CLEERS: Aftertreatment Modeling and Analysis

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ACE023

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Overview

Timeline

- Status: On-going core R&D
- Originated FY03 with DPF
- Now also includes LNT, SCR and DOC technologies

Budget

- FY10 funding \$750K
- FY11 funding allocation \$750K
 - Split between SCR, LNT, DOC and DPF focus areas



Barriers

- Limitations on:
 - available modeling tools
 - chemistry fundamentals
 - knowledge of material behavior
- Effective dissemination of information
- Technical "Valley of Death"

Partner

- Diesel Crosscut Team
- 21CT partners
- USCAR partners
- Oak Ridge National Lab



CLEERS PNNL Subprogram Goal

Working closely with our National Lab partners, the CLEERS industrial/academic team and in coordination with our CRADA portfolio, PNNL will...

...provide the practical & scientific understanding and analytical base required to enable the development of efficient, commercially viable emissions control solutions and modeling tools for ultra high efficiency vehicles.

- VT program goals are achieved through these project objectives:
 - interact with technical community to indentify relevant technological gaps
 - understand fundamental underlying mechanisms and material behavior
 - develop analytical and modeling tools, methodologies, and best practices
 - apply knowledge and tools to advance technologies leading to reducing vehicle emissions while improving efficiency
- Specific work tasks in support of the objectives are arrived at through:
 - focus group industrial monthly teleconferences, diesel x-cut meetings
 - yearly workshops and surveys
 - submission of SOW to the VT office



Technical Milestones & Approach

- The overall performance measure of the project is inextricably linked to the interests of industry
 - PNNL CLEERS activities have resulted in the formation of new CRADAs
 - Tremendous success of the annual workshops
 - Strong participation in the monthly teleconferences
- Specific performance measures are developed with the industrial/academic partners and captured in SOW
 - Specific technical targets and major milestones are described in our AOPs and annual reports to VT
- Approach "Science to Solutions"



PNNL FY11 Portfolio

CLEERS activity

Integrated Systems – John Lee

- DPF subtasks* Mark Stewart
- SCR subtasks* John Lee
- LNT subtasks Chuck Peden

CRADA activities

DPF – DOW Automotive (Stewart)**

SCR/DPF - PACCAR (Rappe)

SCR, HC – Ford Motor Company (Peden, Lee)

SCR, DOC – General Motors (Peden)

LNT – Cummins Inc. (Peden)

Oxidation Catalysts

- General Motors (Lee)
- SDC Materials (Herling)
- Caterpillar (Rappe)**



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*PNNL-led subteam **Past activities

FY2010/2011 Scope Objectives

Selective Catalytic Reduction (SCR)

- Update our SCR model for the state-of-the-art Cu SCR catalyst, and develop kinetics models to describe the performance degradation due to the competitive adsorption and catalyst aging
- Conduct detailed characterization of the Cu SCR catalyst with emphasis on the active sites and its deactivation
- Lean NOx Trap (LNT)

6

- Complete the investigation of CO₂ and H₂O effects on BaO morphology changes and NOx storage properties
- Fundamental studies of novel high-temp LNT catalyst materials
- Diesel Particulate Filter (DPF)
 - Evaluate the accuracy of unit collector model with respect to nano-sized particulates, and improve the accuracy of mico-scale model for prediction of soot-catalyst contact
 - Characterize soot chemistry and structure relevant to exhaust system performance and regeneration



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Technical Accomplishments Outline

SCR

- Developed single site kinetic models to describe the effects of competitive adsorption on NOx reduction on Fe SCR catalyst
- Currently developing Cu SCR model with ORNL, and conducting detailed characterization of the state-of-the-art Cu SCR catalyst with emphasis on the active sites

LNT

- Demonstration of significantly enhanced high temperature performance for a K/MgAl₂O₄ LNT material
- Fundamental studies of complex morphology changes in K-based LNT materials have been initiated

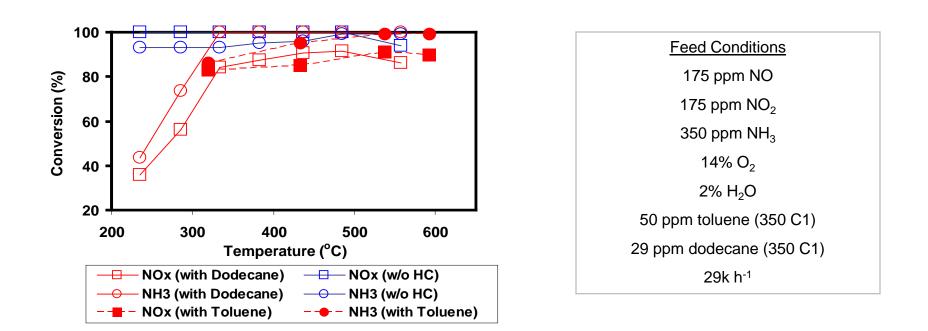
DPF

- Currently evaluating the accuracy of unit collector model with respect to nano-sized particulates
- Recently collected soot from HDD engine at MTU for characterization of soot chemistry and structure



Selective Catalytic Reduction

Effect of Hydrocarbon on SCR Reactions



- The effects of HC on Fe SCR catalyst examined using model HC species for combustion products and unburned fuel
- Detrimental effects of toluene & n-dodecane on SCR reactions
- Models developed to investigate the effects of HC on SCR reaction pathways quantitatively

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Overview of PNNL 1-D SCR Model

- Gas phase, surface phase concentrations and NH₃ storage as states
- Coded as 'C' S-functions and developed in Matlab/Simulink
- Optimized and validated using steady state and transient reactor data

No	Reaction Name	Reaction	Reaction Rate		
1	NH ₃ Adsorption	$NH_3 + S \rightarrow NH_3^*$	$R_1 = k_1 C_{s, NH3} (1 - \theta) \mathcal{Q}$		
2	NH ₃ Desorption	$NH_{3}^{*} \rightarrow NH_{3} + S$	$R_2 = k_2 \theta \mathcal{Q}$		
3	Fast SCR	$2NH_3 + NO + NO_2 \rightarrow 2N_2 + 3H_2O$	$R_3 = k_3 C_{NO} C_{NO2} \theta \Omega$		
4	Standard SCR	$4NH_3 + 4NO + O_2 \rightarrow 4N_2 + 6H_2O$	$R_4 = k_4 C_{NO} \theta \Omega$		
5	NO ₂ -SCR	$4NH_3+3NO_2 \rightarrow 3.5N_2+6H_2O$	$R_5 = k_5 C_{NO2} \mathcal{O} \mathcal{Q}$		
6	NH ₃ Oxidation	$2NH_3+3/2O_2 \rightarrow N_2+3H_2O$	$R_6 = k_6 C_{O2} \theta \Omega$		
7	NO-NO ₂ Oxidation	NO+1/2O ₂ ⇔NO ₂	$\mathbf{R}_{7} = \mathbf{k}_{7,f} \mathbf{C}_{NO} \mathbf{C}_{O2}^{1/2} - \mathbf{k}_{7,b} \mathbf{C}_{NO2}$		



Modeling Competitive Adsorption

- Single site storage model was first developed, and parameters were obtained from the Langmuir isotherms.
- Assuming the adsorbates, such as NH₃, NO, NO₂, H₂O and hydrocarbons (toluene, n-dodecane), competing for the same active site, single site kinetics for each species and the respective surface coverage are defined as follows:

$$\frac{d\theta_{i}}{dt} = \frac{1}{\Omega_{i}} [k_{ads,i}(1 - \theta_{i} - \sum_{j=1}^{N-1+P} \theta_{j})c_{s,i}\Omega_{i} - k_{des,i}\theta_{i}\Omega_{i} - \sum_{k=1}^{M} n_{i,k}r_{i,k}]$$

$$\varepsilon \frac{\partial c_{g,i}}{\partial t} = -\varepsilon u \frac{\partial c_{g,i}}{\partial x} - \beta_{i}A_{g}(c_{g,i} - c_{s,i})$$

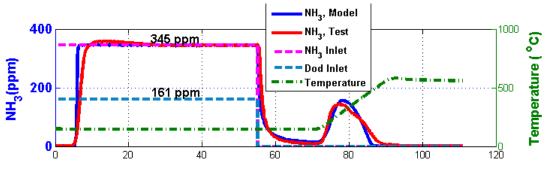
$$(1 - \varepsilon) \frac{\partial c_{s,i}}{\partial t} = \beta_{i}A_{g}(c_{g,i} - c_{s,i}) + \sum_{k} r_{i,k}n_{i,k}$$



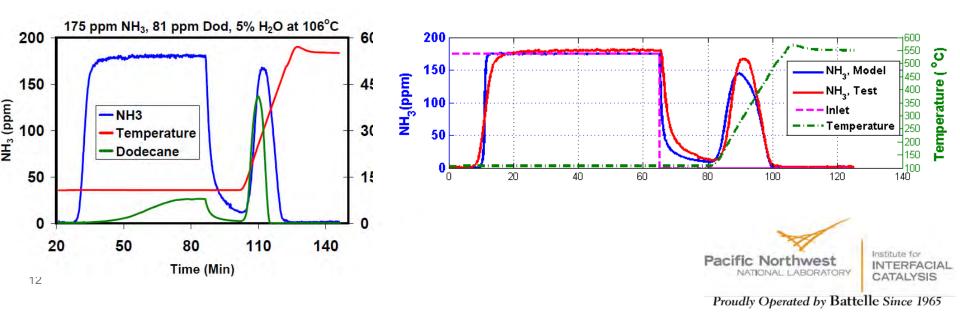
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CA Model Validation: NH₃, H₂O, Dodecane

 Competitive adsorption (CA) model development using data for NH₃ vs. H₂O, NH₃ vs. HC



 CA model validation for full competitive adsorption of NH₃, H₂O, and dodecane

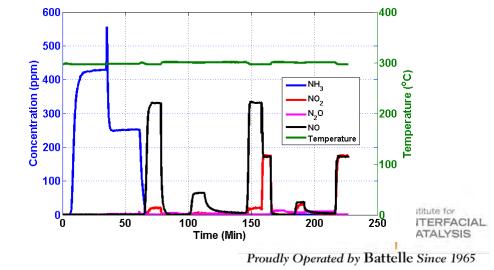


CLEERS Cu SCR Model Development

Commercial Cu SCR catalyst evaluation

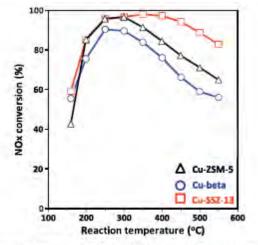
- Cu SCR being used by most OEMs in North America
- Lab reactor testing currently being conducted at ORNL using the CLEERS Transient reactor test protocol
- PNNL's 1-D Cu SCR catalyst model
 - Same strategy used for the Fe SCR catalyst model
 - Cu catalyst model being developed entirely in Matlab/Simulink
 - Autonomie being considered as platform to share CLEERS SCR catalyst model with the others

Step	Description	NO (ppm)	NO₂ (ppm)	NH₃ (ppm)	O₂ (%)	H₂O (%)	CO₂ (%)
а	cool + stabilize	0	0	0	10	5	5
b	NH ₃ adsorption	0	0	420	0	5	5
с	NH ₃ oxidation	0	0	420	10	5	5
d	NH ₃ desorption	0	0	0	10	5	5
е	NO oxidation	350	0	0	10	5	5
f	SCR α=1.2	350	0	420	10	5	5
g	SCR α=0.8	350	0	280	10	5	5
h	SCR α=1.0	350	0	250	10	5	5
i	NH ₃ inventory	350	0	0	10	5	5
j	NO2/NOx = 1	175	175	0	10	5	5
k	SCR α=1.2	175	175	420	10	5	5
I	SCR α=0.8	175	175	280	10	5	5
m	SCR α=1.0	175	175	250	10	5	5
n	NH ₃ inventory	175	175	0	10	5	5



State-of-the-art Cu SCR Catalyst Research

- First open literature studies of the latest Cu SCR catalyst
 - The current production Cu SCR catalyst is based on CHA zeolite.
 - Cu-SSZ-13 prepared and evaluated for various SCR reactions → J. Catal. 275 (2010)187-190
- Detailed characterization of active sites in progress
 - TPD, TPR of model Cu-zeolite catalysts
 - In situ XRD and EXAFS experiments at Brookhaven's NSLS



Hg. 1. NO₂ conversion profiles for Cu-SSZ-13 (squares.) Cu-beta (circles), and Cu-ZSM-5 (triangles) at various temperatures in a gas mixture containing 330 ppm NO. 350 ppm NH₂. 148 O₂ and 2% H₂O with a balance of N₂.

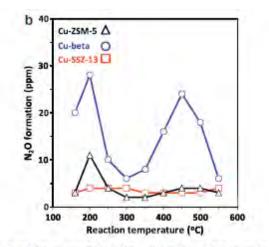


Fig. 2. NO₂ (a) and N₂O (b) formation profiles during NH₂ SCR on Cu-SS2-13 (squares). Cu-beta (circles) and Cu-ZSM-5 (triangles) at various temperatures in a gas mixture containing 350 ppm NO, 350 ppm NH₂ 14% O₂, and 23° H₂O with a balance of N₂.



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Lean NOx Traps

FY2010/2011 Scope Objectives

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- Conduct detailed characterization of the Cu SCR catalyst with emphasis on the active sites and its deactivation

Lean NOx Trap (LNT)

- Investigate CO₂ and H₂O effects on BaO morphology changes and NOx storage properties
- Fundamental studies of novel high-temp LNT catalyst materials

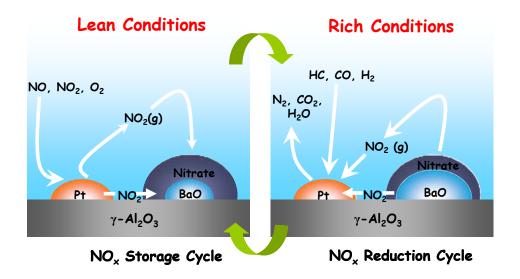
Diesel Particulate Filter (DPF)

- Evaluate the accuracy of unit collector model with respect to nano-sized particulates, and improve the accuracy of mico-scale model for prediction of soot-catalyst contact
- Characterize soot chemistry and structure relevant to exhaust system performance and regeneration



Approach

- Higher temperature NOx reduction performance required for:
 - Difficult to meet "not to exceed" regulations during desulfations
 - Possible use of LNTs for lean-gasoline applications

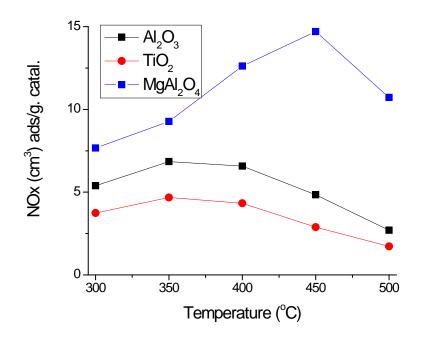


- PNNL/Cummins/JM CRADA focusing on degradation of possible materials for next-generation high temperature LNTs.
- CLEERS studies are addressing more fundamental issues of these potential new LNT materials related to composition, morphology, and chemical reaction kinetics and mechanisms.
- For these studies, PNNL has prepared a range of materials based on literature and prior CLEERS work at PNNL.



High Temperature LNT Catalyst Materials

- K-based LNTs known to exhibit higher temperature performance
- Recent literature reports suggest titania (TiO₂) may be a better support for K-based LNTs than alumina (Al₂O₃)
- Prior CLEERS studies on Ba-based LNTs at PNNL have suggested MgAl₂O₄ as a promising support material for high temperature application

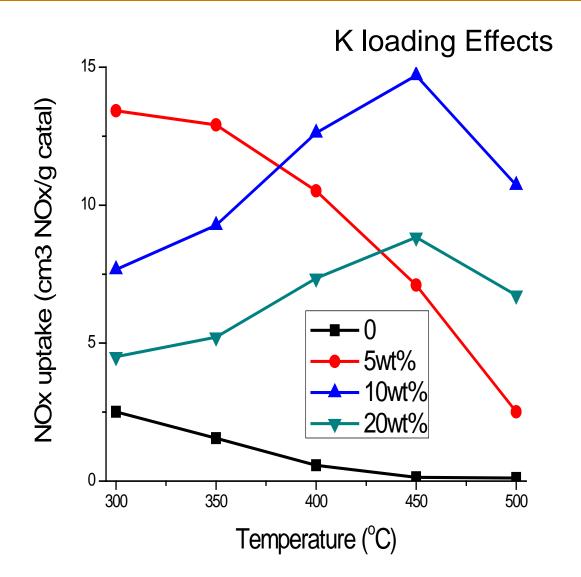


- Superior activity of MgAl₂O₄supported LNT relative to Al₂O₃- and TiO₂-supported samples over all temperatures.
- Moreover, maximum NOx uptake activity at a considerably higher temperature of 450 °C.



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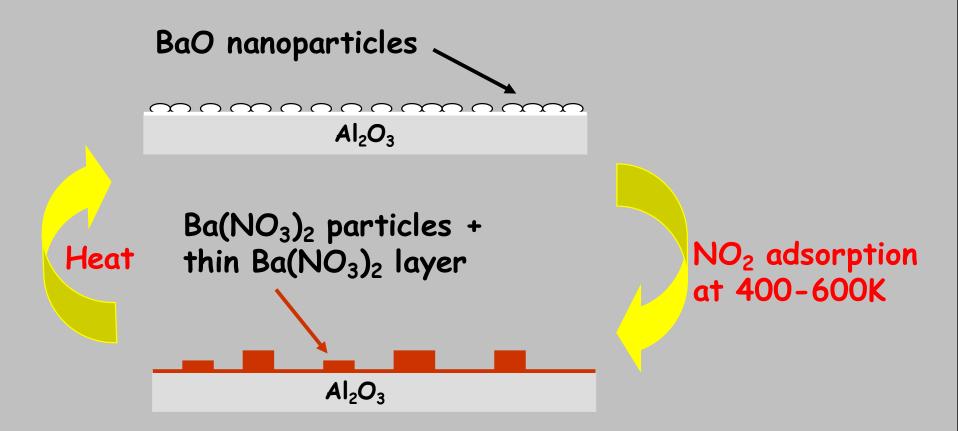
K-loading effects on MgAl₂O₄ support materials



- We're not aware of prior systematic studies of K-loading.
- Negligible MgAl₂O₄ contribution in NOx uptake at high temperature
- Drastic difference between 5 wt% and 10 wt%
- Higher loading than 10 wt% does not improve the activity.



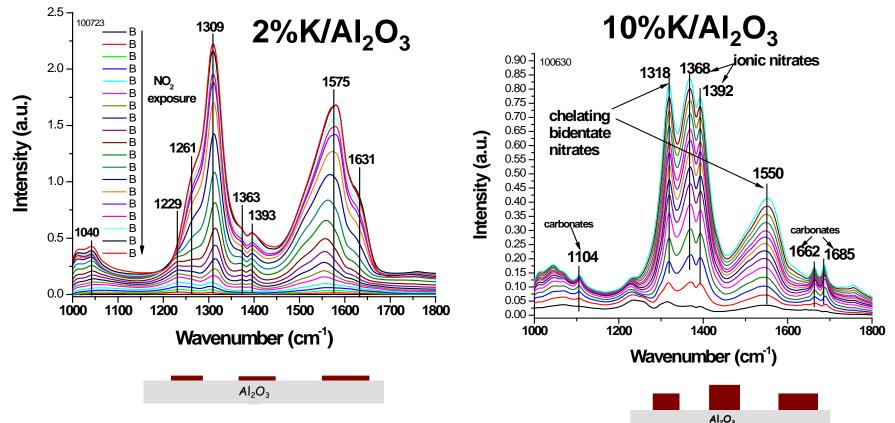
Significant Morphology Changes During Operation of Ba-Based LNTs Were Observed in our Prior CLEERS Studies



Szanyi, Kwak, Hanson, Wang, Szailer, Peden, J. Phys. Chem. B 109 (2005) 7339-7344.

Similar to Ba, FTIR Spectral Changes Consistent with Multiple K-oxide Phases

FTIR spectra of NO₂ adsorbed on K(2 or 10)/Al₂O₃ samples

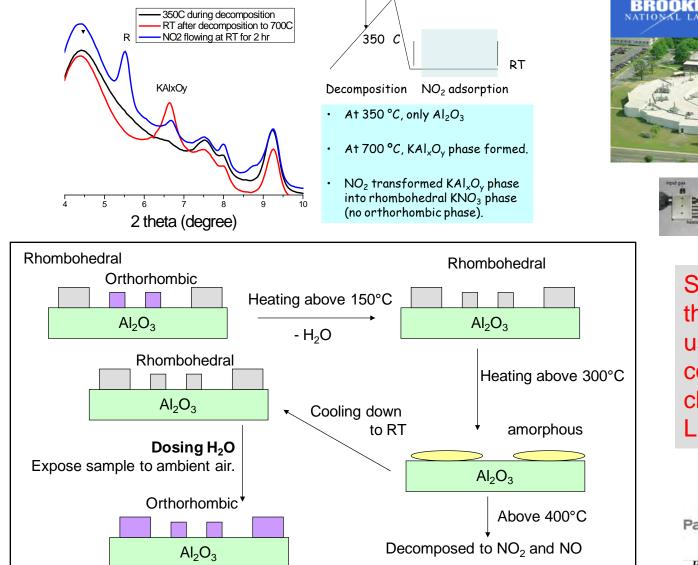


2 wt% K: mostly bidentate nitrates \rightarrow surface nitrates? 10 wt% K: ionic nitrates and bidentate nitrates \rightarrow surface and bulk nitrates?

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Complex Morphology Changes in K/Al₂O₃

Morphology change of K phase during decomposition and formation of nitrate



To mass spectrometer X-ray Synchrotron XRD at the NSLS is being



used to study the

LNTs

complex morphology

changes in K-based

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Diesel Particulate Filter

Approach

- Conduct bench reactor soot experiments
 - Soot oxidation w/ NOx correlated by surface area
 - HC Absorption
- Examine soot nano-structure using TEM and advanced image analysis techniques
- Improve pore-scale filter dynamics tools through:
 - Characterization of necessary microstructure resolution and sample size for accurate predictions in various media
 - Validation and enhancement of particle capture mechanics
 - Validation of particle motion Brownian dynamics algorithm
- Collect structural data for filter substrates using:
 - Micro/nano X-ray computed tomography
 - Porosimetry
- Carry out fundamental single cell filtration experiments using reproducible lab-generated aerosols

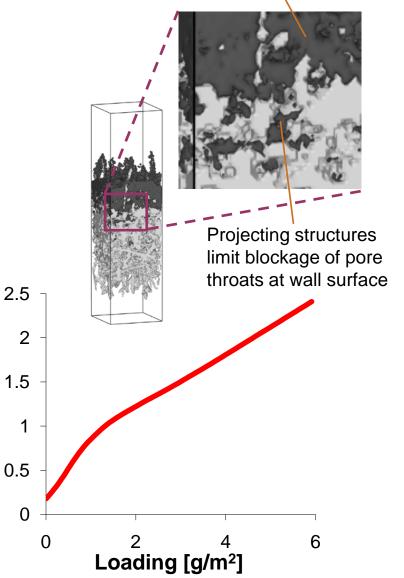


Pore-Scale DPF Model Development

Back-pressure [kPa]

- Numerous pore-scale DPF simulations have been carried out to date using the Lattice-Boltzmann method
- Pore networks have been resolved down to a few microns
- Qualitative features in backpressure, deposit morphology, and soot penetration into filter walls have been reproduced
- Insight into pore-scale mechanisms has assisted in the development of systems and new materials

Structure supports soot cake



Pore-Scale DPF Model Development (cont'd)

- Recent studies suggest requirements for quantitative pore-scale simulations:
 - Sintered granular materials (SiC) are easier to simulate than cordierite
 - Cordierite may require resolutions < 1 um and domain sizes of over 1 mm (>0.5E+9 computational cells)
- Adjustment of lattice-Boltzmann parameters may allow lower resolutions at the expense of shorter time-steps

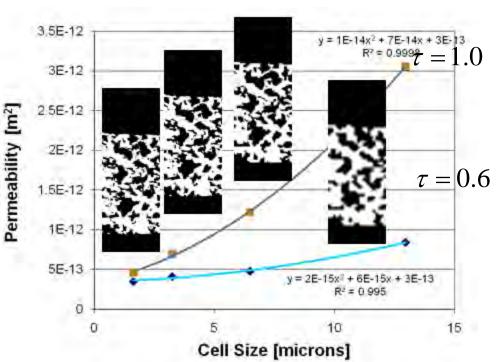
Fortunately, cost and availability of massively parallel computational resources are rapidly becoming more favorable

 \rightarrow Precise quantitative performance predictions from pore-scale simulations may be just around the corner



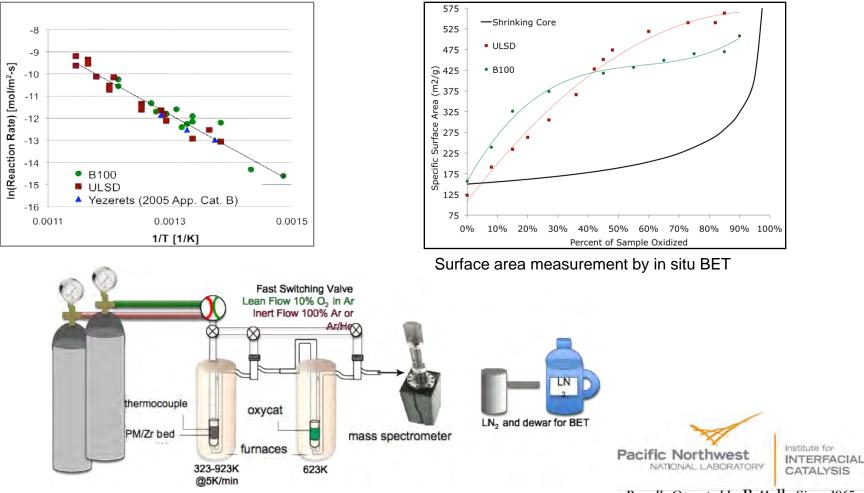
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Predicted permeability vs. computational cell size for cordierite



Kinetic Measurements & Surface Area

- Bench reactor soot experiments (in progress)
 - Soot oxidation w/ NOx correlated by surface area
 - HC adsorption



Conclusion & Future Work

Conclusions

SCR

- Developed single site kinetic models to describe the effects of competitive adsorption on NOx reduction on Fe SCR catalyst
- Currently developing Cu SCR model using data obtained by CLEERS transient reactor protocol with ORNL
- First open literature investigation of the state-of-the-art Cu SCR catalyst with emphasis on the active sites

LNT

- Mg/Al₂O₄ identified as a very promising support material for Kbased LNTs in high-temperature applications
- Initial structural studies of K-based LNTs indicate very complex morphology changes

DPF

Recently collected soot from HDD engine at MTU for characterization of soot chemistry and structure relevant to exhaust system performance and regeneration



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Future Work

SCR

- Update the kinetics for Cu SCR model with ORNL's data, and develop kinetics models to describe the performance degradation due to the competitive adsorption and catalyst aging
- Continue to conduct detailed characterization of the Cu SCR catalyst with emphasis on the active sites and its deactivation

LNT

- Complete studies of CO₂ and H₂O effects on BaO morphology changes and NOx storage properties
- Continue fundamental studies of morphology changes and NOx uptake mechanisms of novel high-temp LNT catalyst materials
- Investigate the formation and stability of PGM particles (also relevant to DOC, TWC)

DPF

- Evaluate the accuracy of unit collector model with respect to nano-sized particulates, and improve the accuracy of mico-scale model for prediction of soot-catalyst contact
- Characterize soot chemistry and structure relevant to exhaust system performance and regeneration

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