TEY/XPS(Mn3p)

$\text{Mn}^{2/3+}$ during charge

$\text{Mn}^{\sim 3.5+}$ discharge



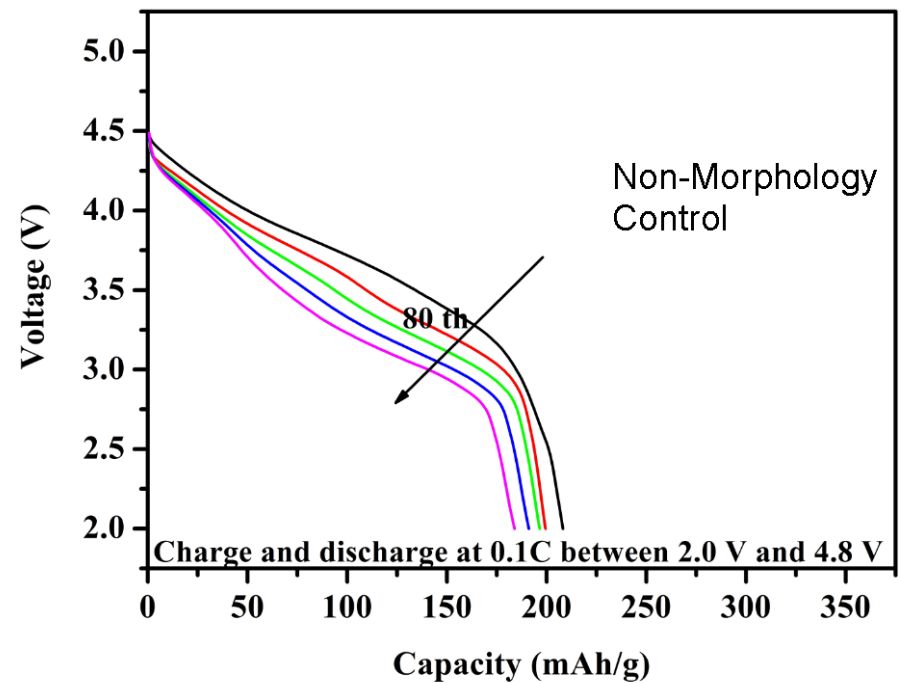
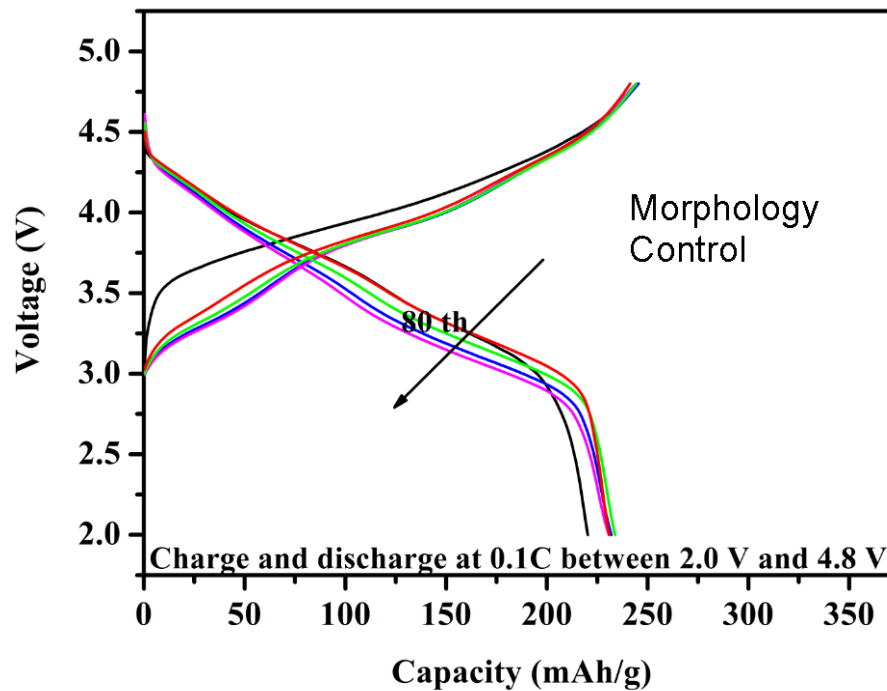
XPS(Mn2p)



Higher Oxidation
Mn after Charge

Impact

Much Improved Voltage Stability



Collaborations

Dr. Miaofang Chi (SHaRe – STEM/EELS)
Dr. An Ke (SNS – in situ Neutron)



Multilayer in situ Neutron Cell



Future Work

- Characterize coating compositions, thickness and morphology and their effect on impedance/ion transport
- Quantitative refinement of in situ Neutron scattering experiment data
- Identify ways to extend the STEM/EELS, XPS and XAS techniques for Si based anode materials

Summary

- Our diagnostic tool suite consisting STEM/EELS, XPS, XAS and FP is effective at identifying the dynamic structural and chemical changes at the interface.
- Probing oxygen activities are possible with in-situ soft X-ray absorption spectroscopy and in-situ Neutron scattering
- Coating, substitutions and morphology control all have profound effect on the “voltage fade”. Strategies for enabling high voltage cathodes need to be optimized.

Responses to Previous Year Reviewers' Comments

NEW PROJECT FY 2013