Can hard coatings and lubricant anti-wear additives work together?

Project ID: FT021

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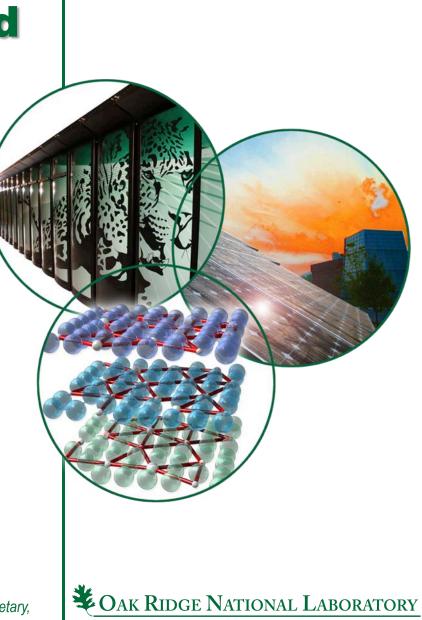
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2014 DOE Vehicle Technologies Program Annual Merit Review, June 19, 2014



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Overview

Timeline

- Project start date: Oct. 1, 2012
- Project direction and continuation are evaluated annually

Budget

- FY13 DOE funding: \$250K
- FY14 DOE funding: \$250K

Partners

- Lubrizol
- Cytec Industries
- Northeast Coating Technologies
- Eaton
- ANL

Barriers

- 10-15% energy generated in an IC engine is lost to parasitic friction.
- Current engine lubricants and their additive packages were designed for ferrous alloy bearing surfaces.
- Compatibility between oil anti-wear additives and non-metallic hard coatings is little known.
- Fundamental understandings gained in this study will help guide future development of engine lubricants.
- A synergistic lubricant-coating combination will provide maximized benefits in fuel economy.



Relevance

- Objective: Investigate the compatibility of engine lubricant anti-wear (AW) additives, specifically conventional ZDDP and newly developed ionic liquids, with selected hard coatings.
- Potential benefits:
 - Fundamental understandings gained in this study will help guide future development of engine lubricants
 - A synergistic lubricant-coating combination will provide maximized benefits in fuel economy.



Milestones

- Demonstrate the lubricant-coating compatibility via tribological testing and analysis at room temperature (June 30, 2013) – complete!
- Reveal the tribo-chemical interactions for selected lubricant-coating combinations at room temperature (September 30, 2013) – complete!
- Tribological testing and analysis of the AW-coating compatibility at 100 °C (June 30, 2014) – on schedule
- Understand the tribochemical interactions of candidate lubricant-coating combinations at 100 °C (September 30, 2014) – on schedule

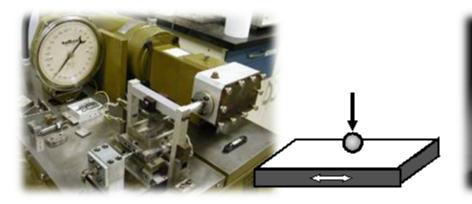


Approach

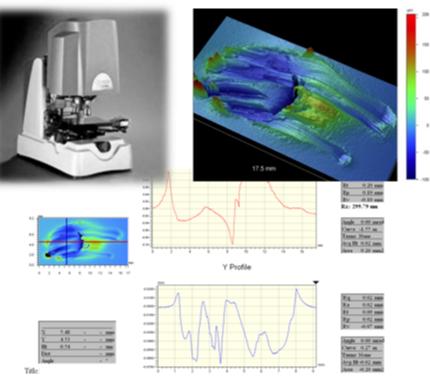
- Experimentally study the friction and wear behavior for selected non-metallic hard coatings lubricated by selected anti-wear additives via tribological bench testing in well-defined conditions.
 - Anti-wear additives: ZDDP and ionic liquid
 - Hard coatings: Borides and DLC
 - Counterface material: AISI 52100 steel
- Mechanistically investigate the tribochemical interactions between the anti-wear additives and the coating surfaces via comprehensive tribofilm characterization.
 - Top surface examination:
 - SEM: worn surface morphology for wear mode analysis
 - EDS: element analysis
 - Tribofilm layered chemical analysis aided by ion sputtering:
 - XPS: composition-depth profile and binding energy spectrum
 - Auger: surface element mapping
 - Tribofilm cross-sectional examination aided by focused-ion-beam (FIB):
 - TEM: nanostructure and tribofilm thickness measurement
 - Electron diffraction: phase determination
 - EDS: element mapping



Tribological testing and wear quantification



- Boundary lubrication ball-on-flat reciprocating sliding test
 - Plint TE-77 tribometer
 - Ball: AISI 52100 steel (10 mm dia)
 - Flat: coatings
 - Ave. sliding speed: 0.2 m/s
 - Load: 50 N
 - Sliding distance: 1000 m
 - Ambient environment



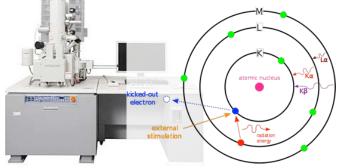
- Wear quantification
 - Wyko NT9100 optical profiler
 - Wear volume/depth
 - Roughness



Comprehensive tribofilm characterization

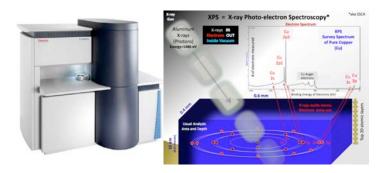
Scanning electron microscopy (SEM)/Energy-dispersive X-ray spectroscopy (EDS)

- ✓ Worn surface morphology
- ✓ Surface element analysis



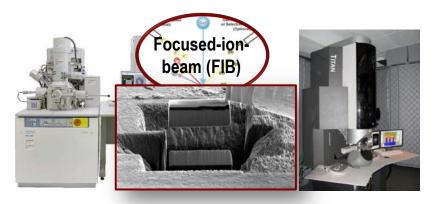
X-ray photoelectron spectroscopy (XPS):

- ✓ Composition-depth profile B
- ✓ Binding energy spectrum



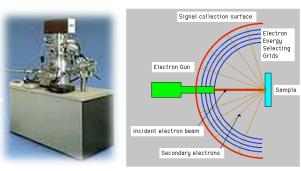
7 Managed by UT-Battelle for the U.S. Department of Energy Transmission electron microscopy (TEM)/Electron Diffraction/EDS

- ✓ Tribofilm nanostructure and thickness
- Cross-sectional element mapping



Auger electron spectroscopy (AES):

Surface element mapping (better spatial resolution than XPS)



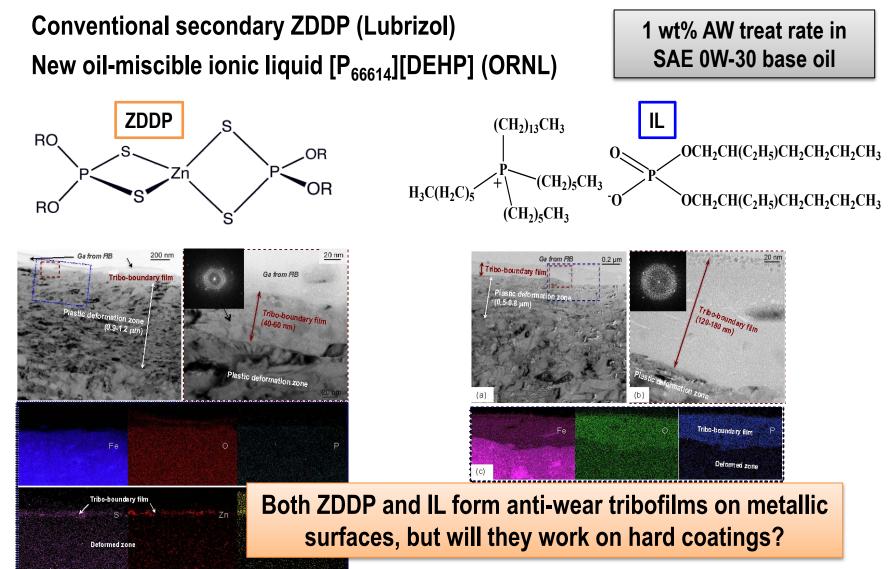


Technical accomplishments – summary

- The mechanism for the ZDDP (and ionic liquid) tribofilm formation on non-metallic coatings has been revealed: ZDDP/IL reacting with metallic wear debris and the new compounds are compressed onto the non-metallic surface.
 - This could be a significant part of the tribofilm formation on a metallic surface as well, in addition to the well-received process of ZDDP/IL directly reacting with the metallic surface.
- The ZDDP and IL formed tribofilms on both boride and DLC coatings with various surface coverage and thicknesses.
- Tribofilms on coatings are composed of reaction products of metal oxides, sulfites (ZDDP only), metal phosphates, and metallic iron (wear debris).
- Tribofilms on boride coatings cover the surface by 80-95% and are up to 60-70 nm thick.
- Tribofilms on DLC have low surface coverage (20-30%) and are <25 nm thick, probably due to poor bonding between tribochemical products and DLC.
- Surprisingly increased wear was observed on the counterface when using the ZDDP (or IL) together with the DLC coating.
 - The IL showed better protection for the steel counterface than the ZDDP though.



Selected lubricant anti-wear additives





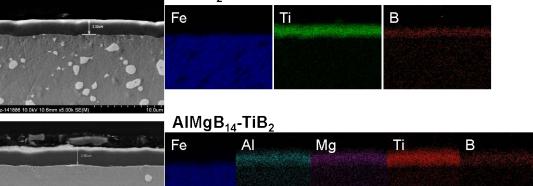
Selected hard coatings

Coating	Composition	Substrate	Process	Thick-ness (μm)	Hardness HK (GPa)	Roughness <i>R</i> _a (μm)	Supplier
TiB ₂	TiB ₂	M2 steel	PVD	2.5	21.2	0.16	Eaton
AIMgB ₁₄ - TiB ₂	AIMgB ₁₄ +50 vol%TiB ₂	M2 steel	PVD	3.0	29.1	0.16	Eaton
DLC	a-C:H	M2 steel	PVD	3.5	18.7	0.16	HEF/NCT

TiB₂

DLC

Fe



C

All three coatings possess high hardness and wear-resistance, but will they work with ZDDP or ionic liquid?



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Friction and wear results

- Boride coatings generated a lightly lower friction coefficient than the DLC in both lubricants
- Similar friction coefficient between the two AW additives
- No measurable wear on coatings.
- The IL-additized oil generated lower ball (counterface) wear than the ZDDP-additized oil for all three coatings suggesting that the IL protects the steel ball better than the ZDDP.

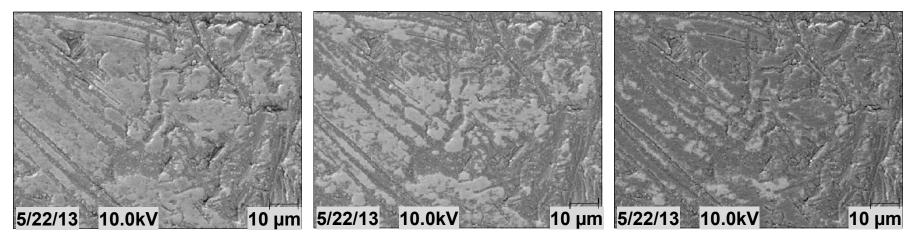
	Wear of coating			of steel ball nm³/N-m)	Steady-state average friction coefficient	
	Oil+ 1%ZDDP	Oil+ 1%IL	Oil+ 1%ZDDP	Oil+ 1%IL	Oil+ 1%ZDDP	Oil+ 1%IL
TiB ₂	Not measurable		7.2	1.3	0.11	0.11
AIMgB ₁₄ -TiB ₂	Not measurable		7.0	3.4	0.11	0.11
DLC	Not measurable		5.3	2.4	0.12	0.12

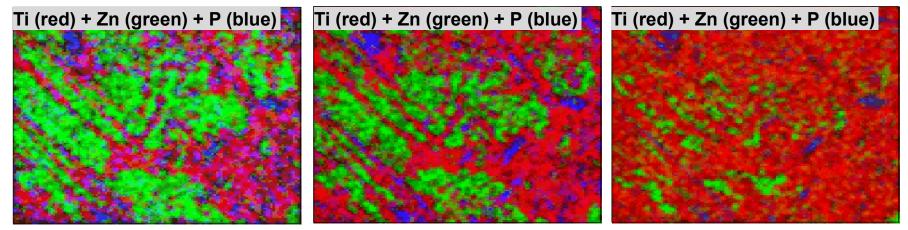


ZDDP-lubricated AIMgB₁₄-TiB₂ – SEM imaging and AES elemental mapping detected a tribofilm

After 30 sec ion sputtering

After 2 min ion sputtering



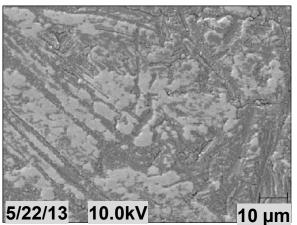




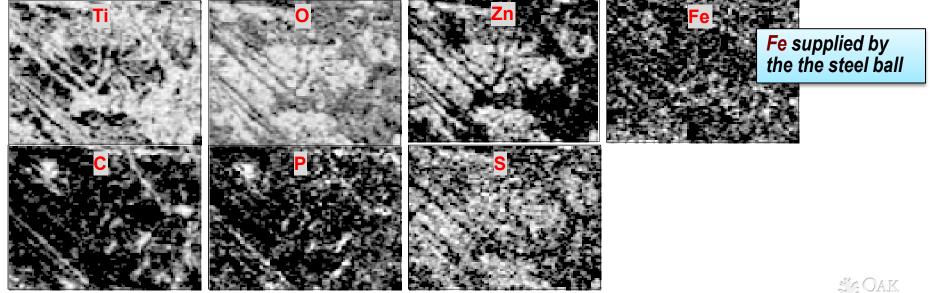
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ZDDP-lubricated AIMgB₁₄-TiB₂ – AES elemental mapping hinted tribofilm composition

After 30 sec ion sputtering



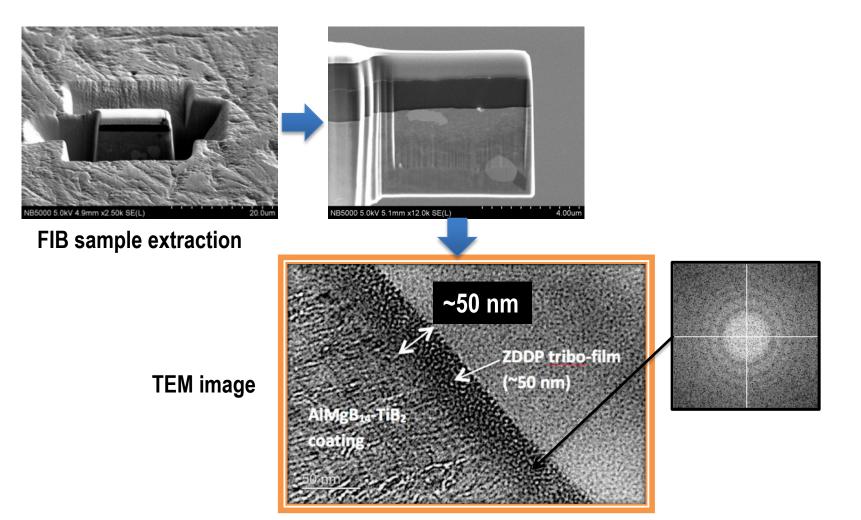
- Zn-O, Zn-S, Fe-S matching maps → zinc oxide, zinc sulfite, and iron oxide(s)
- Fe-P-C-O → maps suggest iron phosphates (inorganic and organic)



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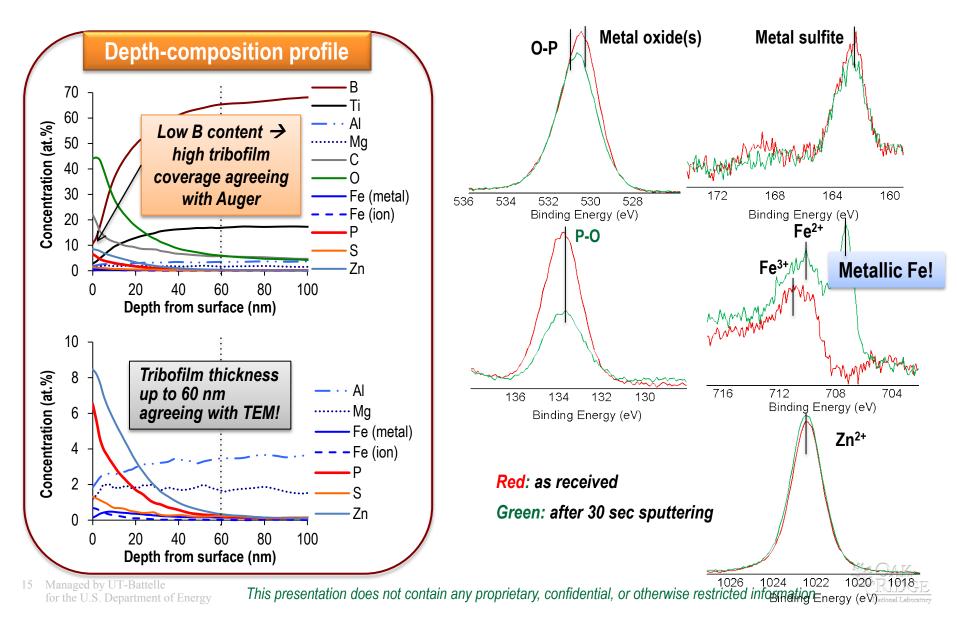
ZDDP-lubricated AIMgB₁₄-TiB₂ – TEM crosssectional imaging revealed the tribofilm ~50 nm thick and dominated by amorphous phases



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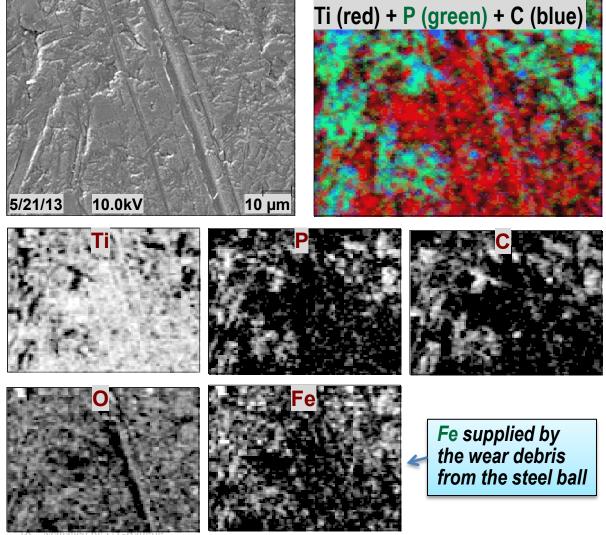


ZDDP-lubricated AIMgB₁₄-**TiB**₂ – XPS analysis provided further info of the tribofilm composition



IL-lubricated AIMgB₁₄-TiB₂ – Auger elemental mapping suggested possible tribofilm composition

After 30 sec ion sputtering

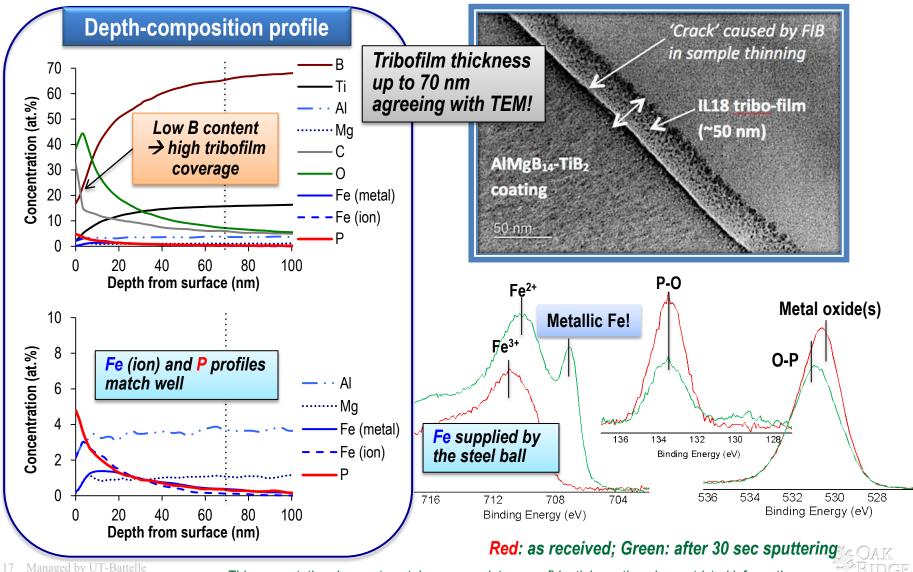


- No AW self-reacted compounds like ZDDP
- All compounds are results of reactions between the IL and wear debris from the steel ball!
- Fe-P-C-O and Fe-O matching maps → iron phosphates (inorganic and organic) and iron oxides
- P-C matching maps → majority of C from non-fully decomposed organophosphate anions



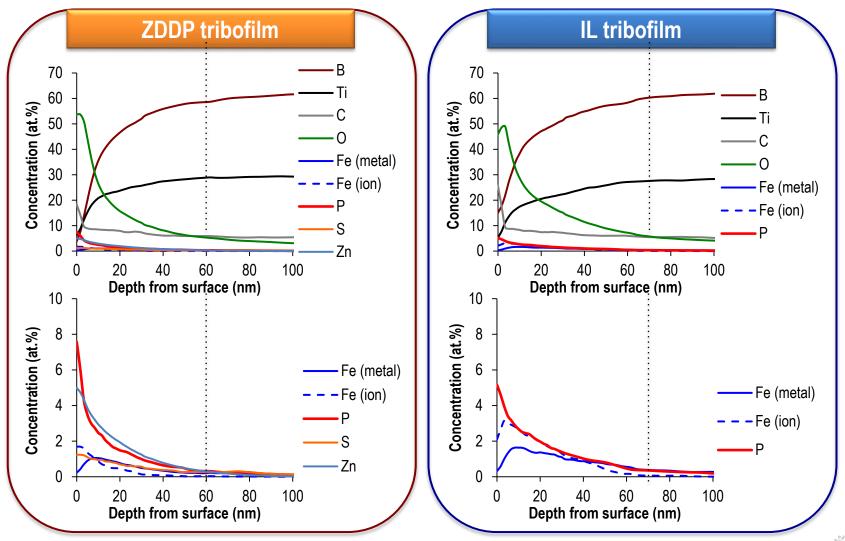
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IL-Iubricated AIMgB₁₄-TiB₂ – TEM crosssectional imaging and XPS analysis of the tribofilm



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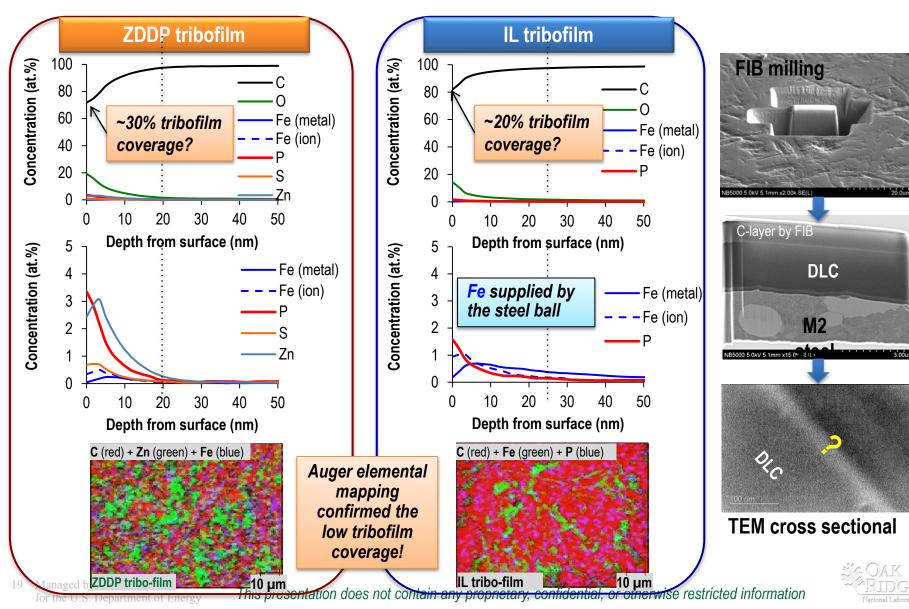
Tribofilms on TiB₂ – similar to those on AlMgB₁₄-TiB₂ (85-95% coverage, up to 60-70 nm thick)



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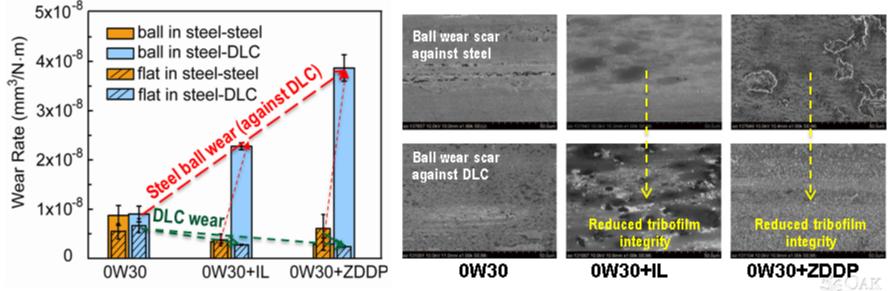


Tribofilms on DLC – lower coverage (20-30%) and thinner (<25 nm)



In progress: surprisingly increased counterface wear when using ZDDP (or IL) and DLC together

- AW additive (either ZDDP or IL) alone in the steel-steel contact (w/o DLC) reduced the wear rates for both the ball and flat, as expected;
- DLC coating alone showed no negative impact on wear or friction;
- Combination of AW (ZDDP or IL) and DLC further reduced the <u>flat wear</u>, <u>however increased</u> the steel <u>ball wear</u>!
 - ZDDP+DLC produced 8X and 4X higher wear on the counter steel ball than using the ZDDP and DLC alone, respectively!
- Hypothesis: competition between AW tribofilm formation and graphite transfer → poor tribofilm integrity → higher wear rate of the steel ball.



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Responses to Previous Year Reviewers' Comments

• Not applicable – this project was not reviewed last year.



Collaboration

- Lubrizol
 - Provided a commercial secondary ZDDP
- Cytec Industries
 - Supplied feed stocks for synthesizing the ionic liquid
- Northeast Coating Technologies
 - Provided two commercial DLC coatings
- Eaton
 - Provided two commercial boride coatings
- ANL
 - Provided two research coatings



Remaining Challenges and Barriers

- Increased counterface wear when using ZDDP (or IL) and DLC together
 - Hypothesis: competition between AW tribofilm formation and graphite transfe → poor tribofilm integrity → higher wear rate of the steel ball.
 - Further characterization involving ultra-high resolution TEM to validate the hypothesis.
- Will the counterface wear increase when using ZDDP (or IL) and other hard coatings?
 - AIMgB₁₄-TiB₂ coating will be used to study this counterface wear issue.
- Lack of understanding of their compatibility on friction behavior in mixed lubrication.
 - Results so far have been focused on boundary lubrication.



Proposed Future Work

Rest of FY 2014

 Further investigation of the issue of increased counterface wear for both DLC and boride coatings

FY 2015

- Investigate the compatibility between ZDDP/IL and hard coatings on friction behavior in mixed lubrication.
 - The majority of literature studies were focused on boundary lubrication.
 - Literature suggests the ZDDP tribofilm commonly increases friction in mixed lubrication for a steel-steel contact. Our IL study showed much lower mixed lubrication friction than ZDDP.
 - ORNL has a newly built Variable Load/Speed Journal Bearing Tester (VLBT), suitable for this task.



Summary

- Relevance: Investigate the compatibility of engine lubricant anti-wear (AW) additives, specifically conventional ZDDP and newly developed ionic liquids, with selected commercial hard coatings to help guide future engine lubricants development.
- Approach/Strategy:
 - Experimentally study the friction and wear behavior for selected non-metallic hard coatings lubricated by selected anti-wear additives via tribological bench testing in well-defined conditions.
 - Mechanistically investigate the tribochemical interactions between the anti-wear additives and the coating surfaces via comprehensive tribofilm characterization.
- Accomplishments:
 - The mechanism for the ZDDP (and IL) tribofilm formation on non-metallic coatings revealed.
 - The AW tribofilms on boride and DLC coatings with various surface coverage and thicknesses.
 - Surprisingly increased wear was observed on the counterface when using the ZDDP (or IL) together with the DLC coating.
- Collaborations:
 - Lubrizol, Cytec Industries Coatings: NCT, Eaton, and ANL
- Proposed Future Work:
 - Rest of FY14: Counterface wear and roughness/temperature effects
 - FY 15: Compatibility on friction behavior in mixed lubrication



Technical Back-up Slides



Ionic liquids (ILs) for lubrication

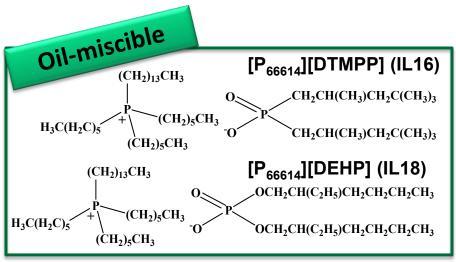
- ILs as neat lubricants or base stocks
 - High thermal stability (up to 500 °C)
 - High viscosity index (120-370)
 - Low EHL/ML friction due to low pressure-viscosity coefficient
 - Wear protection by tribo-film formation
 - Suitable for specialty bearing components

ILs as oil additives

- Potential multi-functions: anti-wear/EP,
 FM, corrosion inhibitor, detergent
- − Ashless \rightarrow low sludge
- Allow the use of lower viscosity oils
- Advantage: cost effective and easier to penetrate into the lubricant market
- Problem: most ILs insoluble in oils

lonic liquids are 'room temperature molten salts', composed of cations & anions, instead of neutral molecules.

• ORNL-developed oil-miscible ILs:



B. Yu, and J. Qu^{*}, et al., Wear (2012) 289 (2012) 58. J. Qu, et al., ACS Applied Materials & Interfaces 4 (2) (2012) 997.

