

# In Situ Electron Microscopy of Electrical Energy Storage Materials

## Raymond R. Unocic

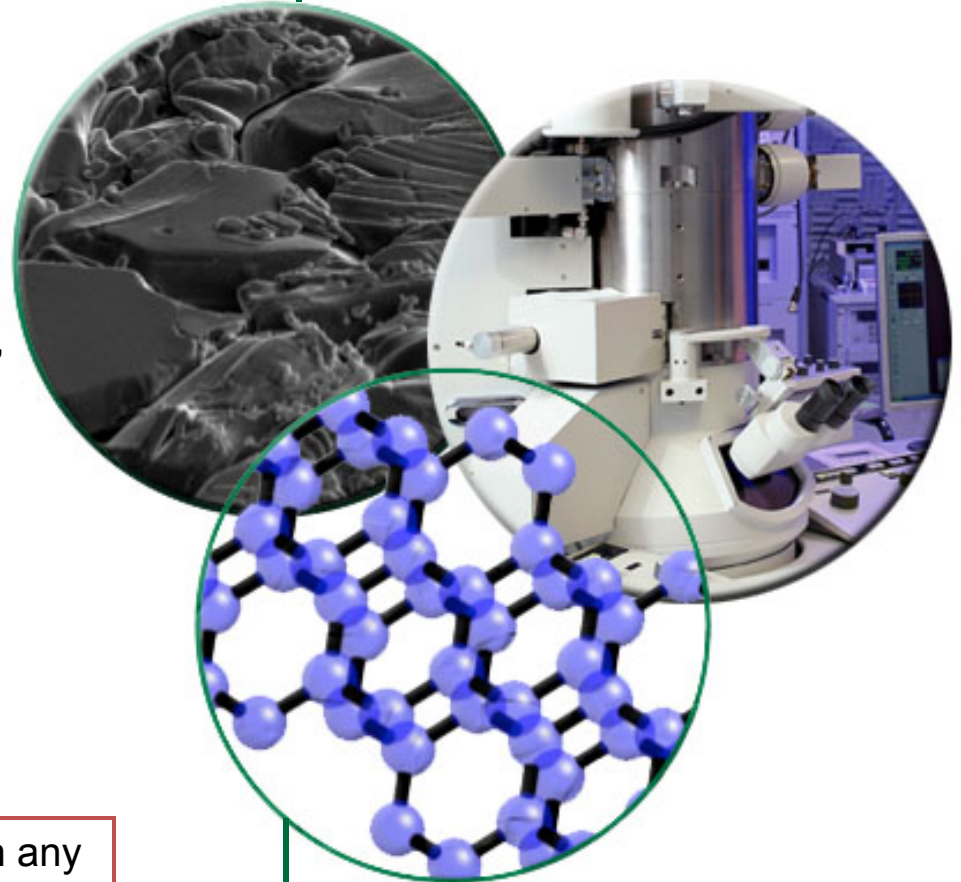
Principal Investigator  
Materials Science and Technology Division  
Oak Ridge National Laboratory

### Contributors:

Loic Baggetto, Robert L. Sacci, Gabriel M. Veith,  
Nancy J. Dudney, Karren L. More  
Materials Science and Technology Division  
Oak Ridge National Laboratory

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# Overview

## Timeline

- Project Start: January 2010
- Project End: Sept 2013
- Percent Complete: 90%

## Technical Barriers

- Limited Cycling Performance
- Limited Stability of Electrodes for High Voltage Batteries
- Safety

## Budget

- Total Project Funding: 100% DOE
- Funding for FY10: \$250k
- Funding for FY11: \$250k
- Funding for FY12: \$300k
- Funding for FY13: \$300k

## Partners/Collaborators

- General Motors Global R&D
  - Drs Xincheng Xiao and Zhongyi Liu
- University of Texas Austin
  - Prof. A. Manthiram
- ORNL FIRST EFRC
  - Dr. David Wesolowski
- Argonne National Laboratory
  - Dr. Daniel Abraham
- ORNL BES ShaRE User Facility

# Objectives and Relevance

- **Objectives**

- To understand nm-scaled microstructural and microchemical changes occurring within energy storage materials using in situ electron microscopy.
- To develop, implement, and validate in situ electrochemical fluid cell microscopy, an **in situ transmission electron microscopy (TEM)-based characterization method**, to perform controlled and **quantitative** nanoscale electrochemistry experiments.



Hitachi HF3300 S/TEM at  
ORNL for in situ TEM  
Characterization

- **Relevance to Technical Barriers**

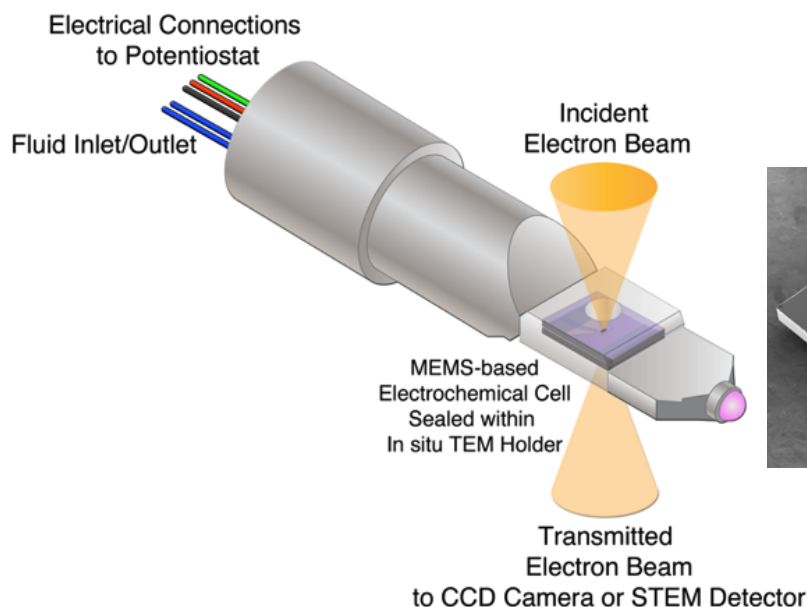
- Understand fundamental issues related to performance, capacity loss, electrode degradation mechanisms and safety.
- Specific technical barriers addressed in this program:
  - SEI Formation Mechanisms (Role of Electrolytes and Electrolyte Additives)
  - Electrode Degradation Mechanisms and Capacity Fading Mechanisms of Relevant High Voltage Cathodes
  - Electrolyte Stability (Reducing and Oxidizing Potentials)
  - Li Dendrite Formation

# Milestones FY13

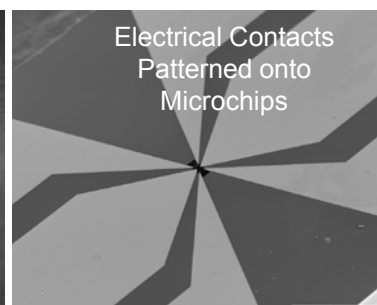
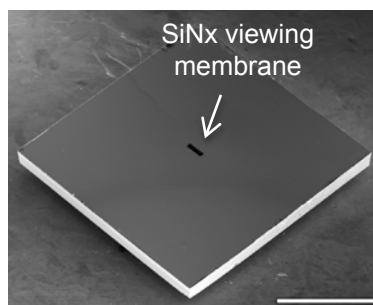
Description:	Target Deadline:	Status:
<b>Milestone 1:</b> Electrolyte decomposition mechanisms and kinetics of SEI on model graphite anodes	(May 2013)	Completed
<b>Milestone 2:</b> Electrolyte stability at high voltage cathodes with different electrolytes.	(May 2013)	On Schedule
<b>Milestone 3:</b> Investigate electrode degradation mechanisms in active electrodes.	(May 2013)	On Schedule
<b>Milestone 4:</b> Evaluate the usefulness of aberration-corrected STEM/EELS/EDS for combined in situ TEM characterization and chemical analysis.	(May 2013)	Completed
<b>Milestone 5:</b> Investigate dendrite formation mechanisms	(August 2013)	On Schedule
<b>Milestone 6:</b> Technique development: Design, develop and optimize 'on-chip' 2- and 3-electrode biasing microchips.	(August 2013)	On Schedule

# Technical Approach and Strategy

- Approach: In situ Electrochemical Liquid Cell Microscopy
  - MEM- based microchips are a platform for sealing electrodes and electrolyte between SiN membranes
  - Quantitative electrochemical measurements are performed
  - In situ characterization of electrochemical reactions  
(via (S)TEM imaging/diffraction/spectroscopy)
  - **No other method allows for the direct characterization of electrochemical process dynamics within native electrolyte environment under realistic testing conditions and while imaging at high spatial resolution with site-specific chemical analysis**

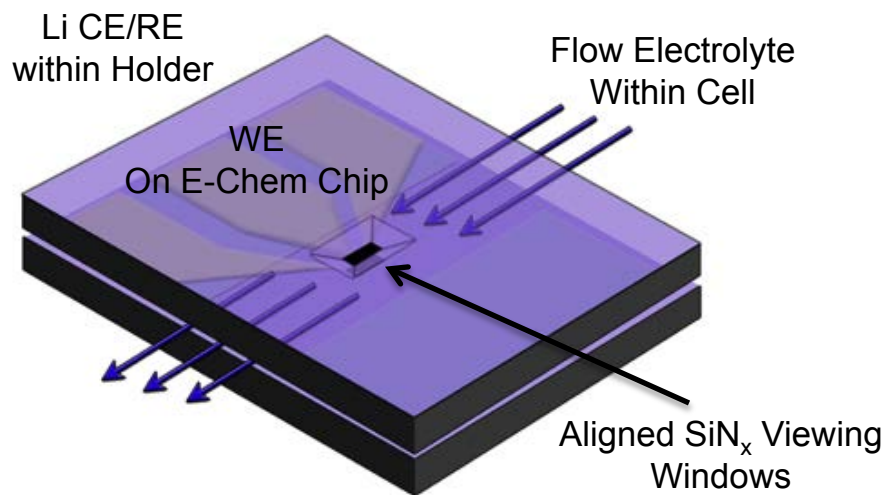


## MicroElectroMechanicalSystems (MEMS) Microchip Platforms for In situ Electrochemistry



# Technical Approach and Strategy

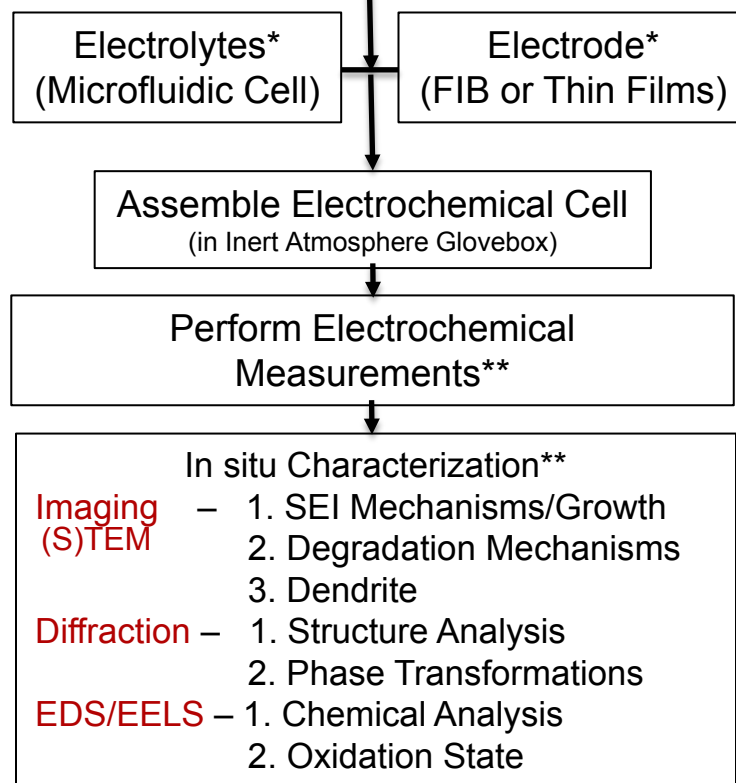
- In situ Electrochemical Test Method: Energy Storage Materials
  - Electrochemical measurements in half-cell configurations, metallic Li as counter (CE) and reference (RE) electrode
  - Working electrode is electrode material (e.g. graphite,  $\text{LiMn}_2\text{O}_4$ ,  $\text{LiMn}_{1.5}\text{Ni}_{0.5}\text{O}_4$ , Li-rich NMC, etc).



\* Materials relevant to VT program

\*\*Unique in situ characterization method developed in this program (FY10-FY12) to study fundamental issues limiting Li and Na ion batteries

## In situ Characterization Methodology



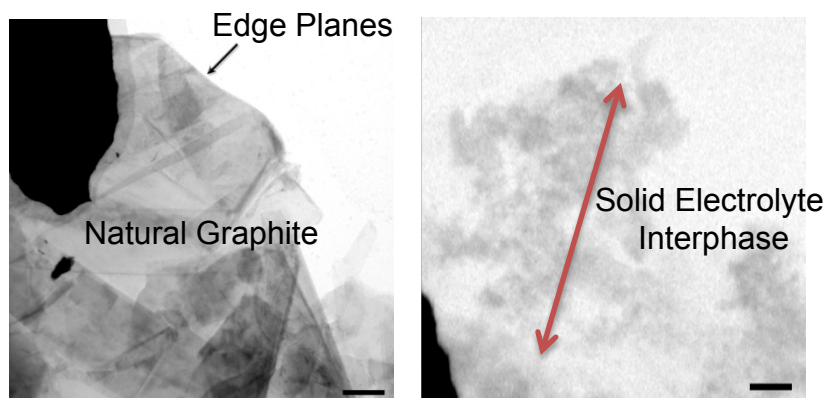


# Technical Accomplishments and Progress

- Executive Summary of Technical Accomplishments
  - Milestone 1: Solid Electrolyte Interphase (SEI)
    - Completed in depth study of in situ SEI formation mechanisms and growth kinetics
  - Milestone 2: Electrolyte Stability for High Voltage Cathodes
    - Developed method to deposit thin film electrodes on MEMS-based E-chips
  - Milestone 3: Electrode Degradation Mechanisms
    - Conducted In situ Investigation of intercalated induced fracture of silicon anodes
  - Milestone 4: Combining Analytical Electron Microscopy Methods
    - Demonstrated feasibility of EELS and EDS in for in situ chemical analysis
  - Milestone 5: Dendrite Formation
    - Conducted preliminary experiments of dendrite formation mechanisms and kinetics
  - Milestone 6: MEMS microchip development
    - Designed future MEMS based E-Chip platforms

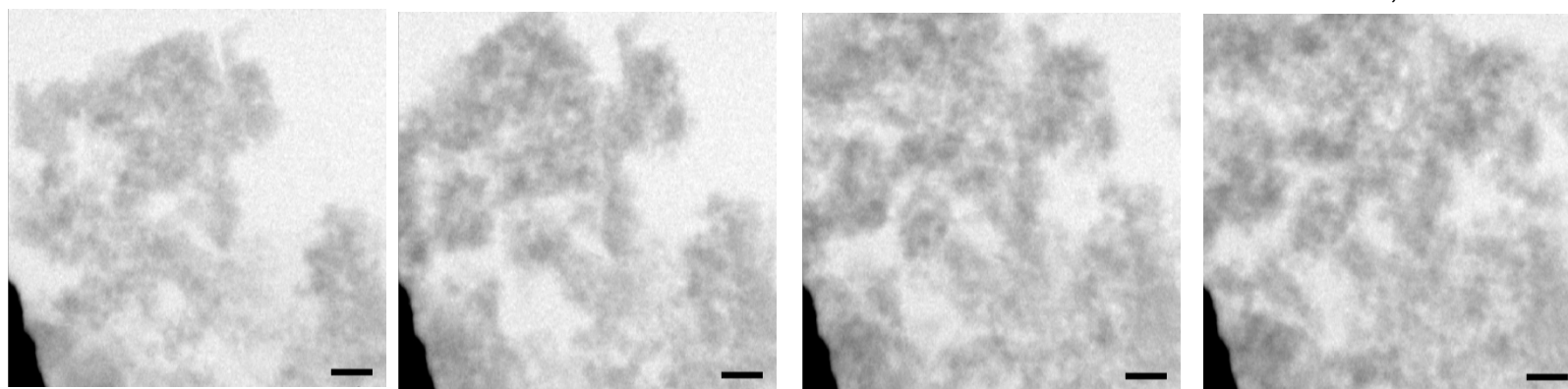
# First Direct Visualization of SEI Formation under realistic conditions

- System is Graphite vs Li/Li<sup>+</sup> and results reveal:
  - Direct observation of SEI growth on natural graphite anode during cell discharge
  - **1M LiPF<sub>6</sub> EC:DEC Electrolyte**
  - Structure of SEI comprised of nanocrystalline species and has a porous morphology.
  - There is sufficient resolution to image SEI within the liquid electrolyte and measure growth rate.



First in situ TEM study of SEI growth mechanisms and kinetics using relevant electrode/electrolyte system and under realistic testing conditions.\*

Direct Imaging of Electrochemical Reactions in Li-ion Batteries. R.R. Unocic et. al., submitted.

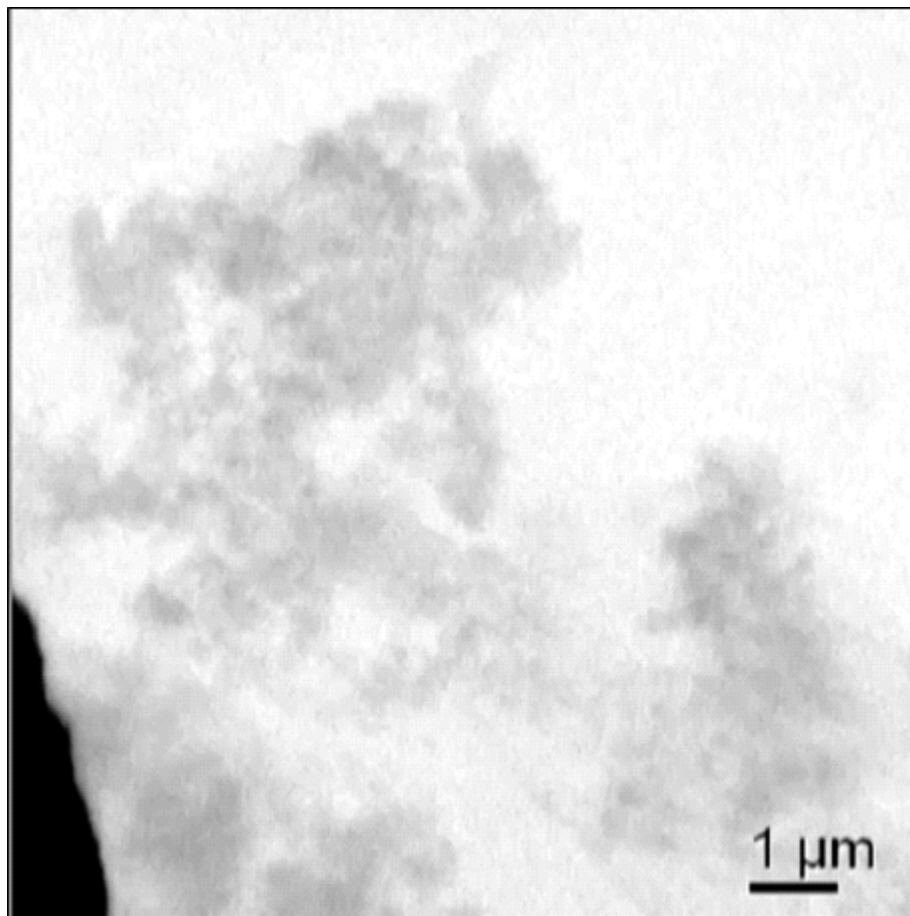


Time-lapsed TEM images of SEI Growth at Constant Cell Potential (1.25V)

**\* In situ SEI studies is crucial for fundamentally understanding irreversible capacity loss, capacity fading, and electrode cyclability issues.**



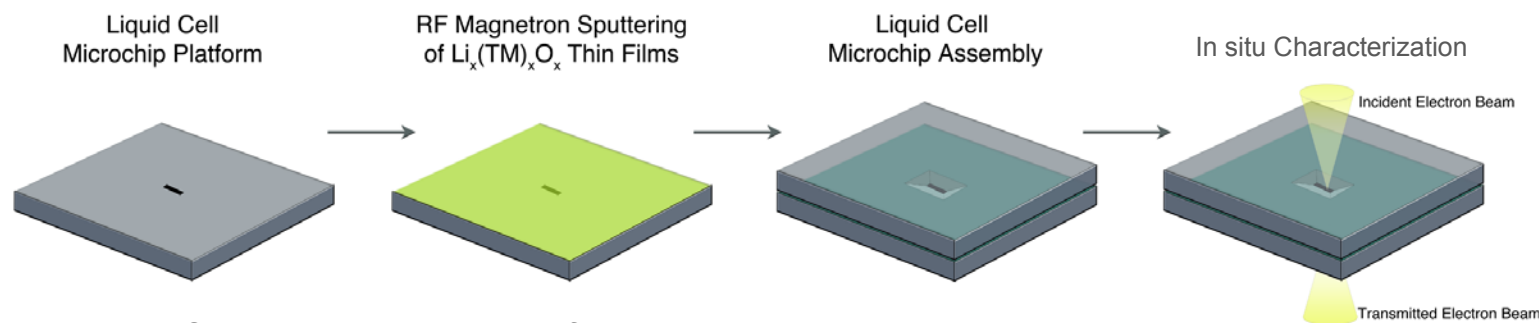
# In situ Video of SEI Growth in Relevant Systems



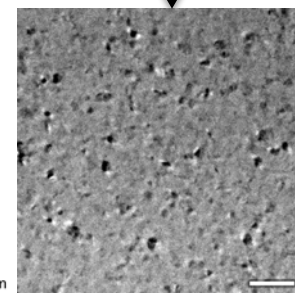
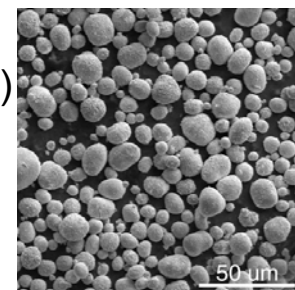
Experimental Details:  
(Graphite vs  $\text{Li/Li}^+$ ) within 1M  $\text{LiPF}_6$  EC:DEC Electrolyte  
Video sequence acquired during potential hold at 1.25V

# Electrode Thin Films as a Model System

- Thin film electrodes are model systems for in situ TEM experiments
  - Major Advantages:
    - Sputtering target directly from electrode material of interest (e.g. Li-rich NMC)
    - Chemical composition can be controlled
    - Electrode thin films directly deposited on E-Chem Chips



Schematic illustration of deposition and characterization methodology



Nanocrystalline Grains

Current Inventory of Thin Film Electrode Materials Successfully Deposited on E-Chem Chips:

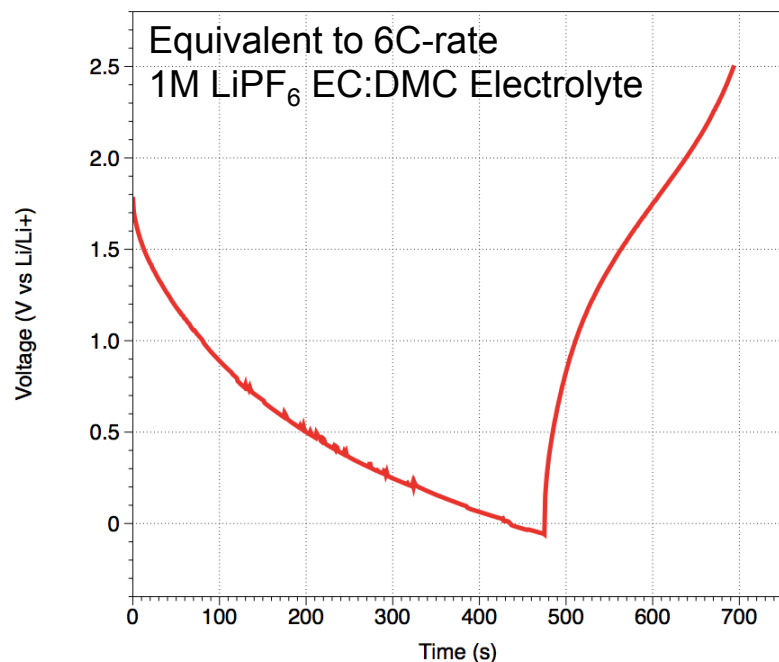
Cathodes: **Li-Rich NMC**,  **$\text{LiMn}_{1.5}\text{Ni}_{0.5}\text{O}_4$** ,  **$\text{LiMn}_2\text{O}_4$**

Anodes:  **$\text{Li}_4\text{Ti}_5\text{O}_{12}$** , **Cu-Sb**, **Cu-Sn**

Note: Coordinating within VT Program for relevant systems to investigate (ORNL, ANL, UT Austin)

# Charge-Discharge Behavior of Cu-Sb Anodes within In Situ Holder

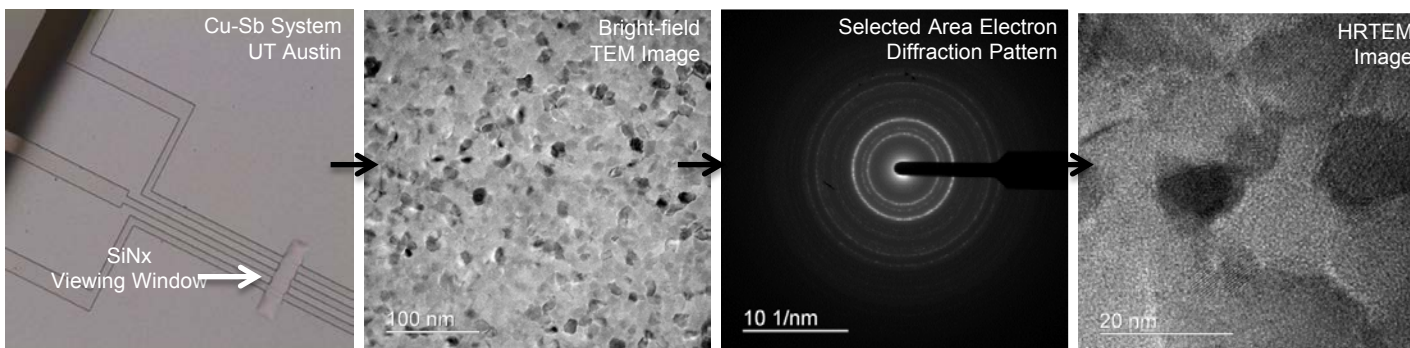
- **Key Question:** Can the thin film electrodes deposited on the E-Chem Chips be cycled?
- Results show that we can conduct constant current charge-discharge experiments (and vary C-rate)



The use of thin film electrodes opens a new pathway to investigate a wide range of materials systems of interest within the VT Program

Results demonstrate feasibility to conduct controlled electrochemical charge-discharge cycling experiments

Rate performance and cycle-life can be assessed while structural and chemical changes can be simultaneously characterized



Promising means for in situ characterization on varying length scales

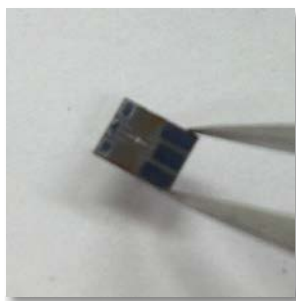
# New Insight into Degradation Mechanisms in High Voltage Cathodes

## • High Voltage Cathodes:

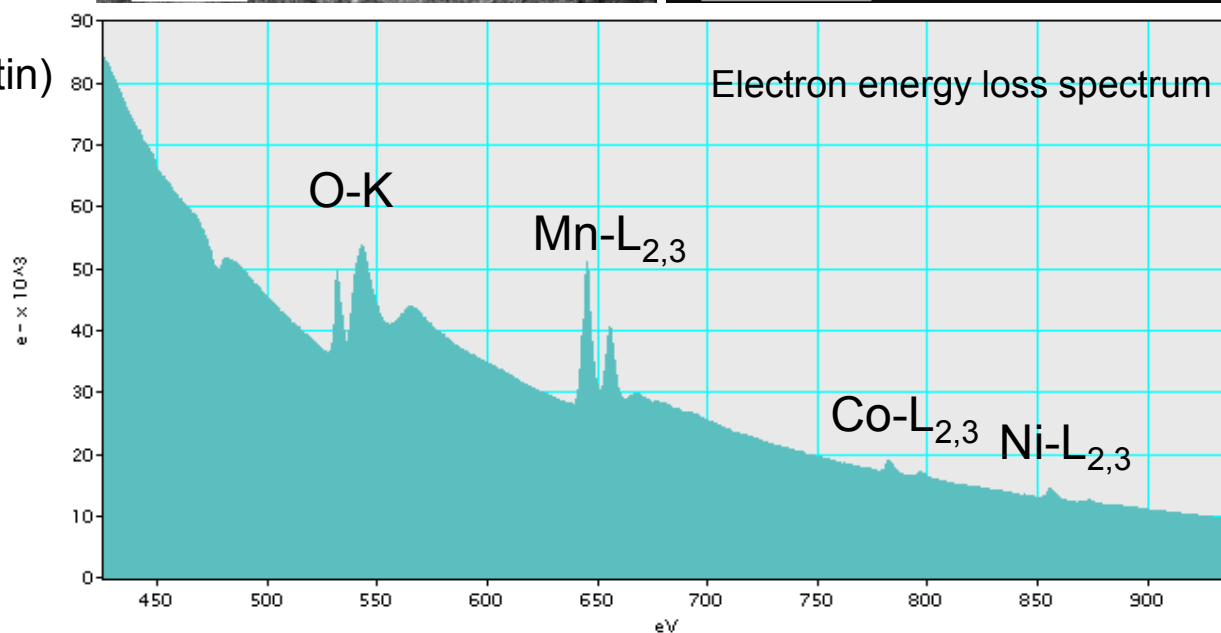
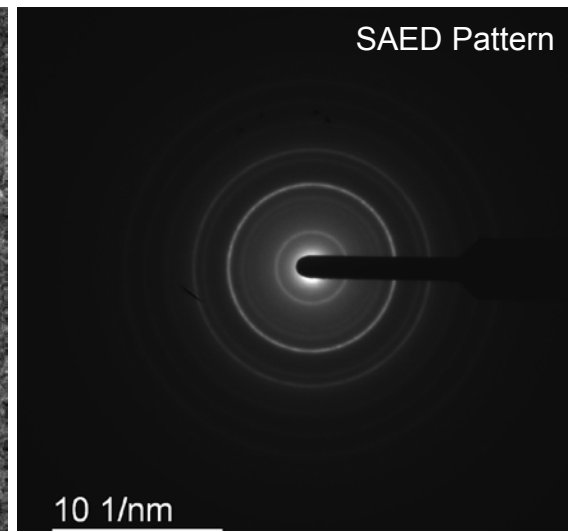
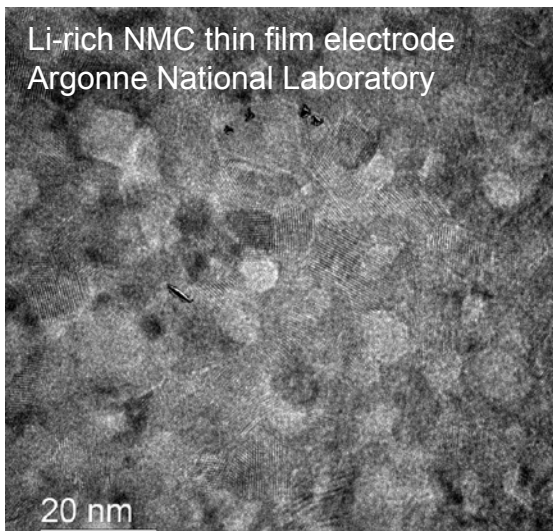
- Can we study phase transformations?  
Layered/Spinel
- Can we image electrolyte stability at High Voltage (5V) in oxidizing conditions?

## • Results:

- Materials
  - Li-rich NMC (ANL)
  - $\text{LiMn}_{1.5}\text{Ni}_{0.5}\text{O}_4$  (UT Austin)



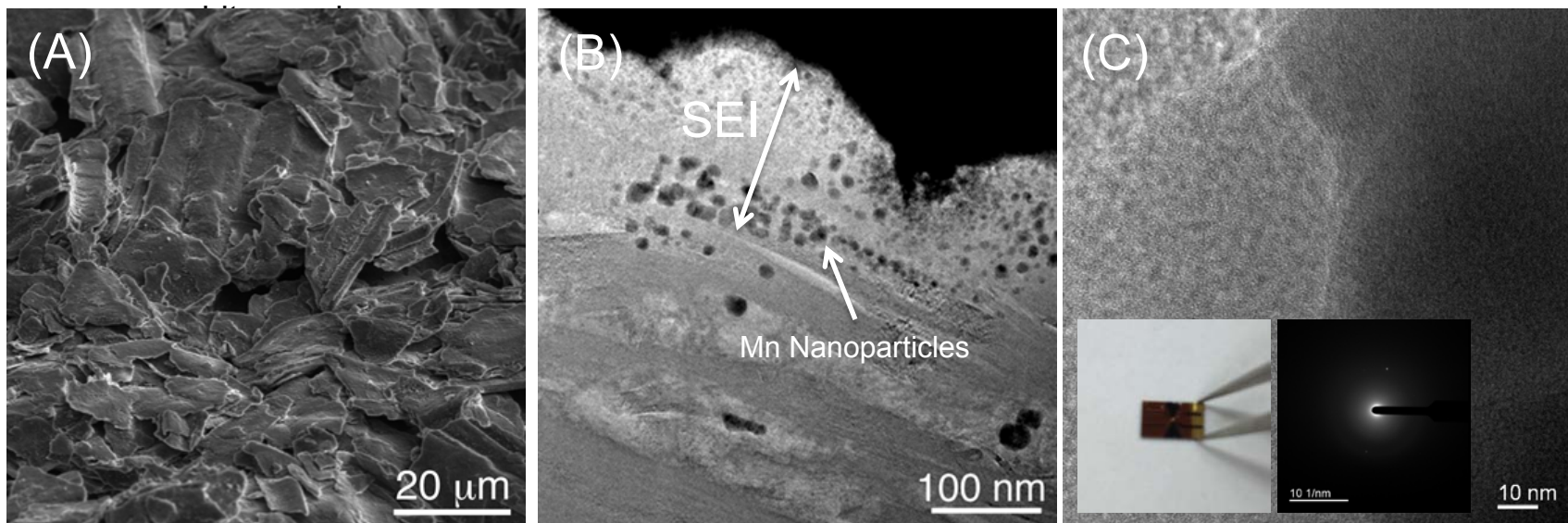
Electrode thin films of High Voltage Cathode Materials successfully deposited on MEMS E-Chem Chips.  
*In situ characterization is in progress.*





# Understanding Capacity Fading Issues in High Voltage Cathodes

- Mn dissolution linked to capacity fade mechanism in High Voltage Cathodes
  - Collaboration with General Motors Global R&D:
    - Cycled graphite anode and LMO cathode.
    - Post-mortem SEM/FIB and STEM/HRTEM analysis shows Mn nanoparticles within SEI on

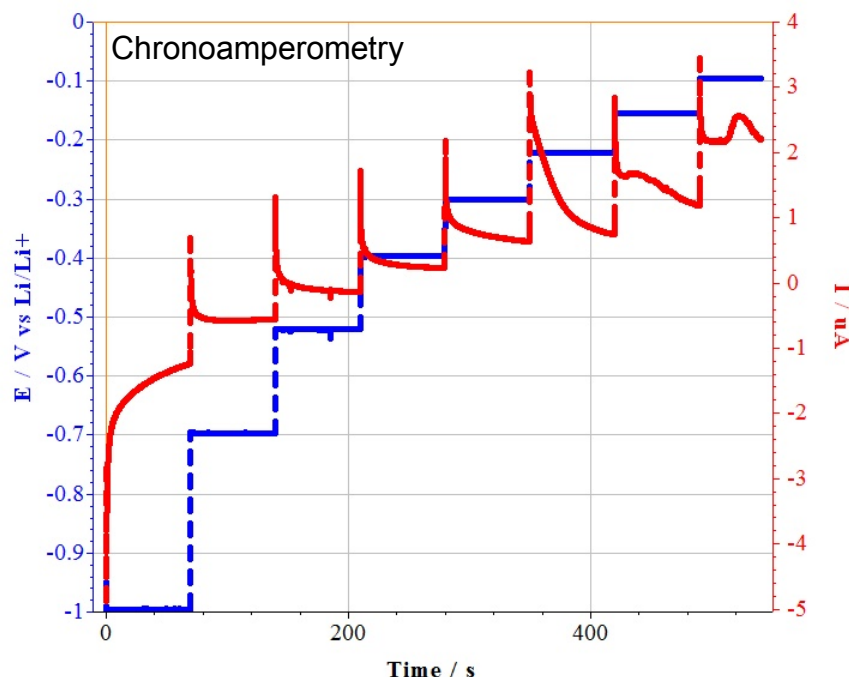
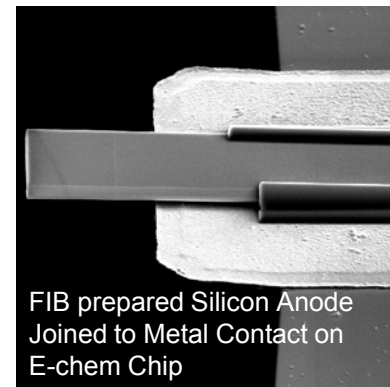


(A) SEM image of graphite anode material extracted from a cycle cell revealing SEI on the electrode surface and (B) TEM Image from FIB X-section showing Mn nanoparticles with SEI segregating locally at the SEI/graphite interface. (C) GM developed method to deposit graphite thin film electrodes on MEMS-based E-Chem Chips. Used as a model substrate for in situ Mn dissolution study.

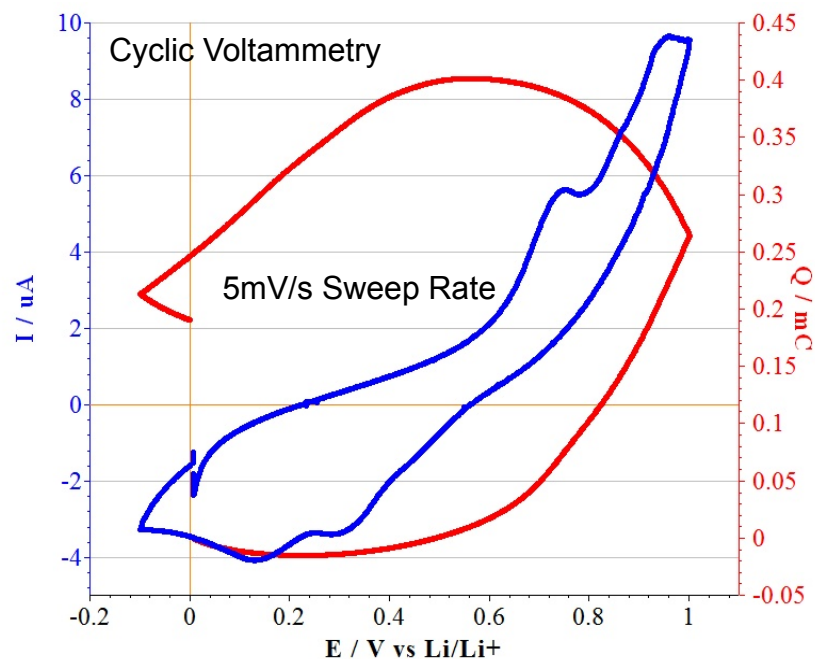
SEI Characterization Study: Facilitated by Hitachi Vacuum Transfer/Air-Tight Argon Glovebox → FIB/SEM → (S)TEM Transfer System (Available at ORNL)

# Continued Development is Providing Quantitative Electrochemical Measurement Results

- **Evaluating In situ System for Quantitative Electrochemistry**
  - Working Electrode: FIB prepared Micron Sized Silicon Anode
  - Li Counter/Reference integrated into tip of TEM holder
  - Results demonstrate feasibility for in situ nanoscale electrochemical measurements



Recording sub micro-amp currents enables the counting of Li atoms into the Si anode along side TEM images of anode swelling and SEI formation

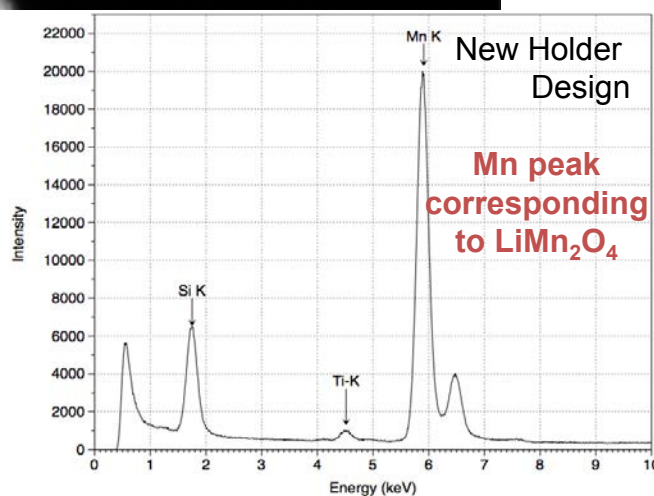
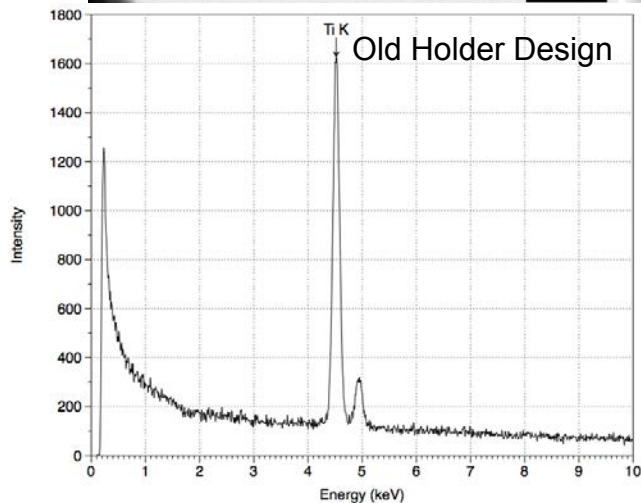
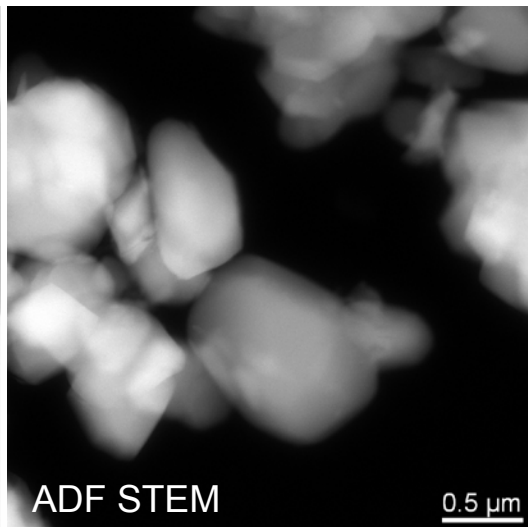
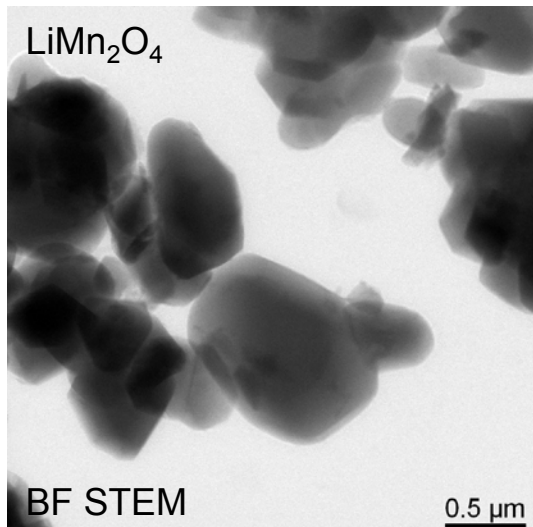


Holder integrated counter/reference minimizes IR loss for better cell control – mimics bench-top cell



# New Holder Design Enables EDS Chemical Analysis w/in Liquid Electrolyte

- **Energy Dispersive X-ray Spectroscopy (EDS) and Electron Energy Loss Spectroscopy (EELS):**
  - Analytical methods used for compositional analysis in combination with (S)TEM imaging
  - Relevant for investigating chemical changes during in situ electrochemical cycling experiments



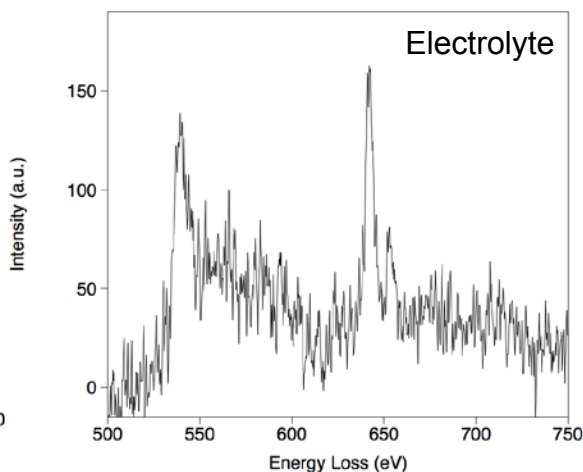
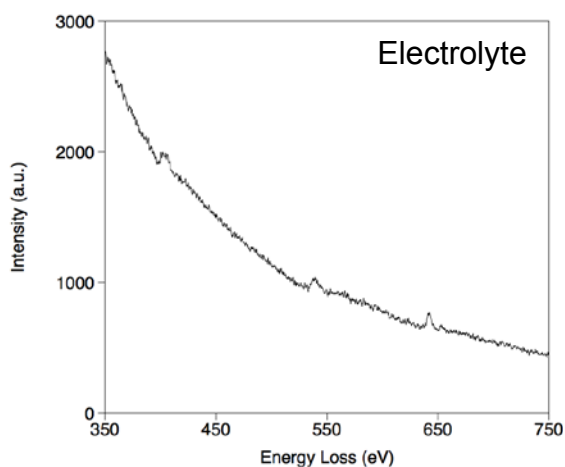
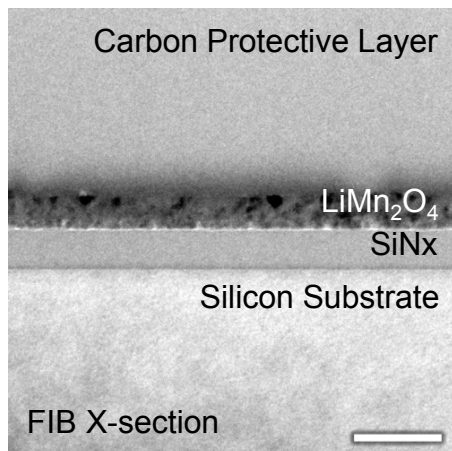
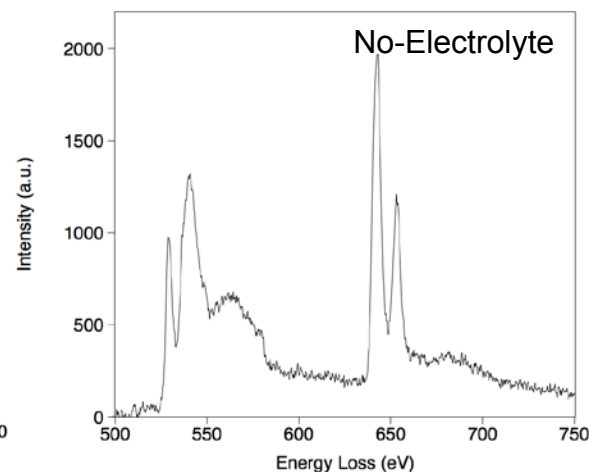
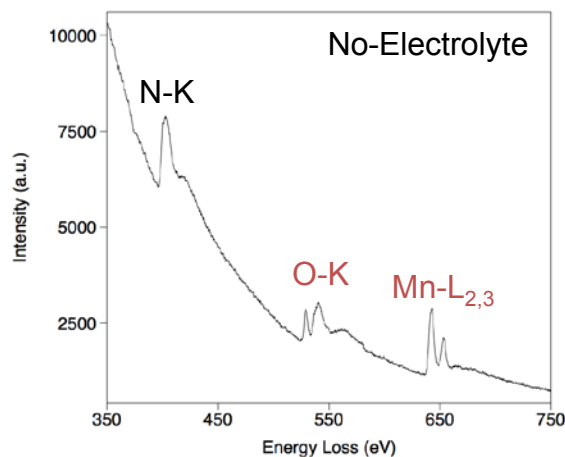
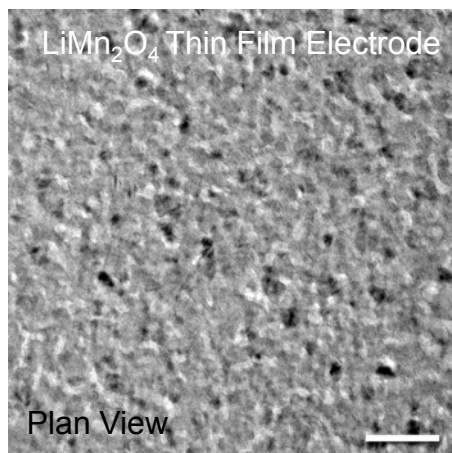
Old Design: X-rays generated within the specimen are effectively blocked by holder components

New Design: Improved holder design allows X-rays to escape specimen and reach EDS detector for chemical analysis

# EELS Chemical Analysis and Oxidation State Determination

## • Electron Energy Loss Spectroscopy (EELS):

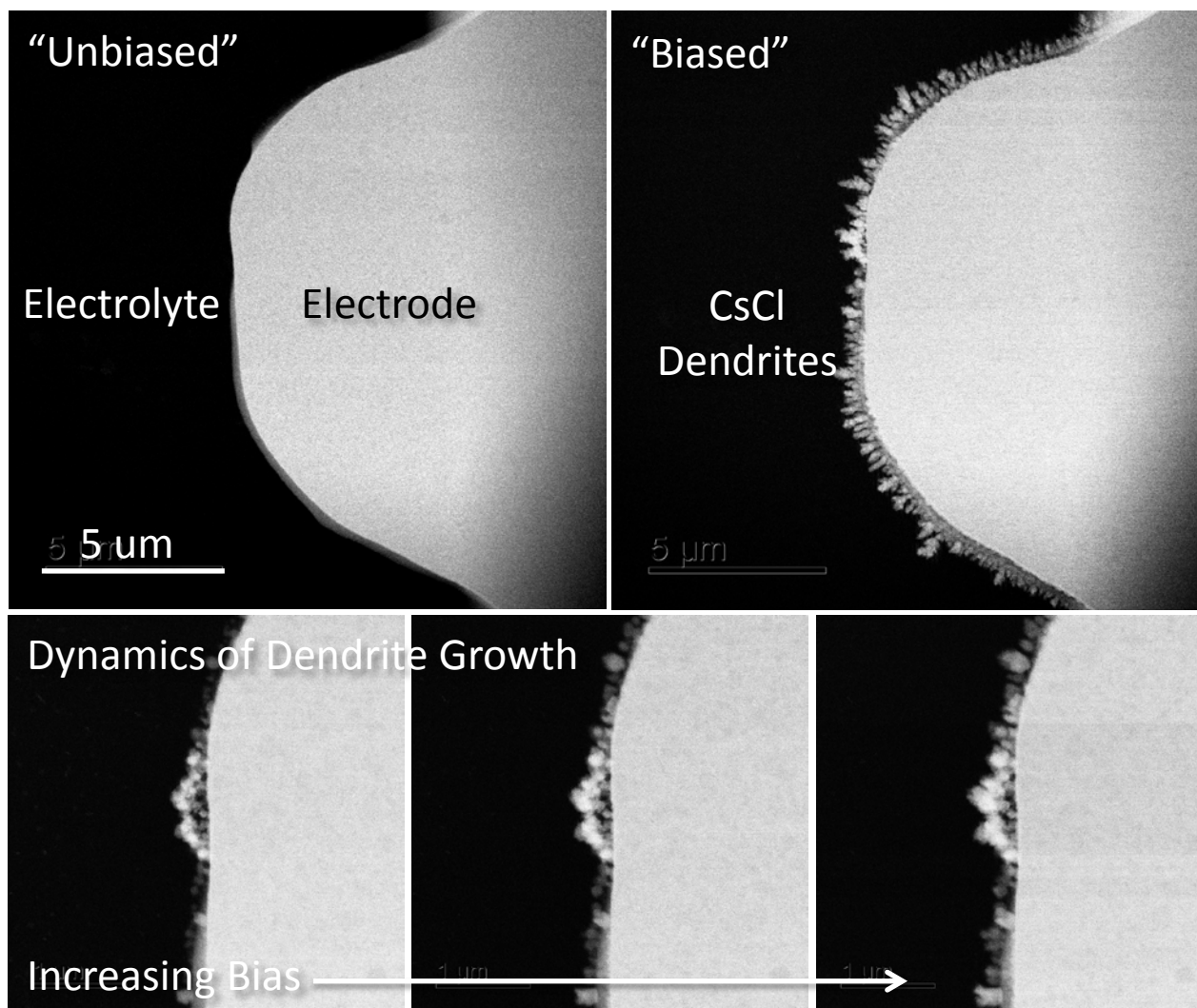
- Method used for chemical composition analysis and to directly determine oxidation state.
- Oxidation state determined through core-loss spectra and transition metal “white-line ratio” method.
- Relevant for in situ investigations of re-dox changes during electrochemical cycling.



Fourier-Log Deconvolution and Background Subtraction

# In Situ Characterization of Dendrites - Safety

- Real time characterization of dendrite formation mechanisms:



The In situ method developed in this program can directly be used to investigate dendrite nucleation and growth mechanisms in Li-ion batteries

— Relevance:

- Safety Concern: Internal short circuiting, overheating, electrolyte flammability
- Better understanding of dendrite formation can lead to new discoveries that will minimize or mitigate dendrite formation during cell operation.
- Li dendrites at graphite, alloy anodes and/or current collector.

# Collaborations and Coordination

- **UT Austin**
  - Cu-Sb Material (Anodes)
    - Supplied Cu-Sb powder → sputtering target → deposited thin film electrodes on MEMS E-Chip
- **Argonne National Laboratory**
  - Li-Rich NMC Material (Cathodes)
    - Supplied Li-Rich NMC → sputtering target → deposited thin film electrodes on MEMS E-Chip
- **General Motors Global R&D**
  - Model thin film graphite electrode, Mn dissolution mechanisms
    - GM developed method to fabricate graphite thin film electrodes on MEMS E-Chip
    - Mn dissolution mechanisms from High V cathodes
- **ORNL FIRST EFRC**
  - SEI Studies, Si Anode Degradation Mechanisms
    - In situ SEI formation mechanisms and growth kinetics on graphite and Si
    - Si electrode fracture mechanisms

# Future Work – FY 13

- **Capacity Fading Issues (High Voltage Cathodes)**
  - Investigate structural changes in situ with electron diffraction and EEL spectroscopy
  - Investigate Mn dissolution mechanisms.
- **Electrode Degradation Mechanisms (Si Anodes)**
  - Continue investigation of Li intercalation mechanisms and electrode degradation and fracture mechanisms
- **Solid Electrolyte Interphase**
  - Conduct further in situ SEI studies on graphite anode with different organic electrolytes
    - $\text{LiPF}_6$  in EC:DMC and  $\text{LiPF}_6$  in PC
    - Other electrolyte and electrolyte additives relevant to VT program
- **Dendrite Formation in Li Metal Anodes**
  - Investigate Li metal dendrites
  - Determine dendrite nucleation mechanisms and quantify growth rates



# Summary

- **Relevance**

- Advanced in situ characterization microscopy methods play a vital role in understanding the basic electrochemical processes that currently limit battery performance

- **Approach**

- In situ Electrochemical Liquid Cell Microscopy method developed
- Apply in situ characterization method to investigate fundamental issues related to performance, capacity loss, electrode degradation mechanisms and safety.

- **Technical Accomplishments and Progress**

- Investigated SEI formation mechanisms and growth kinetics
- Demonstrated the quantitative electrochemistry capabilities of system
- Developed method to deposit thin film electrodes on MEMS-based E-Chem Chips and evaluated the cycling behavior
- Demonstrated feasibility to conduct in situ chemical analysis with EDS/EELS

- **Future Work**

- Continue SEI studies with different electrolytes/electrolyte additives
- Continue work on capacity fading issues related to High Voltage Cathodes
- Continue studies of Li dendrite growth mechanisms



# Acknowledgements

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- General Motors Global Research
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