# Development of High Energy Density Li-Sulfur Cells

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#### Project ID # ES 125

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### **OVERVIEW**

#### Timeline

#### 12.5% completed

*Start:* Sep. 30, 2011 *End:* Jan. 15, 2015

#### **Barriers Addressed**

- Power and energy density
- Cycle and calendar life
- Abuse tolerance

#### **Budget**

- DOE \$ 5M
- Cost Share: \$2.29M
- FY 2012: \$1.626M (DOE)

#### **Partners**

- Johnson Controls
- EC Power
- Argonne National Lab
- Idaho National Lab

#### **O**BJECTIVES

#### Develop a **full lithium-sulfur battery system** with high energy density and efficiency, good cycle life and safe operation.

**Project scope** 

Design of full Lithium-Sulfur cell:

- Cathode: Nanocomposite Sulfur high energy / power
- Advanced Lithium anode stable
- Electrolyte nonflammable, stable
- Optimized cell design

**Performance targets** 

**4Ah cells** 

600 Wh/L (cell level)

Cycle life 500+ cycles

**Excellent safety characteristics** 

## **MILESTONES**

**Phase I: Advanced material development and characterization** (10/2011-01/2013)

- Evaluate baseline cells
- Develop and optimize cathode, anode, and electrolyte
- Thermal and failure mechanism studies

# **Phase II: Material scale up and 1Ah pouch cell development** (01/2013-01/2014)

- Cathode and anode scale-up, continued improvement, and 1 Ah pouch cell design and testing
- Continued investigation of electrolytes and failure mechanisms

#### **Phase III: Large format 4 Ah prismatic cell design** (01/2014-01/2015)

- Continued scale-up, failure mechanism analysis, and 4 Ah prismatic cell design and testing
- Cell modeling and optimization

## **A**PPROACH

Anode	<ul> <li>Investigating / optimizing composite Li-based anodes</li> <li>Exploring polymer electrolyte coatings</li> </ul>
Cathode	<ul> <li>Investigating nanocomposites</li> <li>Pursuing methods for increased sulfur loading</li> </ul>
Electrolyte	<ul> <li>Scanning electrolytes and additives</li> <li>Assessing and developing novel solvents</li> <li>Optimizing electrolytes for poly-sulfide dissolution</li> </ul>
Cell design	<ul> <li>Optimizing cells design for chemistry requirements</li> <li>Conducting modeling and experimental testing</li> </ul>

## APPROACH - CATHODE

- Generate new, well-structured carbon frameworks to improve volumetric energy density
- Optimize framework pore geometry and investigate new framework materials and structures to maximize sulfur loading and thus energy density
- Make intelligent use of additives to prevent polysulfide dissolution and improve cycling and overall performance
- Materials process optimization to identify best production condition.

### **APPROACH - ANODE**

- Design composite lithium-based anode to suppress dendrite growth, promote stable SEI formation, improve anode stability
- Develop effective anode coating to prevent dendrite growth and lithium polysulfide deposition

### **Approach - Electrolyte**

- Develop ionic liquid electrolytes and electrolyte additives to improve performance and safety
- Develop new electrolyte systems to improve safety, rate capability, and anode SEI formation, and decrease polysulfide dissolution
- Investigate the mechanism of polysulfide dissolution to provide complementary insight

## APPROACH – CELL DESIGN

- Optimize prismatic full cell parameters electrode size, electrode matching, number of electrodes in stack, etc
- Synergistically leverage cell modeling and experimental testing to iteratively improve design

## TECHNICAL ACCOMPLISHMENT AND PROGRESS

- Developed 1 Ah NCM-based baseline cells
- Developed metal oxide adsorbent and verify its effect to improve capacity and efficiency in graphene-metal oxide-sulfur nanocomposite cathodes.
- Developed carbon-sulfur composite cathode with high sulfur loading.
- Electrolyte additive development and testing

## **1** AH BASELINE CELLS CYCLING UPDATE



- 1C cycling: 88% capacity retention in 418 cycles
- 2C cycling: 98% capacity retention in 197 cycles

#### **GRAPHENE-METAL OXIDE-SULFUR CATHODES**



- Graphene-metal oxide-sulfur composite with 1<sup>st</sup>-cycle capacity ~1100 mAh/g, 92% efficiency, and capacity retention of 86% after 50 cycles.
- Metal oxides show adsorbing effect to soluble polysulfides.

#### SULFUR-CARBON NANOCOMPOSITE CATHODES



 Nanocomposite sulfur cathodes with sulfur loading increased to 80 wt%, 1<sup>st</sup>-cycle capacity ~1100 mAh/g, 90% efficiency, and capacity retention of 85% after 100 cycles at C/2.

## Sulfur Loading Effects in Nanocomposite Cathodes

- Higher sulfur loading lowers initial specific capacity of sulfur .
- Overall initial capacity of nanocomposite sulfur cathodes still increases.



## ELECTROLYTE ADDITIVES IMPROVE CYCLING PERFORMANCE



- Electrolyte additives improve capacity retention of Lisulfur cells.
- Addition of electrolyte additive also mitigates the growth of mossy lithium or other deposits.

#### Surface morphology of Li metal after 100 cycles in Li-Sulfur Cells



With electrolyte additives



Without electrolyte additives

## FUTURE WORK

- Further tune carbon framework properties (ex. pore size, morphology, etc)
- Investigate new cathode additive to decrease polysulfide migration.
- Investigate promising lithium-based anode systems to mitigate safety and stability issues of lithium metal anodes
- Investigate novel electrolyte systems and additives
- Optimize battery fabrication parameters

### **SUMMARY**

- Designed graphene-sulfur composite cathode with metal oxide adsorbent
- Designed carbon-sulfur composite electrode with 80 wt% sulfur loading
- Investigated potential electrolyte additives and compositions
- Testing baseline 1 Ah cells